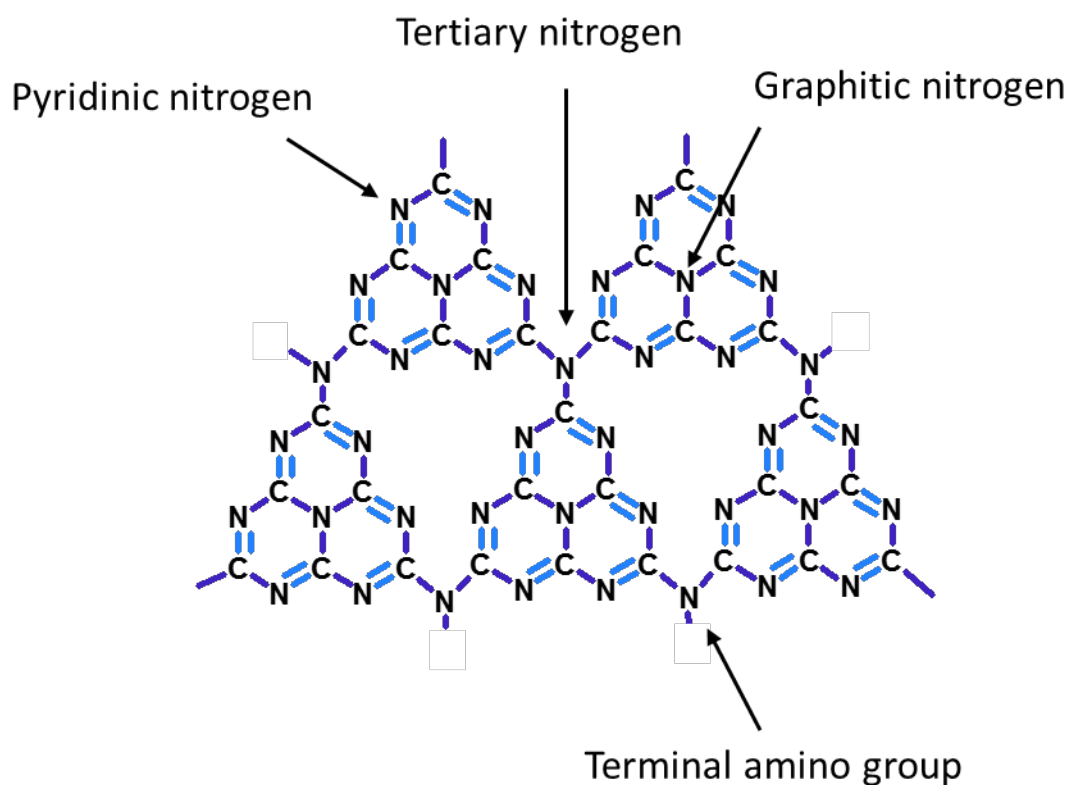


Supporting Information

g-C₃N₄/Dendritic Fibrous Nanosilica Doped with Potassium for Photocatalytic CO₂ Reduction

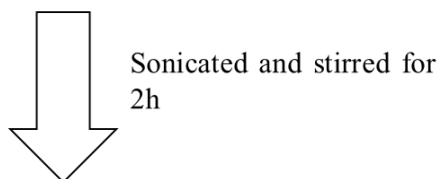
Sushma A. Rawool, Yusuf Kar, and Vivek Polshettiwar*,

Department of Chemical Sciences, Tata Institute of Fundamental Research (TIFR), Mumbai, India. Email: vivekpol@tifr.res.in, Phone: (+91) 8452886556

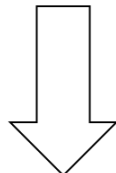


Scheme 1. Types of nitrogen present in the framework of g-C₃N₄

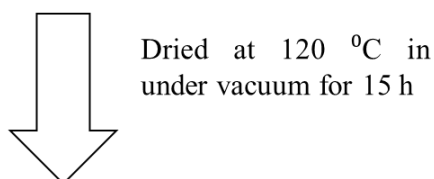
KOH (x mg, x=8.33, 83.33, 208.33) + Melamine (250 mg) + Water (10 mL)



