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# **Supporting Information for**

# Discovery and potential ramifications of reduced iron-bearing nanomaterials—magnetite, wüstite, and zero valent iron—in wildland-urban interface fire ashes

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Sample number	Sampling date	Ash Source	Sample description							
Lightning Complex Fire										
A11	10/7/2020	Structure/Residential	Burned shed							
A12	10/7/2020	Structure/Residential	Burned shed - replicate							
A13	10/7/2020	Vehicle	Burned trailer - south end							
A14	10/7/2020	Vehicle	Burned trailer - north end							
A21	10/7/2020	Structure/Residential	Back room							
A22	10/7/2020	Structure/Residential	Front room							
A23	10/7/2020	Structure/Residential/kitchen	Kitchen area							
A24	10/7/2020	Structure/Foundation	Dark green material on foundation (Cu arsenate?)							
A25	10/7/2020	Structure/Residential	Tool shed - separated from house							
A91	10/15/2020	Structure/Residential	Hot tub / spa							
A92	10/15/2020	Structure/Residential	Red soil - artificial wood from deck							
A93	10/15/2020	Structure/Residential/kitchen	Kitchen area, copper wires							
A94	10/15/2020	Structure/Garage	Garage-1, electronics, copper wires, other blue material, misc. ash							
A95	10/15/2020	Structure/Garage	Garage-2, metal wires, misch ash							
A96	10/15/2020	Structure/Residential	Other parts of house, some electronics (Cu), some CCA treated wood.							
A121	10/15/2020	Structure/Residential	Burned residence - some copper wire, mostly white ash (wallboard)							
A122	10/15/2020	Structure/Residential/paint	Studio - some burned paint (acrylic and watercolor) blues, yellow, red							
A123	10/15/2020	Structure/Residential	Studio - misc. ash, some wires							
A124	10/15/2020	Structure/Residential/paint	Shed - house paint, some wires							
A125	10/15/2020	Structure/Foundation	Olive green ash from foundation, some yellow crystals replacing wallboard							
A131	10/16/2020	Structure/vehicle	Burned barn - redwood, galvanized steel, tires, Cu wire							
A132	10/16/2020	Structure/vehicle	Burned tool storage shed - tires, white & black ash, misc. bags & containers							
A133	10/16/2020	Structure/Residential	Burned farm worker residence - white & black ash; mattress, sink, refrigerator							
A134	10/16/2020	Vehicle	Burned boat & trailer - fiberglass, aluminum, tires							
A135	10/16/2020	Structure/vehicle	Burned farm workshop - tools, Cu wire, tractor, tires; some blue pigments 2 shades), black, white, red & brown ash; white powder							
A136	10/16/2020	Vehicle	Burned trailer (for 2 horses?) - tires, Cu wire, aluminum, book (manual?)							
A137	10/16/2020	Structure/Residential	Burned farmworker residence - black, white & brown ash							
A138	10/16/2020	Structure/Residential	Burned 1-story residence (occupied) white, gray & red ash; burned guns							
A139	10/16/2020	Structure/Residential	Burned 2-story residence (unoccupied) white, brown & red ash (kitchen)							
A141	10/16/2020	Structure/Commercial	North end of store - gray & white ash, fiberglass, dry wall, beverage containers							
A142	10/16/2020	Structure/Commercial/Electronics	South end of store - Electronics (PC and calculator?), Cu wires, white & gray ash							
A143	10/16/2020	Structure/Commercial	Storage building - Gray & white ash, Cu wiring (2 shades of green)							
A31	10/8/2020	Vegetation	Burned chaparral (charred manzanita and some pine) site on gentle slopes; serpentinite unit							
A51	10/8/2020	Vegetation	Burned chapparal (manzanita), white ash with some black ash; sandstone unit							
A81	10/8/2020	Vegetation	Burned oak, white ash with some black ash; bedrock undetermined							

**Table S1.** Description of ashes collected following the 2020 fire season. Samples analyzed by TEM are highlighted in gray.

