

Supporting information for

Photothermal catalytic CO₂ oxidation dehydrogenation of propane to propylene
over BiOX (X = Cl, Br, I) nanocatalysts

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Table S1 Photothermal catalytic CO₂-ODHP over different photocatalysts

Entry	Catalyst	Conv. (%)	Activity ($\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$)					C ₃ H ₆ Sele. ^c (%)
			CH ₄	C ₂ ^a	C ₃ H ₆	C ₄ ^b	CO	
1	None	<0.1	0.4	0.4	5.6	2.4	0.2	60.9
2	TiO ₂ ^d	0.1	0.4	0.7	26.4	14.6	202.4	56.8
3	BN	0.1	0.3	0.7	20.8	12.8	19.8	54.2
4	g-C ₃ N ₄	0.1	0.7	0.3	9.4	13.9	32.2	33.1
5	Co ₇ Cu ₁ Mn ₁ ^e	0.1	4.6	2.2	22.2	11.5	16.6	54.6
6	GaN	0.1	3.6	0.2	9.6	13.3	18.3	33.4
7	Bi	0.2	0.6	2.8	80.8	8.0	24.9	86.4
8	Bi ₂ O ₃	0.1	0.5	0.2	47.9	9.3	52.6	79.0
9	BiOCl nanosheets	0.1	1.2	0.7	25.0	8.5	10.0	67.2
10	BiOBr nanosheets	0.1	1.6	1.6	38.9	8.8	13.9	74.5
11	BiOI nanosheets	0.7	3.6	11.6	282.6	20.4	525.2	89.0
12	BiOI nanoflowers	0.2	1.8	2.6	75.6	13.1	31.9	79.2
13	BiI ₃	<0.1	0.3	0.1	1.3	7.4	20.3	11.5
14	Bi ₂ S ₃	0.1	6.9	0.2	21.1	8.2	76.5	61.2
15	Bi ₂ Fe ₄ O ₉	<0.1	1.0	0.4	10.3	6.7	7.6	51.9
16	Bi ₂ WO ₆	<0.1	2.2	0.5	2.0	9.5	57.8	12.5
17	BiVO ₄	0.1	1.0	1.1	32.4	12.4	80.9	64.9
18	Bi ₂ MoO ₆	<0.1	1.2	0.2	2.9	13.8	24.1	13.4
19	Bi ₁₂ TiO ₂₀	0.2	0.8	3.9	65.2	11.3	45.2	78.5
20	BiOI nanosheets ^f	-	-	-	-	-	121.7	-
21	BiOI nanosheets ^g	-	-	-	-	-	11.4	-
22	KI	<0.1	0.6	0.4	6.3	1.3	0.3	74.1
23	KI+H ₂ O ^h	<0.1	0.9	-	5.9	-	0.8	95.1
24	I ₂	0.5	1.8	28.3	196.3	7.7	105.5	86.9

25	BiOI nanosheets (UV, 300 - 420 nm) ⁱ	0.4	2.1	8.9	167.4	10.2	361.7	89.2
26	BiOI nanosheets (420-1100 nm) ^j	0.2	0.9	14.8	69.2	9.5	130.1	75.2

^aMainly ethane and ethylene. ^bIncluding *n*-butane, isobutane, *n*-butene, and isobutylene. ^cThe selectivity of C₃H₆ in the total hydrocarbon products. ^dPrepared according to ref. 1. ^ePrepared according to ref. 2. Conditions: catalyst 30 mg, C₃H₈/CO₂ molar ratio 1/5, total volume 180 mL, 270 °C, 1 h, and full irradiation under 300 W Xe lamp, 621 mW·cm⁻². ^fThe reaction gases were 15 ml H₂, 15 ml CO₂, and 150 ml Ar. ^gThe reaction gases were 15 ml CO₂ and 165 ml Ar. ^h2 mL of water were added. ⁱThe light intensity was 412 mW·cm⁻². ^jThe light intensity was 264 mW·cm⁻².

