

ELECTRONIC SUPPLEMENTARY INFORMATION

“Towards hydrogen-rich ionic $(\text{NH}_4)(\text{BH}_3\text{NH}_2\text{BH}_2\text{NH}_2\text{BH}_3)$ and related molecular $\text{NH}_3\text{BH}_2\text{NH}_2\text{BH}_2\text{NH}_2\text{BH}_3$ ”

R. Owarzany, T. Jaroń, K. Kazimierzczuk, P. J. Malinowski, W. Grochala, K. J. Fijalkowski

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VBH)[$\text{B}(\text{C}_6\text{H}_5)_4$]
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 $(\text{NH}_4)(\text{B}_3\text{N}_2) \cdot 3(\text{B}_3\text{N}_3)$ $(\text{NH}_4)(\text{B}_3\text{N}_2)$ (B_3N_3) $(\text{NH}_4)(\text{BH}_4)$ NH_3BH_3

1. Records of reporting synthesis of novel amidoborane and M(B3N2) salts and related compounds:

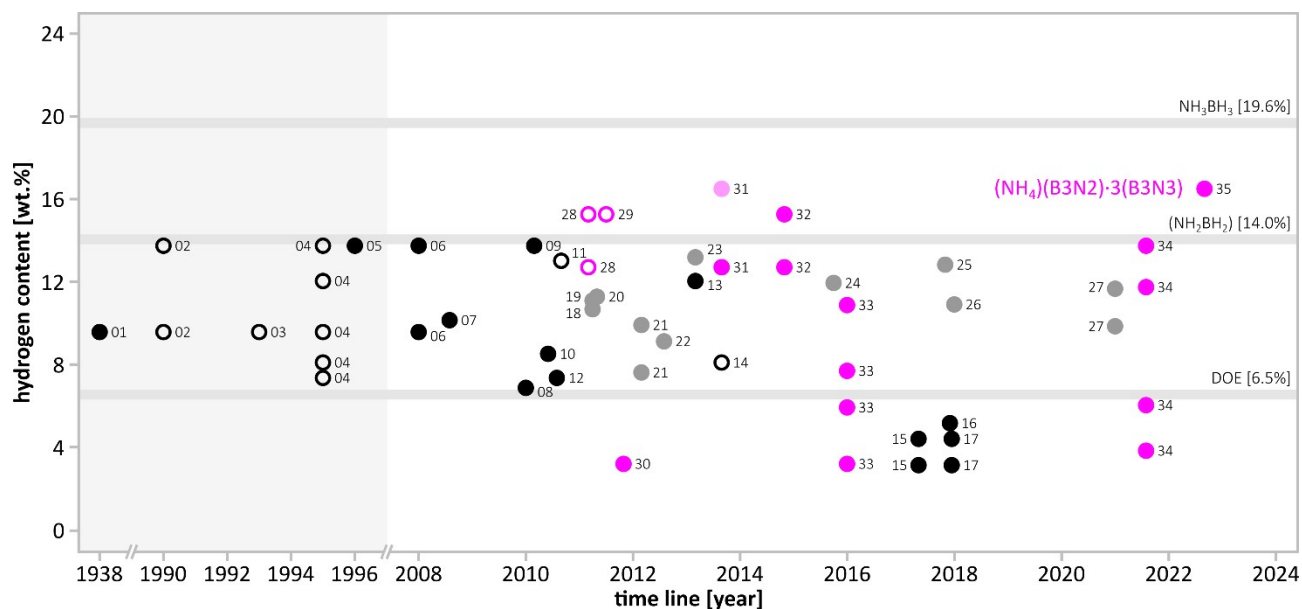


Fig. S1.1. Hydrogen content of monometallic amidoboranes (black), bimetallic amidoboranes (grey), M(B3N2) salts (magenta) and as a function of reporting date. Hydrogen content of NH_3BH_3 (19.6%), polymeric (NH_2BH_2) (14.0%) and DOE ultimate target (6.5%) given as a reference. Reports and theses marked with hollow circles.

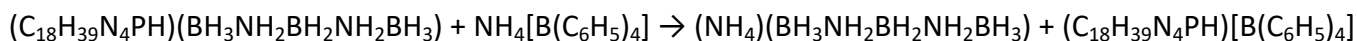
References:

- 01 NaAB (9.5%) – H. I. Schlesinger, A.B. Burg, *J. Am. Chem. Soc.* 60 (1938) 290–299.
- 02 α -LiAB (13.7%), NaAB (9.5%) – P. M. Niedenzu, *Ph.D. Thesis*, Ohio State University, 1990.
- 03 NaAB (9.54%) – T. Salupo, *Ph.D. Thesis*, Ohio State University, 1993.
- 04 α -LiAB (13.7%), NaAB (9.5%), KAB (7.3%), $\text{Mg}(\text{AB})_2$ (12.0%), $\text{Zn}(\text{AB})_2$ (8.1%) – A. L. DeGraffenreid, *Ph.D. Thesis*, Ohio State University, 1995.
- 05 α -LiAB (13.7%) – A. G. Myers, *et al.*, *Tetrahedron Lett.* 37 (1996) 3623–3626.
- 06 α -LiAB (13.70%), NaAB (9.54%) – Z. Xiong, *et al.*, *Nature Mater.* 7 (2008) 138–141.
- 07 $\text{Ca}(\text{AB})_2$ (10.10%) – J. Spielmann, *et al.*, *Angew. Chem. Int. Ed.* 47 (2008) 6290–6295.
- 08 $\text{Sr}(\text{AB})_2$ (6.84%) – Q. Zhang, *et al.*, *J. Phys. Chem. C* 114 (2010) 1709–1714.
- 09 β -LiAB (13.70%) – C. Wu, *et al.*, *Inorg. Chem.* 49 (2010) 4319–4323.
- 10 $\text{Y}(\text{AB})_3$ (8.47%) – R. V. Genova, *et al.*, *J. Alloys. Comp.* 499 (2010) 144–148.
- 11 $\text{Al}(\text{AB})_3$ (12.97%) – M. F. Hawthorne, *et al.*, *Final Report*, University of Missouri, 2010.
- 12 KAB (7.31%) – H. V. K. Diyabalanage, *et al.*, *J. Am. Chem. Soc.* 132 (2010) 11836–11837.
- 13 $\text{Mg}(\text{AB})_2$ (12.00%) – J. Luo, *et al.*, *Energy Environ. Sci.* 6 (2013) 1018–1025.
- 14 $\text{Zn}(\text{AB})_2$ (8.06%) – R. Owarzany, *M.Sc. Thesis*, University of Warsaw, 2013.
- 15 LT-RbAB (4.37%), LT-CsAB (3.1%) – I. V. Kazakov, *et al.*, *Polyhedron* 127 (2017) 186–190.
- 16 $\text{Ba}(\text{AB})_2$ (5.12%) – N. A. Shcherbina, *et al.*, *Rus. J. Gen. Chem.* 87 (2017) 2875–2877.
- 17 HT-RbAB (4.37%), HT-CsAB (3.1%) – R. Owarzany, *et al.*, *Dalton Trans.* 46 (2017) 16315–16320.
- 18 $\text{Na}_2\text{Mg}(\text{AB})_4$ (10.63%) – H. Wu, *et al.*, *Chem. Commun.* 47 (2011) 4102–4104.
- 19 $\text{NaMg}(\text{AB})_3$ (11.05%) – X. Kang, *et al.*, *Dalton Trans.* 40 (2011) 3799–3801.
- 20 $\text{LiNa}(\text{AB})_2$ (11.24%) – K. J. Fijalkowski, *et al.*, *Dalton Trans.* 40 (2011) 4407–4413.
- 21 $\text{KMg}(\text{AB})_3$ (9.88%), $\text{RbMg}(\text{AB})_3$ (7.58%) – X. Kang, *et al.*, *Int. J. Hydrog. Energy* 37 (2012) 4259–4266.
- 22 $\text{K}_2\text{Mg}(\text{AB})_4$ (9.08%) – Y. S. Chua, *et al.*, *Chem. Mater.* 24 (2012) 3574–3581.
- 23 $\text{LiAl}(\text{AB})_4$ (13.15%) – G. Xia *et al.*, *Mater. Chem. A* 1 (2013) 1810–1820.
- 24 $\text{NaAl}(\text{AB})_4$ (11.9%) – I. Dovgaliuk *et al.*, *Chem. Eur. J.* 21 (2015) 14562–14570.
- 25 $\text{Li}_2\text{Mg}(\text{AB})_4$ (12.79%) – N. Biliškov *et al.*, *Chem. Eur. J.* 23 (2017) 16274–16282.
- 26 $\text{KAl}(\text{AB})_4$ (10.87%) – K. T. Møller, *et al.*, *Int. J. Hydrog. Energy* 43 (2018) 311–321.
- 27 $\text{Li}_2\text{Ca}(\text{AB})_4$ (11.63%), $\text{Na}_2\text{Ca}(\text{AB})_4$ (9.81%) – I. Milanovic, *et al.*, *ACS Sustainable Chem. Eng.* 9 (2021) 2089–2099.
- 28 $\text{Li}(\text{B3N2})$ (15.22%), $\text{Na}(\text{B3N2})$ (12.66%) – I. C. Evans, *Ph.D. Thesis*, University of Birmingham, 2011.
- 29 $\text{Li}(\text{B3N2})$ (15.22%) – K. R. Ryan, *Ph.D. Thesis*, University of Oxford, 2011.
- 30 α -VB(B3N2) (3.15%) – W. C. Ewing, *et al.*, *J. Am. Chem. Soc.* 133 (2011) 17093–17099.
- 31 $\text{Na}(\text{B3N2})$ (12.66%), $(\text{NH}_4)(\text{B3N2})/3(\text{B3N3})$ (16.4%) – W. C. Ewing, *et al.*, *Inorg. Chem.* 52 (2013) 10690–10697.
- 32 $\text{Li}(\text{B3N2})$ (15.22%), $\text{Na}(\text{B3N2})$ (12.66%) – K. J. Fijalkowski, *et al.*, *Phys. Chem. Chem. Phys.* 16 (2014) 23340–23346.
- 33 $\text{K}(\text{B3N2})$ (10.83%), $\text{Rb}(\text{B3N2})$ (7.65%), $\text{Cs}(\text{B3N2})$ (5.89%), β -VB(B3N2) (3.15%) – R. Owarzany, *et al.*, *Inorg. Chem.* 55 (2016) 37.
- 34 $(\text{Bu}_4\text{N})(\text{B3N2})$ (3.80%), $(\text{Et}_4\text{N})(\text{B3N2})$ (6.00%), $[\text{C}(\text{N}_3\text{H}_6)](\text{B3N2})$ (13.70%), $[\text{C}(\text{N}_3\text{H}_5\text{CH}_3)](\text{B3N2})$ (11.7%) – X. M. Chen, *Chem Asian J.* 16 (2021) 1–7.
- 35 $(\text{NH}_4)(\text{B3N2})/3(\text{B3N3})$ (16.4%) – this study

2. Synthesis of alkali metal M(B3N2) salts:

All operations were performed under inert Ar atmosphere inside gloveboxes, MBRAUN Labmaster DP or Vigor SG1200 (O₂, H₂O < 1.0 ppm). Commercially available reagents and solvents were used: NH₃BH₃ (98%, JSC Aviabor), NH₄B(C₆H₅)₄ (99%, Sigma-Aldrich (later denoted as SA), C₄H₈O (99%, SA), CH₂Cl₂ (99%, SA).

Metathetic synthesis was performed using (C₁₈H₃₉N₄PH)(BH₃NH₂BH₂NH₂BH₃) and NH₄[B(C₆H₅)₄] in anhydrous THF at room temperature under argon atmosphere:



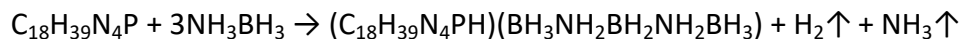
A follow-up process of dehydrogenation of NH₄(BH₃NH₂BH₂NH₂BH₃) occurs leading to neutral linear molecule NH₃BH₂NH₂BH₂NH₂BH₃:



Obtained mixture of products was well soluble in THF. Side product (C₁₈H₃₉N₄PH)[B(C₆H₅)₄] was precipitated by washing with anhydrous DCM.

The main product crystallises in *P*2₁/*c* unit cell with the lattice parameters of: a = 13.401(11) Å, b = 13.196(8) Å, c = 17.828(12) Å, β = 128.83(4)°, V = 2556(3) Å³ and Z = 16. The crystalline product contains two compounds: NH₄(BH₃NH₂BH₂NH₂BH₃) and NH₃BH₂NH₂BH₂NH₂BH₃ in molar ratio 1:3. In the manuscript, the product is denoted as “main product” or “(NH₄)(B3N2)·3(B3N3)”.

