

Electronic Supporting Information (ESI)

A semiconductive copper iodobismuthate hybrid: Structure, optical property and photocurrent response

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1. Selected bond lengths and bond angles

Table S1 Selected bond lengths (\AA) and bond angles ($^\circ$) for compound **1**.

Bi(1)–I(6)	2.9288(16)	I(1)–Cu(1) ^{#1}	2.700(4)
Bi(1)–I(2)	2.9460(16)	I(4)–Cu(1) ^{#1}	2.666(4)
Bi(1)–I(3)	2.9698(15)	I(5)–Cu(1) ^{#1}	2.682(4)
Bi(1)–I(4)	3.1721(16)	Cu(1)–I(4) ^{#1}	2.666(4)
Bi(1)–I(5)	3.2798(16)	Cu(1)–I(5) ^{#1}	2.682(4)
Bi(1)–I(1)	3.3066(16)	Cu(1)–I(1) ^{#1}	2.700(4)
I(1)–Cu(1)	2.560(3)	Cu(1)–Cu(1) ^{#1}	2.597(8)
I(6)–Bi(1)–I(2)	91.83(5)	I(4)–Bi(1)–I(1)	82.69(4)
I(6)–Bi(1)–I(3)	90.41(4)	I(5)–Bi(1)–I(1)	77.04(4)
I(2)–Bi(1)–I(3)	92.70(5)	Cu(1)–I(1)–Cu(1) ^{#1}	59.10(15)
I(6)–Bi(1)–I(4)	91.68(5)	Cu(1)–I(1)–Bi(1)	124.07(10)
I(2)–Bi(1)–I(4)	94.55(5)	Cu(1) ^{#1} –I(1)–Bi(1)	65.15(8)
I(3)–Bi(1)–I(4)	172.39(5)	Cu(1) ^{#1} –I(4)–Bi(1)	67.59(8)
I(6)–Bi(1)–I(5)	99.88(4)	Cu(1) ^{#1} –I(5)–Bi(1)	65.73(8)
I(2)–Bi(1)–I(5)	168.05(5)	I(1)–Cu(1)–I(4) ^{#1}	107.08(13)
I(3)–Bi(1)–I(5)	89.60(4)	I(1)–Cu(1)–I(5) ^{#1}	116.58(14)
I(4)–Bi(1)–I(5)	82.83(4)	I(4) ^{#1} –Cu(1)–I(5) ^{#1}	105.91(14)
I(6)–Bi(1)–I(1)	173.85(4)	I(1)–Cu(1)–I(1) ^{#1}	120.85(15)
I(2)–Bi(1)–I(1)	91.08(4)	I(4) ^{#1} –Cu(1)–I(1) ^{#1}	105.84(13)
I(3)–Bi(1)–I(1)	94.87(4)	I(5) ^{#1} –Cu(1)–I(1) ^{#1}	99.30(12)

Symmetry codes: #1 $-x+1, y, -z+1/2$.

Table S2 Hydrogen bonds (\AA) and angles ($^\circ$) for compound **1**.

D–H \cdots A	d(D–H)	d(H \cdots A)	d(D \cdots A)	\angle (DHA)
C(1)–H(1C) \cdots I(2) ^{#4}	0.96	3.31	3.98(2)	128.8
C(1)–H(1C) \cdots I(6) ^{#4}	0.96	3.25	3.71(2)	111.5
C(3)–H(3A) \cdots I(6) ^{#3}	0.97	3.06	3.984(18)	159.5
C(3)–H(3B) \cdots I(5) ^{#2}	0.97	3.16	3.93(2)	137.6

C(5)–H(5B)···I(2)#5	0.97	3.02	3.91(2)	152.6
C(6)–H(6A)···I(1)#5	0.97	3.32	3.948(19)	124.1
C(6)–H(6A)···I(2)#5	0.97	3.09	3.95(2)	147.5

Symmetry transformations used to generate equivalent atoms: #2 $x, y, z+1$; #3 $x, y-1, z+1$; #4 $-x+1/2, -y+3/2, -z+1$; #5 $x, -y+2, z+1/2$.

