

Supporting Information (SI)

Near-infrared light induced adsorption-desorption cycle for VOC recovery by integration of metal-organic frameworks with graphene oxide nanosheets

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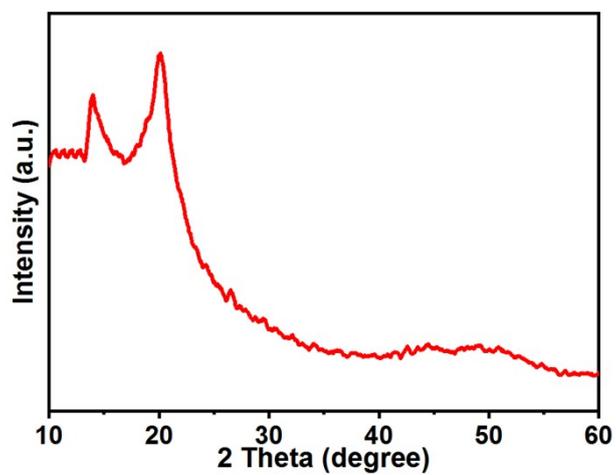


Fig. S1 Powder XRD patterns of GO nanosheets.

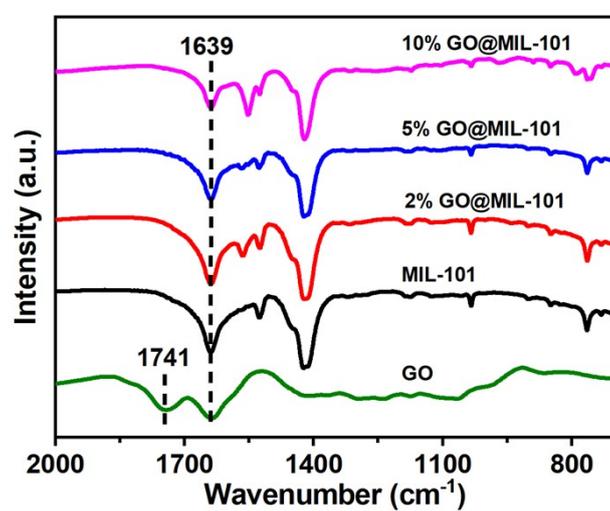


Fig. S2 FT-IR spectra of pure MIL-101 and GO@MIL-101 with GO nanosheets loading amount of 2, 5 and 10 wt%, respectively.

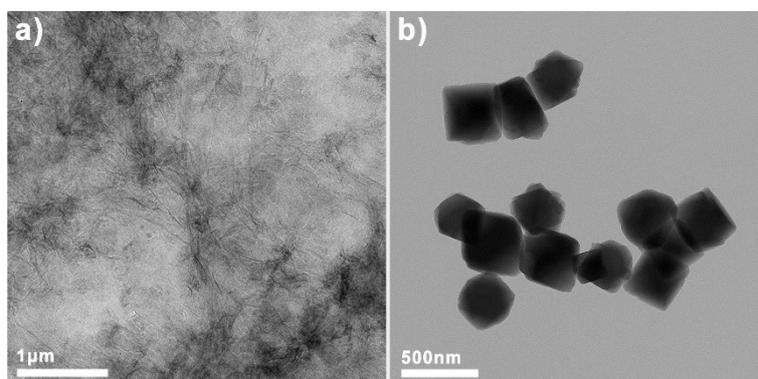


Fig. S3 TEM images of GO nanosheets (a) and pure MIL-101 (b).

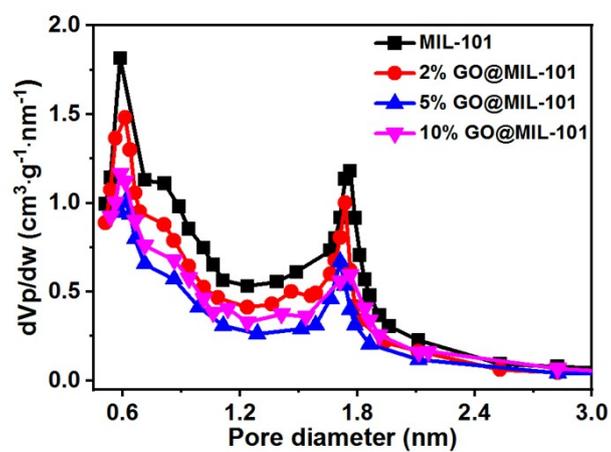


Fig. S4 Pore-size distribution curves of pure MIL-101 and GO@MIL-101 with GO nanosheets loading amount of 2, 5 and 10 wt%, respectively.

