Supplementary Information-Synthesis, Structure and Electrochemical Properties of a New Cation Ordered Layered Li-Ni-Mg-Mo Oxide

Bo Dong^{*,1,2}, Javier Castells-Gil^{1,2}, Pengcheng Zhu^{2,3}, Laura L. Driscoll^{1,2}, Emma Kendrick^{2,3}, Phoebe K. Allan^{1,2}, Peter R. Slater^{*,1,2}

- 1. School of Chemistry, University of Birmingham, Birmingham B15 2TT, U.K.
- 2. The Faraday Institution, Harwell Science and Innovation Campus, Didcot OX11 0RA, U.K.
- 3. School of Metallurgy and Materials, University of Birmingham, Birmingham B15 2TT, U.K.

* b.dong@bham.ac.uk; p.r.slater@bham.ac.uk



Figure S1 Charge/discharge curves of $Li_{1.2}Ni_{0.4}Ti_{0.4}O_2$ between at 5 mAg⁻¹ at

40 °C.



Figure S2. The initial 1st formation cycle of $Li_{1.2}Ni_{0.4}Mo_{0.2}Mg_{0.2}O_2$ at 5 mA/g at 40 °C.



Figure S3. Charge/discharge curves of $Li_{1.1}Ni_{0.6}Mo_{0.15}Mg_{0.15}O_2$ between 1.5

and 4.7 V at 10 mAg⁻¹ at room temperature.



Figure S4. Charge/discharge curves of Li_{1.2}Ni_{0.4}Mo_{0.2}Mg_{0.2}O₂ A {5 mAg⁻¹ (≈C/40) at 40 °C} and Li_{1.1}Ni_{0.6}Mo_{0.15}Mg_{0.15}O₂ {10 mAg⁻¹ (≈C/25) at room temperature} between 1.5 and 4.7 V.