Supporting Information

Preparation of York-Shell Urchin-Like Porous Co₃O₄/NiO@C Microspheres with Excellent Lithium Storage Performance

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Fig.S1 SEM images of (a) YSUCNO@C, and (b) CNO microspheres



Fig.S2 (a) Nitrogen adsorption/desorption curves of YSUCNO@C, YSUCNO, and CNO microspheres, (b) pore size distribution of YSUCNO@C microspheres



Fig.S3 HRTEM image of YSUCNO@C microspheres



Fig.S4 Raman curves of YSUCNO, YSUCNO@C, CNO microspheres



Fig.S5 Thermogravimetry curve of YSUCNO@C microspheres



Fig.S6 (a) Low magnification, (b) high magnification SEM images of YSUCNO@C microspheres after 1000 cycles at 5 C



microspheres



microspheres after 100 cycles

Table	S 1	The	rate	capability	comparison	of the	reported	cobalt	nickel
oxide	mate	erials	and	YSUCNO	@C microspl	neres			

Matariala	Specific capacity	Current density	Reference	
Materials	(mA h g⁻¹)	(mA g ⁻¹)		
Co ₃ O ₄ /NiO/C	421	4000	30	
Co ₃ O ₄ /NiO/NC	493	5000	37	
CoO/NiO/CoNi	267	2000	38	
YSUCNO@C	656	8040	This work	

Materials	Specific capacity	Current density	Reference	
Materials	(mA h g ⁻¹)	(mA g ⁻¹)		
TiO ₂ @C@MnO ₂	186	6700	40 <u>1</u>	
3DG/Fe ₂ O ₃ aerogel	534.2	5000	41 <u>2</u>	
CSHCo ₃ O ₄ @C microspheres	332.6	8900	4 <u>23</u>	
CoO nanoflakes	494	5000	43 <u>4</u>	
MnO-doped Fe₃O₄@C	430	800	44 <u>5</u>	
CoO@N-C nanocubes	309	1000	45 <u>6</u>	
YSUCNO@C	656	8040	This work	

Table	S2	The	TMO-based	cell	performance	comparison	between	
YSUCNO@C microspheres and other reported work at high rate. ⁴⁰¹⁻⁴⁵⁶								

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