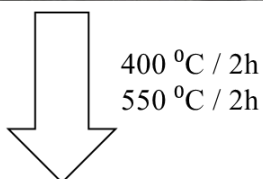
Evaporated to dryness at 80 °C in oven



DFNS (500 mg) + K-Melamine (250 mg) + Ethanol (1.25 mL)



Sample placed in vacuum sealed tube



K-g-C₃N₄ coated DFNS (K-CN/DFNS)

Fig. S1 Synthesis procedure for the production of K-CN/DFNS using a vacuum-sealed quartz tube

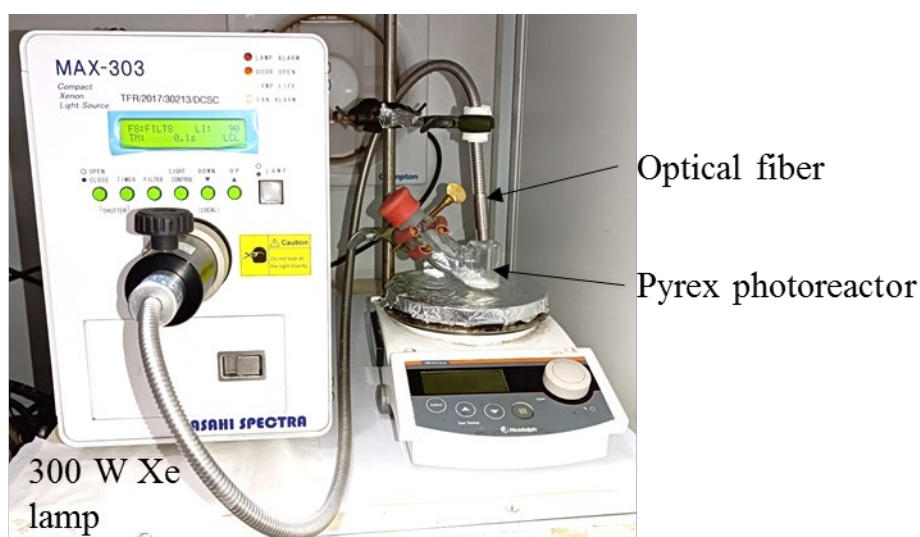


Fig. S2 Photograph of the experimental setup for the photocatalytic CO₂ conversion.

