

Support information for:

## **Peptide-Templated Biomineralization of Titanium Dioxide toward Improved Light Absorption and Photodegradation Activity**

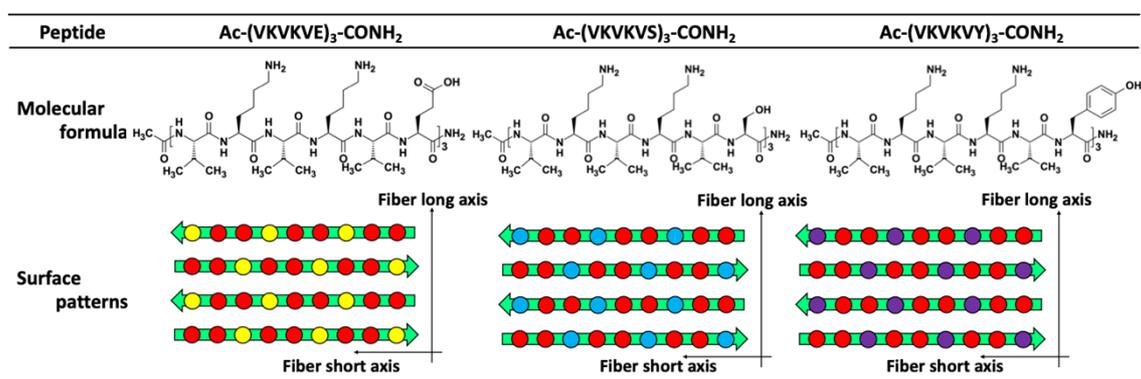
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**Figure S1.** Molecular formula and functional group patterns of the  $\beta$ -sheet peptide templates. Red, yellow, blue, and purple balls indicate the amino, carboxyl, hydroxyl, and tyrosyl groups, respectively.