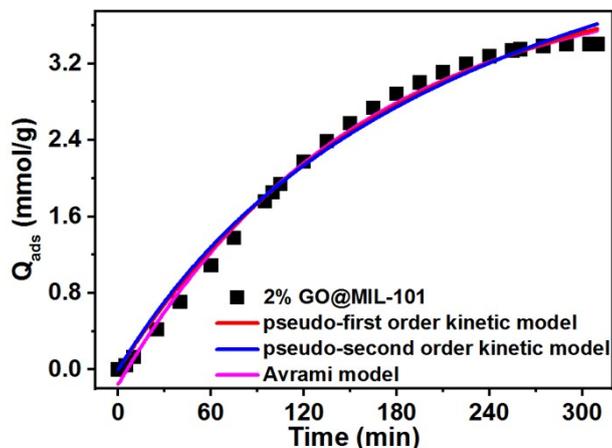


Fig. S5 The experimental results for ethyl acetate uptake (symbols) and corresponding fitting curves (lines) by pseudo-first-order, pseudo-second-order and Avrami kinetic models of GO@MIL-101 with GO nanosheets loading amount of 2 wt%.

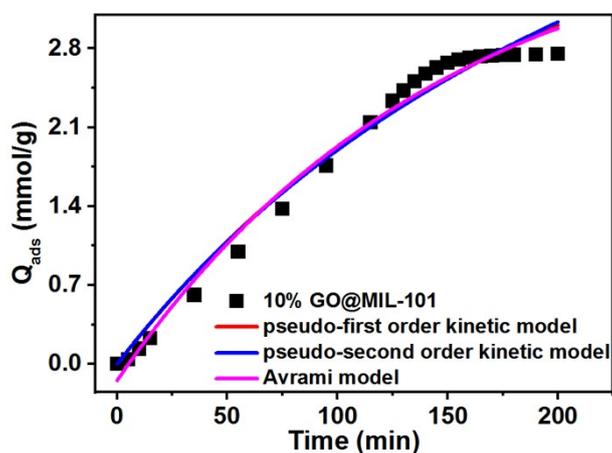


Fig. S6 The experimental results for ethyl acetate uptake (symbols) and corresponding fitting curves (lines) by pseudo-first-order, pseudo-second-order and Avrami kinetic models of GO@MIL-101 with GO nanosheets loading amount of 10 wt%.

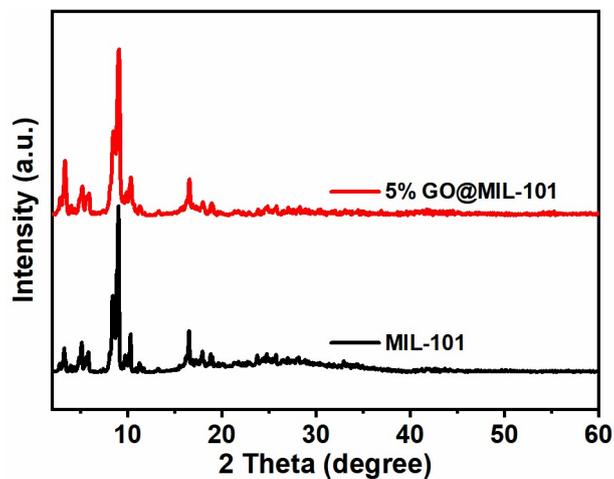


Fig. S7 Powder XRD patterns of pure MIL-101 and 5% GO@MIL-101 after five adsorption-desorption cycles under UV-vis light irradiation.