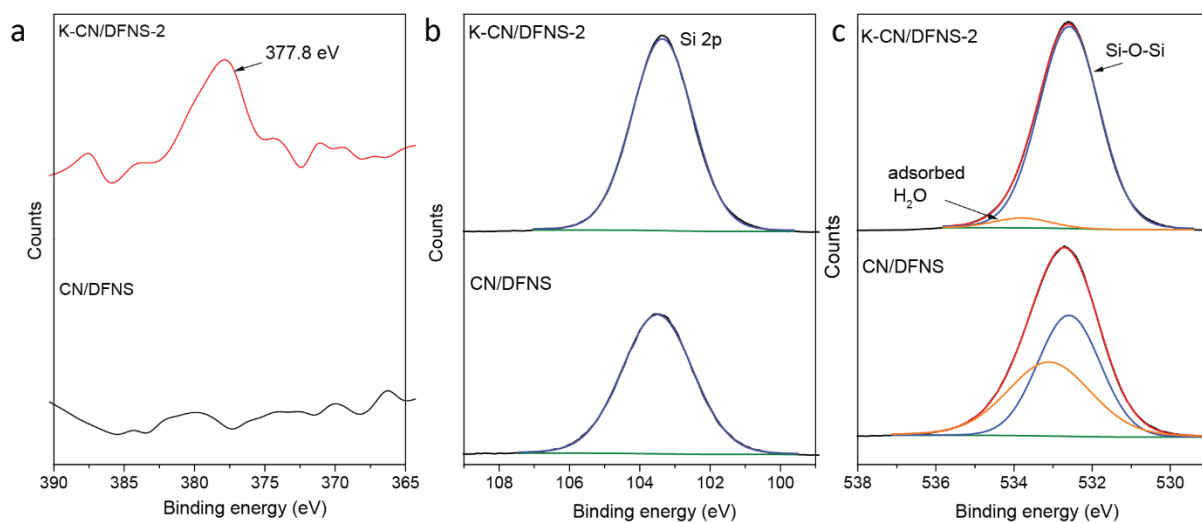


Fig. S3 High-resolution XPS spectra of a. K 1s, b. Si 2p and c. O 1s of CN/DFNS and K-CN/DFNS-2