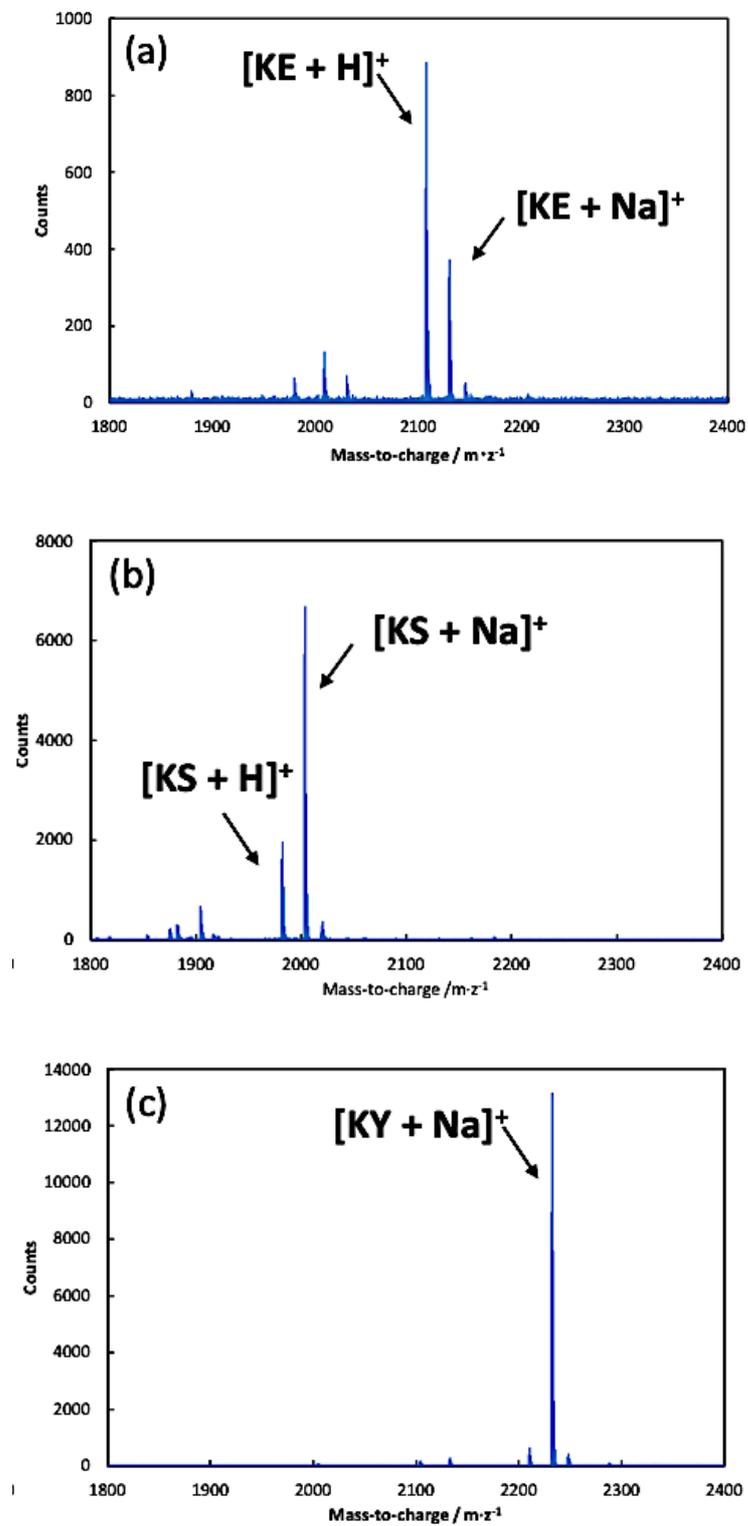
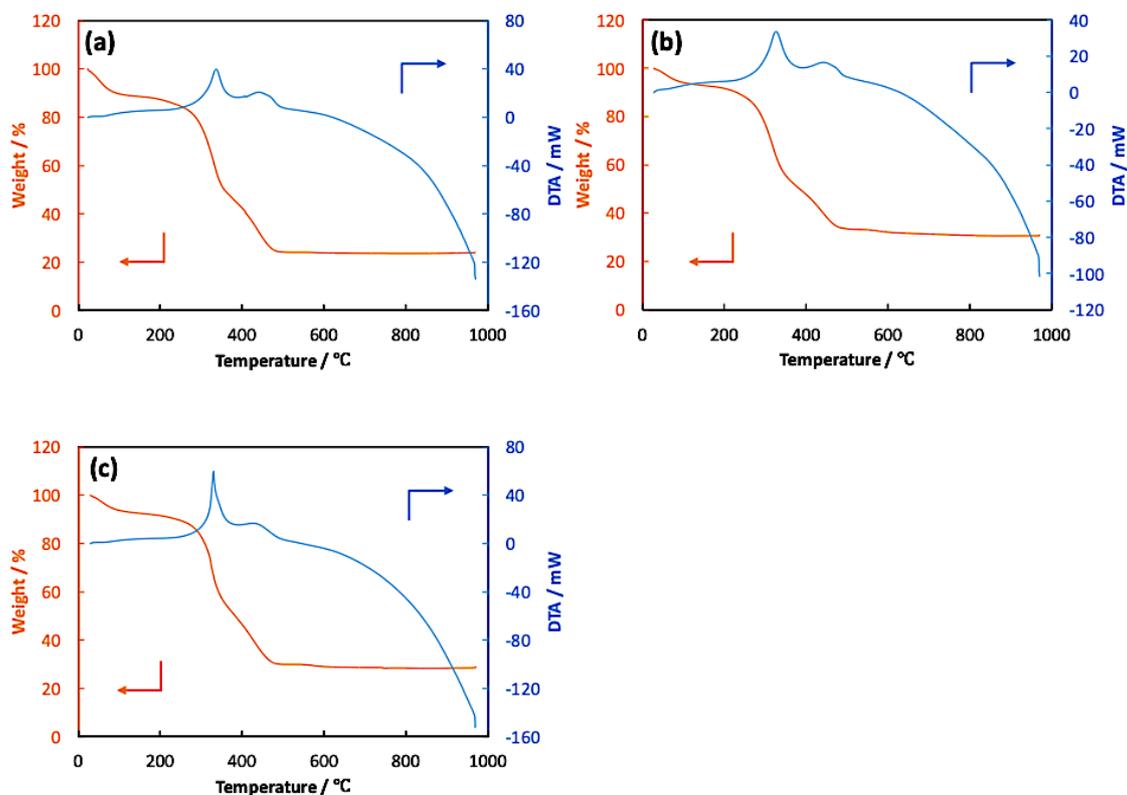