North Complex Fire								
AD		Atmospheric deposition	Ash deposited on a car windshield					
NC-01	10/7/2020	Vegetation	White ash from edge of Lake Madrone (west shore)					
NC01-A	10/7/2020	Vegetation	White ash appears to be remnants of dead tree. Very light and fluffy.					
NC01-B	10/7/2020	Vegetation	Black ash from surface of soil. Appears to be duff layer					
NC01-C	10/7/2020	Soil	Scraped ash away from surface soil and collected 0-2 cm. Soil wa clay rich, red-orange color, very dense. We could not dig down w the plastic scoop (dry clay)					
NCWZ-02	10/7/2020	Structure/Residential	House "ash" collected from bedroom or living room area. Removed large tile material and collected from below large pieces					
NC04-A	10/7/2020	Structure/Garage	White ash from auto garage containing many tools. Collected from a large pile of sifted ash in center of garage					
NC04-B	10/7/2020	Vehicle	Composite car ash					
NC04-C	10/7/2020	Vehicle	Tire material (collected in glass for organic analysis)					
NC06-A	10/7/2020	Soil	Soil 0-2 cm					
NC06-B	10/7/2020	Soil	Soil 10-15 cm					
NC06-C	10/7/2020	Vegetation	Ash from soil surface					
NC-08	10/7/2020	Structure/Residential	Composite sample of cabin built in 1930's era					
NC-09	10/7/2020	Vehicle	Golf cart ash composite					
NC10-A	10/7/2020	Structure/Residential	Cabin. Composite sample collected from inside foundation.					
NC10-B	10/7/2020	Soil	Surface soil/ash collected 5 ft outside of foundation, adjacent to building					
NC11-A	10/7/2020	Vehicle	Car ash, composite from ground, early 2000's Ford Mustang					
NC11-B	10/7/2020	Vehicle	Car ash, interior					
NC12-A	10/7/2020	Vegetation	Ash from pine forest. High severity. Above reservoir.					
NC12-B	10/7/2020	Soil	Soil 0-2 cm					
NC12-C	10/7/2020	Soil	Soil 10-15cm					

#### Soil description

#### Soils in the North Complex Fire area

Most soils around Madrone Lake are in the Islandbar series (primarily decomposed granite) 94

# NCWZ 12: Soil classification 209 – ISLANDBAR SERIES

The Islandbar series consists of very deep, somewhat excessively drained soils that formed in residuum and colluvium from coarse grained intrusive igneous rocks, mainly trondhjemite or quartz diorite. Islandbar soils are on ridge tops and side slopes on granitic Sierra Nevada. Slopes range from 2 to 70%. The mean annual precipitation is about 1473 mm and the mean annual temperature is about 13 °C.

TAXONOMIC CLASS: Coarse-loamy, mixed, active, mesic Typic Dystroxerepts

**TYPICAL PEDON:** Islandbar sandy loam, on a northwest facing 5 percent slope under a cover of conifers, hardwoods, and shrubs at an elevation of 677 m. When described on 7/22/1998, the soil was dry to a depth of 91 cm and was slightly moist from 91 to 183 cm. (Colors are for dry soil unless otherwise noted).

**Oi**--0 to 5 cm; slightly decomposed plant material; abrupt smooth boundary.

A1--5 to 13 cm; light brownish gray (10YR 6/2) sandy loam, dark grayish brown (10YR 4/2) moist; 12% clay; weak fine subangular blocky parting to weak fine granular structure; slightly hard, very friable, nonsticky, nonplastic; many very fine and fine roots; many very fine and fine irregular pores; 5 percent mica flakes; 5 percent rounded gravel; moderately acid, pH 6.0 by Hellige-Truog; abrupt smooth boundary.

#### NCWZ 1, NCWZ 6, and NCWZ 10B: Soil classification 202 – Holland Series

The Holland series consists of very deep, well drained soils that formed in material weathered from granitic rock. Holland soils are on foothills and low mountains and have slopes of 2 to 75%. The mean annual precipitation is about 940 mm and the mean annual air temperature is about 13 °C.

TAXONOMIC CLASS: Fine-loamy, mixed, active, mesic Ultic Haploxeralfs

**TYPICAL PEDON:** Holland sandy loam - on an east facing, linear, slope of 25% under ponderosa pine, incense cedar, and mountain misery at an elevation of 1145 m. (Colors are for dry soil unless otherwise noted. When described August 24, 1977, the soil was slightly moist throughout).