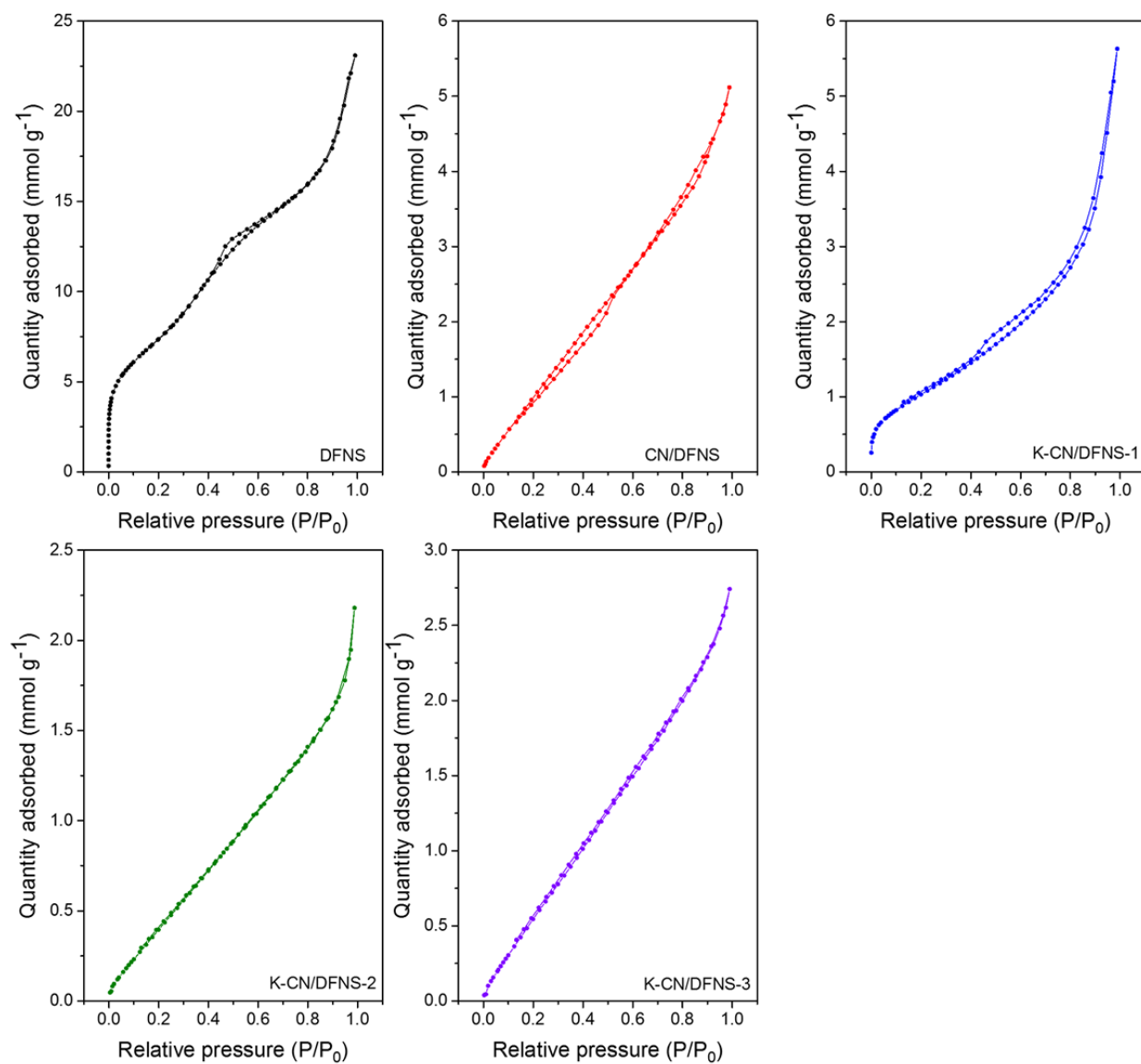


Fig. S4 N_2 sorption isotherms of DFNS, $g\text{-C}_3\text{N}_4$, CN/DFNS and K-CN/DFNS