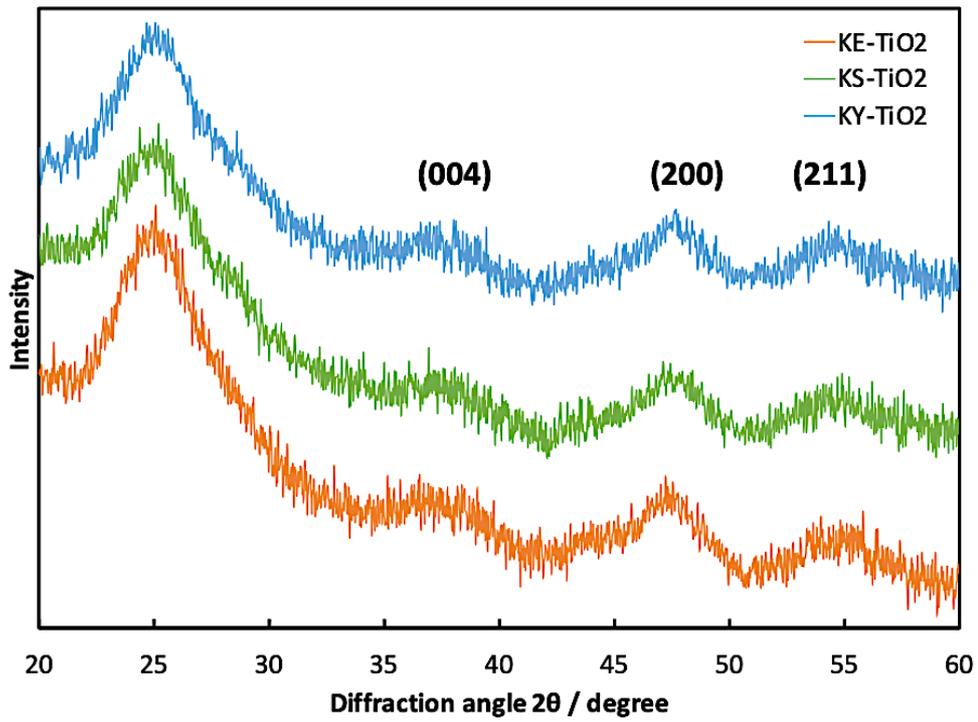
Figure S2. MALDI-TOF-MS spectra of (a) KE, (b) KS, and (c) KY peptides.



**Figure S3.** TG-DTA profiles of peptide-TiO<sub>2</sub> hybrids (a: KE-TiO<sub>2</sub>, b: KS-TiO<sub>2</sub>, and c: KY-TiO<sub>2</sub>).