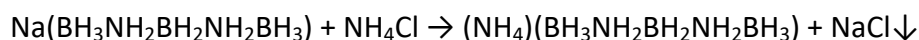
The synthesis of (C₁₈H₃₉N₄PH)(BH₃NH₂BH₂NH₂BH₃) was performed according to the route described in our earlier paper (R. Owarzany, *et al.*, *Inorg. Chem.* 55 (2016) 37/) in a direct reaction of Verkade’s Base with 3 equivalents of ammonia borane in toluene at room temperature:



Different route of metathetic synthesis between Na(BH₃NH₂BH₂NH₂BH₃) and NH₄Cl in glyme at room temperature for 24 hours was reported earlier (W. C. Ewing *et al.* *Inorg. Chem.* 52 (2013) 10690.), however, the authors were aiming NH₃BH₂NH₂BH₂NH₂BH₃ according to the following equation:



Judging from the comparison of NMR data presented by Ewing *et al.* to our own data we strongly believe that this process leads to (NH₄)(B3N2)·3(B3N3) according to the following reaction equations:



3. Table of ^{11}B NMR @ THF- d_8 chemical shifts of $\text{M}(\text{B3N2})$ salts and ammonia borane:

Table S3. Chemical shifts, positions of multiplets, excitation frequencies and J-coupling values observed in ^{11}B NMR spectra in deuterated THF solution (δ [ppm]) of $(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$ at room temperature. Results for ammonia borane [AB], precursor [β -VBH(B3N2)] and alkali metal $\text{M}(\text{B3N2})$ salts: [Li(B3N2), Na(B3N2), K(B3N2), Rb(B3N2), Cs(B3N2)] at RT are shown for comparison.

	NH_3BH_3	VBH(B3N2)	M(B3N2) salts					$(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$
			Li(B3N2)	Na(B3N2)	K(B3N2)	Rb(B3N2)	Cs(B3N2)	
BH₂ triplet	–	–6.590	–6.743	–7.155	–7.499	–7.410	–7.792	–9.95 –11.78
	–	–8.184	–8.360	–8.582	–8.568	–8.424	–8.384	–10.40 –12.26
	–	–9.716	–9.966	–10.227	–9.591	–9.491	–9.042	–10.85 –12.70
position	–	–8.163	–8.356	–8.654	–8.553	–8.442	–8.406	–10.4 –12.3
1J (B,H)	–	100 Hz	103 Hz	99 Hz	101 Hz	100 Hz	101 Hz	101 Hz 102 Hz
freq.	96.32 MHz	64.16 MHz	96.32 MHz	96.32 MHz	96.32 MHz	96.32 MHz	160.48MHz	224.62 MHz

	NH_3BH_3	VBH(B3N2)	M(B3N2) salts					$(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$
			Li(B3N2)	Na(B3N2)	K(B3N2)	Rb(B3N2)	Cs(B3N2)	
BH₃ quartet	–18.184	–19.462	–20.465	–20.264	–20.613	–20.294	–20.279	–21.60
	–19.632	–20.933	–21.836	–21.693	–21.556	–21.226	–20.902	–22.01
	–21.144	–22.342	–23.314	–23.202	–22.499	–22.168	–21.479	–22.42
	–22.610	–23.752	–24.634	–24.582	–23.382	–23.105	–22.025	–22.83
position	–20.393	–21.622	–22.562	–22.435	–22.013	–21.698	–21.171	–22.21
1J (B,H)	95 Hz	91 Hz	90 Hz	91 Hz	89 Hz	90 Hz	94 Hz	91 HZ
freq.	96.32 MHz	64.16 MHz	96.32 MHz	96.32 MHz	96.32 MHz	96.32 MHz	160.48MHz	224.62 MHz

4. Table of bands appearing in the IR spectra of M(B3N2) salts and ammonia borane:

Table S4. Absorption bands detected in IR spectra (wavenumber [cm^{-1}]) of $\text{NH}_4(\text{B3N2}) \cdot 3(\text{B3N3})$ at room temperature. Results for ammonia borane [AB] and alkali metal M(B3N2) salts: [Li(B3N2), Na(B3N2), K(B3N2), Rb(B3N2), Cs(B3N2)] at RT. Absorption bands of ammonia borane at RT are shown for comparison. (ν = stretching, δ = deformation: bending and torsional modes).

Band	NH_3BH_3	M(B3N2) salts					$(\text{NH}_4)(\text{B3N2}) \cdot 3(\text{B3N3})$	
		Li(B3N2)	Na(B3N2)	K(B3N2)	Rb(B3N2)	Cs(B3N2)		
$\nu(\text{NH})$	3311 vs	3310 s	3302 vs	3305 vs	3308 m 3295 m	3313 w 3287 m	3306 vs 3288 vs 3268 sh	
	3253 vs	3273 m	3256 m	3261 m	3261 w 3252 w	3261 w 3235 m	3259 m 3239 m 3223 sh	
	3196 s							
$\nu(\text{BH})$				2420 sh			2439 sh 2407 s 2357 s 2317 vs 2302 sh	
	2347 vs	2350 vs 2322 s	2364 s 2315 s	2352 m	2390 sh 2346 s	2389 sh 2329 m		
	2289 s	2282 vs 2245 s	2286 vs	2304 s 2279 s 2259 s	2294 vs 2263 s 2248 sh	2291 vs	2260 m	
	2118 m			2210 sh	2204 sh	2248 s 2189 sh		
$\delta(\text{NH})$	1611 m				1617 vw		1604 sh 1572 w 1564 m 1556 m	
		1571 vs	1576 w 1556 m	1583 w 1568 m	1562 sh 1557 m	1579 vw 1565 w 1557 vw	1480 w 1428 w 1415 m 1392 m 1375 w	
$\delta(\text{BH})$	1163 vs	1283 m 1226 s 1201 s	1248 m	1244 m		1233 s 1206 m 1193 sh	1259 m 1231 s 1205 s	1262 s 1244 s 1203 vs 1179 s
		1148 s 1135 m	1199 vs 1175 m	1202 s 1194 sh	1182 m 1128 w	1167 s 1134 m	1188 s 1171 s 1127 m	1168 sh 1134 m 1118 w
		1067 s		1074 m	1073 w	1065 w	1063 w	1078 w
			1044 m 1013 w	1055 m 999 w	1056 w 997 w	1038 w 1012 vw	1043 m 1001 w	1056 w 983 w
$\nu(\text{BN})$ and other		916 w	893 vw	893 vw 875 w	901 vw 873 vw	902 w 881 w	861 w	
		874 vw 799 vw	870 w 785 vw	781 vw 727 vw	854 vw 811 vw 727 vw	856 w 791 vw 723 vw	804 vw 749 w 711 w	

5. Table of bands appearing in the RAMAN spectra of M(B3N2) salts and ammonia borane:

Table S5. Absorption bands detected in Raman spectra (wavenumber [cm^{-1}]) of $\text{NH}_4(\text{B}_3\text{N}_2) \cdot 3(\text{B}_3\text{N}_3)$ at room temperature. Results for ammonia borane [AB] and alkali metal M(B3N2) salts: [Li(B3N2), Na(B3N2), K(B3N2), Rb(B3N2), Cs(B3N2)] at RT. Absorption bands of ammonia borane at RT are shown for comparison. (ν = stretching, δ = deformation: bending and torsional modes).

Band	NH_3BH_3	M(B3N2) salts					$(\text{NH}_4)(\text{B}_3\text{N}_2) \cdot 3(\text{B}_3\text{N}_3)$
		Li(B3N2)	Na(B3N2)	K(B3N2)	Rb(B3N2)	Cs(B3N2)	
$\nu(\text{NH})$					3304 sh	3307 sh	
	3314 m	3314 m	3302 s	3306 s	3293 w 3288 w	3303 w 3292 sh 3279 w	3307 m 3288 m
	3253 vs 3177 m	3272 s	3265 vs	3263 vs	3256 m 3245 m	3258 m 3232 m	3260 vs 3241 vs 3041 s
$\nu(\text{BH})$	2378 vs	2418 vw 2370 w	2403 w 2373 w	2382 m 2347 m	2379 m 2341 m	2379 m 2343 m	2475 sh 2442 w 2394 m 2320 vs 2261 s
	2284 vs	2282 m 2250 s	2275 s 2243 s 2214 sh	2301 s 2274 s	2291 vs 2243 vs	2286 vs 2250 vs	
		2166 vw		2186 sh	2192 sh		
$\delta(\text{NH})$	1598 m 1583 m	1567 m	1539 w 1519 vw	1649 vw 1585 vw 1569 vw	1608 vw 1565 w 1550 w	1645 vw 1566 w 1534 w	1578 s 1559 m
	1190 sh 1168 m 1069 vw	1281 w 1259 vw 1226 w 1206 m	1212 w	1227 w 1193 w 1168 w	1207 w 1188 m 1155 w 1122 vw 1056 vw	1206 sh 1183 m 1163 m 1036 vw	1201 w 1184 w 1173 w 1155 w 1102 w 1027 m 998 vs
$\nu(\text{BN})$ and other	800 w 785 m 729w	895 vw 873 w 806 w	856 vw 835 w 749 w 614 vw	892 vw 871 w 778 w	895 w 862 w 847 w 779 w 721 w 653 w 639 vw	915 w 899 vw 876 vw 851 w 783 vw 724 w 715 vw 643 vw	857 w 796 w 667 m 618 w 604 s

6. Comparison of IR and Raman spectra of alkali metal M(B3N2) salts:

IR and Raman spectra of $\text{NH}_4(\text{B3N2})/3(\text{B3N3})$ and alkali metal M(B3N2) salts: [Li(B3N2), Na(B3N2), K(B3N2), Rb(B3N2), Cs(B3N2)]. NH and BH stretching and NH bending regions highlighted and magnified in separate figures.

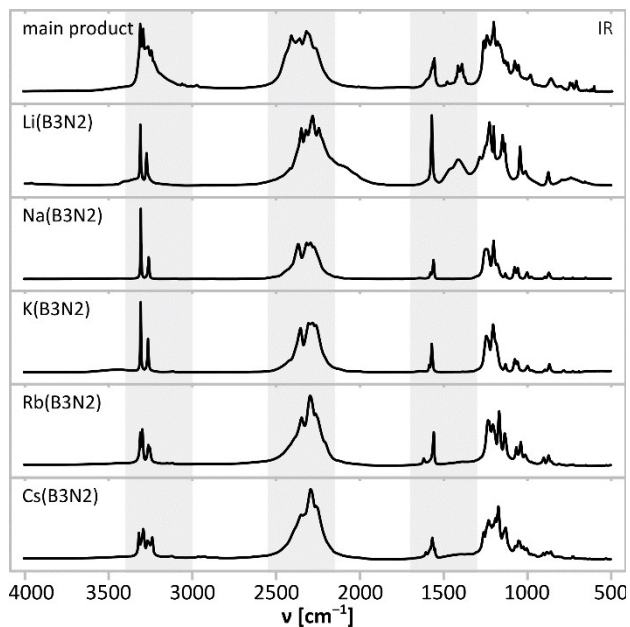


Fig. S6.1. Comparison of IR spectra of M(B3N2) salts.

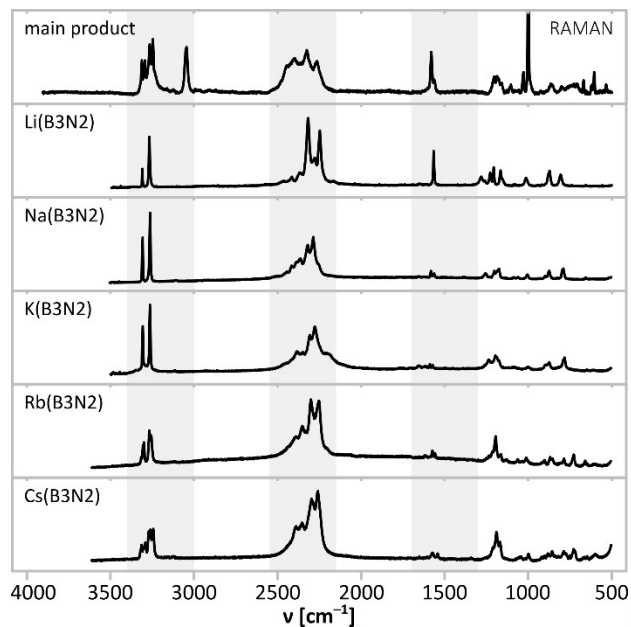


Fig. S6.2. Comparison of Raman spectra of M(B3N2) salts.

