juice (FLJ)				
Basic indicators	LJ	FLJ		
Viable count (log CFU/mL)	\	$9.25\pm0.10^{\boldsymbol{*}}$		
GABA (mg/100 mL)	108 ± 4	$125 \pm 6*$		
Total carbohydrate (mg/mL)	194 ± 5	$154 \pm 6*$		
Glucose (mg/mL)	90 ± 4	$76 \pm 1.3*$		
Fructose (mg/mL)	108 ± 5	$90 \pm 1.5*$		
Sucrose (mg/mL)	43.5 ± 0.8	$11.9\pm0.4*$		
Titratable acid (mg/mL)	1.23 ± 0.03	$7.63\pm0.24*$		
pH	5.39 ± 0.07	$3.70\pm0.01\text{*}$		
Total phenolics (mg/mL)	0.65 ± 0.12	0.67±0.13		
Soluble protein (mg/mL)	1.24 ± 0.02	1.02 ± 0.5		

Supplementary Table 1 Physicochemical parameters in litchi juice (LJ) and fermented litchi 1

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Values are expressed as mean \pm SD, n=3; *p < 0.05, compared with LJ

NI4:4:		D124	450B	D12492	
Nutrition	Ingredient	Content	Calories	Content	Calories
component		(g/kg)	(kcal%)	(g/kg)	(kcal%)
	Casein	200	20	200	20
Protein	L-cystine	3	3	20	
	Corn starch	315		0	
Caulta haadaataa	Maltodextrin	35	70	125	20
Carbohydrates	Sucrose	350	70	70 <u>6808</u> 20	20
	Cellulose	50		50	
Γ.	Soya-bean oil	25	10	25	(0)
Fat	Lard oil	20	10	245	60
	Mineral	10		10	
	Hydryoxyapetite	13		13	
Otherm	Calcium carbonate	5.5		5.5	
Others	Potassium citrate	16.5		16.5	
	Vitamin	10		10	
	Choline tartrate	2		2	
	Total calories		3.85		5.24

8	Supplementary	Table 3 Rep	presentative MS	fragments of	the major	phenolic ac	ids in litchi juice

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(LJ) and fermented litchi juice (FLJ).

Peak no.	RT (min)	Compound name	Formula	Mass (m/z)	LJ (ng/mL)	FLJ (ng/mL)
1	0.92	Gallic acid	C ₇ H ₆ O ₅	170.02	69.40±0.97	39.62±2.09
2	1.09	Phenylalanine	$C_9H_{11}NO_2$	165.08	36.24±3.44	14.89±5.03
3	1.83	3,4-Dihydroxybenzoic acid	$C_7H_6O_4$	154.03	366.99±14.16	201.33±21.45
4	3.08	Protocatechualdehyde	$C_7H_6O_3$	138.03	428.06±0.14	293.57±36.31
5	3.30	4-Hydroxybenzoic acid	$C_7H_6O_3$	138.03	130.27±15.87	82.09±3.24
6	4.07	Vanillic acid	$C_8H_8O_4$	168.04	49.35±7.89	15.30±1.06
7	4.24	Caffeic acid	$C_9H_8O_4$	180.04	8.14±0.52	8.55±0.34
8	4.44	Syringic acid	$C_9H_{10}O_5$	198.05	16.16±1.83	10.56±0.40
9	5.03	Vanillin	$C_8H_8O_3$	152.05	1.47±0.05	0.83±0.07
10	5.23	p-Hydroxycinnamic acid	$C_9H_8O_3$	164.05	49.72±1.51	29.08±0.40
11	5.38	Syringaldehyde	$C_9H_{10}O_4$	182.06	2.21±0.28	1.73±0.05
12	5.49	Salicylic acid	$C_7H_6O_3$	138.03	246.58±2.50	339.90±5.82
13	5.59	Trans-Ferulic acid	$C_{10}H_{10}O_4$	194.06	11.18±0.42	5.82±0.04
14	5.64	Sinapic acid	$C_{11}H_{12}O_5$	224.07	3.66±0.24	2.27±0.14
15	5.92	Benzoic acid	$C_7H_6O_2$	122.04	246.28±8.20	193.29±4.44
16	7.04	Hydrocinnamic acid	$C_{9}H_{10}O_{2}$	150.07	$1.34{\pm}0.00$	2.69±0.45
17	7.15	Trans-Cinnamic acid	$C_9H_8O_2$	148.05	10.92±0.89	5.12±0.07
Total					1677.98	1246.63

12	Supplementary	y Table 4 Re	presentative MS	fragments of the	major flavo	noids in litchi	juice (LJ)	

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and fermented litchi juice (FLJ).

