Supporting Information

Tetranuclear [Cu<sub>3</sub>Ln] complexes derived from

a tetraketone-type ligand

Takuya Shiga,\*a Haruka Miyamoto,a Yukiko Okamoto,a

Hiroki Oshio,<sup>a</sup> Nozomi Mihara<sup>a</sup> and Masayuki Nihei\*<sup>a</sup>

 Degree Programs in Pure and Applied Sciences, Graduate School of Science and Technology, University of Tsukuba, Tennodai 1-1-1, Tsukuba, Ibaraki 305-8571, Japan.

Corresponding author Dr. Takuya Shiga Degree Programs in Pure and Applied Sciences, Graduate School of Science and Technology, University of Tsukuba Tennodai 1-1-1, Tsukuba, Ibaraki 305-8571 (Japan) TEL: (+81)29-852-4426 FAX: (+81)29-852-4426 E-mail:

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## **Explanation of alerts in CIFCHECK**



Figure S1 Experimental (red) and simulated (black) powder X-ray diffraction data of 1'.



Figure S2 Experimental (red) and simulated (black) powder X-ray diffraction data of 2'.



Figure S3 Molecular structures of complex 2'. Top view (top) and side view (bottom)



Figure S4 Field dependences of magnetization for complexes 1 (red), 2 (green), and 3 (blue).



**Figure S5** The AC magnetic susceptibilities of complex **3** under an applied dc magnetic field of 1300 Oe.



**Figure S6** Debye plots of complex **3** for frequencies 1000 (red), 750 (blue), 500 (green) Hz. The solid lines correspond to the fit of the data.

	Comp. 1'	Comp. 2'	Comp. 3'
Formula	$C_{29}H_{44}Cu_3GdN_3O_{26}$	$C_{29}H_{44}Cu_3N_3O_{26}Tb$	$C_{27}H_{36}Cu_3DyN_3O_{24}$
$M \ / \ g \ mol^{-1}$	1198.54	1200.21	1139.71
Temp. / K	100(2)	100(2)	100(2)
Crystal system	Monoclinic	Monoclinic	Triclinic
Space group	$P2_1/c$	$P2_1/c$	PError!
<i>a</i> / Å	8.9376(8)	8.9281(8)	7.9231(6)
<i>b</i> / Å	27.147(2)	27.150(2)	11.5750(10)
<i>c</i> / Å	17.9174(16)	17.9180(17)	22.3077(18)
α/o	_	_	93.0390(10)
β/°	103.2910(10)	103.5200(10)	91.7730(10)
γ/ °	_	_	105.8840(10)
$V/\text{\AA}^3$	4230.8(7)	4222.9(7)	1962.8(3)
Z	4	4	2
d / g cm <sup>-3</sup>	1.882	1.888	1.928
$\mu$ / mm <sup>-1</sup>	3.130	3.240	3.578
F(000)	2392	2396	1128
Reflections			
collected / unique	24193 / 9682	20642 / 7777	11429 / 8616
$R_{\rm int}$	0.0312	0.0309	0.1627
GOF	1.254	1.267	1.032
$R1 (I > 2\sigma(I))$	0.0558	0.0848	0.0629
$R_{\rm w}2 \ (I > 2\sigma(I)]$	0.1223	0.1955	0.1716
$\Delta  ho_{ m max}$ / e Å <sup>-3</sup>	1.365	3.081	6.271
$\Delta  ho_{ m min}$ / e Å <sup>-3</sup>	-2.086	-2.742	-4.831
CCDC No.	2223281	2223282	2223283

**Table S1.** Crystal parameters of 1', 2', and 3'.

Complexes	$-\Delta S_{\rm m}({\rm J~kg^{-1}~K^{-1}})$	$\Delta H(\mathbf{T})$	$T_{\max}$ (K)	Ref.
Cu <sub>3</sub> Gd	16.4	5	2.4	This work
$Cu_5Gd_4$	31	9	3.0	1
$Cu_2Gd_7$	34.6	9	2.7	2
Cu <sub>3</sub> Gd <sub>6</sub>	26.9	7	3.0	3
Cu <sub>6</sub> Gd <sub>2</sub>	11.9	7	2.0	4
Cu <sub>5</sub> Gd <sub>2</sub>	16.3	5	2.0	5
Cu <sub>5</sub> Gd <sub>2</sub>	15.7	5	2.0	6
Cu <sub>4</sub> Gd <sub>2</sub>	22.9	7	3.0	7
Cu <sub>4</sub> Gd	10	7	3.0	8
Cu <sub>2</sub> Gd <sub>2</sub>	23.5	7	3.0	9
Cu <sub>2</sub> Gd <sub>2</sub>	22	7	3.0	10
Cu <sub>2</sub> Gd <sub>2</sub>	17	7	3.0	10
Cu <sub>2</sub> Gd	17	7	3.5	11
CuGd	23.5	7	2.3	12

**Table S2**Maximum values of the entropy changes for the previously reported MCE materials based onCu(II)-Gd(III) systems with up to nonanuclear clusters.