Oi--0 to 5 cm; fresh and decomposing pine needle and leaf litter.

A1--5 to 13 cm; brown (10YR 5/3) sandy loam, very dark grayish brown (10YR 3/2) moist; 17% clay; strong fine granular structure; soft, very friable, slightly sticky and slightly plastic; many very fine and fine roots; many very fine interstitial, common fine tubular pores; moderately acid (pH 5.7); abrupt wavy boundary. (5 to 15 cm thick)

#### NCWZ 13: Soil classification – Hartsmill Series

The Hartsmill series consists of very deep, well drained soils that formed in colluvium and residuum from metavolcanic rocks, mainly greenschist. Hartsmill soils are on ridge tops and side slopes on metamorphic Sierra Nevada foothills. Slopes range from 2 to 90%. The mean annual precipitation is about 43 inches, (109 cm) and the mean annual temperature is about 14 °C.

TAXONOMIC CLASS: Loamy-skeletal, mixed, superactive, thermic Ultic Palexeralfs

**TYPICAL PEDON:** Hartsmill gravelly loam, on a southwest facing 46 percent slope under a cover of whiteleaf manzanita (*Arctostaphylos manzanita*), toyon (*Heteromeles arbutifolia*), interior live oak (*Quercus wislizeni*), California black oak (*Quercus kelloggii*), Pacific poison oak (*Toxicodendron diversilobum*) and very scattered ponderosa pine (*Pinus ponderosa*) at an

elevation of 1,840 feet, (561 m). When described on 7/27/1999, the soil was dry throughout. (Colors are for dry soil unless otherwise noted).

**Oi**--0 to 3 cm; slightly decomposed plant material; abrupt smooth boundary.

A--3 to 8 cm; reddish brown (5YR 4/4) gravelly loam, dark reddish brown (5YR 3/3), moist; 14% clay; strong fine granular structure; soft, very friable, nonsticky, slightly plastic; common very fine roots; many very fine irregular pores; 30 percent angular gravel; slightly acid, pH 6.5 by Hellige-Truog; abrupt smooth boundary. 5 to 8 cm thick).



Figure S1. Land use classification for the North Complex fire and approximate locations of ash samples.



**Figure S2.** Land use classification for the LNU Lightning Complex fire and approximate locations of ash samples.



Figure S3. Collection of residential structure ash and soil samples

# **Synthesis of Goethite**

Goethite was synthesized using an aqueous aging of ferrihydrite modified method after Schwertmann et al. (1985) <sup>51</sup>. Sodium hydroxide solution (5 M, 120 ml) was added dropwise to an iron (III) nitrate solution (1 M, 200 ml). The resulting ferrihydrite suspension was basified by dropwise addition of sodium hydroxide (5 M) until pH 12 was achieved. The bottle was then closed and heated in a furnace for 3 days. The initial temperature was 50 °C, however, this was increased to 60 °C after the first day of the reaction. The product was recovered by centrifugation. The product was then washed four times and let to dry.

# Synthesis magnetite and maghemite

Magnetite nanoparticles were synthetized by aqueous coprecipitation of ferric and ferrous iron in an anaerobic chamber in strict anaerobic conditions in pH 12 deionized water. Magnetite nanoparticles were then vacuum-dried on 0.2  $\mu$ m cellulose filter and washed with O<sub>2</sub>-free deionized water. Average diameter is 10.0 ± 1.4 nm, determined by High Resolution Transmission Electron Microscopy (HR-TEM). X-ray diffraction (XRD) was used to check the purity of the phase, it indicated a pure magnetite phase with a mean coherent domain size of 12.7 nm.

Maghemite nanoparticles were obtained by heating the previously described nanoparticles at 200°C during one night in strict anaerobic conditions. XRD indicated a pure maghemite phase with a 11.4 nm mean coherent domain size and an average diameter size of  $10.3 \pm 2.7$  nm (HR-TEM).

#### Synthesis of Hematite

Hematite was synthesized by aqueous precipitation of ferric iron. Sodium hydroxide solution (5 M) was added dropwise to an iron (III) nitrate solution (1 M) until pH 7 was achieved. The resulting hematite solution was washed with Milli-Q water several times. The pH is then

adjusted again to 7 using NaOH (1M). The bottle was then closed and heated in a furnace at 80 °C. The product was recovered by centrifugation. The product was then washed four times and let to dry.