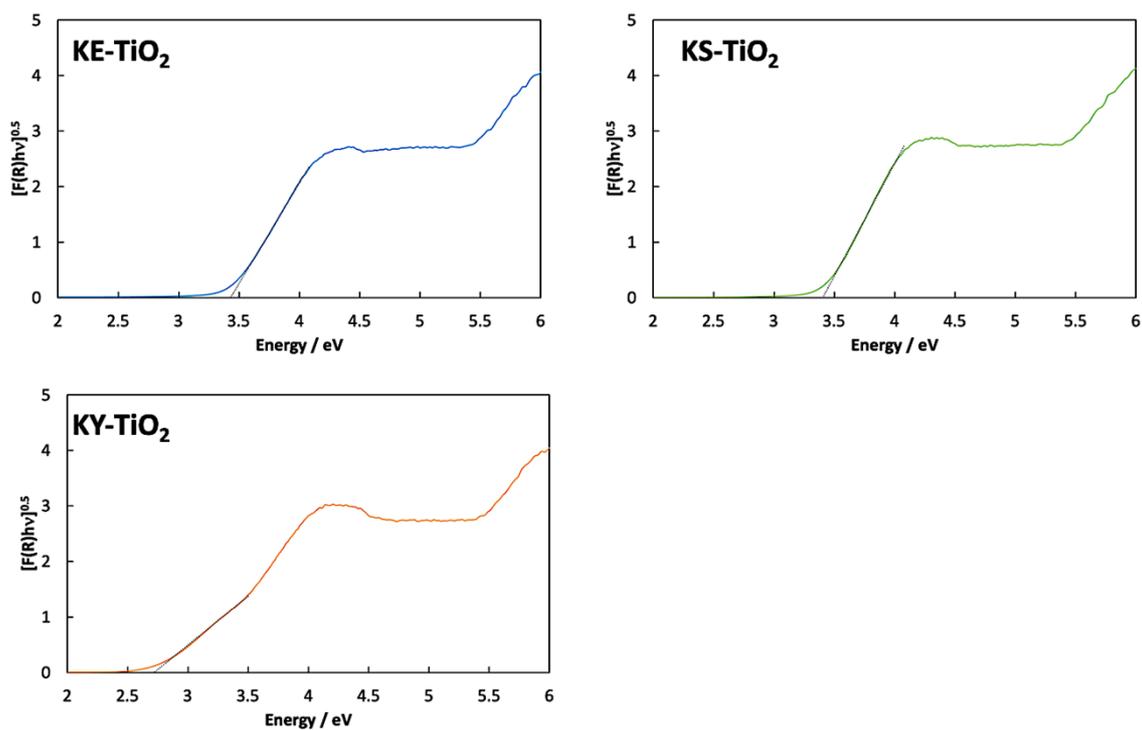
**Table S1.** Compositions, precipitation amounts, and catalytic activities of peptide-TiO<sub>2</sub> hybrids. The composition and amount of precipitation were estimated using TGA.

Sample	Composition / wt%			Precipitation amount of TiO <sub>2</sub> (mg/ peptide solution mL)	Catalytic activity of peptide (TiO <sub>2</sub> mg/peptide mmol)
	H <sub>2</sub> O	Peptide	TiO <sub>2</sub>		
KE-TiO <sub>2</sub>	12	43	45	0.84 ± 0.09	432 ± 45
KS-TiO <sub>2</sub>	9	43	48	0.67 ± 0.12	352 ± 63
KY-TiO <sub>2</sub>	8	44	48	0.59 ± 0.08	307 ± 43

