

Supporting Information

Interfacial Effect induced by Rare Earth Oxide in Boosting Conversion of CO₂ to Formate

Lianpeng Song, Zhong Liang, Mingzi Sun, Bolong Huang*, Yaping Du*

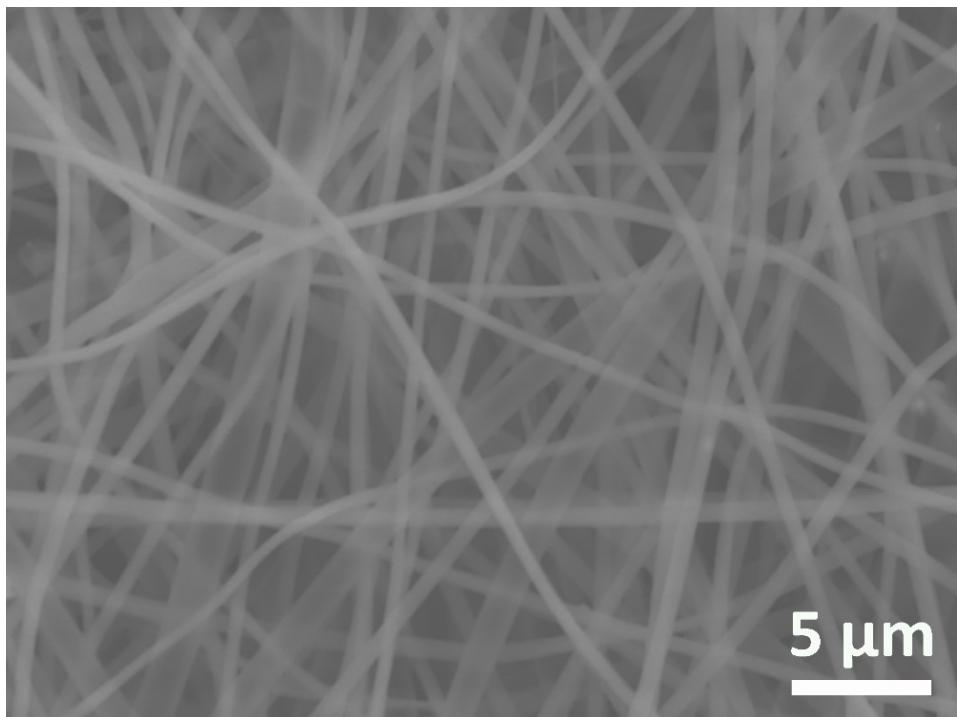


Fig. S1 SEM image of as-spun fiber of CeO₂/Bi₃NbO₇.

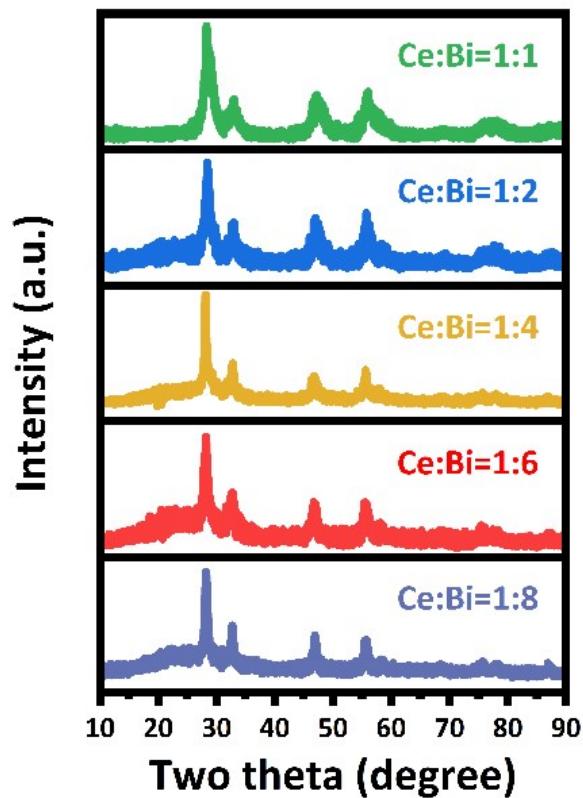


Fig. S2 XRD pattern comparison of CeO₂/Bi₃NbO₇ with different Ce: Bi ratios.

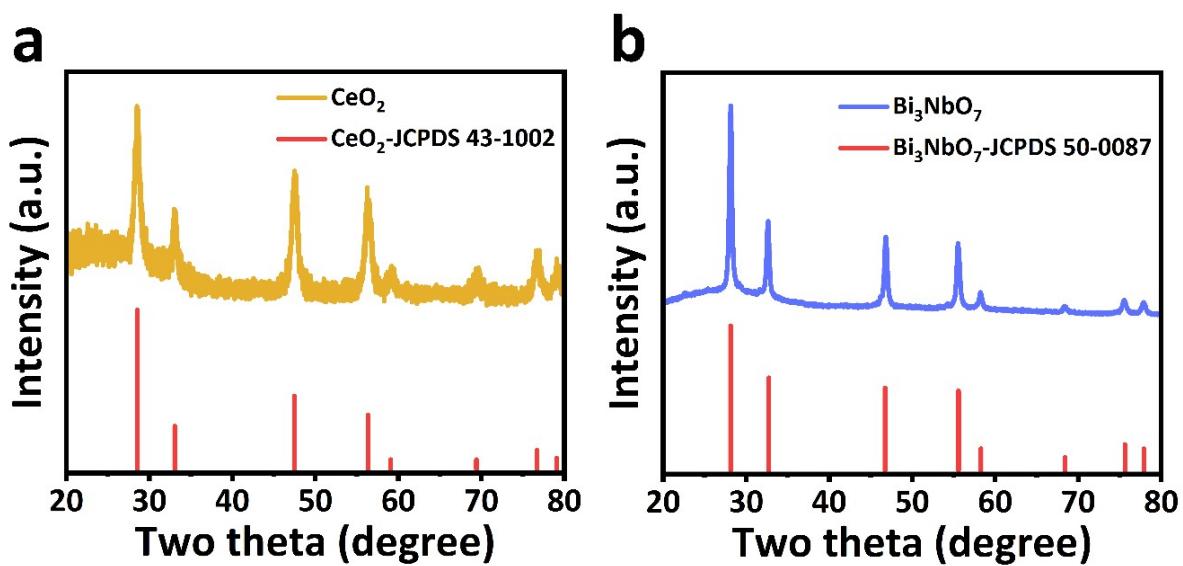


Fig. S3 **a** XRD pattern of CeO_2 and **b** Bi_3NbO_7 .