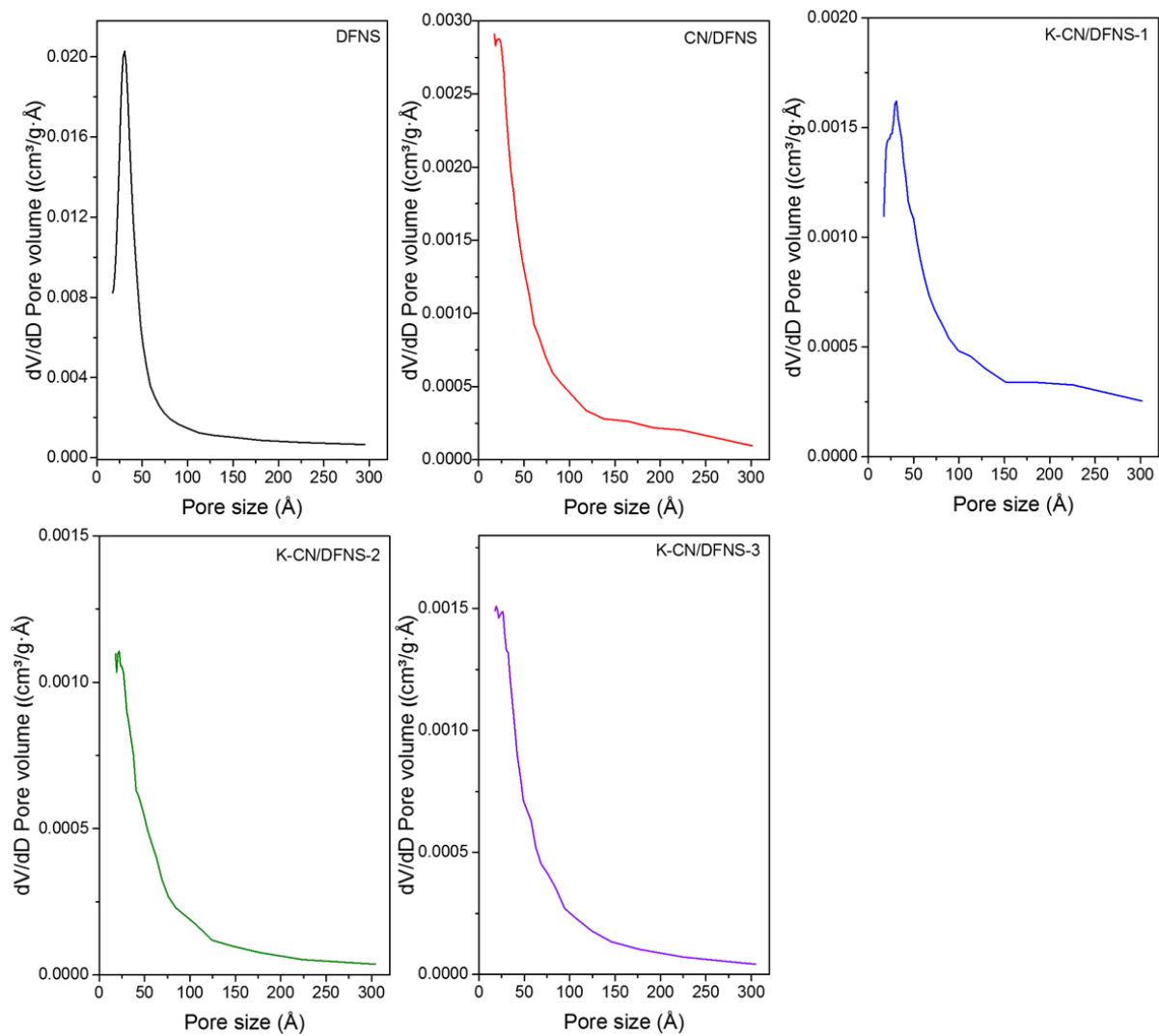


Fig. S5 BJH pore size distribution of DFNS, $g\text{-C}_3\text{N}_4$, CN/DFNS and K-CN/DFNS

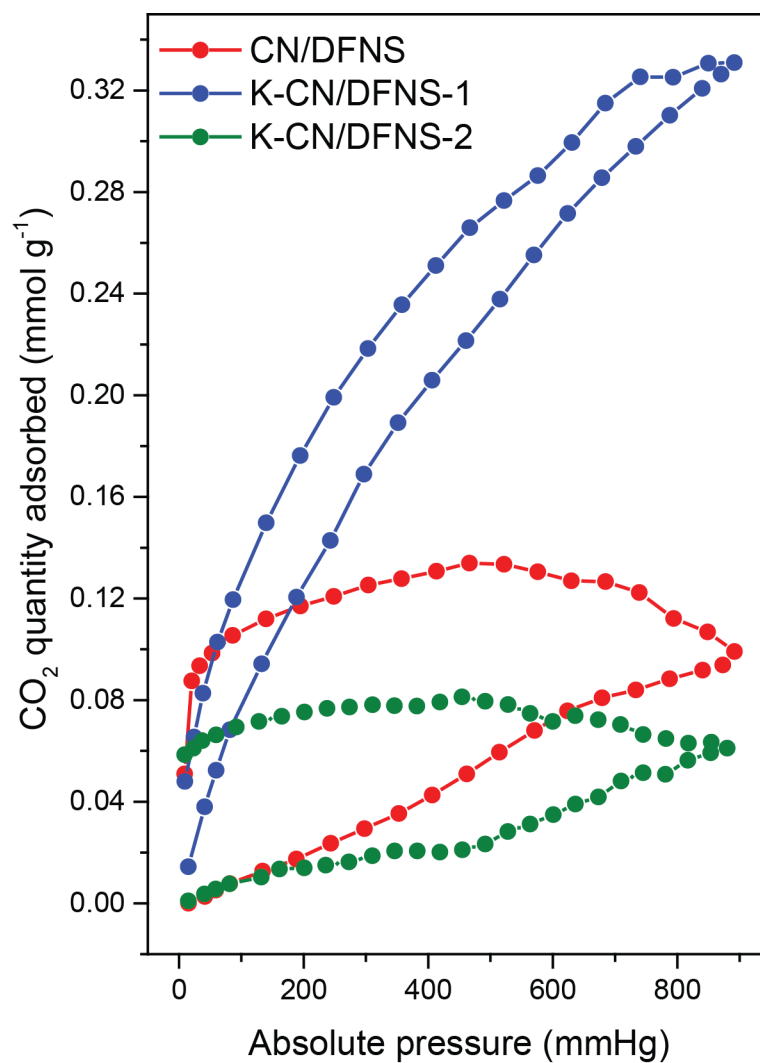


Fig. S6 CO₂ adsorption-desorption isotherm recorded at 298 K for CN/DFNS, K-CN/DFNS-1 and K-CN/DFNS-2.