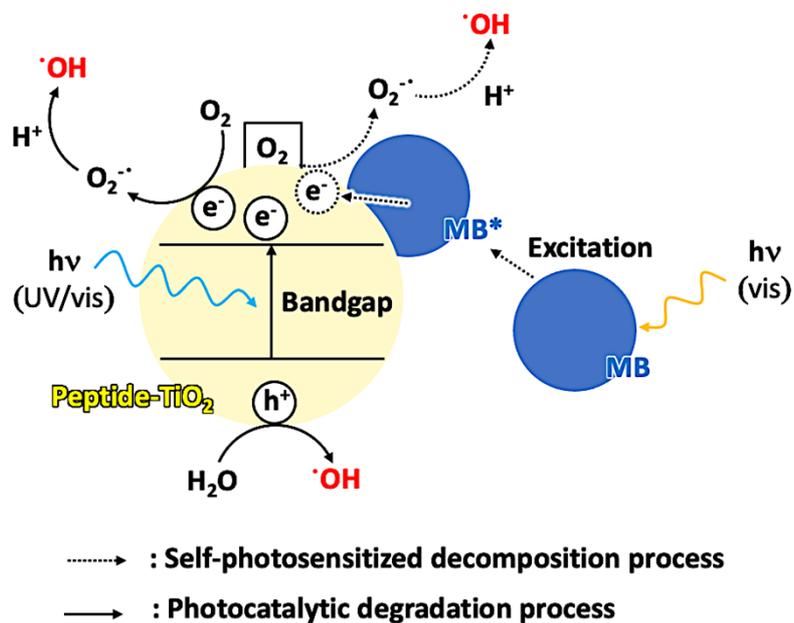


**Figure S4.** XRD profiles of mineralized peptide-TiO<sub>2</sub> hybrids.

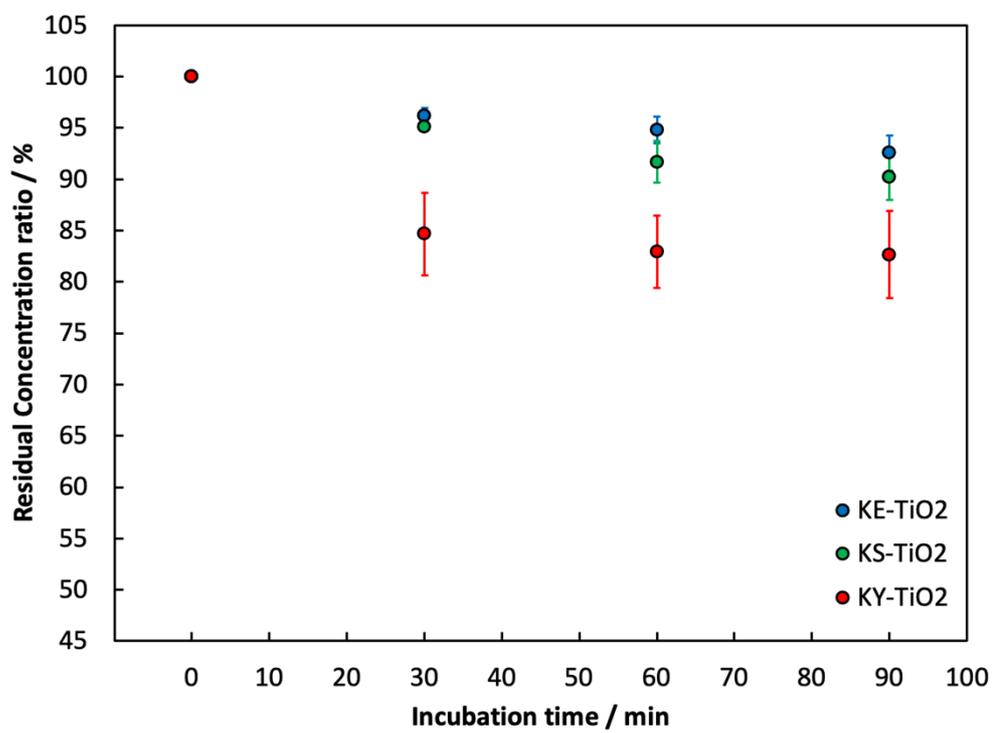
Samples	Bandgap energy / eV
KE-TiO <sub>2</sub>	3.42
KS-TiO <sub>2</sub>	3.40
KY-TiO <sub>2</sub>	2.72



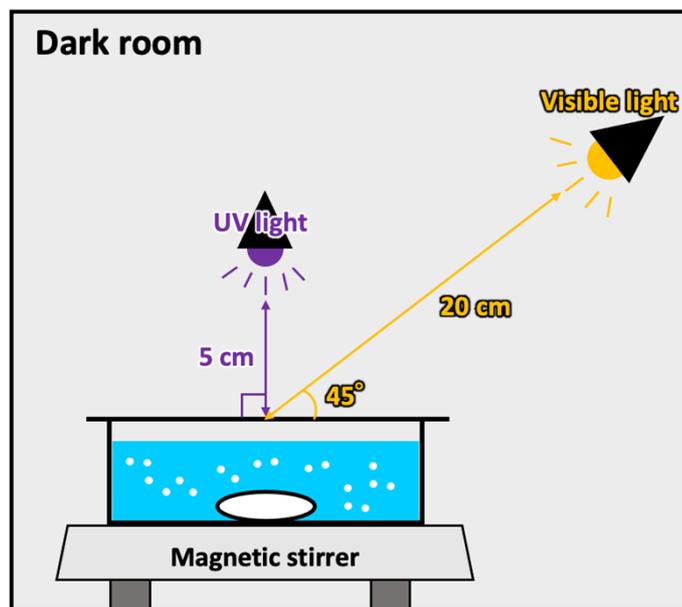
**Figure S5.** Bandgap energy analysis of the peptide-TiO<sub>2</sub> hybrid materials.



**Figure S6.** Photodegradation mechanism of MB with peptide-TiO<sub>2</sub> hybrids under UV/vis lights.



**Figure S7.** MB absorption profiles on the peptide-TiO<sub>2</sub> surface.



**Figure S8.** Schematic image of the photodegradation of methylene blue utilizing mineralized peptide-TiO<sub>2</sub> hybrids.