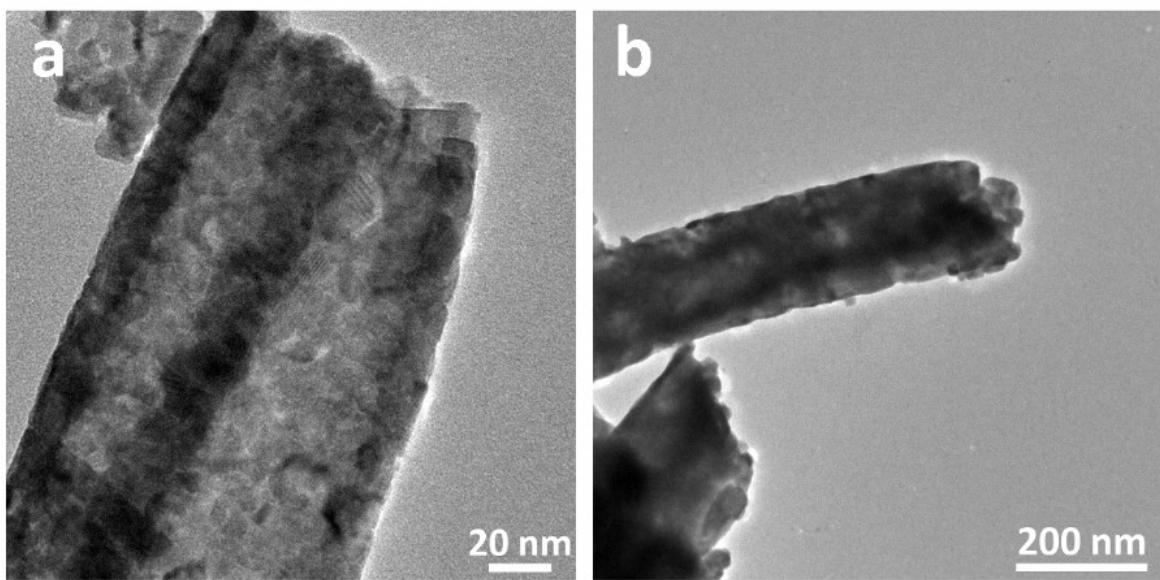


Fig. S4 Morphologic characterization of **a** CeO₂ and **b** Bi₃NbO₇

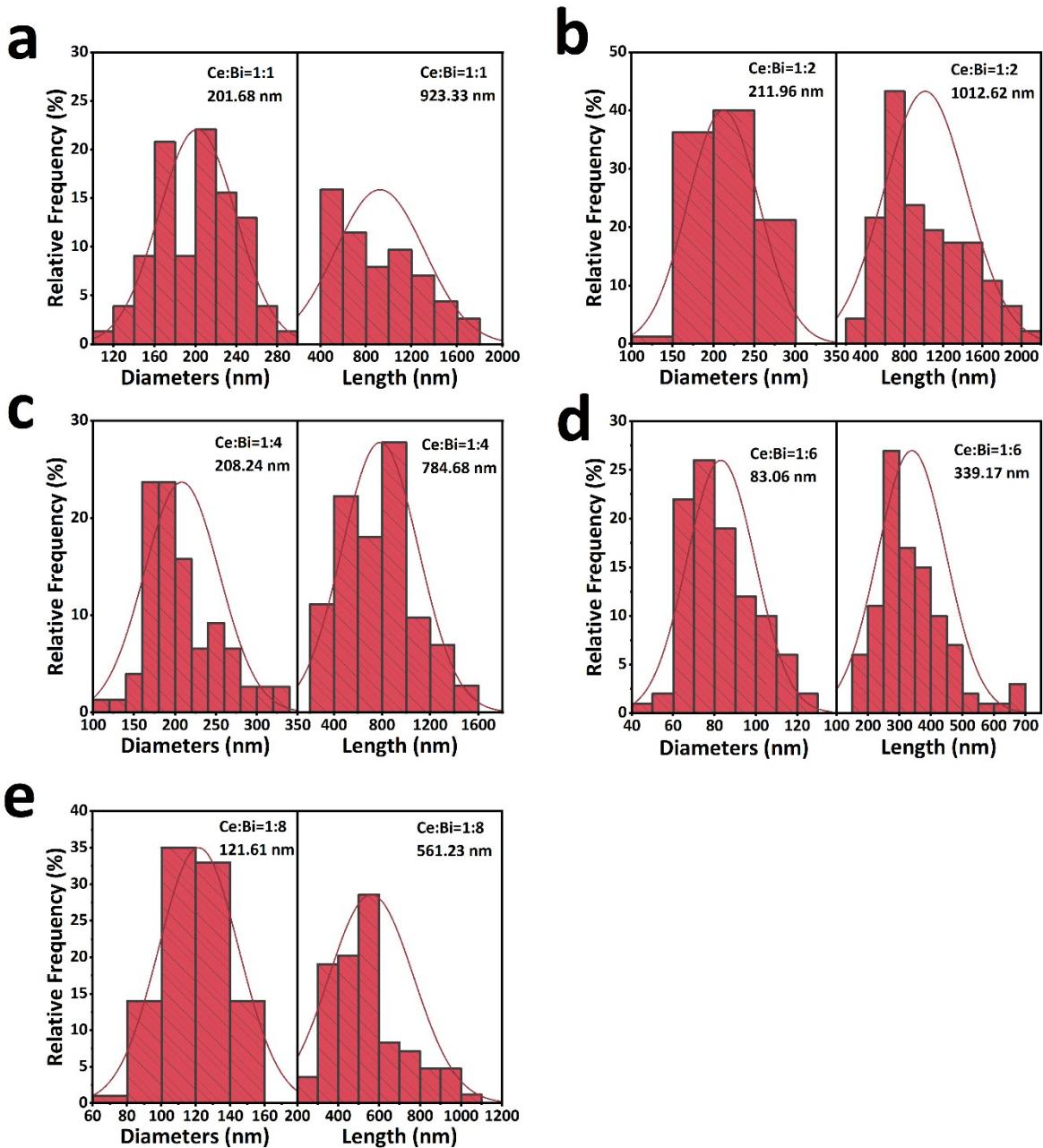


Fig. S5 The comparisons of the size distributions (diameters and length) for $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ with different ratios.

