A bimetallic $Ag_{15}Cu_{12}(S-c-C_6H_{11})_{18}(CH_3COO)_3$ nanocluster featuring an

irregular Ag₁₂ kernel

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Fig. S1 The photos of (A) $Ag_{15}Cu_{12}$ cluster dissolved in dichloromethane and crystals (insert of A). (B) Representative TEM images of $Ag_{15}Cu_{12}$. Insert of (B) the corresponding histogram of $Ag_{15}Cu_{12}$ nanocluster.



Fig. S2 The UV-vis spectrum of product synthesized without phosphorus ligands during the synthesis process.



Fig. S3 The UV-Vis spectra of crude product synthesized via adding phosphine ligands with different substituting group.



Fig. S4 Illustration of the C₃ axial symmetry in $Ag_{15}Cu_{12}$. For clarity, hydrogen atoms and a part of carbon atoms are omitted. Color labels: sky blue = Ag; blue= Cu; red = S; yellow = O; and gray = C. All hydrogen atoms are omitted for clarity.



Fig. S5 Detailed structure analysis of (A) Ag_{12} core, two Ag_6 facets are cross interconnected in a slightly angular twist. (B) Ag₃Cu₁₂(SR)₁₈(CH₃COO)₃ shell, two Cu₃(SR)₆ moieties, three monomeric [-SR-Ag-SR-] units and three Cu₂(CH₃COO) staples. Color labels: sky blue and light green = Ag, blue= Cu; red = S; yellow = O; and gray = C. All hydrogen atoms are omitted for clarity.



Fig. S6 Bond lengths of the $Ag_{15}Cu_{12}$ nanocluster. Color label: sky blue = Ag; blue= Cu; red = S; yellow = O; and gray = C.



Fig. S7 The detailed angles in the Cu₃(SR)₃ structures of $Ag_{15}Cu_{12}$ nanocluster. Color labels: sky blue = Ag; blue = Cu; red = S; yellow = O; and gray = C.



Fig. S8 The entire structure of $Ag_{15}Cu_{12}$ nanocluster with one hexane solvent molecule around. Color labels: sky blue = Ag; blue= Cu; red = S; yellow = O; and gray = C.



Fig. S9 Molar extinction coefficients of Ag₁₅Cu₁₂.



Fig.S10 (A) ESI mass spectrum in full width and (B) matching of isotope patterns between the experimental and simulated spectra of $Ag_{15}Cu_{12}$.



Fig.S11 (A) The High-resolution XPS spectra of Ag 3d of $Ag_{15}Cu_{12}$ (B) The High-resolution XPS spectra of Cu 2p of $Ag_{15}Cu_{12}$.



Fig.S12 Temperature-dependent UV-vis absorption of Ag₁₅Cu₁₂ in CH₂Cl₂.



Fig.S13 Time-dependent UV-vis spectra (A) UV-vis spectra at room temperature (dissolved in CH_2Cl_2), and (B) at 45°C (dissolved in $CHCl_3$).



Fig.S14 Kohn–Sham (KS) orbital energy levels and the electron density profile of the HOMO and LUMO of $Ag_{15}Cu_{12}$. Color labels: sky blue = Ag; lake blue= Cu; orange = S; yellow = O; and gray = C.



Fig.S15 Transition involved frontier molecular orbitals of $Ag_{15}Cu_{12}^{T}$.



Fig.S16 Superatomic orbitals of $Ag_{15}Cu_{12}^{T}$.

Table S1. The comparison of the bond length of the icosahedral Ag_{13} metal core and Ag_{12} core of $Ag_{15}Cu_{12}$.

Cluster	Core	Average bond length
$[Ag_{25}(SR)_{18}]^{-1}$	icosahedral Ag ₁₃	2.866 Å
$Ag_{29}(SR)_{12}(PPh_3)_4$	icosahedral Ag ₁₃	2.868 Å
Ag ₁₅ Cu ₁₂ (SR) ₁₈ (CH ₃ COO) ₃	irregular Ag ₁₂	2.893Å
$Ag_{27}H_{11}(SR)_{12}(DPPM)_{6}$	twisted Ag ₁₃	2.993 Å

Table S2. Crystal Date and Structure Refinement of the $Ag_{15}Cu_{12}(SR)_{18}(CH_3COO)_3$ (C₆H₁₄) nanocluster.

Ellipsoid plot	ADPs
1000/E PORCEO 1 esp * 120 1 control of the state of the	H Ag Cu S
Identification code	lzz-revised_sq
Empirical formula	$C_{120}H_{221}Ag_{15}Cu_{12}O_6S_{18}$
Formula weight	4717.57
Temperature/K	120
Crystal system	triclinic
Space group	P-1
a/Å	17.4880(12)
b/Å	17.6703(12)
c/Å	30.473(2)
α/°	90.275(5)
β/°	96.680(5)
$\gamma/^{\circ}$	93.511(5)
Volume/Å ³	9334.6(11)
Ζ	2
$\rho_{calc}g/cm^3$	1.678
µ/mm ⁻¹	15.873
F(000)	4656.0
Crystal size/mm ³	$0.02 \times 0.01 \times 0.01$
Radiation	$CuK\alpha (\lambda = 1.54186)$
2Θ range for data collection/°	6.924 to 124.996
Index ranges	$-19 \le h \le 20, -20 \le k \le 7, -35 \le l \le 33$
Reflections collected	56338
Independent reflections	$28534 [R_{int} = 0.0693, R_{sigma} = 0.0818]$

Data/restraints/parameters	28534/913/1543
Goodness-of-fit on F ²	0.993
Final R indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0772, wR_2 = 0.1997$
Final R indexes [all data]	$R_1 = 0.1055, wR_2 = 0.2341$
Largest diff. peak/hole / e Å ⁻³	2.67/-2.47