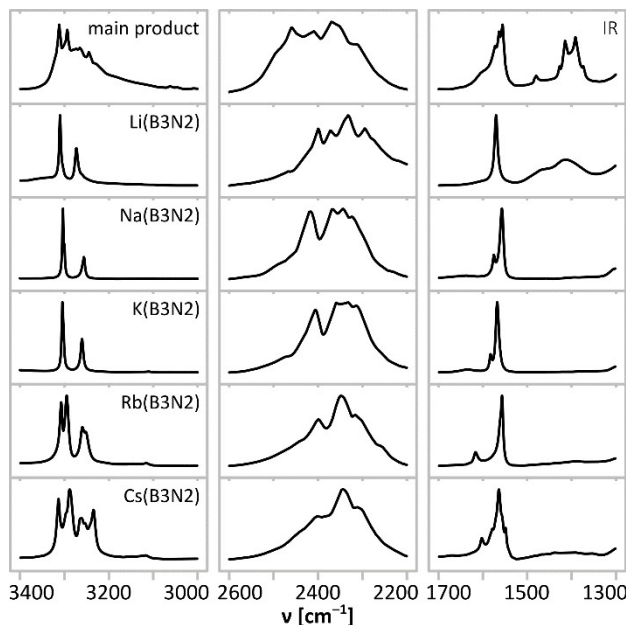


Fig. S6.3. Comparison of NH and BH stretching and NH bending regions of IR spectra of M(B3N2) salts.

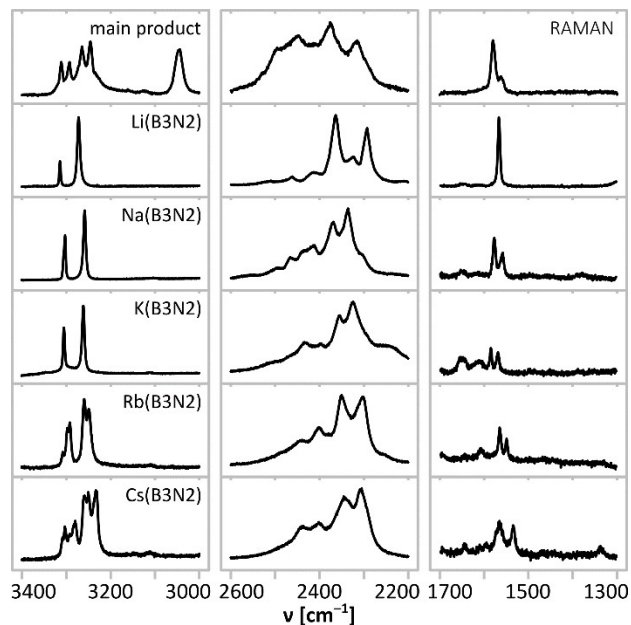


Fig. S6.4. Comparison of NH and BH stretching and NH bending regions of Raman spectra of M(B3N2) salts.

7. Thermal decomposition (TGA curves) of M(B3N2) salts:

The thermal decomposition of $\text{NH}_4(\text{B3N2})/3(\text{B3N3})$ and alkali metal $\text{M}(\text{B3N2})$ salts occurs at the temperature range of 120–180°C.

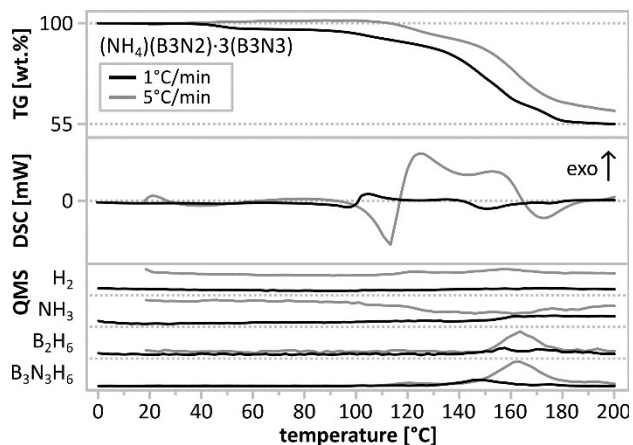


Fig.S7.1. TGA/DSC experiments of $(\text{NH}_4)(\text{B3N2}) \cdot 3(\text{B3N3})$ with scanning rates (1 K/min -black, 5 K/min -grey).

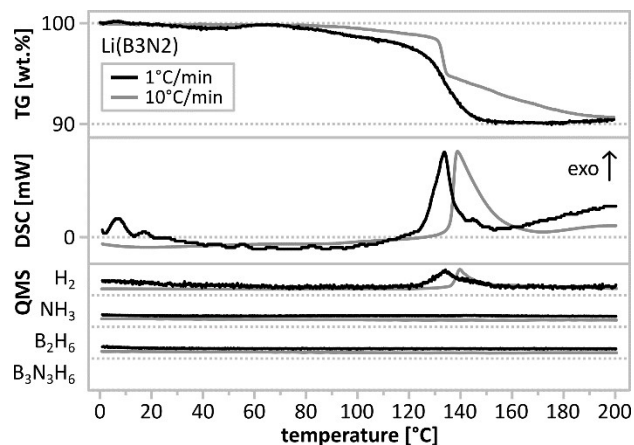


Fig.S7.2. TGA/DSC experiments of $\text{Li}(\text{B3N2})$ sample with different scanning rates (1 K/min -black, 10 K/min -grey).

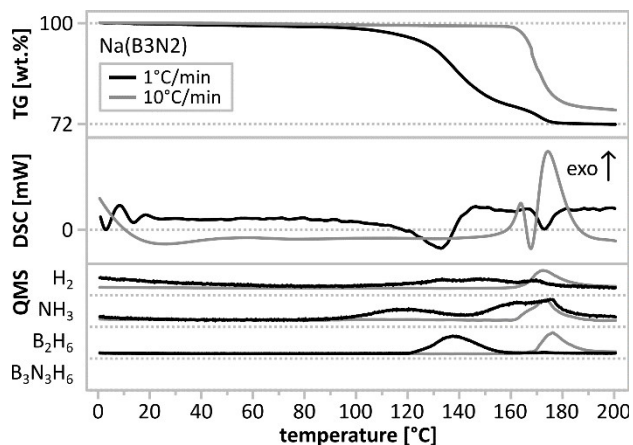


Fig.S7.3. TGA/DSC experiments of $\text{Na}(\text{B3N2})$ sample with different scanning rates (1 K/min -black, 10 K/min -grey).

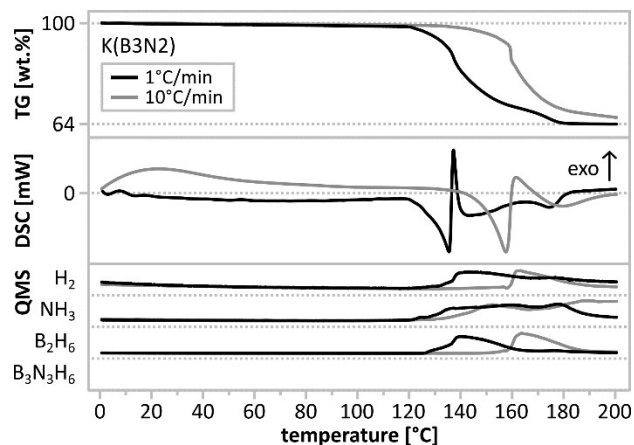


Fig.S7.4. TGA/DSC experiments of $\text{K}(\text{B3N2})$ sample with different scanning rates (1 K/min -black, 10 K/min -grey).

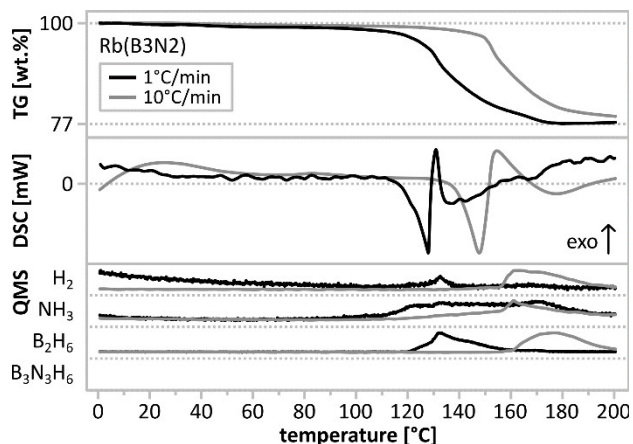


Fig.S7.5. TGA/DSC experiments of $\text{Rb}(\text{B3N2})$ sample with different scanning rates (1 K/min -black, 10 K/min -grey).

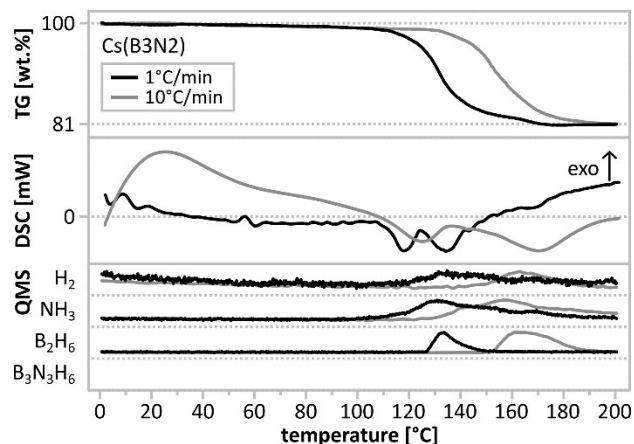


Fig.S7.6. TGA/DSC experiments of $\text{Cs}(\text{B3N2})$ sample with different scanning rates (1 K/min -black, 10 K/min -grey).

8. IR spectra of the products of thermal decomposition of M(B₃N₂) salts:

The thermal decomposition of (NH₄)(B₃N₂) leads to formation of boron nitride while decomposition of alkali metal M(B₃N₂) salts leads to formation of respective borohydrides.

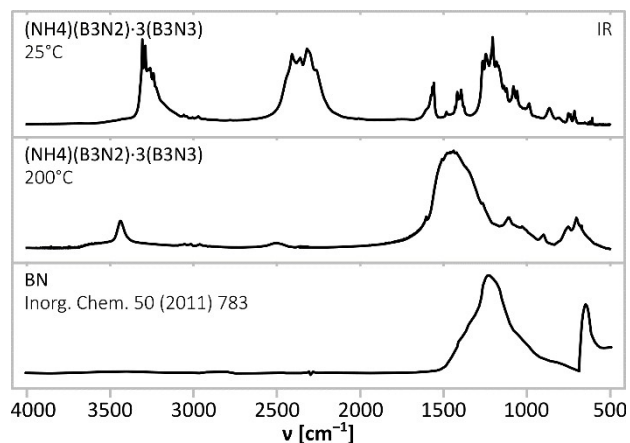


Fig. S8.1. IR spectra of the product of thermal decomposition of (NH₄)(B₃N₂)·3(B₃N₃) sample.

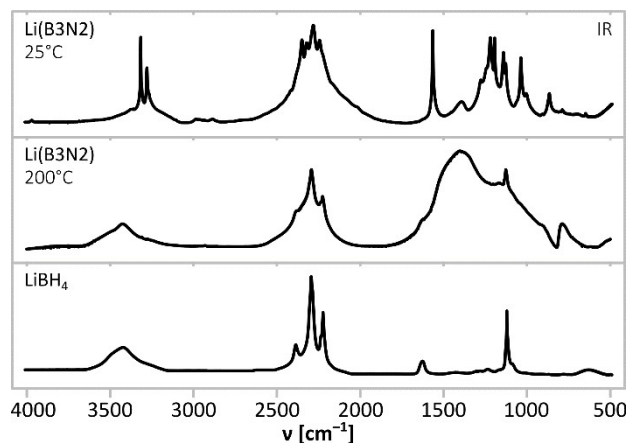


Fig. S8.2. IR spectra of the product of thermal decomposition of Li(B₃N₂) sample.

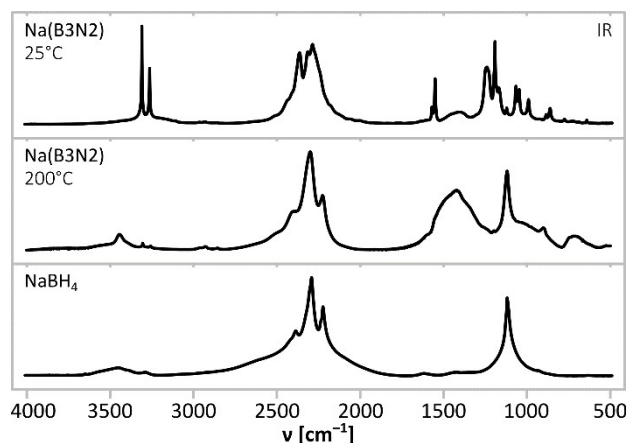


Fig. S8.3. IR spectra of the product of thermal decomposition of Na(B₃N₂) sample.