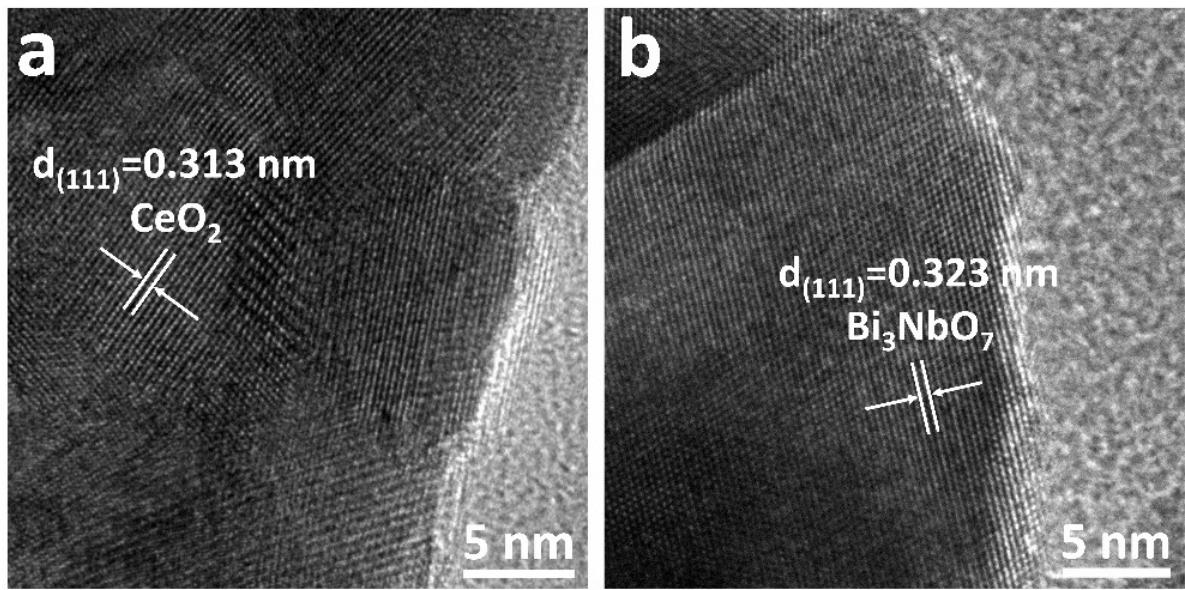


Fig. S6 HRTEM images of **a** CeO_2 and **b** Bi_3NbO_7 .

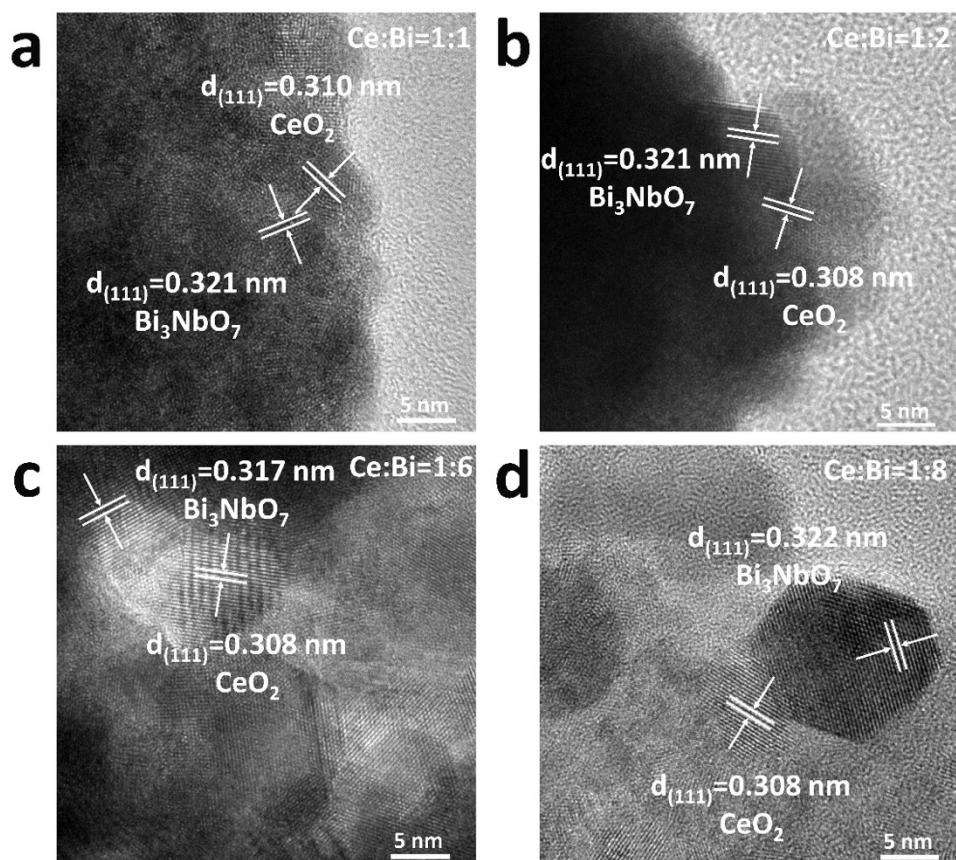


Fig. S7 HRTEM of the interfaces in $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ with different Ce: Bi ratios.

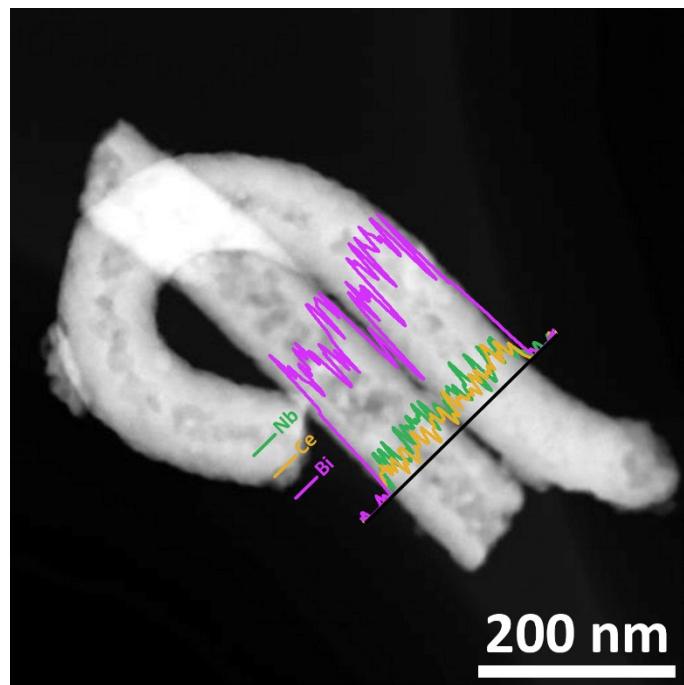


Fig. S8 The line scan of different elements of $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$.