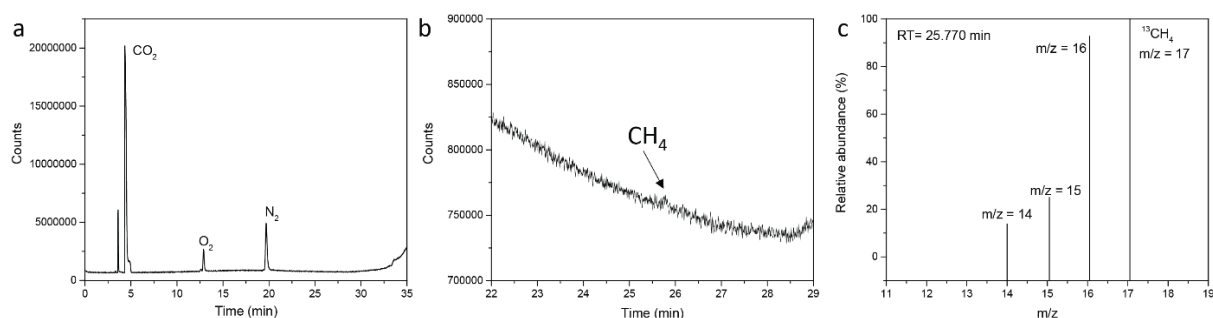


Fig. S7 a. Gas chromatogram recorded using mass spectrometer (MS) detector, b. enlarged view of chromatogram at 25.77 min and c. mass spectrum at 25.77 min.