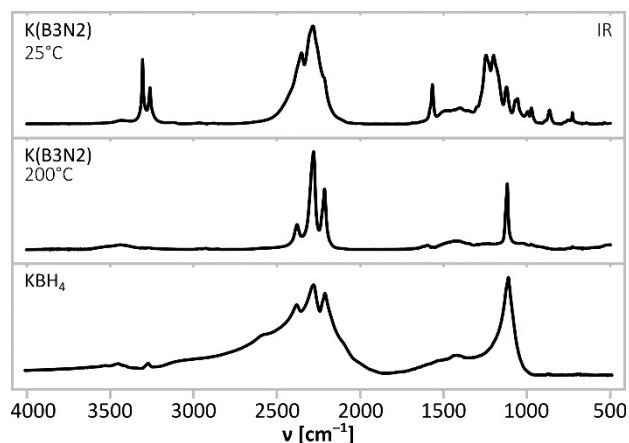


Fig. S8.4. IR spectra of the product of thermal decomposition of K(B₃N₂) sample.

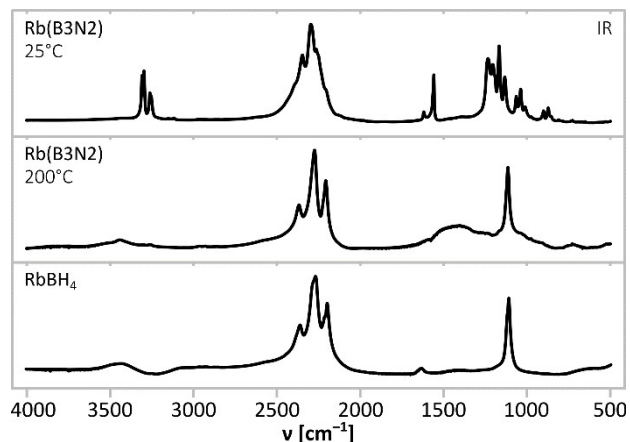


Fig. S8.5. IR spectra of the product of thermal decomposition of Rb(B₃N₂) sample.

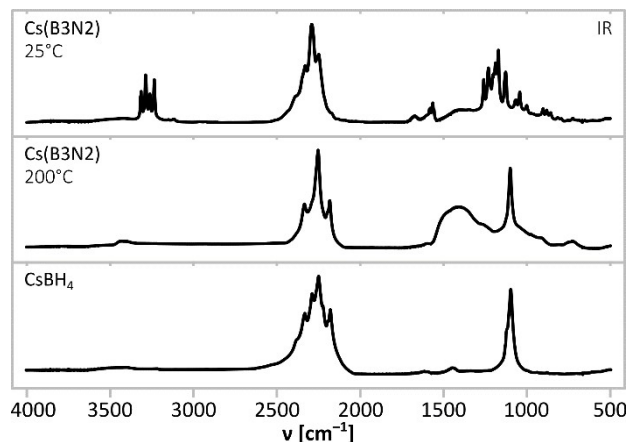


Fig. S8.6. IR spectra of the product of thermal decomposition of Cs(B₃N₂) sample.

9. Experimental crystal structure and Rietveld fit for $(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$:

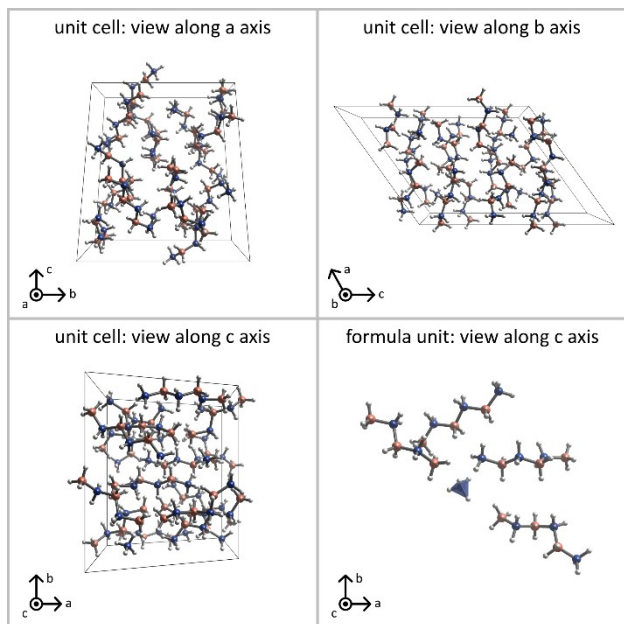


Fig. S9.1. Visualisation of the unit cell and formula unit of the main product: $(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$

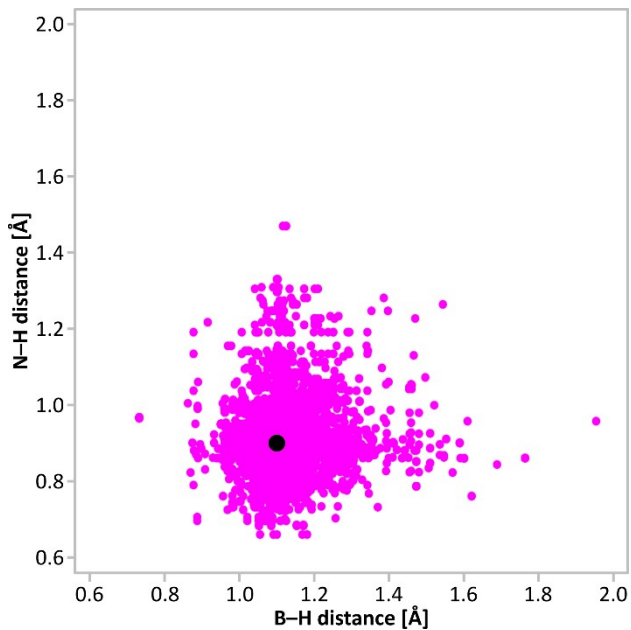


Fig. S9.2. Distribution of N-H and B-H distances in systems comprising both $[\text{NH}_x]$ and $[\text{BH}_x]$ groups found in structures in CSD database. Value of N-H and B-H distances in $(\text{NH}_4)(\text{B3N2})\cdot 3(\text{B3N3})$ marked with a dot.

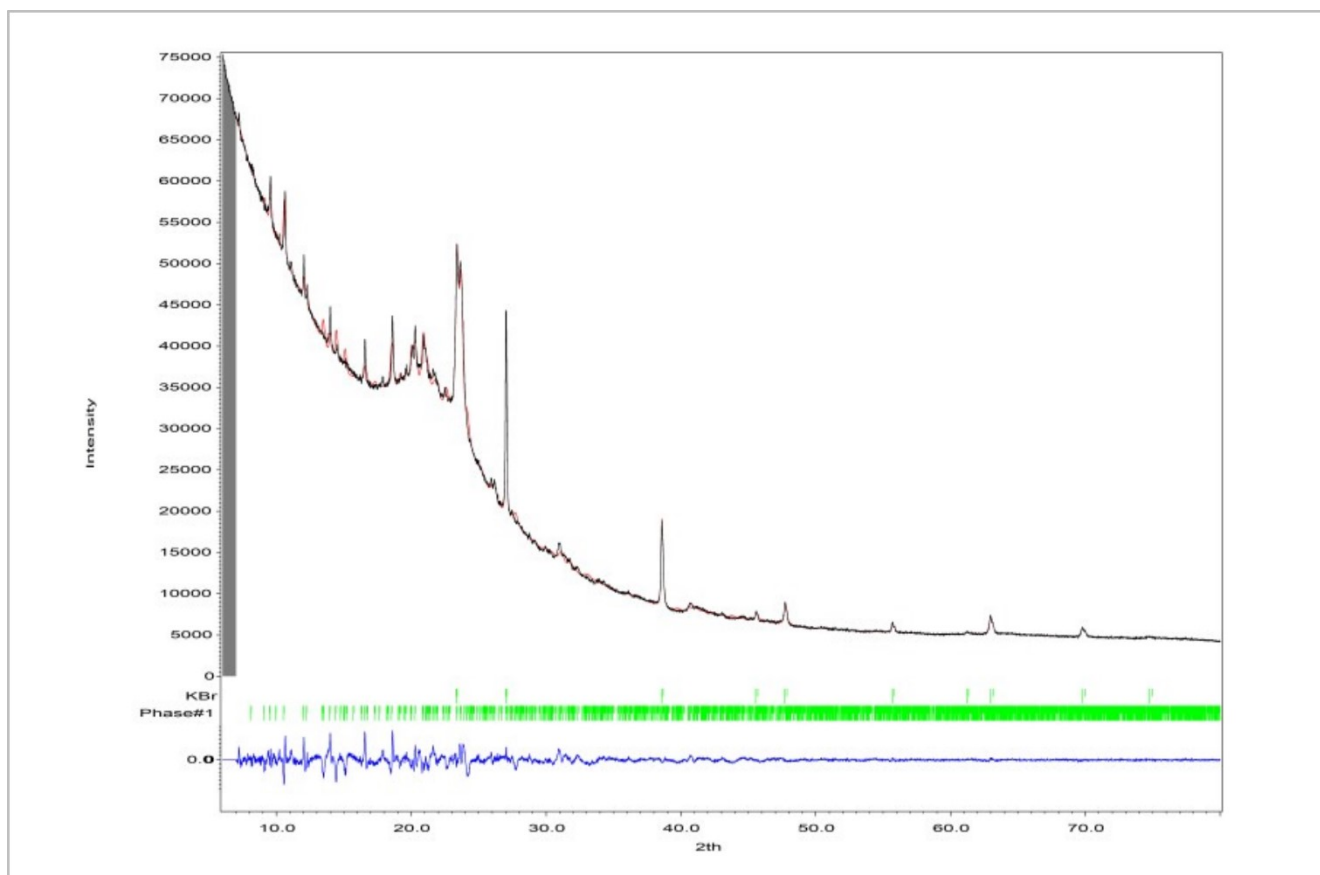


Fig. S9.3. Rietveld analysis of $(\text{NH}_4)(\text{B3N2})/3(\text{B3N3})$ powder pattern. $\text{CoK}_{\alpha 1,2}$, $\lambda = 1.78901 \text{ \AA}$.

10. Table with the closest H...H distances in the crystal structure of (NH₄)(B₃N₂)·3(B₃N₃):

Table S10. List of the closest H...H distances in experimental crystal structure of (NH₄)(B₃N₂)·3(B₃N₃). Listed only strong dihydrogen bonds, < 2 Å.

H atom1	H atom2	Length [Å]	Length-VdW [Å]	Neighboring groups
H8	H39	1.927	-0.473	B-H...H-N
H12	H29	1.928	-0.472	B-H...H-N
H6	H52	1.931	-0.469	B-H...H-N
H12	H53	1.935	-0.465	B-H...H-N
H15	H34	1.948	-0.452	B-H...H-B
H54	H42	1.949	-0.451	B-H...H-N
H1	H51	1.951	-0.449	B-H...H-N
H50	H64	1.971	-0.429	N-H...H-N
H19	H31	1.984	-0.416	B-H...H-B
H52	H64	1.987	-0.413	N-H...H-N
H17	H47	1.993	-0.407	B-H...H-N

11. Experimental and modelled NMR spectra for various possible compositions of the main product: NMR spectra were simulated for various discussed possible compositions of the main product to ease visual examination of the experimental spectra obtained by us and reported earlier by Ewing *et al.*, *Inorganic Chemistry* 52 (2013) 10690.

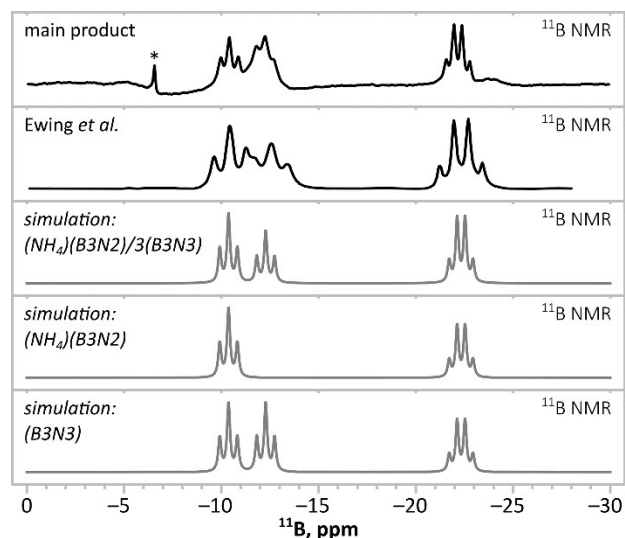


Fig. S11.1. Comparison of the experimental ^{11}B NMR spectra obtained here and reported by Ewing *et al.* (*Inorganic Chemistry* 52 (2013) 10690) with spectra simulated for various possible compositions of the main product: $(\text{NH}_4)(\text{B}_3\text{N}_2)\cdot 3(\text{B}_3\text{N}_3)$, $(\text{NH}_4)(\text{B}_3\text{N}_2)$ and (B_3N_3) .