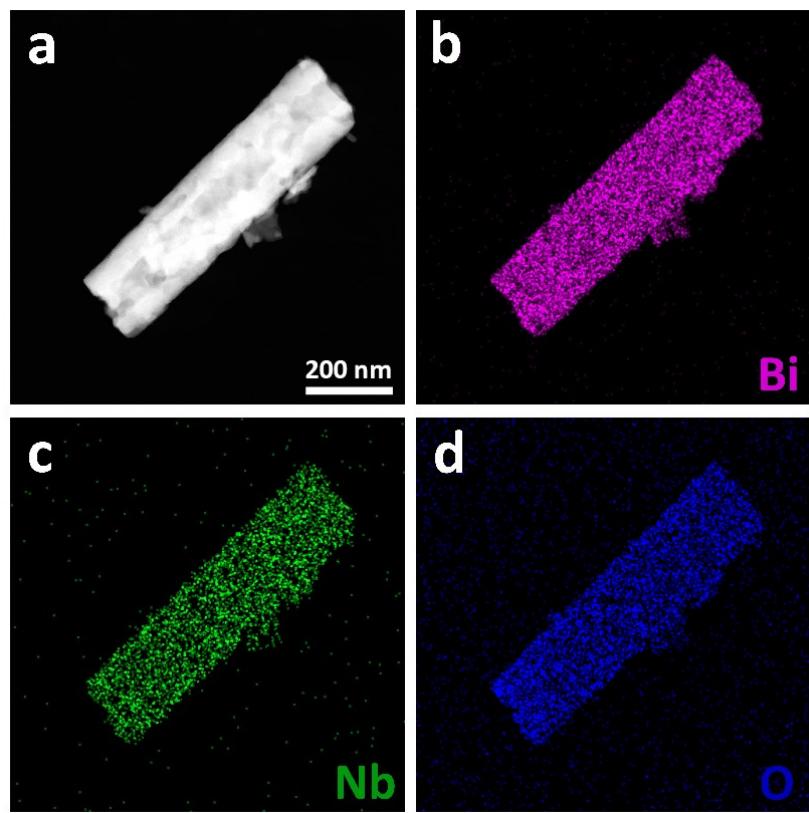


Fig. S9 a-d Corresponding elemental mapping images of Bi_3NbO_7 .

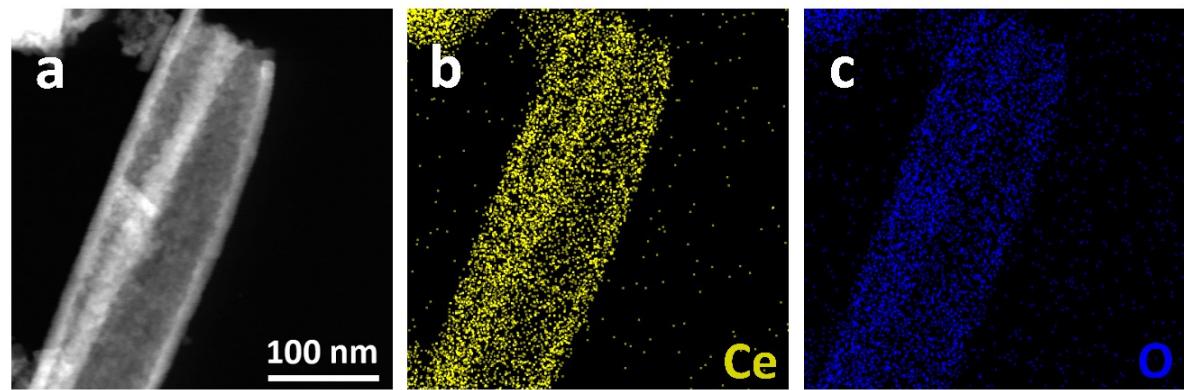


Fig. S10 a-c Corresponding elemental mapping images of CeO_2 .

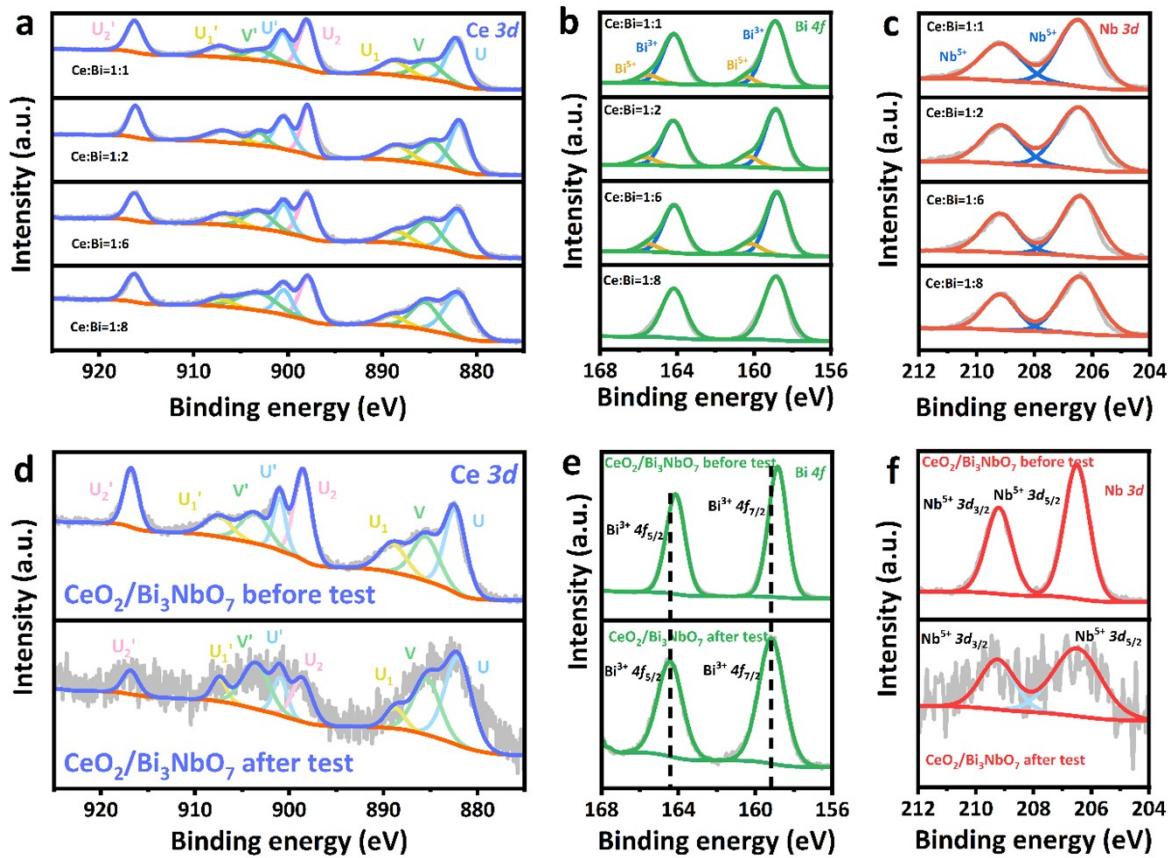


Fig. S11. a-c XPS of CeO₂/Bi₃NbO₇ with different ratios. **d-f** XPS of CeO₂/Bi₃NbO₇ before and after tests.