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Peak no.	RT (min)	Compound name	Formula	Mass(m/ z)	LJ (ng/mL)	FLJ (ng/mL)
1	3.89	Catechin	C ₁₅ H ₁₄ O ₆	290.08	3.56±0.30	17.66±1.18
2	4.76	L-Epicatechin	$C_{15}H_{14}O_{6}$	290.08	3.61±0.42	393.75±7.80
3	4.77	Dihydromyricetin	$C_{15}H_{12}O_8$	320.05	17.68±0.05	17.92±0.07
4	5.57	Rutin	$C_{27}H_{30}O_{16}$	610.15	2303.92±139.50	2192.30±8.22
5	5.71	Quercetin 3-β-D- glucoside	$C_{21}H_{20}O_{12}$	464.09	2.85±0.33	2.89±0.35
6	5.78	(+)-Dihydroquercetin	$C_{15}H_{12}O_{7}$	304.06	2.70±0.33	9.73±0.35
7	6.05	Kaempferol-3-O- glucoside	C ₂₁ H ₁₉ O ₁₁	448.10	8.29±0.28	7.00±0.02
8	6.38	(+)-Dihydrokaempferol	$C_{15}H_{12}O_{6}$	288.06	3.15±0.32	7.85±0.04
9	7.00	Luteolin	$C_{15}H_{10}O_{6}$	286.05	0.65 ± 0.46	0.77±0.21
10	7.03	Quercetin	$C_{15}H_{10}O_7$	302.04	2.75±0.99	8.94±0.55
11	7.52	Apigenin	$C_{15}H_{10}O_5$	270.05	/	0.48 ± 0.06
12	7.54	Naringenin	$C_{15}H_{12}O_5$	272.07	$0.69{\pm}0.06$	1.13±0.04
13	7.62	Kaempferol	$C_{15}H_{10}O_{6}$	286.05	0.74±0.39	0.99±0.20
14	7.69	Isorhamnetin	$C_{16}H_{12}O_7$	316.06	2.33±0.67	3.25±0.04
Total					2352.92	2664.65

	Supplementary Table 5 HPLC au	tosampler program for	OPA derivatization of GABA
Step	Action	Vial ^a	Volume (µL)
1	Draw	1	10
2	Draw	Sample	2
3	Mix ^b		12
4	Draw	2	0
5	Draw	3	2
6	Mix ^c		14
7	Draw	2	
8	Inject		
	1 2 3 4 5 6 7	StepAction1Draw2Draw3Mix ^b 4Draw5Draw6Mix ^c 7Draw	1Draw12DrawSample3Mix ^b 44Draw25Draw36Mix ^c 77Draw2

Supplementary Table 5 HPLC autosampler program for OPA derivatization of GABA

17 ^a Vial 1: borate buffer solution; vial 2: rinse solvent(acetonitrile); vial 3: OPA derivatization reagent.

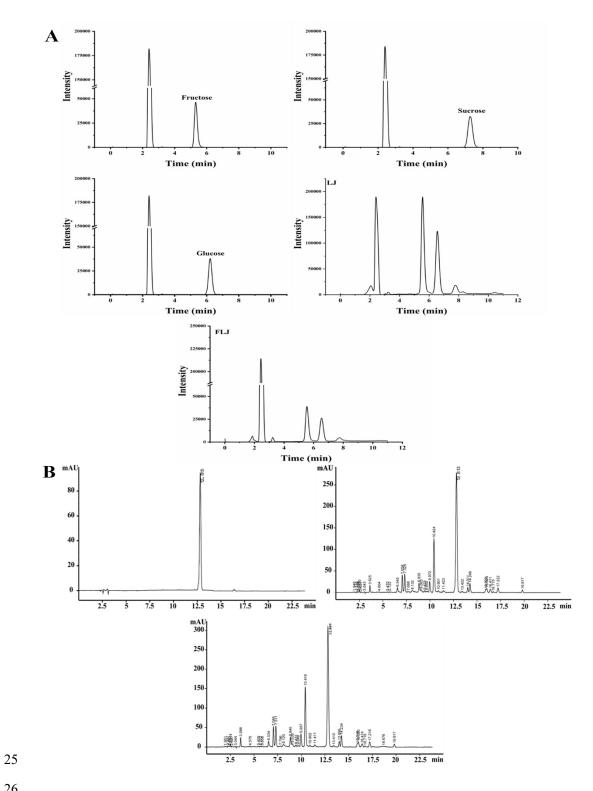
18 ^b In air; default speed; three times.

19 ° In air; default speed; fifteen times.