2. Crystal structure

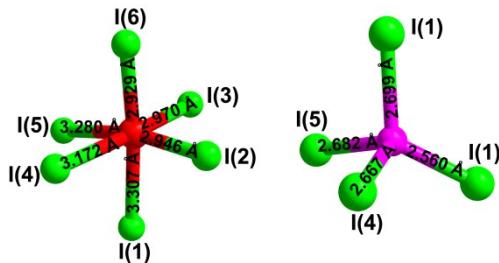


Fig. S1 The primary building units of BiI_6 (left) and CuI_4 (right) in compound **1**.

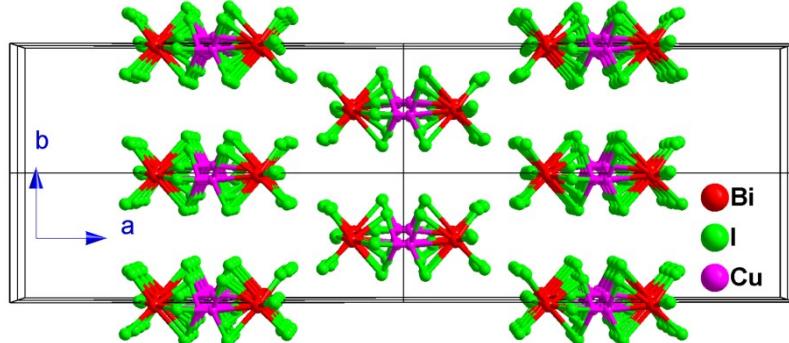


Fig. S2 Perspective view of compound **1** viewed along the c -axis.

3. Hirshfeld surface

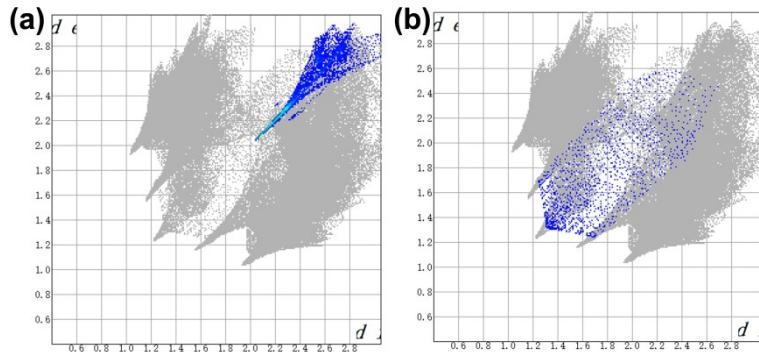


Fig. S3 Fingerprint plots: resolved into $\text{I} \cdots \text{I}$ (a) and $\text{Cu} \cdots \text{Cu}$ (b) contacts for compound **1**.

4. Physical measurements

4a). EDX

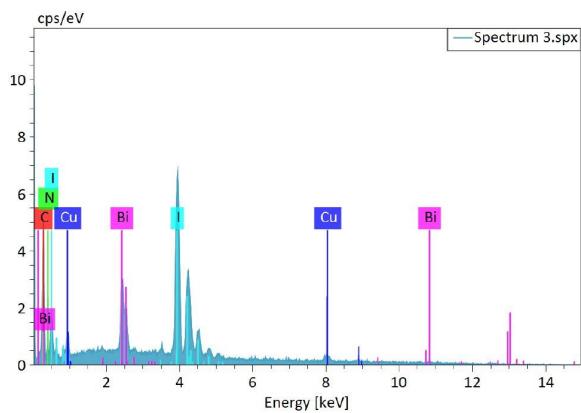


Fig. S4 EDX spectrum of compound 1.