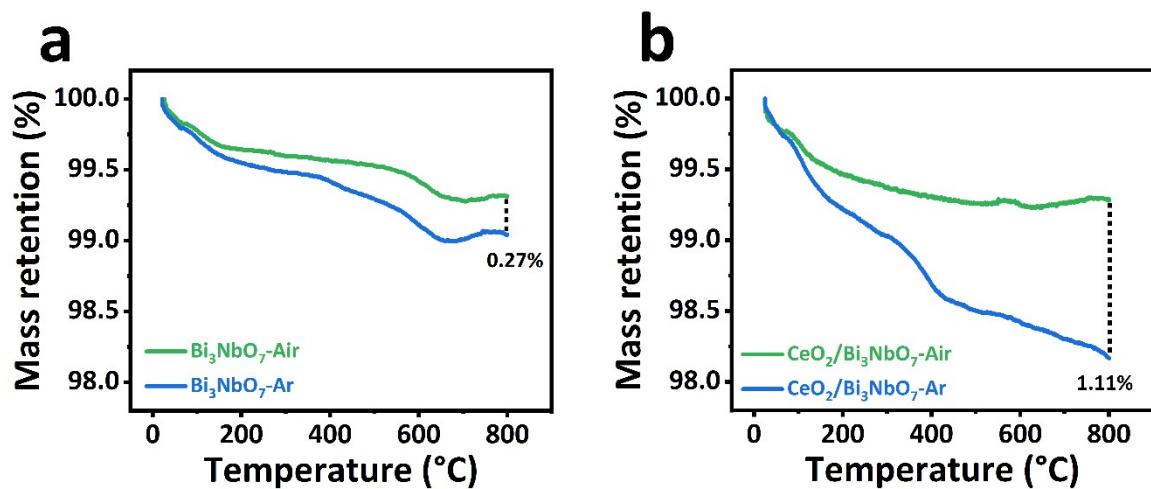


Fig. S12. a-b TGA analysis of Bi_3NbO_7 and $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ in air and argon atmosphere. (Flow of 100 mL/min and a ramping rate of 10 $^{\circ}\text{C}/\text{min}$). The concentration of oxygen vacancies was calculated from the difference in weight decrease between the two TGA traces.



Fig. S13 H-type cell for electrocatalytic carbon dioxide reduction.

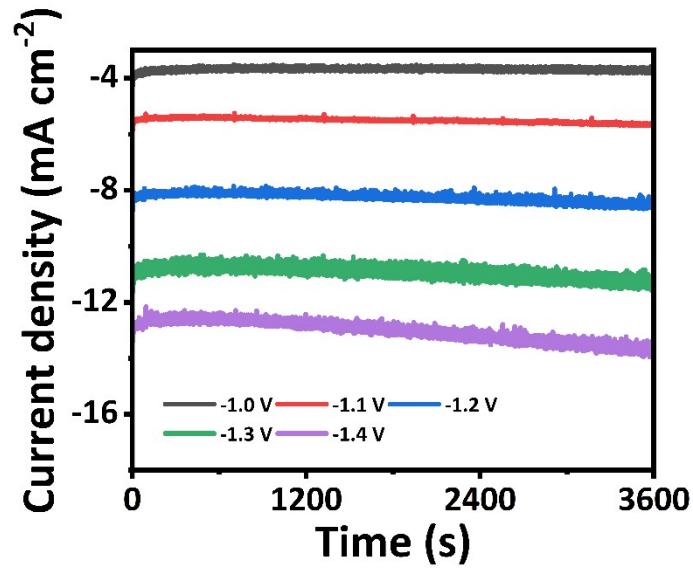


Fig. S14 Chronoamperometric curves of $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ (1: 4) at different working potentials.

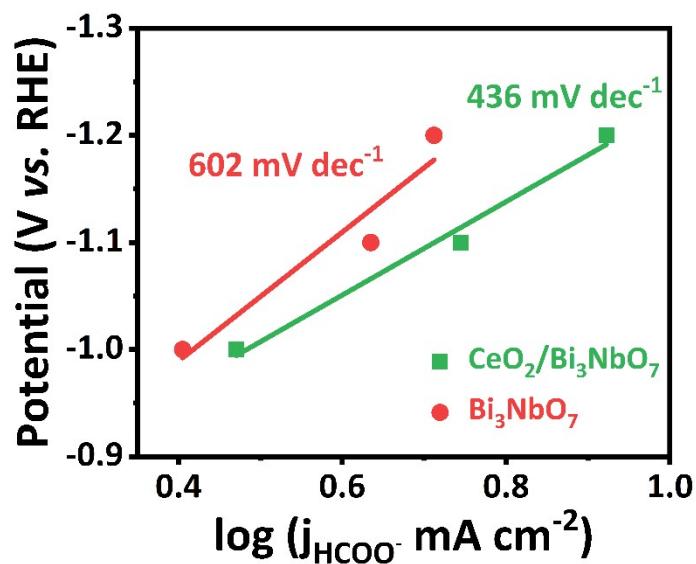


Fig. S15 Tafel plot of Bi_3NbO_7 and $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$.

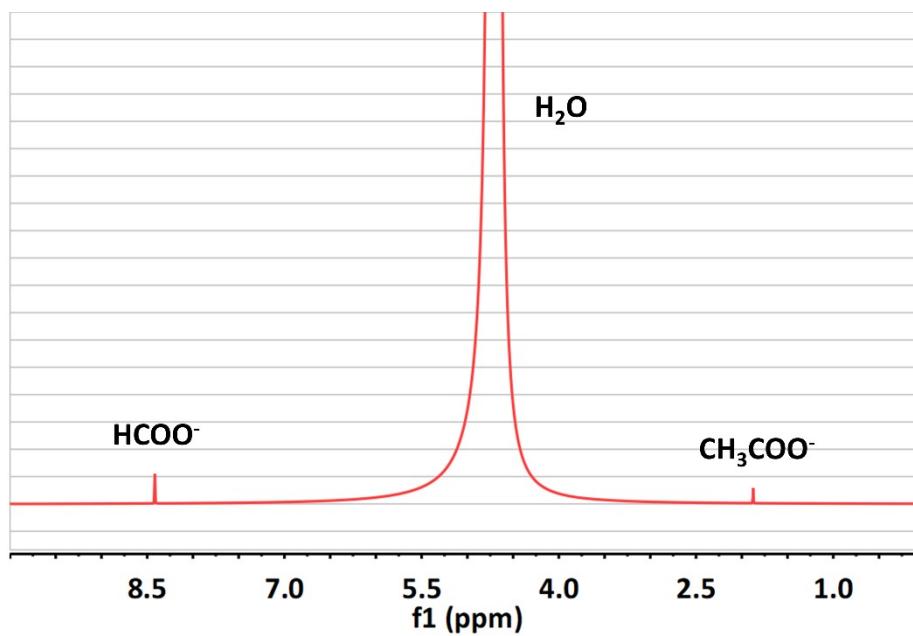


Fig. S16 Representative NMR spectrum of $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ after electrolysis at -1.2 V versus RHE for 6 h.