#### References

- 1 S.K. Langley, N.F. Chilton, B. Moubaraki, T. Hooper, E.K. Brechin, M. Evangelisti, K.S. Murray, *Chem. Sci.*, 2011, **2**, 1166.
- 2 S.K. Langley, B. Moubaraki, C. Tomasi, M. Evangelisti, E.K. Brechin, K.S. Murray, *Inorg. Chem.*, 2014, **53**, 13154.
- 3 E. Moreno Pineda, C. Heesing, F. Tuna, Y.-Z. Zheng, E.J.L. McInnes, J. Schnack, R.E.P. Winpenny, *Inorg. Chem.*, 2015, **54**, 6331.
- 4 S. Xue, Y.-N. Guo, L. Zhao, H. Zhang, J. Tang, *Inorg. Chem.*, 2014, **53**, 8165.
- 5 D. Dermitzaki, V. Psycharis, Y. Sanakis, T.C. Stamatatos, M. Pissas, C.P. Raptopoulou, *Polyhedron*, 2019, **169**, 135.
- 6 D. Dermitzaki, O. Bistola, M. Pissas, V. Psycharis, Y. Sanakis, C.P. Raptopoulou, *Polyhedron*, 2018, **150**, 47.
- 7 D.I. Alexandropoulos, L. Cunha-Silva, J. Tang, T.C. Stamatatos, *Dalton Trans.*, 2018, 47, 11934.
- 8 P. Richardson, D.I. Alexandropoulos, L. Cunha-Silva, G. Lorusso, M. Evangelisti, J. Tang, T.C. Stamatatos, *Inorg. Chem. Front.*, 2015, **2**, 945.

- 9 S. Maity, A. Mondal, S. Konar, A. Ghosh, *Dalton Trans.*, 2019, **48**,15170.
- 10 B. Dey, S. Roy, A.K. Mondal, A. Santra, S. Konar, *Eur. J. Inorg. Chem.*, 2018, 2018, 2429.
- 11 M.K. Singh, T. Rajeshkumar, R. Kumar, S.K. Singh, G. Rajaraman, *Inorg. Chem.*, 2018, 57, 1846.
- 12 A.S. Dinca, A. Ghirri, A.M. Madalan, M. Affronte, M. Andruh, *Inorg. Chem.*, 2012, **51**, 3935.

## Explanation of alerts in CIFCHECK

#### Datablock: Comp1\_Cu3Gd-L1 Alert level B

The following B-level alerts arise from the diffraction data being slightly weak. PLAT220\_ALERT\_2\_B NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range 6.4 Ratio PLAT972\_ALERT\_2\_B Check Calcd Resid. Dens. 1.23Ang From C9 -2.53 eA-3

#### Datablock: Comp2\_Cu3Tb-L1 Alert level A

The following A level alert arise from the imperfect absorption correction of heavy metal (Tb ion) compound.

# PLAT971\_ALERT\_2\_A Check Calcd Resid. Dens. 0.96Ang From Tb1 3.70 eA-3

## Alert level B

The following B-level alerts arise from the diffraction data being slightly weak. PLAT214\_ALERT\_2\_B Atom O24 (Anion/Solvent) ADP max/min Ratio 5.3 prolat PLAT220\_ALERT\_2\_B NonSolvent Resd 1 C Ueq(max)/Ueq(min) Range 6.4 Ratio PLAT415\_ALERT\_2\_B Short Inter D-H..H-X H40 ..H42 . 1.99 Ang. 1+x,y,z = 1\_655 Check PLAT420\_ALERT\_2\_B D-H Bond Without Acceptor O25 --H40 . Please Check PLAT934\_ALERT\_3\_B Number of (lobs-lcalc)/Sigma(W) > 10 Outliers .. 2 Check

The following B level alert arise from the imperfect absorption correction of heavy metal (Tb ion) compound.

PLAT971\_ALERT\_2\_B Check Calcd Resid. Dens. 1.79Ang From O25 2.58 eA-3 PLAT972\_ALERT\_2\_B Check Calcd Resid. Dens. 1.09Ang From C17 -3.31 eA-3 PLAT972\_ALERT\_2\_B Check Calcd Resid. Dens. 1.59Ang From C8 -2.86 eA-3

## Datablock: Comp3\_Cu3Dy-L1

## Alert level A

The following A level alert arise from the imperfect absorption correction of heavy metal (Dy ion) compound.

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PLAT971_ALERT_2_A Check Calcd Resid. Dens. 0.89Ang From Dy1 6.31 eA-3
PLAT971_ALERT_2_A Check Calcd Resid. Dens. 0.77Ang From Dy1 5.73 eA-3
PLAT972_ALERT_2_A Check Calcd Resid. Dens. 0.85Ang From Dy1 -4.98 eA-3
PLAT972_ALERT_2_A Check Calcd Resid. Dens. 0.85Ang From Dy1 -4.79 eA-3
PLAT972_ALERT_2_A Check Calcd Resid. Dens. 0.74Ang From Dy1 -4.41 eA-3
PLAT972_ALERT_2_A Check Calcd Resid. Dens. 0.67Ang From Dy1 -4.23 eA-3
```

Alert level B

The following B-level alerts arise from the diffraction data being slightly weak. PLAT213\_ALERT\_2\_B Atom O6 has ADP max/min Ratio ..... 4.1 prolat PLAT213\_ALERT\_2\_B Atom O7 has ADP max/min Ratio ..... 4.4 prolat PLAT213\_ALERT\_2\_B Atom O10 has ADP max/min Ratio ..... 5.0 prolat PLAT213\_ALERT\_2\_B Atom C14 has ADP max/min Ratio ..... 4.2 prolat

The following B level alert arise from the imperfect absorption correction of heavy metal (Dy ion) compound.

PLAT971\_ALERT\_2\_B Check Calcd Resid. Dens. 0.96Ang From Dy1 2.70 eA-3 PLAT971\_ALERT\_2\_B Check Calcd Resid. Dens. 1.01Ang From O19 2.56 eA-3 PLAT972\_ALERT\_2\_B Check Calcd Resid. Dens. 1.39Ang From Dy1 -3.01 eA-3 PLAT972\_ALERT\_2\_B Check Calcd Resid. Dens. 1.37Ang From Dy1 -3.01 eA-3