4b). XPS

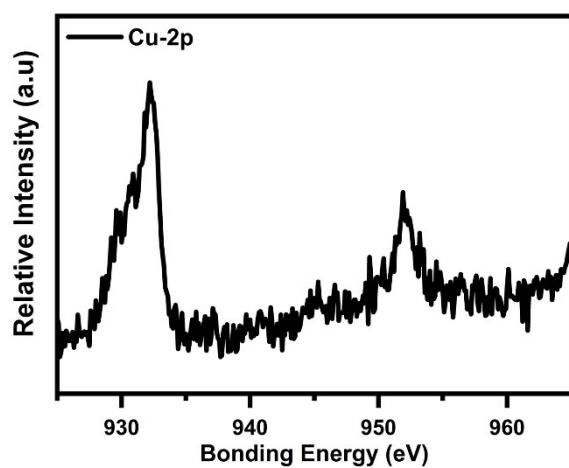


Fig. S5 High-resolution Cu-2p spectrum of compound 1.

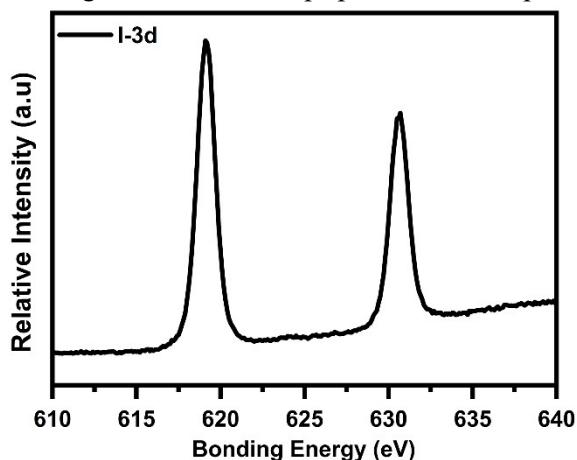


Fig. S6 High-resolution I-3d spectrum of compound 1.

5. DFT calculations

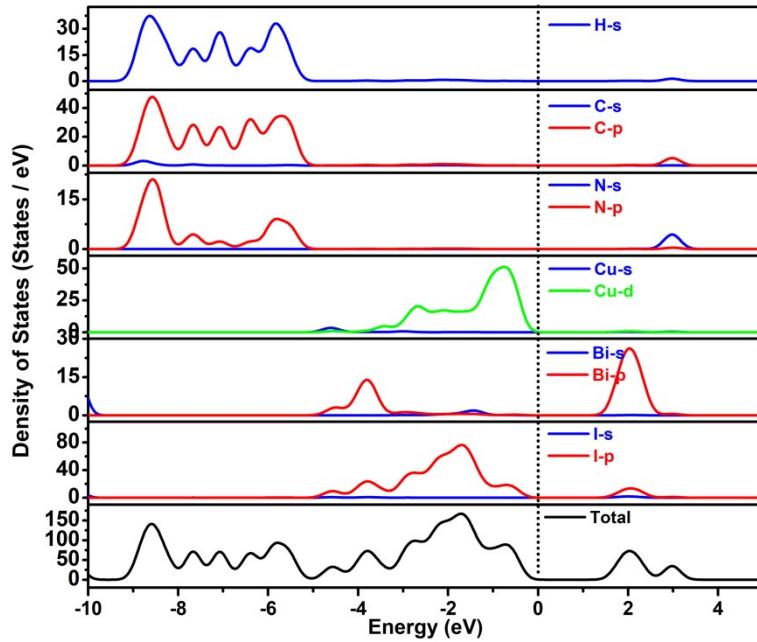


Fig. S7 The total density of states and partial density of states of compound **1**. The valence-band maximum (VBM) is set at 0 eV.

Table S3 Summary of band gaps of representative halobismuthates in the literature.

