Supporting Information

Targeted-regulating the selectivity of cascade synthesis towards imines/secondary amines by carbon-coated Co-based catalysts

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Figure S1. Representative SEM images of PtCo/SiO₂@CN



Figure S2. TEM images of the effect of ethanol, water and ammonia water on the morphology of SiO₂ balls and the degree of metal dispersion when the ratio of ethanol, water and ammonia water was 12.5 : 1.65 : 0.5. a, b) Co/SiO₂@CN; c, d)) PtCo/SiO₂@CN.



Figure S3. TEM images of the effect of ethanol, water and ammonia water on the morphology of SiO_2 balls and the degree of metal dispersion when the ratio of ethanol, water and ammonia water was 24 : 80 : 1. a, b, c, d) PtCo/SiO₂@CN.



Figure S4. a,b,c) Representative TEM images of Pt/SiO₂@CN.



Figure S5. a,b,c) Representative TEM and HRTEM images of PtCo/SiO₂@CN-HCl-21day and the corresponding EDS elemental mapping of Co and Pt.



Figure S6. a,b,c) Representative TEM and HRTEM images of PtCo/SiO₂-NaBH₄ and the corresponding EDS elemental mapping of Co and Pt.



Figure S7. a,b,c) Representative HRTEM images of Co/SiO₂@CN-Pt and the corresponding EDS elemental mapping of Co and Pt.



Figure S8. XRD images of Co/SiO₂@CN-Pt, PtCo/SiO₂@CN-HCl-21day and PtCo/SiO₂-NaBH₄.



Figure S9. Co 2p XPS spectrum images of Co/SiO₂@CN-Pt, PtCo/SiO₂@CN-HCl-1day, PtCo/SiO₂@CN-HCl-21day and PtCo/SiO₂-NaBH₄.



Figure S10. Pt 4f XPS spectrum images of PtCo/SiO₂@CN-HCl-1day, PtCo/SiO₂@CN-HCl-21day.



Figure S11. N 1s XPS spectrum images of Co/SiO₂@CN-HCl-1h and Co/SiO₂@CN-HCl-1day.



Figure S12. N 1s XPS spectrum images of Pt/SiO₂@CN, Co/SiO₂@CN-Pt, PtCo/SiO₂@CN-HCl-1day and PtCo/SiO₂@CN-HCl-21day.



Figure S13. Activity test of PtCo/SiO₂@CN at different temperatures a) room temperature, b) 50 °C.



PtCo/SiO₂@CN

Figure S14. Physical mixed solution of $PtCo/SiO_2@CN$ and WO_3 before and after the reaction. Reaction conditions:1.02 g mixture catalyst ($PtCo/SiO_2@CN : WO_3=1:50$), 600 r/min, 5 mL solvent, 1 h,1 Mpa H₂.





Figure S15. Physical mixed solution of $Pt/SiO_2@CN$ and WO_3 before and after the reaction. Reaction conditions:1.02 g mixture catalyst ($PtCo/SiO_2@CN : WO_3=1:50$), 600 r/min, 5 mL solvent, 1 h,1 Mpa H₂.



Figure S16. Solution of WO₃ before and after the reaction. Reaction conditions:1 g WO3, 600 r/min, 5 mL solvent, 4 h,1 Mpa H_2 .



Co/SiO₂@CN

Figure S17. Physical mixed solution of $Co/SiO_2@CN$ and WO_3 before and after the reaction. Reaction conditions:1.02 g mixture catalyst ($Co/SiO_2@CN : WO_3=1:50$), 600 r/min, 5 mL solvent, 4.5 h,1 MPa H₂.



Figure S18. a, b, c) Representative TEM and HRTEM images of the used $Co/SiO_2@CN$.



Figure S19. a, b, c) Representative TEM and HRTEM images of the used $PtCo/SiO_2@CN$.



Figure S20. a, b, c) Representative TEM images of the Co/SiO₂@CN-HCl-1day.





Figure S22. Effect of Pt in and out of carbon layer on reaction rate. a) hydrogenation of nitrobenzene, b) hydrogenation of benzaldehyde.



Figure S23. Pt 4f XPS spectra of PtCo/SiO₂@CN before and after NB adsorption.

Figure S21. Hot filtration experiments of Co/SiO2@CN.



Figure S24. Pt 4f XPS spectra of Co/SiO₂@CN-Pt before and after NB adsorption.



Figure S25. FT-IR spectra of PtCo/SiO₂@CN after treatment with BA and NB.



Figure S26. FT-IR spectra of Co/SiO₂@CN-Pt after treatment with BA and NB.



Figure S27. FT-IR spectra of $PtCo/SiO_2@CN$ and $Co/SiO_2@CN-Pt$ after treatment with the mixture of BA and NB.

Sample	Metal loading				
	Pt (wt.%)	Co (wt.%)			
PtCo/SiO2@CN	0.8	8.4			
Co/SiO2@CN-Pt	1.4	10.2			
PtCo/SiO ₂ -NaBH ₄	1.1	10.8			
Co/SiO ₂ @CN	/	9.0			
Pt/SiO2@CN	0.8	/			

 Table S1. ICP testing of each catalyst

Entry	Catalyst	T [ºC]	Pre. [MPa]	TOF[h ⁻	Selectivity[%]		
					Imine	Amine	Ref.
1	Co/SiO ₂ @CN	60	1	64	98.0		This work
2	PtCo/SiO2@CN	60	1	296		100	This work
3	CAT-450	120	2	-	0.7	85.2	[1]
4	Pd_3/γ - Al_2O_3	80	2	9.2	94.5		[2]
5	CoOx@NC-800	110	5	-	100		[3]
6	CoS ₂ @MoS ₂ 180-0.75	60	1.5	-	93	3.8	[4]
7	Ni ₃ Sn ₂ /TiO ₂	150	1	2.2	100		[5]
8	NiCo ₅	90	1	10.0	83		[6]
9	Co/mCN-900	120	1	12.3	0.2	99.8	[7]
10	Pd ₁ Ag ₁ @MIL- 101	r.t.	0.2	20.2	3	90	[8]
11	AuPd-Fe ₃ O ₄	r.t.	0.1	93.3		93	[9]
12	GA-Pd	r.t.	0.1	-		88	[10]
13	Fe ₂ O ₃ /NGr@C	120	5	1.48		89	[11]
					100 ,		
14	CoOx@NCNTs	110	3	8.3	Hydrogenation of nitrobenzene		[12]
15	Co-SiCN	110	5	1.4	82	_	[13]
					86.	6,	
16	Ni/C	140	1	6.3	hydrogen	ation of o-	[14]
	N: W. Change				chloronitrobenzenes 100 ,		
17	INI-W ₂ C/mpg- CNx	220	2.5	28.1	Hydrogenation of nitrobenzene		[15]
18	Co/NC-600	170	4	-	99.4		[16]

 Table S2. Comparison of the catalytic performance of PtCo/SiO2@CN and Co/SiO2@CN with other catalytic system in references

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