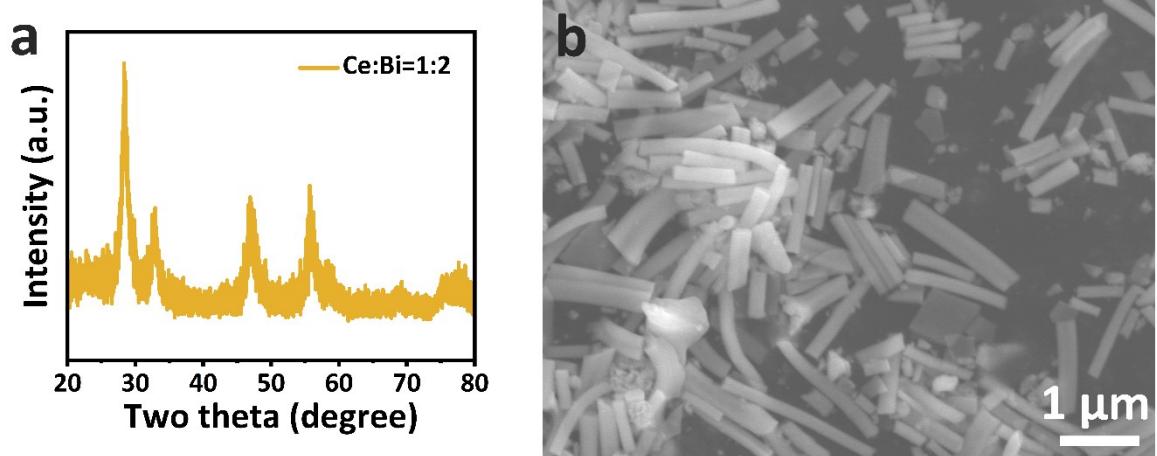


Fig. S17 **a** XRD pattern, **b** SEM image of the complex with a molar ratio of Ce and Bi of 2.

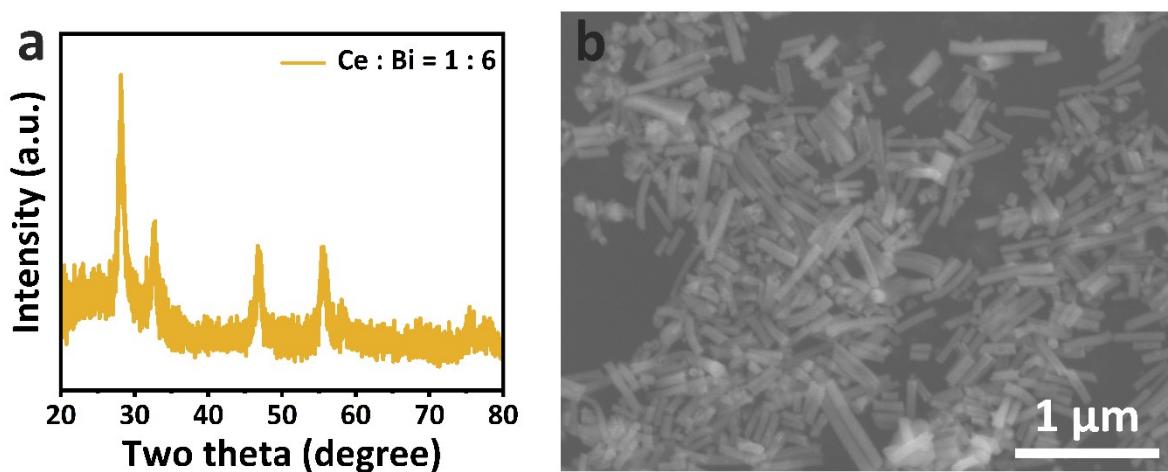


Fig. S18 **a** XRD pattern, **b** SEM image of complex with a molar ratio of Ce and Bi of 6.

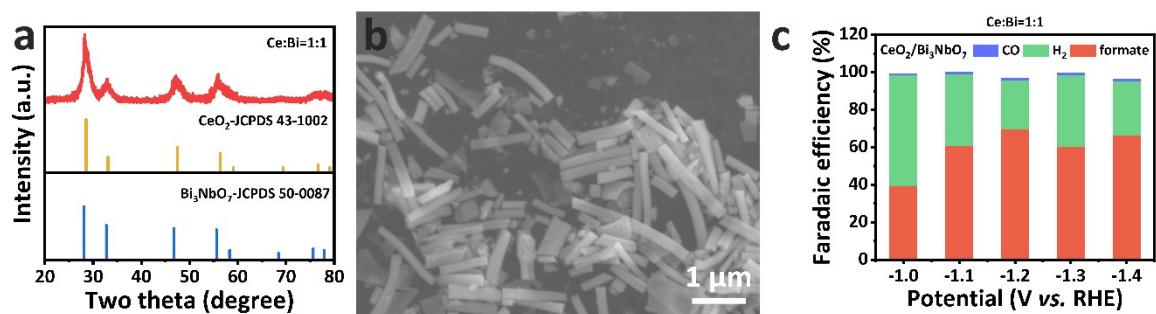


Fig. S19 a XRD pattern, b SEM image and c FE of formate, CO₂ and H₂ of CeO₂/Bi₃NbO₇ (Ce/Bi=1/1) at selected voltages.

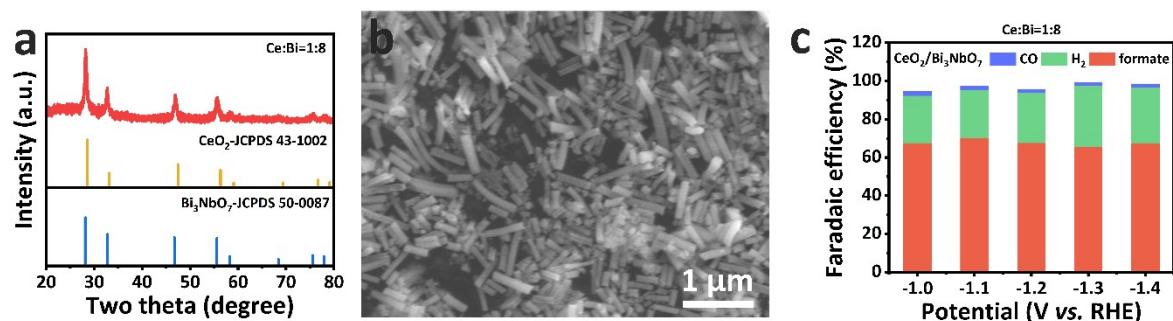


Fig. S20 **a** XRD pattern, **b** SEM image and **c** FE of formate, CO₂ and H₂ of CeO₂/Bi₃NbO₇ (Ce/Bi=1/8) at selected voltages.