**Figure S4.** Iron K-edge XANES spectra of the model compound used to fit the iron K-edge XANES spectra of the fire ashes.

Iron oxidation state		Ee(III)					Fe(III)/Fe(II) Fe(II)					Fe(0)		
Iron species	Ferrihvdrite	Maghemite	Goethite	Hematite	Fe(III)chlorite	Fe(III)nitrate	Fe(III)sulfate	Magnetite	FeS	Fe(II)chlorite	FeO	Fe(II)sulfate	Elemental Fe	X2
A11	25 ± 3	75 ± 3	0	0	0	0	0	0	0	0	0	0	0	0.021
A12	0	84 ± 2	0	0	0	0	16 ± 2	0	0	0	0	0	0	0.053
A13	65 ± 7	0	0	0	0	0	0	35 ± 6	0	0	0	0	0	0.039
A14	21 ± 4	26 ± 2	0	0	16 ± 1	0	0	37 ± 4	0	0	0	0	0	0.015
A21	51 ± 2	35 ± 2	0	0	0	0	0	0	0	0	0	0	14 ± 1	0.030
A22	0	53 + 2	23 + 2	0	0	0	0	0	14 + 1	10 + 1	0	0	0	0.018
A23	63 + 4	14 + 4	0	0	0	0	0	0	13 + 1	0	0	0	10 + 1	0.028
A24	0	78 + 1	0	0	22 + 1	0	0	0	0	0	0	0	0	0.034
A25	0	69 + 3	0	0	165 + 1	0	0	15 + 2	0	0	0	0	0	0.004
A91	0	0010	19 + 2	0	0	0	0	81 + 2	0	0	0	0	0	0.022
A92	0	90 + 1	0	0	0	0	0	0	0	10 + 1	0	0	0	0.049
A03	0	15 + 2	0	0	0	0	0	46 + 2	15 + 1	23 + 1	0	0	0	0.040
A93	0	17 + 2	0	0	0	0	0	40±2 53+2	0	0	0	0	0	0.011
A05	13 + 3	38 + 3	0	0	0	0	0	50 ± 2	0	0	0	0	0	0.023
A06	0	25 ± 2	0	0	0	0	0	24 ± 4	0	0	15 ± 1	26 ± 1	0	0.027
A121	0	35 ± 3	0	0	0	0	0	18 + 3	0	17 + 1	0	2011	0	0.040
A121	0	30±3	0	0	0	0	0	40 ± 3	0	0	0	16 ± 1	0	0.029
A122	40 1 2	40 ± 2	0	0	0	0	0	44 ± 3	0	10.1.1	0	10 ± 1	10 1 1	0.013
A123	49 ± 3	30 ± 3	0	50 J 1	0	0	0	0	0	10 ± 1	0	0	10 ± 1	0.020
A124	0	50 ± 1	0	50 ± 1	0	0	0	0	0	0	0	0	0	0.006
A125	0	51 ± 2	0	0	0	0	0	31 ± 2	10 ± 1	0	0	0	0	0.019
A131	21 ± 2	50 ± 2	0	0	0	0	0	29 ± 1	0	0	0	0	0	0.010
A132	0	38 ± 4	0	0	10 ± 1	0	0	52 ± 3	0	0	0	0	0	0.041
A133	0	45 ± 2	0	0	10 ± 1	0	0	32 ± 3	0	0	0	13±1	0	0.015
A134	0	53 ± 2	26 ± 2	0	0	0	0	0	21 ± 1	0	0	0	0	0.024
A135	0	46 ± 2	36 ± 2	0	0	0	0	0	19 ± 1	0	0	0	0	0.027
A136	0	21 ± 2	44 ± 1	0	0	0	0	35 ± 2	0	0	0	0	0	0.017
A137	0	57 ± 3	0	0	0	0	0	32 ± 3	0	11 ± 1	0	0	0	0.014
A138	0	34 ± 2	20 ± 2	0	0	0	0	46 ± 2	0	0	0	0	0	0.029
A139	0	44 ± 2	0	0	0	0	0	45 ± 3	0	0	0	12 ± 1	0	0.018
A141	0	0	0	0	0	0	0	65 ± 1	22 ± 1	23 ± 1	0	0	0	0.011
A142	25 ± 3	26 ± 3	0	0	0	0	0	27 ± 3	13 ± 1	0	0	10 ± 1	0	0.013
A143	42 ± 2	14 ± 2	0	0	0	0	0	35 ± 2	10 ± 1	0	0	0	0	0.011