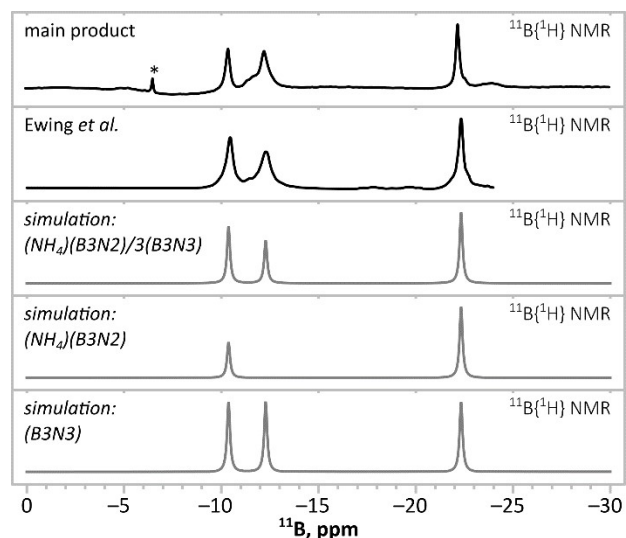


Fig. S11.2. Comparison of the experimental $^{11}\text{B}\{^1\text{H}\}$ NMR spectra obtained here and reported by Ewing *et al.* (*Inorganic Chemistry* 52 (2013) 10690) with spectra simulated for various possible compositions of the main product: $(\text{NH}_4)(\text{B}_3\text{N}_2)\cdot 3(\text{B}_3\text{N}_3)$, $(\text{NH}_4)(\text{B}_3\text{N}_2)$ and (B_3N_3) .

12. Results of DFT optimisation of modelled crystal structures:

NH₄(BH₃NH₂BH₂NH₂BH₃)-3(NH₃BH₂NH₂BH₂NH₂BH₃), unit cell optimised

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H60	H	0.69072	0.81080	0.32069	0.00000	Uiso	1.00
H61	H	0.98211	1.02676	0.62065	0.00000	Uiso	1.00
H62	H	1.15010	0.96615	0.72590	0.00000	Uiso	1.00
H63	H	1.06616	1.01664	0.76417	0.00000	Uiso	1.00
H64	H	1.03969	0.84687	0.74802	0.00000	Uiso	1.00
H65	H	1.00510	0.83410	0.64268	0.00000	Uiso	1.00
H66	H	0.79154	0.90640	0.53074	0.00000	Uiso	1.00
H67	H	0.83251	0.93665	0.65811	0.00000	Uiso	1.00
H68	H	0.83316	0.76230	0.68178	0.00000	Uiso	1.00
H69	H	0.81562	0.73071	0.58623	0.00000	Uiso	1.00
H70	H	0.61644	0.70373	0.54393	0.00000	Uiso	1.00
H71	H	0.60103	0.79436	0.45185	0.00000	Uiso	1.00
H72	H	0.61714	0.84255	0.56348	0.00000	Uiso	1.00
H73	H	0.23237	1.01081	0.84819	0.00000	Uiso	1.00
H74	H	0.37678	0.98260	0.95540	0.00000	Uiso	1.00
H75	H	0.33389	1.09739	0.93189	0.00000	Uiso	1.00
H76	H	0.27302	1.02156	0.95924	0.00000	Uiso	1.00
H77	H	0.92845	0.34726	0.29651	0.00000	Uiso	1.00
H78	H	0.85564	0.34756	0.17184	0.00000	Uiso	1.00
H79	H	0.83467	0.43444	0.22113	0.00000	Uiso	1.00
H80	H	0.97006	0.42799	0.25162	0.00000	Uiso	1.00
H81	H	0.40974	-0.08366	0.57357	0.00000	Uiso	1.00
H82	H	0.40191	-0.05445	0.48091	0.00000	Uiso	1.00
H83	H	0.32424	0.01191	0.49763	0.00000	Uiso	1.00
H84	H	0.47946	0.01857	0.58567	0.00000	Uiso	1.00
H85	H	0.17156	-0.02595	0.56907	0.00000	Uiso	1.00
H86	H	0.08359	0.07275	0.53422	0.00000	Uiso	1.00
H87	H	0.14981	0.01190	0.64117	0.00000	Uiso	1.00
H88	H	0.23755	0.07955	0.63623	0.00000	Uiso	1.00

(NH₃BH₂NH₂BH₂NH₂BH₃), unit cell optimised

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_audit_creation_method 'Materials Studio'
_symmetry_space_group_name_H-M 'P21/C'
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loop_

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-x,-y,-z

x,-y+1/2,z+1/2

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loop_

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_atom_site_occupancy

B1	B	0.61234	0.14161	0.69593	0.05000	Uiso	1.00
N1	N	0.74985	0.13798	0.78118	0.05000	Uiso	1.00
B2	B	0.80335	0.24827	0.83148	0.05000	Uiso	1.00
N2	N	0.94601	0.24008	0.90513	0.05000	Uiso	1.00
B3	B	1.00612	0.33672	0.97390	0.05000	Uiso	1.00
H1	H	0.54946	0.17792	0.71315	0.05000	Uiso	1.00
H2	H	0.60187	0.18975	0.63663	0.05000	Uiso	1.00
H4	H	0.76311	0.08260	0.82641	0.05000	Uiso	1.00
H5	H	0.80196	0.11174	0.76378	0.05000	Uiso	1.00
H6	H	0.78333	0.31438	0.77695	0.05000	Uiso	1.00
H7	H	0.75652	0.27053	0.86549	0.05000	Uiso	1.00
H8	H	0.98548	0.23333	0.87482	0.05000	Uiso	1.00
H9	H	0.97069	0.17067	0.94111	0.05000	Uiso	1.00
H10	H	1.00173	0.31692	1.03399	0.05000	Uiso	1.00
H11	H	0.95188	0.41887	0.93541	0.05000	Uiso	1.00
H12	H	1.11486	0.34492	1.00700	0.05000	Uiso	1.00
B4	B	0.50969	0.43569	0.71360	0.05000	Uiso	1.00
N3	N	0.46702	0.38819	0.62292	0.05000	Uiso	1.00
B5	B	0.33020	0.34237	0.55254	0.05000	Uiso	1.00
N4	N	0.25702	0.34528	0.59044	0.05000	Uiso	1.00
B6	B	0.17134	0.44485	0.56525	0.05000	Uiso	1.00
H13	H	0.50674	0.36814	0.75773	0.05000	Uiso	1.00
H15	H	0.45024	0.51405	0.70061	0.05000	Uiso	1.00
H16	H	0.47623	0.44544	0.58905	0.05000	Uiso	1.00
H17	H	0.52384	0.32667	0.63483	0.05000	Uiso	1.00
H18	H	0.33803	0.24984	0.53711	0.05000	Uiso	1.00
H19	H	0.27691	0.39738	0.48626	0.05000	Uiso	1.00
H20	H	0.31598	0.33436	0.65830	0.05000	Uiso	1.00
H21	H	0.20393	0.27761	0.56705	0.05000	Uiso	1.00
H22	H	0.22991	0.52690	0.58489	0.05000	Uiso	1.00
H23	H	0.12913	0.44167	0.60495	0.05000	Uiso	1.00

H24	H	0.08772	0.44511	0.48540	0.05000	Uiso	1.00
B7	B	0.27345	0.57022	-0.00885	0.05000	Uiso	1.00
N5	N	0.26635	0.66818	-0.06284	0.05000	Uiso	1.00
B8	B	0.22211	0.77997	-0.05360	0.05000	Uiso	1.00
N6	N	0.33300	0.83259	0.03608	0.05000	Uiso	1.00
B9	B	0.44363	0.86757	0.03827	0.05000	Uiso	1.00
H25	H	0.31804	0.49323	-0.01834	0.05000	Uiso	1.00
H26	H	0.33068	0.59327	0.06945	0.05000	Uiso	1.00
H28	H	0.34950	0.68144	-0.04834	0.05000	Uiso	1.00
H29	H	0.20852	0.64938	-0.12983	0.05000	Uiso	1.00
H30	H	0.19464	0.83592	-0.11562	0.05000	Uiso	1.00
H31	H	0.13695	0.76681	-0.05285	0.05000	Uiso	1.00
H32	H	0.36163	0.78334	0.08860	0.05000	Uiso	1.00
H33	H	0.29991	0.89825	0.04691	0.05000	Uiso	1.00
H34	H	0.41205	0.94339	-0.01110	0.05000	Uiso	1.00
H35	H	0.53224	0.88970	0.11504	0.05000	Uiso	1.00
H36	H	0.47107	0.79357	0.01180	0.05000	Uiso	1.00
B10	B	1.01084	1.09682	0.78046	0.05000	Uiso	1.00
N7	N	1.00571	0.98490	0.81574	0.05000	Uiso	1.00
B11	B	0.89496	0.91200	0.74411	0.05000	Uiso	1.00
N8	N	0.88698	0.81275	0.79116	0.05000	Uiso	1.00
B12	B	0.76080	0.75282	0.73529	0.05000	Uiso	1.00
H37	H	0.91116	1.14046	0.73976	0.05000	Uiso	1.00
H38	H	1.08492	1.15377	0.84346	0.05000	Uiso	1.00
H39	H	1.03831	1.08470	0.73181	0.05000	Uiso	1.00
H40	H	1.00795	0.99774	0.86972	0.05000	Uiso	1.00
H41	H	1.08598	0.94429	0.84318	0.05000	Uiso	1.00
H42	H	0.90823	0.88165	0.69059	0.05000	Uiso	1.00
H43	H	0.79993	0.96281	0.70640	0.05000	Uiso	1.00
H44	H	0.95736	0.76075	0.81376	0.05000	Uiso	1.00
H45	H	0.90141	0.83918	0.84788	0.05000	Uiso	1.00
H46	H	0.68411	0.81054	0.72636	0.05000	Uiso	1.00
H48	H	0.76990	0.66919	0.77091	0.05000	Uiso	1.00
N9	N	0.64892	0.47383	0.77073	0.05000	Uiso	1.00
H50	H	0.65955	0.53699	0.74190	0.05000	Uiso	1.00
H51	H	0.70708	0.41277	0.78183	0.05000	Uiso	1.00
H52	H	0.68082	0.49842	0.83280	0.05000	Uiso	1.00
N10	N	0.13759	0.54192	-0.04916	0.05000	Uiso	1.00
H53	H	0.10005	0.59620	-0.03170	0.05000	Uiso	1.00
H54	H	0.12833	0.46828	-0.02971	0.05000	Uiso	1.00
H55	H	0.07927	0.54020	-0.11808	0.05000	Uiso	1.00
N11	N	0.56770	0.02149	0.66238	0.05000	Uiso	1.00
H57	H	0.48257	0.01895	0.60065	0.05000	Uiso	1.00
H58	H	0.62717	-0.02152	0.65896	0.05000	Uiso	1.00
H59	H	0.55927	-0.01953	0.70490	0.05000	Uiso	1.00
N12	N	0.27916	0.22761	0.86024	0.05000	Uiso	1.00
H61	H	0.36599	0.19486	0.90001	0.05000	Uiso	1.00
H63	H	0.28219	0.29363	0.89304	0.05000	Uiso	1.00
H64	H	0.22046	0.17569	0.85841	0.05000	Uiso	1.00

13. Crystal structure (VBH)[B(C₆H₅)₄]

Table 13.1. Crystal structure parameters of (C₁₈H₃₉N₄PH)[B(C₆H₅)₄].

