





**Figure S1.** Schematics of the procedures for sample preparation for: A) Cu EXC fraction; B) Bulk fraction.

## Calculation of isotope ratios

For mass bias correction, a combination of internal normalization according to the exponential model described by Maréchal *et al.*<sup>41</sup> using the  $^{62}\text{Ni}/^{60}\text{Ni}$  ratio, following the equation (1S) and (2S) and the standard-sample-standard bracketing sequence with the NIST SRM 3114 were used.

$$R_{Cu} = r_{Cu} * \frac{1}{\left(\frac{65}{63}\right)^f} \quad (1S)$$

Where  $R_{Cu}$  is the real  $^{65}\text{Cu}/^{63}\text{Cu}$  isotope ratio,  $r_{Cu}$  is the measured isotope ratio and  $f$  is the exponential fractionation coefficient calculated following the equation (2S)

$$f = \frac{\ln \frac{r_{Ni}}{R_{Ni}}}{\ln \frac{62}{60}} \quad (2S)$$

Where  $R_{Ni}$  is the real of  $^{62}\text{Ni}/^{60}\text{Ni}$  value and  $r_{Ni}$  is the measured value (calculated by LRS).

Finally, the results were expressed as delta values ( $\delta$ ), *i.e.*, as a relative difference (in per mil) versus a reference (NIST SRM 3114) following equation (3S):

$$\delta^{65}\text{Cu} (\text{‰}) = \frac{(R_{sample} - R_{STD})}{R_{STD}} * 1000 \quad (3S)$$

where  $R_{sample}$  is the  $^{65}\text{Cu}/^{63}\text{Cu}$  isotope ratio determined for a given sample and  $R_{STD}$  is the average  $^{65}\text{Cu}/^{63}\text{Cu}$  isotope ratio determined for the standards measured before and after that particular sample.

**Table S1.** Protocol for Cu isolation with the anion exchange Cu specific resin (Triskem).

<b>Step</b>	<b>Volume (mL)</b>	<b>Medium</b>
Resin loading	0.5	Cu-specific
Cleaning	10	6 mol L <sup>-1</sup> HCl
Conditioning	10	5 mmol L <sup>-1</sup> HCl
Sample load	4	5 mmol L <sup>-1</sup> HCl
Matrix elution	40	5 mmol L <sup>-1</sup> HCl
Cu elution	10	6 mol L <sup>-1</sup> HCl