2. GC spectra

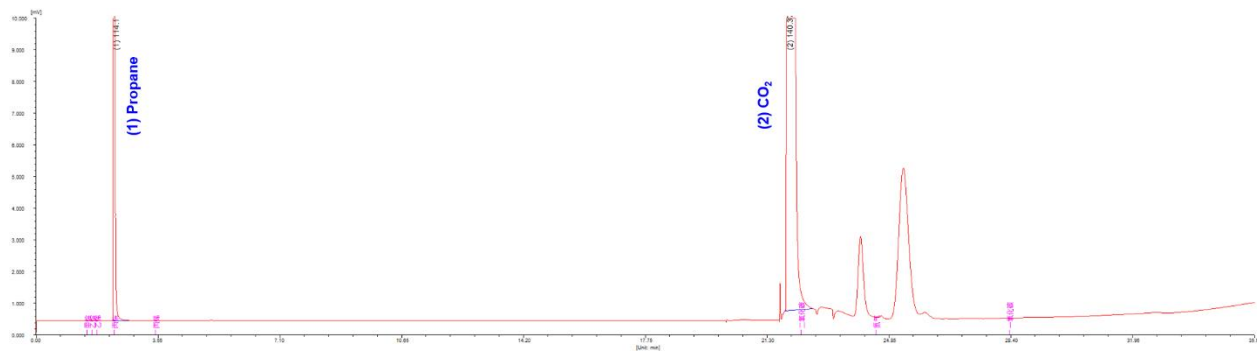


Fig. S1 The GC spectrum of the reaction gas.

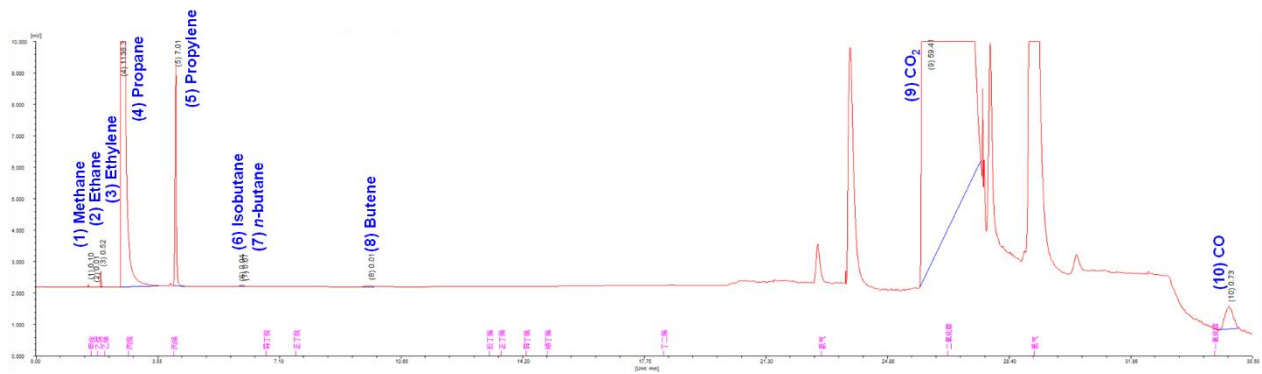


Fig. S2 The GC spectrum of photothermal catalytic CO₂-ODHP reaction over the BiOI nanosheets catalyst.

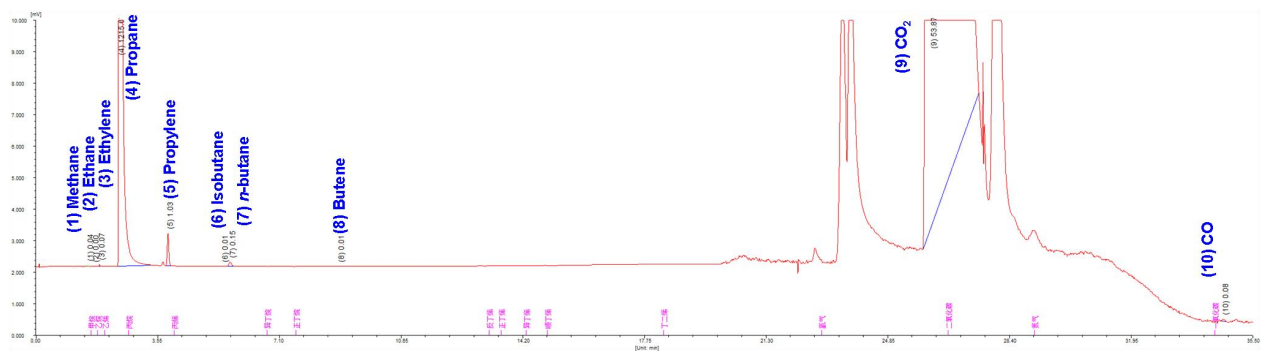


Fig. S3 The GC spectrum of the photothermal catalytic CO₂-ODHP reaction over the BiOI nanoflowers catalyst.