A031	17 ± 3	33 ± 4	0	0	0	0	0	33 ± 4	0	17 ± 2	0	0	0	0.040
A051	0	44 ± 2	0	0	13 ± 1	0	0	28 ± 2	0	15 ± 1	0	0	0	0.013
A081	0	43 ± 2	0	0	10 ± 1	0	0	47 ± 1	0	0	0	0	0	0.012
AD	0	33 ± 3	0	0	0	0	0	57 ± 3	0	0	0	10 ± 1	0	0.025
NC-1	0	32 ± 3	0	18 ± 2	0	0	0	50 ± 2	0	0	0	0	0	0.029
NC-1A	0	50 ± 2	33 ± 1	0	0	0	0	17 ± 2	0	0	0	0	0	0.022
NC-1B	24 ± 5	20 ± 4	0	31 ± 4	0	0	0	25 ± 5	0	0	0	0	0	0.038
NC-1C	16 ± 7	43 ± 3	0	25 ± 4	15 ± 2	0	0	0	0	0	0	0	0	0.033
NC-2	33 ± 3	0	0	0	0	0	0	31 ± 4	0	0	21 ± 1	15 ± 1	0	0.020
NC-4B	20 ± 2	0	0	0	0	0	0	63 ± 2	17 ± 1	0	0	0	0	0.018
NC-4C	0	34 ± 2	0	31 ± 3	35 ± 1	0	0	0	0	0	0	0	0	0.028
NC-6A	0	34 ± 2	0	31 ± 3	35 ± 1	0	0	0	0	0	0	0	0	0.028
NC-6B	0	26 ± 2	0	48 ± 3	25 ± 1	0	0	0	0	0	0	0	0	0.015
NC-6C	0	29 ± 3	16 ± 5	35 ± 6	0	0	0	21 ± 2	0	0	0	0	0	0.037
NC-8	0	60 ± 3	0	0	0	0	0	29 ± 4	0	0	0	11 ± 1	0	0.038
NC-9	0	36 ± 3	0	0	0	0	0	54 ± 4	0	10 ± 2	0	0	0	0.044
NC-10A	15 ± 3	39 ± 4	0	0	0	0	0	33 ± 4	0	0	0	13 ± 1	0	0.034
NC-10B	0	25 ± 3	0	0	0	0	0	58 ± 4	0	0	0	18 ± 1	0	0.042
NC-11A	0	0	0	0	0	0	0	48 ± 1	17 ± 2	0	20 ± 3	16 ± 2	0	0.045
NC-11B	0	0	0	0	0	0	0	26 ± 1	21 ± 1	0	53 ± 1	0	0	0.016
NC-12A	0	0	0	0	0	11 ± 1	0	58 ± 2	0	30 ± 1	0	0	0	0.018
NC-12B	19 ± 6	16 ± 2	42 ± 4	0	0	0	0	0	0	23 ± 1	0	0	0	0.008
NC-12C	0	25 ± 1	54 ± 1	0	0	0	0	0	0	21 ± 1	0	0	0	0.009

 Table S2. Spectral weight of the different iron oxide phases and oxidation states together with the best fit error



Figure S5. Iron phases as a function of ash color



**Figure S6.** Percentage distribution of Fe species in ash samples determined by linear combination fitting analysis of Fe K-edge XANES (a). Estimated relative percentage of Fe oxidation states based on individual components determined from linear combination of XANES (b). Veg: vegetation, St: structure, Veh: vehicle, and AD: atmospheric deposition.



Figure S7. Iron phases as a function of ash source





**Figure S8.** Iron speciation in ashes generated as a result of fire at the wildland-urban interface (a) maghemite, (b) magnetite, (c) wüstite, and (d) zero valent iron. Veg: vegetation, St: structure, Veh: vehicle, and AD: atmospheric deposition.

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