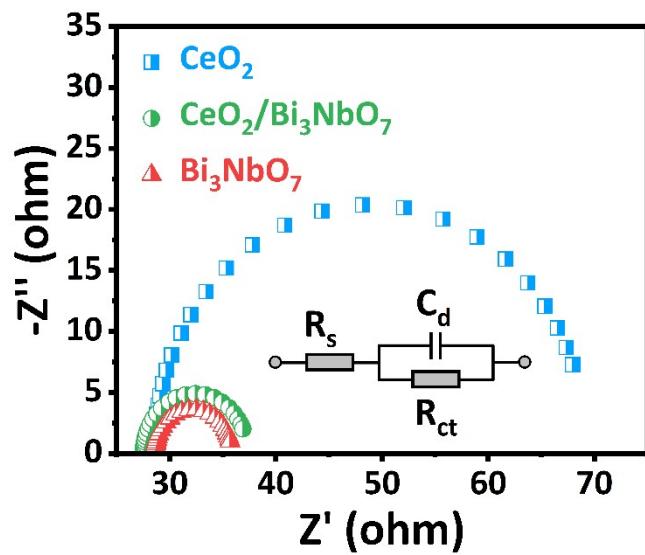


Fig. S21 Electrochemical impedance spectra (EIS) of CeO_2 , Bi_3NbO_7 and $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ at -1.2 V versus RHE, after fitting.

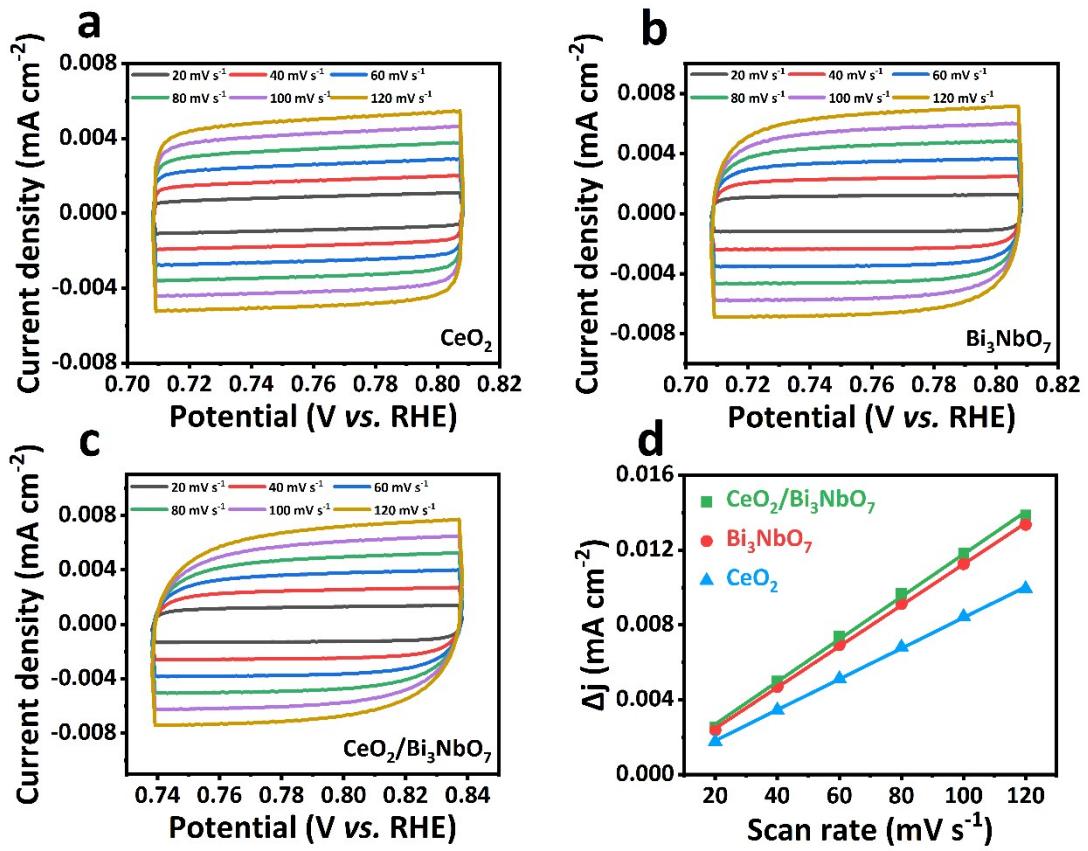


Fig. S22 Cyclic voltammetry curves of a CeO_2 , b Bi_3NbO_7 , c $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ and d different samples plotted against scan rate.

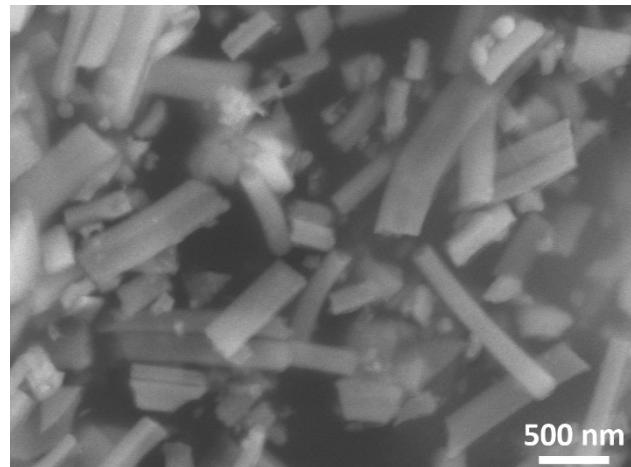


Fig. S23 SEM images of the $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ dispersed on carbon paper after 12 h stability test.

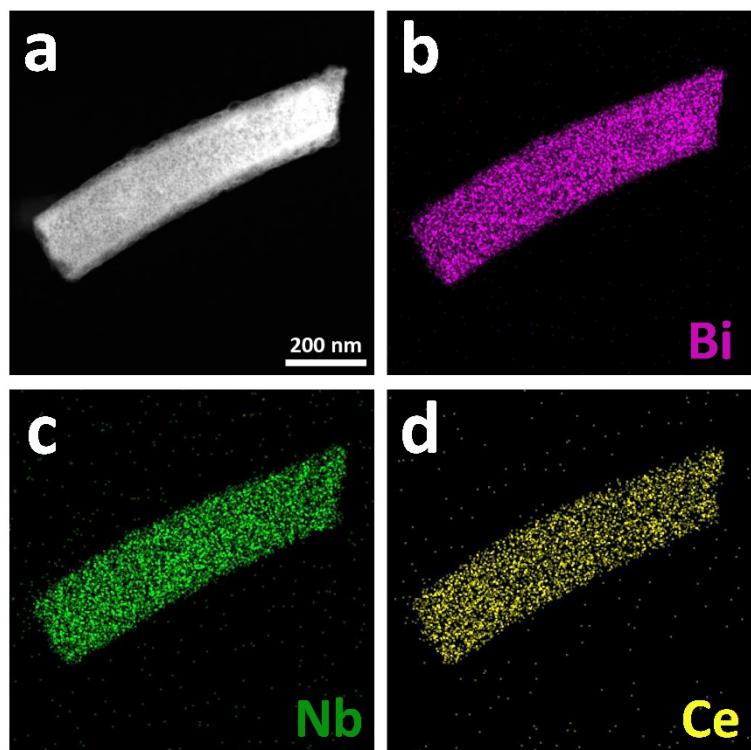


Fig. S24 a-d Corresponding elemental mapping images of $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ after stability test.