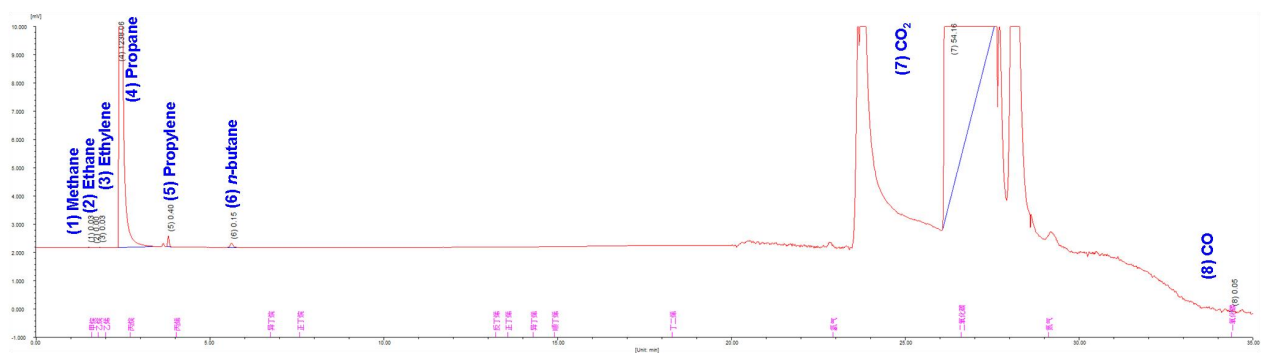


Fig. S4 The GC spectrum of photothermal catalytic CO₂-ODHP reaction over the BiOBr nanosheets catalyst.

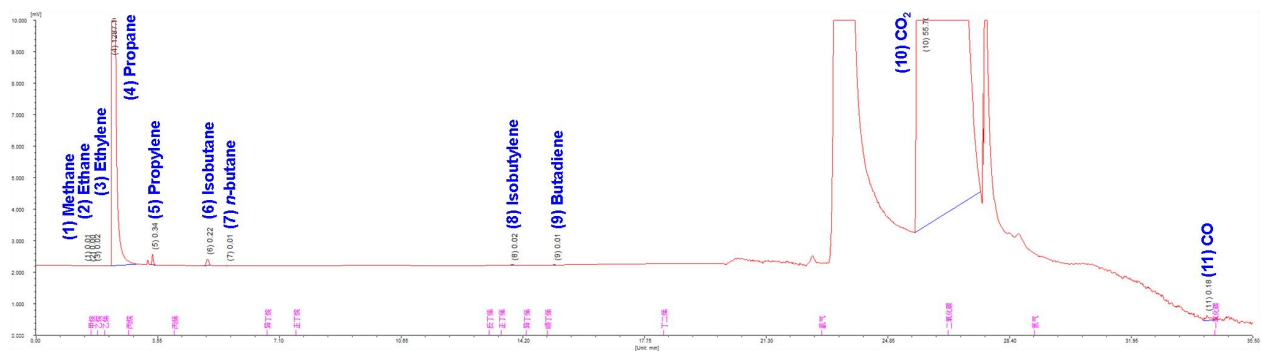


Fig. S5 The GC spectrum of photothermal catalytic CO₂-ODHP reaction over the BiOCl nanosheets catalyst.

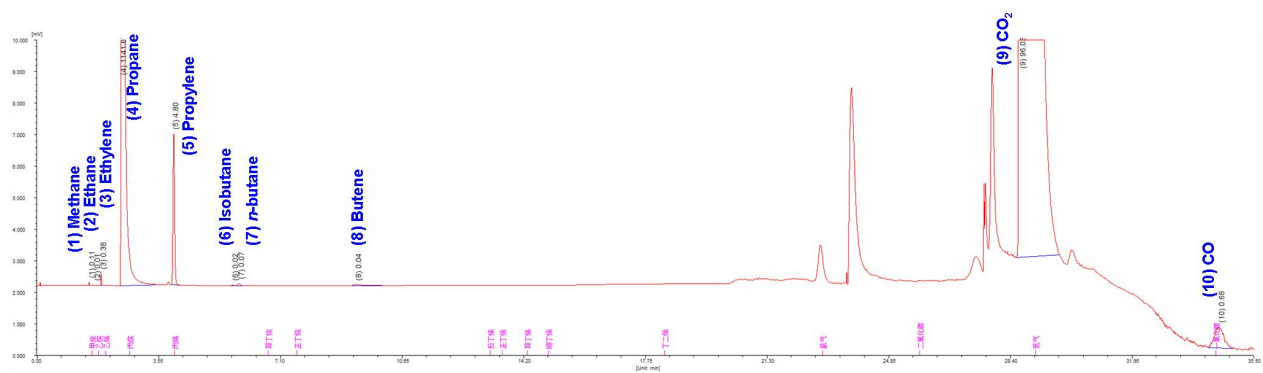


Fig. S6. The GC spectrum of photothermal catalytic DHP reaction over the BiOI nanosheets catalyst.

