## **Supporting Information**

## Modulation of Nucleation and Crystallization in PbI<sub>2</sub> Films Promoting

## Preferential Perovskite Orientation Growth for Efficient Solar Cells

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**Figure S1.** Statistical distribution of *J-V* parameters of devices containing 0 (control), 0.5, 1, and 2 mg/mL of PFAT.



Figure S2.  $PbI_2$  solutions in DMF/DMSO (950/50, volume) w/o and with PFAT, and the corresponding Tyndall effect optical images.



Figure S3. Top-view SEM images of the a)  $PbI_2$ -Control and b)  $PbI_2$ -Target unannealed films deposited on ITO/SnO<sub>2</sub> substrates. The scale bars represent 1 $\mu$ m.



**Figure S4.** AFM 2D images of the a) PbI<sub>2</sub>-Control and b) PbI<sub>2</sub>-Target unannealed films deposited on ITO/SnO<sub>2</sub> substrates.



**Figure S5.** AFM 3D images of the a) PbI<sub>2</sub>-Control and b) PbI<sub>2</sub>-Target unannealed films deposited on ITO/SnO<sub>2</sub> substrates.



**Figure S6.** The statistics of cluster sizes in the PbI<sub>2</sub>-Control and PbI<sub>2</sub>-Target films from the topview SEM images.



Figure S7. AFM images of the a) PbI<sub>2</sub>-Control and b) PbI<sub>2</sub>-Target annealed films deposited on

ITO/SnO2 substrates.



Figure S8. Cross-sectional SEM images of the a)  $PbI_2$ -Control and b)  $PbI_2$ -Target unannealed films deposited on ITO/SnO<sub>2</sub> substrates. The scale bars represent 1 $\mu$ m.



**Figure S9.** Cross-sectional SEM images of the a) PbI<sub>2</sub>-Control and b) PbI<sub>2</sub>-Target annealed films deposited on ITO/SnO<sub>2</sub> substrates. The scale bars represent 500 nm.



Figure S10. Top-view SEM image of a PbI<sub>2</sub>-PFAT (3 mg/mL) annealed film deposited on an  $ITO/SnO_2$  substrate. The scale bar represents 1 $\mu$ m.



**Figure S11.** Contact angles of the PbI<sub>2</sub>-Control films with a) diiodomethane, b) ethylene glycol, c) glycerol, and d) water solution droplets.



**Figure S12.** Contact angles of the PbI<sub>2</sub>-Target films with a) diiodomethane, b) ethylene glycol, c) glycerol, and d) water solution droplets.



Figure S13. Evolution of diffraction peak intensity for (001) planes of different  $PbI_2$  films with various annealing times.



Figure S14. AFM images of the perovskite films a) without and b) with PFAT incorporation.



Figure S15. KPFM image of a highly oriented pyrolytic graphite (HOPG) as a standard sample.



**Figure S16.** UPS spectra in the secondary electron cut-off regions (left) and valence band (VB) regions (right) of the control and target perovskite films. The laser energy source was He I (21.22 eV) for excitation.



**Figure S17.** UV-vis absorbance spectra of the control and target perovskite films deposited on glass substrates.



Figure S18. Tauc plots and measured bandgaps of the control and target perovskite films.



Figure S19. Normalized bleaching kinetics for different perovskite films.



**Figure S20.** PL spectra of different perovskite films in the a) glass/SnO<sub>2</sub>/perovskite and b) glass/perovskite/spiro-OMeTAD configurations.



Figure S21. XPS full spectra of different perovskite films.



**Figure S22.** Comparison of photovoltaic parameters of the best-performing target device measured with reverse-scan without (w/o) and with a PDMS antireflective (AR) film.



Figure S23. Reverse and forward-scanned J-V curves of the best-performing target device with a PDMS antireflective film.



**Figure S24.** a) Reverse and forward-scanned *J*-*V* curves of the best-performing control device. b) EQE spectrum and integrated curve of the best-performing control cell. The integrated  $J_{SC}$  is 24.46 mA/cm<sup>2</sup>. c) SPO test of the best-performing control device measured at a constant bias voltage of 0.94 V within 200 s. The steady-state PCE is ~22.2%.



Figure S25. SCLC measurements for the hole-only a) control and b) target devices.



Figure S26. Nyquist plots of the control and target PSCs.