Compound	(C ₁₈ H ₃₉ N ₄ PH)[B(C ₆ H ₅) ₄]
K _α (Å)	1.54184 (Cu)
Temperature (K)	100(2)
Space group	<i>P</i> ¹
Z	4
<i>a</i> (Å)	11.7376(3)
<i>b</i> (Å)	19.5388(5)
<i>c</i> (Å)	20.5479(4)
α (°)	61.751(2)
β (°)	73.618(2)
γ (°)	89.605(2)
<i>V</i> (Å ³)	3937.71(18)
$\rho_{calc.}$ (g cm ⁻³)	1.118
$\mu_{exp.}$ (mm ⁻¹)	0.856
ϑ_{max} (°)	75.2030
<i>R</i> ₁	0.0695
<i>wR</i> ₂	0.2094
<i>Goof</i>	1.048
Crystal size (mm×mm×mm)	0.06 x 0.16 x 0.20
Crystal colour	colorless
CCDC No.	2195203

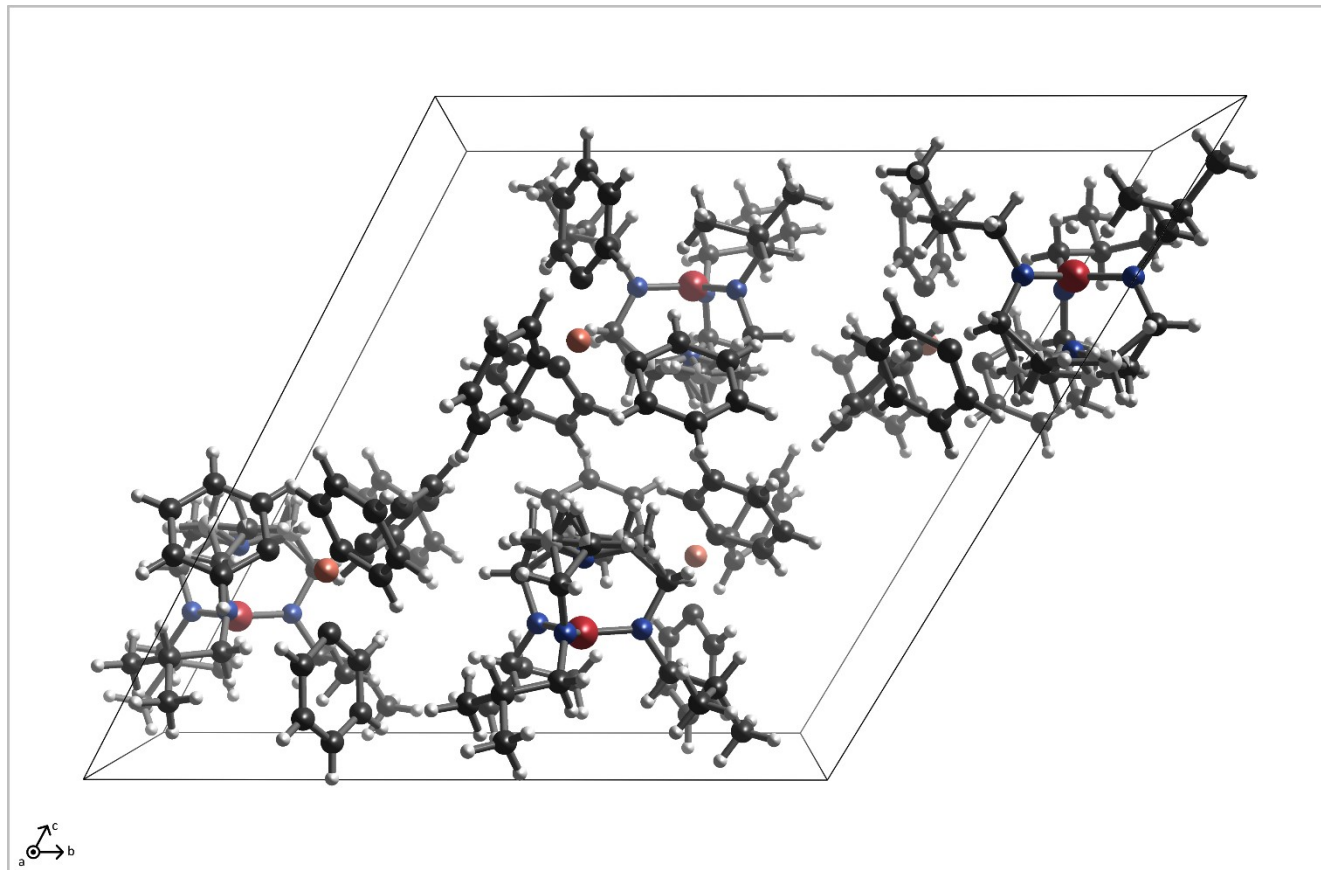


Fig. S13.2. Visualisation of the unit cell of the side product: (C₁₈H₃₉N₄PH)[B(C₆H₅)₄].

14. Molecular DFT calculations

Summary of energies and coordinates of investigated species. Three top energy values (yellow) are for r2SCAN-3c method, while the bottom (green) is single point energy obtained from ω B97X-V/def2-TZVP/D4 method. ν_{im} denotes imaginary mode in transition state. All values are calculated with CPCM solvent model.

1. Decomposition of $\text{NH}_4^+(\text{B3N2})^-$ into $\text{N}_3\text{B}_3\text{H}_{14}$ and H_2 in THF:

$\Delta\Delta G^* = 69 \text{ kJ mol}^{-1}$ (17 kcal mol⁻¹)

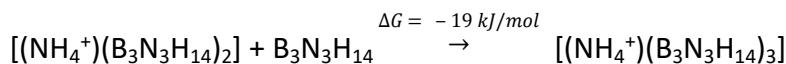
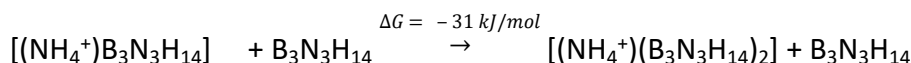
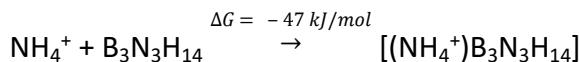
Substrates: $\text{NH}_4^+(\text{B3N2})^-$			Transition state; $\nu_{im} = -320 \text{ cm}^{-1}$			Product: $\text{B}_3\text{N}_3\text{H}_{14} + \text{H}_2$		
E	-248,4092007			-248,37501			-248.43450178	
H	-248,2034667			-248,1777712			-248.23417146	
G	-248,25242			-248,2261315			-248.28262085	
E	-248,7570143			-248,7228562			-248.78438156	
H	3.90244901624217	9.42504815708469	H	0.80170974429476	-0.58938294659471	H	3.33113309198864	10.46453928912834
10.04906651434456			0.92112336173798			9.56444211326653		
H	6.83353381108372	12.91820843573510	H	2.66969558973920	-0.70433922319661	H	5.02572135591715	9.88106982686070
9.73238715788029			0.78167122769226			11.00317297272730		
H	4.43719534906216	11.69986703130003	H	0.36495594143596	1.86894382787090	H	3.84762900907083	12.88688025383359
9.00119983920167			0.43906370692944			9.07733965028903		
N	3.35862674414461	10.22123538619771	N	0.00846434717645	0.05845216322118	N	2.69579559798272	11.13627610950967
10.38092877505048			0.95595287658311			10.00054716362470		
H	2.08664145131618	10.34421890053314	H	-0.46661747709990	-0.45576680629441	H	1.60845648636007	11.68106625038090
8.45810330345879			1.23733567211541			8.05540244641130		
H	3.22180904508244	10.04317038224818	H	-0.41729221181681	-0.01334737771859	H	2.49963888331001	10.74498040627554
11.37568405931101			1.87670189060266			10.92350538263724		
B	4.20224212731816	11.56612718028035	B	0.58122003140657	1.57117577845870	B	3.51259161818743	12.49774016115045
10.18623053147915			0.72123857557494			10.17790436497004		
B	6.55213280741590	12.71639843934119	B	2.97933120513288	0.36151179986559	B	5.65843982182757	10.86344232062214
10.90468983874320			0.25811913111074			10.63727985825128		
B	1.9392706092242	10.20317854563982	B	-0.94353633117531	-0.10910243011772	B	1.36893817199600	11.28740801966533
9.66256757334818			0.20385095468940			9.17325291368432		
H	1.40424007204310	9.10583082681292	H	-1.49265637421284	-1.68457994034317	H	3.48359085994021	7.92991002197611
9.88380960719910			0.39895354380707			9.88980668229961		
H	7.56276675279290	12.45915048747947	H	4.16824500072783	0.57030839062130	H	6.73919742229469	10.89951332418051
11.55101467613406			0.43956561509864			11.20954289014780		
N	5.56960451671451	11.43711896889493	N	2.13623029173253	1.54375508330920	N	4.81971458964296	12.18469419893546
10.97525341642985			0.96807802207883			11.02628277852379		
H	6.09013476343527	10.62675613008629	H	2.33890850007003	1.54463056086151	H	4.60121352721698	12.15291522456638
10.64128772462874			1.96733462842189			12.02103156725538		
H	5.96807450372491	13.67664017222668	H	2.69610438759599	0.35743408948199	H	5.80706285350062	10.84941951855814
11.38440241473068			0.92804788099191			9.42243569715587		
H	3.56066563888645	12.48763495472371	H	0.00579516741906	2.30510966664070	H	2.80830056613616	13.29993025663493
10.65131369665503			1.49538594886243			10.76433177874059		
H	1.23545113672535	11.09142464973338	H	-1.96854284261449	0.49577010842976	H	0.57012404565763	11.97005305809256
10.11752298391511			0.14690711616890			9.77256642902258		
H	5.38796126729011	11.23812726099694	H	2.50422044235709	2.43460208115741	H	5.43257301727170	12.99249149886208
11.95850258491732			0.63870427237763			10.93386988021092		
H	0.03017525741386	9.41187502392585	H	-1.97107733977835	-1.68635946701670	H	2.96846927381031	7.43417532233740
8.99311863252098			0.21018432273963			9.68509172992716		
N	-0.84655043457016	9.62021124500224	N	-3.32481367679018	-1.94571655131046	N	0.70770447481291	9.81174262907880
8.47317357925754			1.83094216696788			9.02105091951508		
H	-0.98311320521337	8.93077090534359	H	-2.99674476800290	-2.63566492277144	H	1.30222300827910	9.15396018162515
7.73492188634211			2.49923875749211			8.51678164259233		
H	-0.77111209349771	10.54927092937775	H	-3.40156928586551	-1.07181182499489	H	-0.16479473110288	9.87145307464581
8.05982943839386			2.34151699858190			8.49817611963089		
H	-1.63920585683336	9.59758391303486	H	-4.27203034173204	-2.21562205955953	H	0.47927038729829	9.37618697907695
9.11384164655721			1.58574151727162			9.91503489951346		

2. Decomposition of $\text{NH}_4^+\text{-BH}_4^-$ into NH_3BH_3 and H_2 in THF

$\Delta\Delta G^* = 58 \text{ kJ mol}^{-1}$ (14 kcal mol⁻¹)

Substrates: $\text{NH}_4^+\text{-BH}_4^-$			Transition state; $\nu_{\text{im}} = -192 \text{ cm}^{-1}$			Product: $\text{NH}_3\text{BH}_3+\text{H}_2$		
E	-84,34346903		-84,30381255		-84.36551277			
H	-84,2516124		-84,21922254		-84.27782184			
G	-84,283986		-84,25915181		-84.31102158			
E	-84,47225417		-84,43542144		-84.49696412			
B	-4.82982252170536	1.72053779034281	B	-1.50287303418764	0.63876813464282	B	-3.98602337625892	2.38548091238253
0.02367363739606			0.44569224703493			0.70841812911015		
N	-1.77367358139469	1.45242795552326	N	1.67096820610126	-0.11608199448868	N	-2.51079344393880	1.86489815834672
0.01414769812632			0.13044477456475			0.25450034458934		
H	-4.45397488619308	2.87264765918443	H	-0.83238957131596	1.55369006602716	H	-3.85675245842353	3.53856657893541
0.17619394703757			0.08036684171849			1.07580410577538		
H	-4.55519637579775	1.38472415937971	H	-1.22784572791200	0.05935014085067	H	-4.35417639142868	1.65869397734892
1.17295743090450			1.44994762688896			1.61284039523883		
H	-4.25235789710763	0.96441599040701	H	-1.06496706453873	-1.52511621206970	H	-3.85295516368822	-0.63689270861993
0.77283311497087			1.52727501597430			1.69803943331986		
H	-6.02965143542113	1.61290340024765	H	-2.48352228729422	0.34162613383935	H	-4.70773376077587	2.29773626490969
0.17909899434801			0.16220977007508			0.26737675717421		
H	-1.29474234440937	0.55767880628941	H	2.58988863001572	-0.54781171926806	H	-2.52845718183990	0.89694632011587
0.07770920647177			0.11238143578636			0.05955104624972		
H	-1.35058478382702	1.98878864032109	H	1.75573313094210	0.74428319630006	H	-2.12916987739779	2.41384498897070
0.77020211262428			0.40121193505143			0.51299544552452		
H	-1.68565617843929	1.97352767555258	H	1.51315884032439	0.16792692146426	H	-1.83480908027833	1.91448540590981
0.85709379486076			1.09195828901368			1.01379819684301		
H	-2.79574901200452	1.29909030865197	H	-0.41815112213492	-1.31663466729790	H	-3.26053828226938	-0.50701751240000
0.21895774962602			1.22009449390636			2.12739803638837		