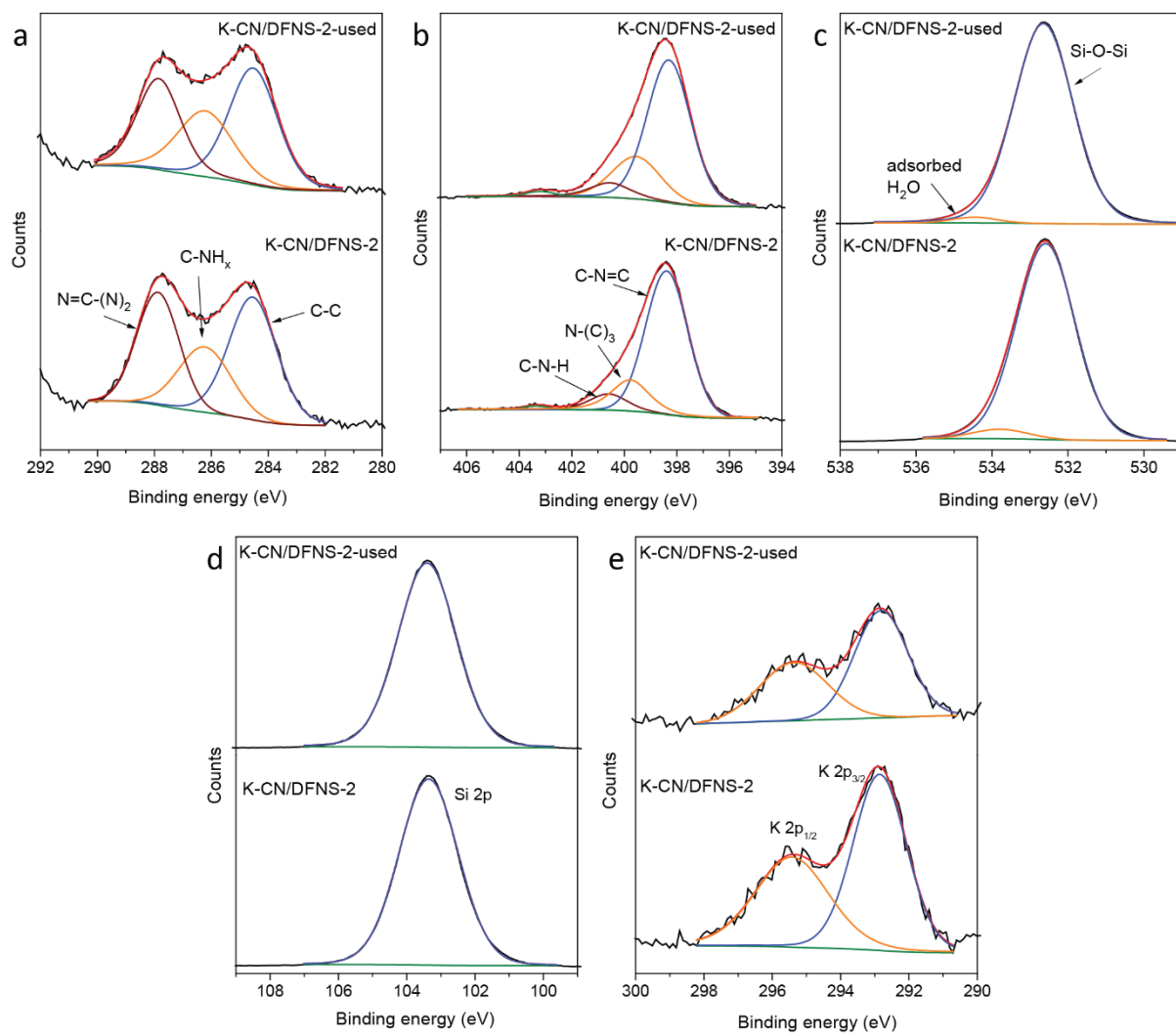


Fig. S8 High-resolution XPS spectra of a. C 1s, b. N 1s, c. O 1s, d. Si 2p, and e. K 2p of k-CN/DFNS and K-CN/DFNS-2 used

Table S1: Comparison of photocatalytic activity of K-CN/DFNS sample with literature

Sr. No.	Sample	Experimental Conditions	Light source	CO ₂ conversion products	Reference
1	K-CN/DFNS-2 (6 wt% potassium ion)	15 mg catalyst, moist CO ₂	300 W Xe lamp (385-740 nm, 318 mW cm ⁻²)	CH ₄ : 1.7 μmol g ⁻¹ in 4 h	This work
2	Potassium doped g-C ₃ N ₄ (1.85 atom% potassium)	Catalyst dispersed in 2 mL of water, CO ₂	Visible light (simulated by a CEAULIGHT CEL-HXF300)	CO: 8.7 μmol g ⁻¹ h ⁻¹	Wang et al [1]

			with 420 nm cut-off filter)		
3	K-incorporated amino-rich g-C ₃ N ₄	15 mg catalyst, water vapour, CO ₂	Xe lamp (1 sun simulated sunlight)	CO: ~ 9.6 μmol g ⁻¹ h ⁻¹ CH ₄ : ~ 8 μmol g ⁻¹ h ⁻¹	Sun et al [2]
4	Amine-functionalized g-C ₃ N ₄	100 mg catalyst, CO ₂ generated inside the reactor (0.084 g NaHCO ₃ + 0.3 mL of 2 M H ₂ SO ₄)	300 W Xe	CH ₄ : 0.34 CH ₃ OH: 0.28	Huang et al [3]
5	Sulfur doped g-C ₃ N ₄	100 mg catalyst, CO ₂ generated inside the photoreactor (NaHCO ₃ + HCl)	300-W simulated solar Xe arc lamp	CH ₃ OH: 1.12 μmol g ⁻¹ in 3h	Wang et al [4]
6	P doped and cyano group incorporated g-C ₃ N ₄	100 mg catalyst + 1 M NaHCO ₃ + CO ₂	300 W Xe lamp equipped with a 420 nm cut-off filter, 200 mW cm ⁻²	CO: 1.17 μmol g ⁻¹ in 3h	Liu et al [5]
6	Nonmetal (B, P, O and S) doped porous g-C ₃ N ₄	50 mg catalyst dispersed in 100 mL of water, CO ₂	UV lamp (100-280 nm)	CH ₄ : 55.1 nmol (ml of H ₂ O) ⁻¹ g ⁻¹ h ⁻¹	Arumugam et al [6]

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