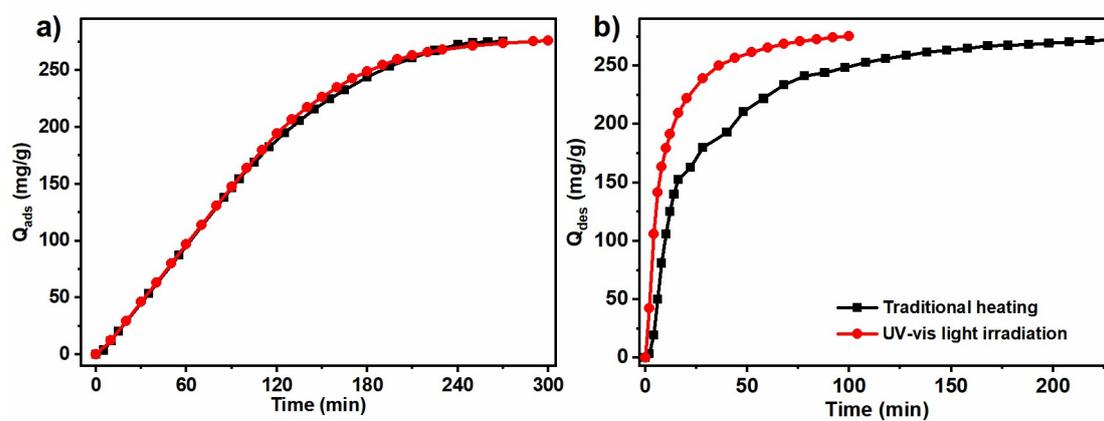


Fig. S8 The ethyl acetate adsorption (a) and desorption performance (b) of the 5% GO@MIL-101 under traditional heating and UV-vis light irradiation.

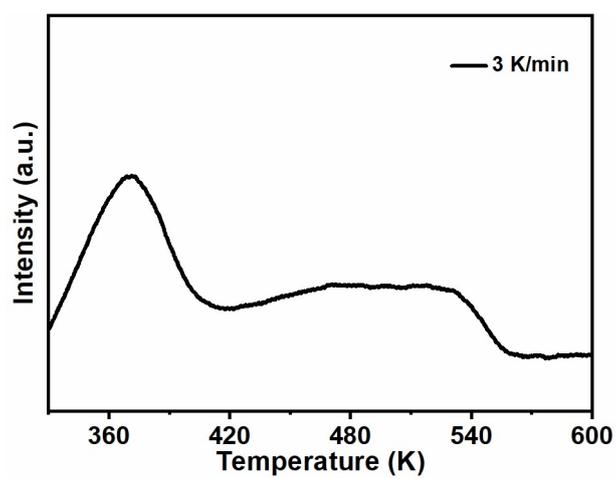


Fig. S9 TPD spectra of ethyl acetate adsorption onto the 5% GO@MIL-101.

Table S1 The saturated adsorption capacity of ethyl acetate onto MIL-101 and GO@MIL-101 as well as their corresponding desorption capacity under UV-vis light irradiation.

Sample	Adsorption capacity (mg/g)	Desorption time (min)	Desorption efficiency (%)
MIL-101	389.5	300	74.4
2% GO@MIL-101	299.9	300	87.2
5% GO@MIL-101	275.5	98	99.9
10% GO@MIL-101	242.3	300	89.1

Table S2 The saturated adsorption capacity of ethyl acetate onto 5 % GO@MIL-101 and the corresponding desorption capacity within 100 min of UV-vis light irradiation during the five adsorption-desorption cycles.

Cycles	Adsorption capacity (mg/g)	Desorption capacity (mg/g)
First	275.4	275.2
Second	275.3	274.0
Third	267.6	266.3
Fourth	265.0	263.7
Fifth	271.6	268.9

Table S3 The saturated adsorption capacity of ethyl acetate onto MIL-101 and the corresponding desorption capacity within 300 min of UV-vis light irradiation during the five adsorption-desorption cycles.

Cycles	Adsorption capacity (mg/g)	Desorption capacity (mg/g)
First	389.5	289.9
Second	314.4	256.7
Third	301.8	253.9
Fourth	294.1	259.4
Fifth	294.9	257.2

Table S4 The saturated adsorption capacity of ethyl acetate onto MIL-101 and GO@MIL-101 as well as the corresponding desorption capacity under UV-vis-NIR light irradiation.

Sample	Adsorption capacity (mg/g)	Desorption time (min)	Desorption capacity (mg/g)
MIL-101	383.88	150	267.19
5% GO@MIL-101	281.05	60	281.43

Table S5 Acidity characteristics of MIL-101 and 5 % GO@MIL-101 adsorbent.

Sample	C_L (umol/g)
MIL-101	111.0
5% GO@MIL-101	29.1

Table S6 Desorption peak temperatures of ethyl acetate at different heating rates with different desorption activation energies (E_d) of ethyl acetate onto MIL-101 and GO@MIL-101 calculated from the TPD results.

Sample	The peak temperature T_p (K) at different heating rates (K/min)					E_d (kJ/mol)
	3	4	5	6	7	
MIL-101	366.7	370.2	374.0	377.0	383.0	53.90
2% GO@MIL-101	371.1	376.9	380.3	383.1	387.9	40.15
5% GO@MIL-101	368.6	373.7	378.9	382.6	385.7	34.82
10% GO@MIL-101	356.8	362.8	367.7	374.4	377.8	25.69