3. Complexation of NH_4^+ with $\text{B}_3\text{N}_3\text{H}_{14}$ in THF:



NH_4^+		$\text{B}_3\text{N}_3\text{H}_{14}$		$(\text{NH}_4^+)(\text{B}_3\text{N}_3\text{H}_{14})$	
E	-56,99274527		-247,2442444		-304,2698601
H	-56,9393111		-247,0574479		-304,0284483
G	-56,9627515		-247,0989134		-304,0796272
B	-1.39495813257824	0.33929967056958	B	-1.39495813257824	0.33929967056958
1.06916287991551			1.06916287991551		
N	-2.20445880584144	-0.36321058405875	N	-2.20445880584144	-0.36321058405875
0.14819201267700			0.14819201267700		
B	-1.59042905375964	-1.64827802036495	B	-1.59042905375964	-1.64827802036495
0.85488220886687			0.85488220886687		
H	-2.42459402852973	0.34918195621964	H	-2.42459402852973	0.34918195621964
0.84294611821610			0.84294611821610		
H	-3.11305065794379	-0.63800421413397	H	-3.11305065794379	-0.63800421413397
0.21929251238882			0.21929251238882		
N	-0.06459922819024	-1.30058957981328	N	-0.06459922819024	-1.30058957981328
1.26023756554426			1.26023756554426		
H	-2.21175721956632	-1.91226478803726	H	-2.21175721956632	-1.91226478803726
1.86369199693538			1.86369199693538		
H	-1.61411457062865	-2.57031270870798	H	-1.61411457062865	-2.57031270870798
0.06271490095861			0.06271490095861		
B	1.10830352290340	-2.09705479656034	B	1.10830352290340	-2.09705479656034
0.54727018566218			0.54727018566218		
H	0.09168472465526	-0.30132058673718	H	0.09168472465526	-0.30132058673718
1.11357154655471			1.11357154655471		
H	0.04509184170150	-1.44980278699727	H	0.04509184170150	-1.44980278699727
2.25974628123964			2.25974628123964		
N	0.79935921599377	-2.03432455998856	N	0.79935921599377	-2.03432455998856
1.03698691841691			1.03698691841691		
H	2.17293231747920	-1.54970286550128	H	2.17293231747920	-1.54970286550128
0.73099839281503			0.73099839281503		
H	1.08618816022553	-3.25721650284473	H	1.08618816022553	-3.25721650284473
0.880846464593162			0.880846464593162		
H	0.61101004728679	-1.07863093837470	H	0.61101004728679	-1.07863093837470
1.36145786307852			1.36145786307852		
H	1.57173493962069	-2.42246771972654	H	1.57173493962069	-2.42246771972654
1.57133490317869			1.57133490317869		
H	-0.04928905465051	-2.55720839413654	H	-0.04928905465051	-2.55720839413654
1.25768150201177			1.25768150201177		
H	-0.28111522606123	0.66358422953666	H	-0.28111522606123	0.66358422953666
0.65998116432115			0.65998116432115		
H	-1.31868792080042	-0.47796636605897	H	-1.31868792080042	-0.47796636605897
1.97406236201134			1.97406236201134		
H	-2.00211658561593	1.32831350371642	H	-2.00211658561593	1.32831350371642
1.42039218717870			1.42039218717870		
B	-3.21977651162233	0.85698108237180	B	-3.21977651162233	0.85698108237180
0.27308569725256			0.27308569725256		
N	-1.66893806115107	0.62845162102579	N	-1.66893806115107	0.62845162102579
0.63072146820986			0.63072146820986		
B	-1.22135497078232	-0.87641467692306	B	-1.22135497078232	-0.87641467692306
0.72255129427725			0.72255129427725		
H	-1.10626921505391	1.12098785937935	H	-1.10626921505391	1.12098785937935
0.06203637374238			0.06203637374238		
H	-1.47643446617394	1.11402250507021	H	-1.47643446617394	1.11402250507021
1.50703374934660			1.50703374934660		
N	0.33749177660046	-0.96859107958974	N	0.33749177660046	-0.96859107958974
1.11226174481475			1.11226174481475		
H	-1.86416002898519	-1.43228023661467	H	-1.86416002898519	-1.43228023661467
1.59823427617944			1.59823427617944		
H	-1.36187114516945	-1.40716194741124	H	-1.36187114516945	-1.40716194741124
0.36253491174973			0.36253491174973		
B	1.32095076580676	-0.20995473369159	B	1.32095076580676	-0.20995473369159
0.14019937475252			0.14019937475252		
H	0.46115017022099	-0.65613878675463	H	0.46115017022099	-0.65613878675463
2.07674620883883			2.07674620883883		
H	0.556942848484783	-1.96517601659965	H	0.556942848484783	-1.96517601659965
1.13566195946173			1.13566195946173		
N	2.83902470620326	-0.55219074176528	N	2.83902470620326	-0.55219074176528
0.58093842560065			0.58093842560065		
H	1.17263846843348	-0.61221966628137	H	1.17263846843348	-0.61221966628137
0.99014988340315			0.99014988340315		
H	1.18657880105997	0.98698169153638	H	1.18657880105997	0.98698169153638
0.25440268250575			0.25440268250575		
H	3.06568043885840	-1.54479368479119	H	3.06568043885840	-1.54479368479119
0.51739517720339			0.51739517720339		
H	3.49739240107152	-0.07026605038333	H	3.49739240107152	-0.07026605038333
0.03028309752868			0.03028309752868		
H	3.06199852597102	-0.25593704344891	H	3.06199852597102	-0.25593704344891
1.53111251277506			1.53111251277506		
H	-3.43171846380942	0.33658371613133	H	-3.43171846380942	0.33658371613133
0.82187866171214			0.82187866171214		
H	-3.89897346060714	0.33851958620337	H	-3.89897346060714	0.33851958620337
1.14504074546465			1.14504074546465		
H	-3.44547790769362	2.05108955515805	H	-3.44547790769362	2.05108955515805
0.19294875098966			0.19294875098966		
N	-4.19853954399310	-2.19938696994099	N	-4.19853954399310	-2.19938696994099
0.20783894429267			0.20783894429267		
H	-3.52529722070539	-2.50246500814299	H	-3.52529722070539	-2.50246500814299
0.49781445334120			0.49781445334120		
H	-3.96879100404545	-1.21767942521280	H	-3.96879100404545	-1.21767942521280
0.46170043759037			0.46170043759037		
H	-4.13204870820608	-2.80368861437980	H	-4.13204870820608	-2.80368861437980
1.02616914480727			1.02616914480727		
H	-5.14203701337525	-2.24475228004498	H	-5.14203701337525	-2.24475228004498
0.17578091901243			0.17578091901243		

[(NH ₄ ⁺)(B ₃ N ₃ H ₁₄) ₂]			[(NH ₄ ⁺)(B ₃ N ₃ H ₁₄) ₃]		
E	-551,5445856		-798,8162026		
H	-551,1147622		-798,1981365		
G	-551,1904521		-798,2966141		
B	-3.11986004524949	0.76215573551633	B	-2.74981161074899	1.13506870533074
	0.02687135332424			0.04989699084143	
N	-1.64743872392956	0.89634785510820	N	-1.34940611630615	1.07385926380832
	0.60709391730230			0.84182842466040	
B	-0.95432825071799	-0.48071330127820	B	-0.81061774660508	-0.38639732420989
	0.91732745999767			1.06315796339265	
H	-1.06995608330084	1.44650845815148	H	-0.66650792187929	1.62932486290522
	0.02712533281041			0.32858165747400	
H	-1.71873901349812	1.46728560692289	H	-1.46284531086275	1.56116119744841
	1.44947723228699			1.73063652499563	
N	0.47033775294366	-0.24051847493337	N	0.57675001228961	-0.35153357885946
	1.63260493716679			1.88714862241926	
H	-1.65428349734687	-1.11751677623843	H	-1.61984288398251	-1.01781398327157
	1.68472030162008			1.71889096085587	
H	-0.75053890243271	-1.07106283611923	H	-0.58374997293807	-0.89816420680780
	0.12435636386542			0.01373223090375	
B	1.48818526612186	0.65818911686037	B	1.72666993876936	0.50228871997024
	0.83582363840581			1.23461000889558	
H	0.31133548030767	0.12699928158070	H	0.38889378217410	-0.04797714095678
	2.57210030911344			2.84396296812920	
H	0.85741975382638	-1.17298510341005	H	0.86026420651351	-1.32832245788036
	1.78399405827553			1.97737864644006	
N	2.88966489630694	0.68060433037163	N	3.07637081888178	0.29208201758057
	1.64646543644201			2.10588302835937	
H	1.70601359493412	0.17279780900042	H	1.95523360987935	0.10359152059523
	0.24979372499201			0.11646842723494	
H	1.09712992782956	1.80254757333990	H	1.46129000554976	1.68212480835814
	0.78917772689553			1.28757635690962	
H	3.31589152690344	-0.24107763898463	H	3.39478899054453	-0.67669817060800
	1.74393456504465			2.12848472101265	
H	3.56994428777303	1.26116133692236	H	3.84011466320670	0.83599593678631
	1.15734733146355			1.70614543593611	
H	2.80549315901031	1.06904083233800	H	2.98442256517201	0.59733059377464
	2.58588646998912			3.07479171446153	
H	-3.03063062003186	0.07694016713273	H	-2.59259312758043	0.56489389994200
	1.04345203659699			1.02455815577740	
H	-3.84589878907643	0.23149787460825	H	-3.59843979733198	0.57430268956494
	0.79955434831813			0.72731733767542	
H	-3.53746828831985	1.86909100803453	H	-3.05426555517580	2.30119239584373
	0.32244897280417			0.14326192690213	
N	-3.45593303413246	-2.55143517266849	N	-3.40436137351559	-2.13644120425613
	0.14033883064426			0.50669335442596	
H	-2.55697076655568	-3.01909366809907	H	-2.57728458146402	-2.73513126376488
	0.06056447026998			0.37275235591159	
H	-3.30673228405240	-1.56067303199716	H	-3.13644860029184	-1.14046752519274
	0.39738395926921			0.54337673378521	
H	-3.90968257808262	-3.05656578803449	H	-3.89997429792008	-2.38875096047791
	0.90168217107346			1.37254631816984	
H	-4.04476736850209	-2.59270181168214	H	-4.04702563559814	-2.27203497061428
	0.69019012531722			0.27312186273353	
H	-3.82192485569185	-5.32746364953546	H	-3.81008640524010	-4.99808027688050
	1.24450761520394			1.43413181206637	
H	-2.46847979736275	-5.01104751883461	H	-2.46627212068912	-4.68273947132610
	1.11916305394685			0.87693890578797	
H	-3.41629846923204	-7.31979296141109	H	-3.52263629591118	-7.00153312302894
	2.68133610226729			2.88477998027164	
H	-1.94876724554137	-6.93734048169796	H	-2.00302555857182	-6.67871063104786
	0.49177651996570			0.67837108099437	
B	-2.75212938380756	-5.44313235101093	B	-2.75469234431700	-5.16300558872806
	1.81800461056669			2.01624256438472	
N	-2.95776717956341	-6.47890574909142	N	-3.01619508975010	-6.18712228916998
	3.03633932452669			3.23563856844959	
H	-3.65171776246029	-6.04848564629801	H	-3.68458287872029	-5.71051183697072
	3.64894436091237			3.84278135751539	
B	-1.42379955279695	-5.09634391840186	B	-1.418884981600118	-4.82631048042547
	0.49231844757904			0.26476025904556	
N	-1.68392455755169	-6.00518496314754	N	-1.70356317503137	-5.76322874828479
	0.81059307599722			1.01359192510141	
H	-0.55217074105882	-5.61356081152254	H	-0.57335403663875	-5.35490918828134
	1.17163746817266			0.97075549683143	
H	-2.52547206719916	-8.65827825841582	H	-2.71628367952942	-8.34801825953255
	4.80551620997253			5.06246571754249	
H	-2.35669341075630	-4.39334375164554	H	-2.31801520910738	-4.13414185268946