Liquid (L)	Surface Tension (mN/m)			Contact	x	у	m	с	
	Total σ	Dispersive σ <sub>L</sub> <sup>d</sup>	Polar σ <sub>L</sub> <sup>p</sup>	angle θ (°)					
PbI <sub>2</sub> -Control film									
Diiodomet hane	50.8	50.8	0	8.64	0	7.088	4.21 5	6.488	
Ethylene glycol	48	29	19	7.649	0. 81	8.874			
Glycerol	63.3	20.22	43.0 8	43.206	1. 46	12.17 0			
Water	72.3	18.7	53.6	42.5	1. 69	14.52 0			
PbI2-Target film									
Diiodomet hane	50.8	50.8	0	28.59	0	6.693			
Ethylene glycol	48	29	19	3.298	0. 81	8.905	4.44 1	6.215	
Glycerol	63.3	20.22	43.0 8	37.073	1. 46	12.65 6			
Water	72.3	18.7	53.6	45.687	1. 69	14.20 3			

Table S1. Surface tension and contact angles between the standard liquids and the  $PbI_2$  films without or with PFAT.

Table S2. The fitted data of time-resolved TA spectra for different perovskite films.

Sample	$A_1$	$\tau_1  [ps]$	A <sub>2</sub>	$\tau_2 [ps]$
Control	2.56E-2	697.2	6.85E-1	7613.3
Target	-5.64E-2	259.6	9.39E-1	11293.4

Sample	A <sub>1</sub> [%]	$\tau_1 [ns]$	A <sub>2</sub> [%]	$\tau_2 [ns]$	$\tau_{ave} [ns]$
Control	5.26	213.23	94.74	1202.96	966.76
Target	4.34	181.74	95.66	2357.98	1551.57

**Table S3.** TRPL lifetime parameters of the control and target perovskite films fitted by a biexponential decay function of  $F(t) = A_1 \exp(-t/\tau_1) + A_2 \exp(-t/\tau_2) + \gamma_0$ .

**Note S1.** Calculation of the total surface tension ( $\sigma$ ) of the PbI<sub>2</sub>-Control and PbI<sub>2</sub>-Target films. The Owens-Wendt-Rabel-Kaelble (OWRK) model:

$$\frac{\sigma_{\rm L}(1+\cos\theta)}{2\sqrt{\sigma_{\rm L}^{\rm d}}} = \sqrt{\sigma^{\rm p}} \sqrt{\frac{\sigma_{\rm L}^{\rm p}}{\sigma_{\rm L}^{\rm d}}} + \sqrt{\sigma^{\rm d}}$$

therein  $y = \frac{\sigma_L(1+\cos\theta)}{2\sqrt{\sigma_L^d}}$ ,  $x = \sqrt{\frac{\sigma_L^p}{\sigma_L^d}}$ , the slope  $m = \sqrt{\sigma^p}$ , and the intercept  $c = \sqrt{\sigma^d}$ . By fitting the line y

= m \* x + c, we obtain  $\sqrt{\sigma^p}$  and  $\sqrt{\sigma^d}$ , then the total surface tension  $\sigma = \sigma^p + \sigma^d$ .

Note S2. Estimation of Fermi energy levels from KPFM data.

The data obtained from KPFM measurements is contact potential difference (CPD, i.e., the mean value here). When DC voltage ( $V_{dc}$ ) is applied to a sample, the relationship between the measured CPD and the work function ( $\Phi$ ) of the sample is  $V_{CPD} = (\Phi_{sample} - \Phi_{tip})/e$ . From Figure 2g, h and S15, we can list the following equations:

$$V_{\text{CPD-HOPG}} = (\Phi_{\text{HOPG}} - \Phi_{\text{tip}}) / e = 0.139 \text{ V},$$
$$V_{\text{CPD-Control}} = (\Phi_{\text{Control}} - \Phi_{\text{tip}}) / e = -0.367 \text{ V},$$
$$V_{\text{CPD-Target}} = (\Phi_{\text{Target}} - \Phi_{\text{tip}}) / e = -0.145 \text{ V},$$

then we obtain  $\Phi_{\text{Control}} - \Phi_{\text{HOPG}} = -0.506 \text{ eV}$ , and  $\Phi_{\text{Target}} - \Phi_{\text{HOPG}} = -0.284 \text{ eV}$ . Since  $\Phi_{\text{HOPG}} = 4.5 \text{ eV}$ ,  $\Phi_{\text{Control}} = 3.994 \text{ eV}$  and  $\Phi_{\text{Target}} = 4.216 \text{ eV}$  can be calculated.