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Time/min —	Mobile	phase/%
Time/min —	А	В
0	90	10
20	55	45
25	90	10

22 Roles: solvent A: 25 mill solitain accure, pri 5.9, solvent B:



Supplementary Figure 1. HPLC analysis of monosaccharides (A) and GABA (B) in litchi juice (LJ) and fermented litchi juice (FLJ).

30 Supplementary methods:

31 1. Chromatography and MS

32 1.1 Chemicals and reagents

Methanol, acetonitrile, acetic acid was purchased from ANPEL. All solvents were of LC-MS grade. Sodium hydroxide, hydrochloric acid, ethyl acetate was purchased from Sinopharm which were AR Grade. And ultra-pure water in-house prepared using a Milli-Q water purification system (Millipore, Bedford, MA, USA).

37 1.2 Sample preparation and extraction

Samples were treated with 2 mL of 4 M aqueous NaOH. The mixed solution was 38 hydrolyzed at 40 °C for 2 h in a gas bath with shaking and protection from light. The 39 pH value was adjusted to 2 by adding 4 M aqueous HCl. The mixture was shaken with 40 2 mL of n-hexane at room temperature for 20 min to remove the n-hexane layer. Ethyl 41 acetate $(2 \times 2 \text{ mL})$ was used to extract the aqueous layer, and the mixed extracts were 42 concentrated to nearly dry on a rotary evaporator at 35 °C under reduced pressure. 43 Before analysis, the residue was dissolved in 200 µl of 50% methanol/water and 44 transferred to insert-equipped vials. 45

46 1.3 UPLC Conditions

The sample extracts were analyzed using an UPLC– Orbitrap-MS system (UPLC, Vanquish; MS, QE). The analytical conditions were as follows, UPLC: column, Waters ACQUITY UPLC HSS T3(1.8 μ m, 2.1 mm*50 mm); column temperature, 40 °C; flow rate, 0.3 mL/min; injection volume, 2 μ L; solvent system, water (0.1% acetic acid): acetonitrile (0.1% acetic acid); gradient program, 90:10 V/V at 0 min, 90:10 V/V at 2.0 min, 40:60 V/V at 6.0 min, 40:60 V/V at 8.0 min, 90:10 V/V at 8.1 min, 90:10 V/V at 12.0 min.

54 1.4 QE

HRMS data were recorded on a Q Exactive hybrid Q–Orbitrap mass spectrometer equipped with a heated ESI source (Thermo Fisher Scientific) utilizing the Full MS acquisition methods. The ESI source parameters were set as follows: spray voltage, -2.8 kV; sheath gas pressure, 40 arb; aux gas pressure, 10 arb; sweep gas pressure, 0 arb; capillary temperature, 320 °C; and aux gas heater temperature, 350 °C.

60 1.5 Data analysis and interpretation

Data were acquired on the Q-Exactive using Xcalibur 4.1 (Thermo Scientific), and
processed using TraceFinder[™]4.1 Clinical (Thermo Scientific). Quantified data were
output into Excel format.

64 2. Measurement of GABA contents

GABA contents were determined by HPLC (Agilent Technologies 1260 Infinity 65 II, Santa Clara, CA, USA) equipped with a Zorbax SB-C18 column (4.6 mm × 250 mm, 66 5 µm, Nacalai Tesque, Japan), using the method described by Moret et al. (2005) with 67 some modification. The samples of fermented litchi juice were centrifuged at 10,000×g 68 for 10 min. The resulting supernatants were filtered through a membrane filter (0.22 69 µm) and the filtrate was used for HPLC analysis. Pre-column derivatization of samples 70 was performed using the Agilent automated derivatization protocol with O-71 72 phthalaldehyde (OPA, Sigma Chemical Industries, Ltd., Saint Louis, MO, USA), and boric acid (pH 10.4) serving as buffer. Use of the Agilent automatic derivatization 73 injection program (Supplementary Table S5) was as follows: 10 µL of boric acid buffer 74 and 2 μ L of the sample were combined and mixed 3 times, the injection needle was 75 washed for 3 s with acetonitrile, 2 µL of the OPA derivatization reagent was added, the 76 injection needle was rewashed for 3 s with acetonitrile, and the 14 µL volume consisting 77 78 of sample, buffer and derivation reagent was mixed 15 times. Sample separation was accomplished using a gradient program (Supplementary Table S6) consisting of mobile 79 phase [solvent A: 25 mM sodium acetate, pH 5.9, solvent B: 100% acetonitrile] with a 80 flow rate of 1 mL/min in a column held at 40 °C. The OPA derivatives of GABA were 81 detected by a UV detector at 323 nm. The GABA contents were quantified based on 82 integrated peak area in comparison with standard curves prepared from known 83 concentrations of GABA standard. 84

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