2.28589791889604		2.49211351480559	
H -1.04287242437314	-3.99147398816093 -	H -0.99875633728959	-3.75020407225487
0.11415438400579		0.14721930785591	
H -0.79313574210667	-6.14293184221405	H -0.81898062998760	-5.93950733225117
1.28627891446148		1.48781885307113	
N -2.07520983715054	-7.78824193218884	N -2.21748039953372	-7.49663483182302
5.09079019380927		5.32151034200395	
B -1.65577389126226	-6.84239116581788	B -1.73874971700232	-6.61860825098380
3.84586197545988		4.04716334847625	
H -2.70328681150698	-7.33322572295090	H -2.82196839827921	-6.98267750923446
5.75332094676226		5.96286312100030	
H -0.89859885484810	-7.49109144329099	H -1.03283448147575	-7.33842487583590
3.15990329694254		3.37678118669928	
H -1.24659939125443	-8.05548721831688	H -1.40887868247612	-7.79272142807358
5.62109895866372		5.86693476910004	
H -1.16051292990348	-5.83916324548888	H -1.17053704595792	-5.64147981190702
4.30372780502526		4.47604367237351	
		H -5.13740420451449	-2.80824886504039
		2.74056852419030	
		H -6.11446342232350	-3.53278884667645
		0.31514530084987	
		B -5.90126918534716	-1.84758834821190
		2.75927419589252	
		H -7.68059612905569	-2.97327718259136
		2.28150627733491	
		H -8.57714974848525	-3.79399619215984 -
		0.06183326675459	
		B -6.84095363776522	-2.55542872736697
		0.34438950948201	
		H -6.28273634409661	-1.64663970002707
		3.90214453335991	
		N -7.17570486272842	-2.20499444496955
		1.83987345291339	
		H -7.90257766667606	-3.17502996978521 -
		1.39241941720272	
		N -8.19720622167069	-2.92734977357646 -
		0.44727410212904	
		H -5.36537056600805	-0.85245632484656
		2.29780838547558	
		H -6.34752394340042	-1.58851122878829 -
		0.20186625243433	
		H -7.82216095844643	-1.41808169367191
		1.88021372791635	
		H -10.97588824126134	-3.16602914209636 -
		1.02037649484354	
		B -9.29305779154673	-1.79719865074552 -
		0.49304922415965	
		H -9.73710725461541	-1.58952701757240
		0.61349924408549	
		N -10.52988722258301	-2.32846759051453 -
		1.39546025808854	
		H -10.26548173713014	-2.53901891352129 -
		2.35811410777236	
		H -8.84889193316731	-0.81207066927045 -
		1.03384562336816	
		H -11.25703022974726	-1.61600716014585 -
		1.44735467844314	

5.22157467557784			3.07783866514610			3.83898227237947		
H -6.96226201463519	-3.41128590407633	-	H -1.69333711161359	-3.22392646232934	-	H -5.97249543001921	-3.21556352408381	-
4.12176369941528			2.80958681859341			3.14767814507552		
H -6.35142528707451	-0.95930807812568	-	H -0.37042604105596	-3.37326859774143	-	H -5.57232851876223	-0.22801784799885	-
5.98222451542640			5.65156596617817			4.01994756819247		
B -6.35172444465070	-2.38313432143847	-	B -0.94567412074293	-2.78513176896134	-	B -5.52449054000253	-2.10223786223792	-
4.35687543646739			3.65586041393652			2.93965571883580		
H -8.60755635056070	-0.56283021894363	-	H -2.69392970169147	-2.33938941223529	-	H -8.07182690560947	-0.57602842304007	-
4.81472656287485			5.97739335689899			3.49243412110827		
N -4.80338287677326	-2.71326440319982	-	N 0.58660425812486	-2.98666254089487	-	N -3.93328496872300	-2.22803975364388	-
4.41927733467855			3.06553309497818			2.90783770775522		
H -4.73486530030004	-4.55508965734203	-	H 0.05949541935957	-2.95950057009183	-	H -3.96825571242119	-4.22214453990551	-
3.03093568304795			0.83286863301398			1.73933168490171		
H -4.27161204533088	-1.85747347804267	-	H 1.27880935911173	-2.59710156943196	-	H -3.49645277261708	-1.30700222432812	-
4.57559666598130			3.70624222431574			2.80554369750814		
B -4.26077632833148	-3.43237833981028	-	B 0.81826460767882	-2.54834375979215	-	B -3.22441228592620	-3.29147538817247	-
3.09408702453409			1.64976872065530			1.94301526405673		
H -3.03461644000549	-3.48509356617226	-	H 1.78057880311700	-1.90607287771543	-	H -2.18392765178568	-3.63658532615937	-
3.12100436489471			1.39264918773973			2.46324193638247		
H -6.51321977521847	-1.53853442835827	-	H -1.09861028289508	-1.61087242559697	-	H -5.94311746890761	-1.60454101943400	-
3.49066042057411			3.88962482684422			1.91208996255461		
H -4.63216393706986	-2.76209094657814	-	H -0.20992353310031	-0.19226734744219	-	H -5.21080190892674	-1.17737105863830	-
2.13999463894711			2.26354519341510			0.13161202749128		

5. Dehydrogenation and cyclisation of B₃N₃H₁₄ in THF:

$$\Delta\Delta G^* = 178 \text{ kJ mol}^{-1} (42 \text{ kcal mol}^{-1})$$

Substrates: B ₃ N ₃ H ₁₄			Transition state; $\nu_{\text{im}} = -204 \text{ cm}^{-1}$			Product: B ₃ N ₃ H ₁₂ + H ₂		
E	-247,2662843		- 247,1909889			-247.2758594		
H	-247,0800285		-247,0140584			-247.09599548		
G	-247,122004		-247,056871			-247.1401522		
E	-247,6042227		-247,5260112			-247.615999430351		
B	-1.54021257597342	0.33518671174295	B	-1.48933548070614	1.69028293241534	B	-1.05546083732234	-0.41491796481028
1.07577370926230			0.90952371791048			0.84291476154349		
N	-2.24767204444348	-0.33042947009478	N	-1.85061940734333	0.92150461988765	N	-2.23684857660377	-0.91969834364003
0.20922419342796			0.31643842195996			0.11330653157109		
B	-1.57323874628140	-1.62812148879007	B	-0.93913373012405	-0.47915801785037	B	-1.80421480810396	-1.92209161250638
0.82251153718234			0.59589234652356			1.28561819864657		
H	-2.36885303399078	0.37840621104163	H	-1.79710832135936	1.48517055476195	H	-2.69636405569349	-0.10002906118230
0.93273756662568			1.16522278719921			0.50417598451900		
H	-3.20055415994328	-0.58220939306202	H	-2.80211009606767	0.55991781489787	H	-2.96033401902721	-1.37634735397911
0.04784336455853			0.25633786656530			0.43780385624553		
N	-0.05306778392926	-1.27039042657532	N	0.48767158852963	-0.05588078907379	N	-0.30373233495498	-1.56140637107804
1.20802912397333			1.05889053135279			1.66317454603026		
H	-2.16130522345302	-1.97512119634289	H	-1.51968856915579	-1.05946462599117	H	-2.53519517305881	-1.77642429940327
1.83132178985886			1.48427034984702			2.24736246474465		
H	-1.59490616256961	-2.50272191323365	H	-0.93573380788072	-1.07462538528650	H	-1.82745726278180	-3.06065148394025
0.02462030764529			0.45178627304930			0.8572865209533		
B	1.08471963587744	-2.16782663650063	B	1.64043202083740	-1.18915138157626	B	0.75196939449638	-1.85385643859558
0.57211131460446			0.70463415973849			0.49921432702027		
H	0.13118542431796	-0.29102437783617	H	0.81827190266837	0.80016205317940	H	-0.24282395674732	-0.59079310583134
0.97948961602105			0.61449110767296			1.96914911118958		
H	0.05330732904628	-1.32118846610820	H	0.50251634779866	0.13922908914621	H	-0.03854229702164	-2.10426766553993
2.21877626243848			2.05815800567083			2.48205568866728		
N	0.95090690216052	-2.02817310679335	N	1.87528485855925	-1.13969342475653	N	0.01646247417678	-1.58592224999171
1.02344567239229			0.80358309791266			0.89373437355813		
H	2.18331899032538	-1.75744906541819	H	2.63650492534253	-0.84869498524101	H	1.68971120870533	-1.08447644258186
0.88629277415234			1.33498074665712			0.59086250003938		
H	0.90558643810501	-3.32963856256439	H	1.19011999695609	-2.25120319562546	H	1.09061109315911	-3.01978437925855
0.86323586078577			1.11988736557730			0.57092239841860		
H	1.10087813560022	-1.07544998540478	H	0.56815563848040	0.38891158646307	H	1.85476940905151	0.81187245722791
1.35674157184943			1.58311457552563			2.92647131209845		
H	1.62488982651624	-2.62421572413340	H	2.83270699754095	-0.97135760170923	H	0.72147721480060	-1.34978556074013
1.49981066964694			1.08219371490947			1.58848093224413		
H	0.01905738657255	-2.29634404797524	H	1.54148565654742	-1.93788083128647	H	-0.41488708302660	-2.44019711608998
1.34632461587986			1.32768021977848			1.24525703793069		
H	-0.41757243547574	0.71491413564573	H	0.10573093295074	0.94800854425153	H	1.23351096842995	0.99093586508606
0.75907621668372			1.85041175125671			2.55996998978200		
H	-1.47267349918011	-0.51662152655766	H	-2.05714121760914	1.41641296341921	H	-1.49213942052244	-0.18211820439285
1.95055380606411			1.91771864565424			1.95457680356297		
H	-2.20666011758149	1.29044227696042	H	-0.80801023596524	2.65751007997451	H	-0.53337765225509	0.56198327914812
1.44479472218780			0.78319169276753			0.33917382607667		

6. Decomposition of NH_3BH_3 via intermolecular H-H coupling in THF:
 $\Delta\Delta G^* = 234 \text{ kJ mol}^{-1}$ (56 kcal mol⁻¹)

Substrates: $\text{NH}_3\text{BH}_3\text{-NH}_3\text{BH}_3$			Transition state; $\nu_{\text{im}} = -398 \text{ cm}^{-1}$			Product: $\text{B}_2\text{N}_2\text{H}_{10} + \text{H}_2$				
E	-166,3762763		-166,2993303			-166.40021646				
H	-166,2248895		-166,1591102			-166.2557570				
G	-166,2652426		-166,2001			-166.29789097				
E	-166,6379647		-166,5376871			-166.64088737				
B	-2.38802484572491	2.91204211672677	B	1.17802562560649	0.24807060287034	-	B	-3.42137861966307	2.93167561462684	
0.95553559629116			1.30111656047350				1.29084413951406			
N	-3.59460107020342	1.93839093891025	N	0.35252917973386	-1.01857255796370	-	N	-4.19200561125328	1.70166452483815	
0.47577935726196			1.12348832007337				0.59083253048053			
H	-1.45246917778288	2.75513917252628	H	2.27450643126923	0.10208199995796	-	H	-2.25332901894857	2.61482981877786	
0.19701513936704			1.73023537420129				1.39324432606630			
H	-2.09842828934953	2.58578716202429	H	2.56580191086223	0.28456456579400		H	-1.27490449489458	4.91424880903867	
2.09496590784631			0.51581958427068				2.89557228167457			
H	-2.81265684261429	4.05505155217783	H	0.60562257623616	1.28094725757088	-	H	-3.59073751003527	3.90470592005182	
0.90788091263165			1.25906457998112				0.58818238373643			
H	-4.40942254583189	2.02860674154634	H	-0.43140303861209	-0.91218596489681	-	H	-5.19382711846080	1.88409759386304	
1.08823878930229			0.42752163615396				0.51739730243204			
H	-3.90956332488206	2.14846201933829	-	H	-0.10431651149796	-1.24745258361255	-	H	-3.83733468608333	1.54077432457612
0.46905010942889			2.01404823641472				0.34995639728083			
H	-3.32587025519320	0.95672283876557	H	0.90382536514622	-1.84239090428124	-	H	-4.09340744473607	0.82261541239944	
0.48383429107018			0.87706687914468				1.09770543299171			
H	-3.62717797010433	3.89629075839835	H	1.85514833259062	0.39251112931152		H	-0.88099751526103	4.87072093951393	
2.87591054037878			0.76161528516905				3.52382414568928			
H	-4.12754750310937	3.78055059674458	H	-0.23641515850994	0.33445073650433		H	-3.71272556542333	2.52931185523065	
4.43348089074959			2.26677275896971				3.39378077751019			
N	-4.44203288198516	3.98850855025422	N	-0.40446822735139	0.66296011652211		N	-4.08940885638064	3.17194519088480	
3.48802707862085			1.32004421685270				2.69836332303267			
H	-4.71021533324017	4.97030642737220	H	-0.38325261470850	1.67613943252819		H	-3.78858894768645	4.08963732705960	
3.47766707344280			1.38248041823231				3.02108422013436			
H	-5.22385536548494	1.87138863528488	H	-1.67375565259136	-1.08951646247121		H	-6.05373233756791	1.95451820506417	
3.05899105321156			0.57562924827017				2.63370464052930			
B	-5.64871137004771	3.01425581314372	B	-1.76287244032122	0.14662480989853		B	-5.70487343612638	3.12072869804726	
3.00986425492581			0.75885676086020				2.75544311693162			
H	-5.93846317948459	3.33891220829129	H	-1.98746499931990	0.67684801971950	-	H	-6.13412063411801	3.79951589798047	
1.87002116122901			0.33406842635950				1.83293283487361			
H	-6.58426543836136	3.17210906069501	H	-2.75151077853243	0.30491980254814		H	-6.07193359655973	3.56153446034606	
3.76819322379979			1.48539174017732				3.83340010218345			