Compound	Space group	Band gap (eV)	Reference
[Et ₄ N] ₂ Ag ₂ Bi ₂ I ₁₀	$P\bar{1}$	2.05	[1]
[Et ₄ N] ₂ Ag ₂ Bi ₄ I ₁₆	$P2_1/c$	1.93	[1]
[SMe ₃] ₂ Ag ₂ Bi ₂ I ₁₀	$P2_1/n$	1.82	[2]
[Bu ₄ N] ₂ [{(PPh ₃) ₂ Bi ₂ Cu ₂ I ₁₀ }]	$P2_1/c$	NA	[3]
[Fe(bipy) ₃]AgBiI ₆	$P2_1/c$	1.92	[4]
[Fe(bipy) ₃]AgBiBr ₆	$C2/c$	1.82	[4]
[(C ₈ H ₁₇ N ₂) ₂ Cu ₂ Bi ₂ I ₁₀]	$P2_1/n$	1.82	[5]
[(C ₆ H ₁₃ N ₂) ₂ BiCu ₂ I ₇]·C ₂ H ₅ OH	$P2_1/c$	1.91	[5]
[Et ₄ N] ₂ Cu ₂ Bi ₂ I ₁₀	$P\bar{1}$	1.89	[6]
[Cu(CH ₃ CN) ₄] ₂ Cu ₂ Bi ₂ I ₁₀	$P\bar{1}$	1.80	[6]
[PPh ₄] ₄ Cu ₂ Bi ₂ I ₁₂	$Pbca$	1.80	[7]
[PPh ₄] ₄ Ag ₂ Bi ₂ I ₁₂	$Pbca$	2.10	[7]
[Et ₄ N] ₄ Hg ₂ Bi ₄ I ₂₀	$P1$	2.08	[8]
[4FPEA] ₄ AgBiI ₈	$P\bar{1}$	2.16	[9]
[4FPEA] ₄ AgBiBr ₈	$P\bar{1}$	2.80	[9]
[4FPEA] ₄ AgBiCl ₈	$P\bar{1}$	3.30	[9]
[C ₃ H ₉ NCl] ₄ AgBiBr ₈	Pc	2.69	[10]
[C ₃ H ₉ NI] ₄ AgBiI ₈	$P\bar{1}$	1.87	[11]
[CH ₃ NH ₃] ₂ AgBiBr ₆	$Fm\bar{3}m$	2.02	[12]
[C ₆ H ₁₆ N ₂] ₂ CuBiI ₈ ·0.5H ₂ O	$P2_1/c$	1.68	[13]
[C ₈ H ₂₀ N ₂] ₂ AgBiBr ₈	$P\bar{1}$	2.70	[14]
[H ₂ EPZ] ₂ AgBiBr ₈	$P2_1/c$	2.77	[15]
[H ₂ MPA] ₂ AgBiBr ₈	$P2_1/n$	2.84	[15]

[AMP] ₄ [AgBiI ₈] ₂ ·H ₂ O	<i>C</i> 2/ <i>c</i>	2.07	[16]
[APP] ₄ [AgBiI ₈]·H ₂ O	<i>C</i> 2/ <i>c</i>	2.12	[16]
[C ₆ H ₁₆ N ₂] ₂ AgBiI ₈ ·H ₂ O	<i>P</i> 2 ₁ / <i>n</i>	1.93	[17]
[C ₆ H ₁₆ N ₂] ₂ CuBiI ₈ ·0.5H ₂ O	<i>P</i> 2 ₁ / <i>n</i>	1.68	[17]
[La(DMSO) ₈]Bi ₂ I ₉	<i>P</i> 1̄	2.21	[18]
[Bi(DMSO) ₈]Bi ₂ I ₉	<i>P</i> 1̄	2.17	[18]
[AmV]BiI ₅	NA	1.54	[19]
[PiC ₂] ₂ Bi ₂ I ₁₀	<i>P</i> 2 ₁ / <i>n</i>	2.08	[20]
PiC ₅ BiI ₅	<i>I</i> 222	1.73	[20]
[PiC ₅] ₂ Bi ₄ I ₁₆	<i>P</i> 2 ₁ / <i>n</i>	2.10	[20]
[HpipeH ₂] ₂ Bi ₂ I ₁₀ ·2H ₂ O	<i>P</i> 1̄	1.80	[21]
LiBiI ₄ ·5H ₂ O	<i>C</i> 2/ <i>c</i>	1.70	[22]
KBiI ₄ ·H ₂ O	<i>P</i> 2 ₁ / <i>n</i>	1.76	[22]
[(Me) ₂ -(DABCO)]CuBiI ₆	<i>C</i> 2/ <i>c</i>	1.80	This work

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