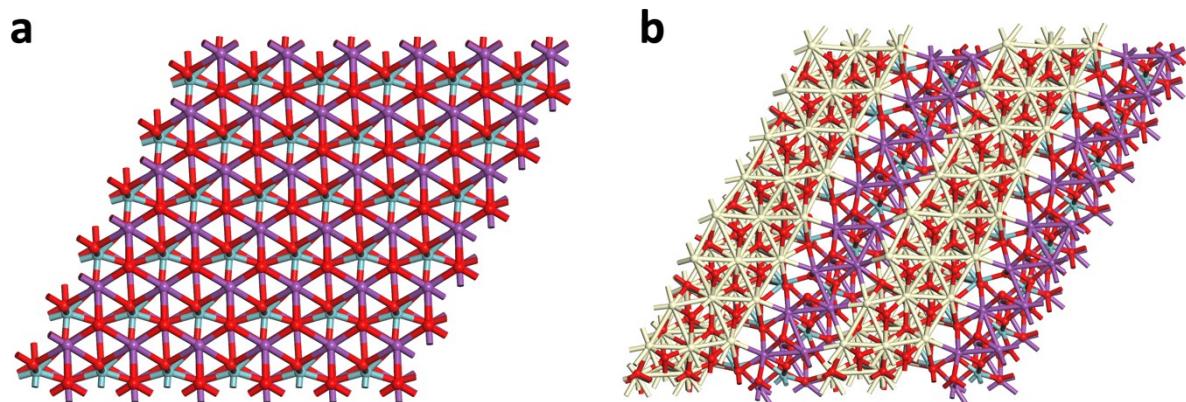


Fig. S25 The DFT calculation models of (a) Bi_3NbO_7 and (b) $\text{CeO}_2/\text{Bi}_3\text{NbO}_7$ after relaxation.

Table S1. The morphology analysis of different CeO₂/Bi₃NbO₇ samples.

Ce: Bi Ratio Sample	diameters (nm)	Length (nm)
1:1	201.68	923.33
1:2	211.96	1012.62
1:4	208.24	784.68
1:6	83.06	339.17
1:8	121.61	561.23

Table S2. The composition ratios of different CeO₂/Bi₃NbO₇ samples.

Ce: Bi Ratio Sample	Ce Ratio (%)	Bi Ratio (%)	Nb Ratio (%)
1:1	41.01	42.23	16.76
1:2	26.26	55.52	18.22
1:4	17.61	60.06	22.33
1:6	13.11	63.48	23.41
1:8	9.22	72.67	18.11

Table S3. Performance comparison of CeO₂/Bi₃NbO₇ with other Bi-based electrocatalysts reported recently.

Catalysts	Electrolyte	Formate FE	j _{formate}	Ref.
2.0 M KHCO ₃	Bi@Sn NPs	92%@-1.0 V vs. RHE	230 mA cm ⁻²	1
0.1 M KHCO ₃	Ag - Bi	88.4%@-1.2 V vs. RHE	18.74 mA cm ⁻²	2
0.1 M KHCO ₃	Bi ₂ O ₂ CO ₃	83%@-0.59 V vs. RHE	0.17 mA cm ⁻²	3
0.1 M KHCO ₃	Bi ₅₂ Sn ₄₆ In ₂	82%@-1.2 V vs. RHE	25 mA cm ⁻²	4
0.1 M HNO ₃	MWCNT@CeO ₂	65%@-0.22 V vs. RHE	0.325 mA cm ⁻²	5
0.1 M HNO ₃	Bi _x Sn _{1-x}	78%@-1.1 V vs. RHE	9.36 mA cm ⁻²	6
0.1 M KHCO ₃	CeO ₂ /Bi ₃ NbO ₇	83.13%@-1.4 V vs. RHE	14.29 mA cm ⁻²	This work

Reference

1. Xing, Y.; Kong, X.; Guo, X.; Liu, Y.; Li, Q.; Zhang, Y.; Sheng, Y.; Yang, X.; Geng, Z.; Zeng, J., Bi@Sn Core-Shell Structure with Compressive Strain Boosts the Electroreduction of CO₂ into Formic Acid. *Adv. Sci.* 2020, 7 (22), 1902989.
2. Zhou, X.; Lu, X.; Yu, T.; Wang, H.; Qian, L.; Lei, P.; Yu, Y.; Liu, L.; Xia, S.; Fang, J., Conformal Shell Amorphization of Nanoporous Ag-Bi for Efficient Formate Generation. *ACS Appl. Mater. Interfaces* 2020, 12 (28), 31319-31326.
3. Lv, W.; Bei, J.; Zhang, R.; Wang, W.; Kong, F.; Wang, L.; Wang, W., Bi₂O₂CO₃ Nanosheets as Electrocatalysts for Selective Reduction of CO₂ to Formate at Low Overpotential. *ACS Omega* 2017, 2 (6), 2561-2567.
4. Allioux, F. M.; Merhebi, S.; Ghasemian, M. B.; Tang, J.; Merenda, A.; Abbasi, R.; Mayyas, M.; Daeneke, T.; O'Mullane, A. P.; Daiyan, R.; Amal, R.; Kalantar-Zadeh, K., Bi-Sn Catalytic Foam Governed by Nanometallurgy of Liquid Metals. *Nano Lett* 2020, 20 (6), 4403-4409.
5. Valenti, G.; Melchionna, M.; Montini, T.; Boni, A.; Nasi, L.; Fonda, E.; Criado, A.; Zitolo, A.; Voci, S.; Bertoni, G.; Bonchio, M.; Fornasiero, P.; Paolucci, F.; Prato, M., Water-Mediated ElectroHydrogenation of CO₂ at Near-Equilibrium Potential by Carbon Nanotubes/Cerium Dioxide Nanohybrids. *ACS Appl. Energy Mater.* 2020, 3 (9), 8509-8518.
6. Tang, J.; Daiyan, R.; Ghasemian, M. B.; Idrus-Saidi, S. A.; Zavabeti, A.; Daeneke, T.; Yang, J.; Koshy, P.; Cheong, S.; Tilley, R. D.; Kaner, R. B.; Amal, R.; Kalantar-Zadeh, K., Advantages of eutectic alloys for creating catalysts in the realm of nanotechnology-enabled metallurgy. *Nat. Commun.* 2019, 10 (1), 4645.