3. Pictures of the fresh BiOI nanosheets catalysts



Fig. S7. Pictures of the fresh BiOI nanosheets catalyst (a), the used BiOI nanosheets catalyst (b), and the catalyst after calcination at 400 °C(c), respectively.

4. XRD and XPS characterizations of the BiOI nanosheets catalyst after 13 runs

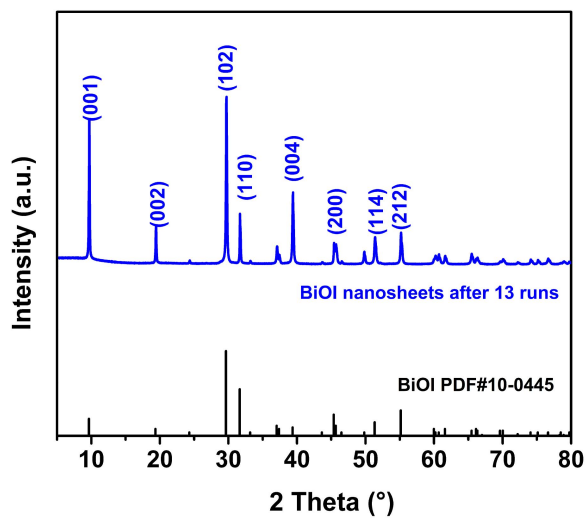


Fig. S8 XRD of the catalyst after 13 runs.

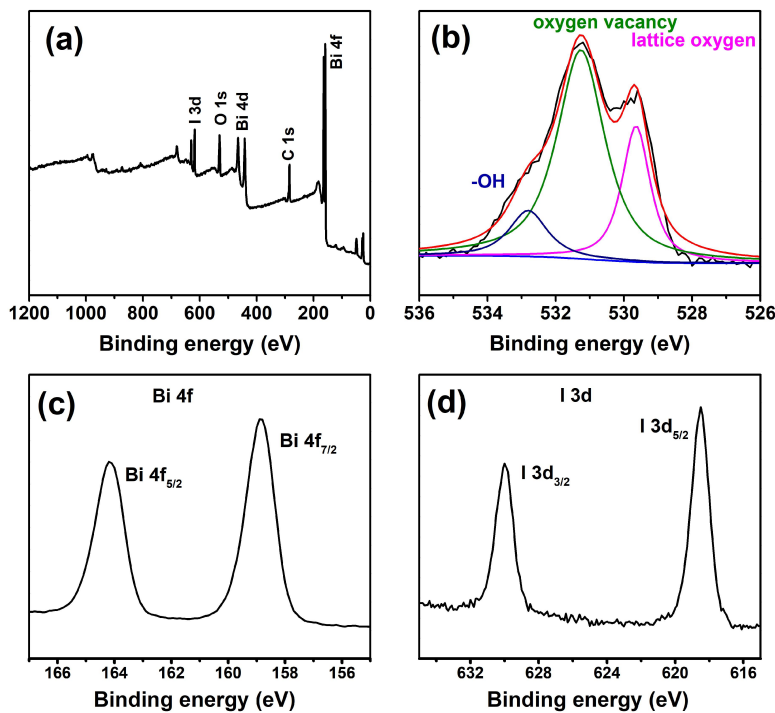


Fig. S9 XPS spectra of the catalyst after 13 runs.

References

- [1] J. Wang, K. Wang, Z.-H. He, S.-S. Li, R.-R. Zhang, P. Guo, W. Wang, Y. Yang, Z.-T. Liu, Solvent-induced synthesis of hierarchical TiO₂ nanoflowers with tunable morphology by monolayer self-assembly for probing the photocatalytic performance, *J. Nanostructure Chem.*, 2021, <https://doi.org/10.1007/s40097-021-00445-2>.
- [2] Z.-H. He, Z.-H. Li, Z.-Y. Wang, K. Wang, Y.-C. Sun, S.-W. Wang, W.-T. Wang, Y. Yang, Z.-T. Liu, Photothermal CO₂ hydrogenation to hydrocarbons over trimetallic Co-Cu-Mn catalysts, *Green Chem.*, 2021, 23, 5775-5785.