Electronic Supplementary Material (ESI) for Molecular Systems Design & Engineering. This journal is © The Royal Society of Chemistry 2022

Supporting Information

An experimental evidence for the key role of diphenylalanine in fibril formation

Santosh Kumar,^a Srayoshi Roy Chowdhury,^a Sahabaj Mondal,^a and Debasish Haldar^{a*} ^aDepartment of Chemical Sciences Indian Institute of Science Education and Research Kolkata Mohanpur 741246, West Bengal, India

E-mail: deba_h76@iiserkol.ac.in, deba_h76@yahoo.com.

Table of contents

2. ESI Table S1S33. ESI Figure S2S44. ESI Table S2S54. ESI Figure S3S66. ESI Figure S4S67. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-S1	1. ESI Figure S1	<i>S3</i>
3. ESI Figure S2S44. ESI Table S2S54. ESI Figure S3S66. ESI Figure S4S67. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-SI	2. ESI Table S1	S3
4. ESI Table S2S54. ESI Figure S3S66. ESI Figure S4S67. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-SI	3. ESI Figure S2	<i>S4</i>
4. ESI Figure S3S66. ESI Figure S4S67. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-SI	4. ESI Table S2	<i>S</i> 5
6. ESI Figure S4S67. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-S1	4. ESI Figure S3	<i>S6</i>
7. ESI Figure S5S78. Synthesis and characterization of peptides 1, 2 and 3S7-S1	6. ESI Figure S4	<i>S6</i>
8. Synthesis and characterization of peptides 1, 2 and 3 S7-SI	7. ESI Figure S5	<i>S</i> 7
	8. Synthesis and characterization of peptides 1, 2 and 3	<i>S7-S17</i>



Fig. S1: The ORTEP diagram of peptide 2 including the atom numbering scheme. Thermal ellipsoids are shown at the level of 50% probability.

ESI Table S1: Crystal data and structure refinement for peptide 2.

Identification code	Npt_xylene_a
Empirical formula	$C_{27}H_{29}N_3O_5$
Formula weight	493.55
Temperature/K	100.00(10)
Crystal system	orthorhombic
Space group	P21
a/Å	4.9730(3)
b/Å	26.2917(17)
c/Å	9.3394(6)
α/°	90
β/°	90.444(2)
$\gamma/^{\circ}$	90
Volume/Å ³	1221.08(13)
Z	2
$\rho_{calc}g/cm^3$	1.342
µ/mm ⁻¹	0.096
F(000)	524.0

$0.20 \times 0.20 \times 0.20$
MoKa ($\lambda = 0.71073$)
2.3 to 28.7
$\text{-}6 \leq h \leq 6, \text{-}35 \leq k \leq 35, \text{-}12 \leq l \leq 12$
35544
4381
1.063
0.44/-0.63
0.8(7)
0.0646
0.1683



Fig. S2: The ORTEP diagram of peptide 3 including the atom numbering scheme. Thermal ellipsoids are shown at the level of 50% probability.

Identification code	NTPOMe
Empirical formula	$C_{27}H_{29}N_3O_5$
Formula weight	475.53
Temperature/K	100.00(10)
Crystal system	Orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
a/Å	8.3717(2)
b/Å	15.2576(4)
c/Å	19.7373(4)
α/°	90
β/°	90
$\gamma/^{\circ}$	90
Volume/Å ³	2521.09(10)
Z	4
$\rho_{calc}g/cm^3$	1.253
μ/mm^{-1}	0.712
F(000)	1008
Crystal size/mm ³	$0.22 \times 0.24 \times 0.27$
Radiation	$CuK\alpha \ (\lambda = 1.54184)$
2Θ range for data collection/°	3.7 to 68.4
Index ranges	$-10 \le h \le 10, -18 \le k \le 18, -17 \le l \le 17$
Reflections collected	13781
Independent reflections	3741
Goodness-of-fit on F ²	1.037
Largest diff. peak/hole / e Å ⁻³	1.50/-0.26
Flack parameter	0.02(15)
R	0.0674
WR2	0.1897

ESI Table S2: Crystal data and structure refinement for Peptide **3**.



Fig. S3. WAXS spectra of peptides 1-3.). Peptides 2 and 3 polydisperse microspheres show sharp reflections in the 5–25° 2θ range whereas only a very broad feature was observed for peptide 1 fibers.



Fig. S4. POM images of peptides 1 showing fibers formation in fresh solution (a) and elongated fibers after 24 h incubation. Peptides 2 (b) and 3 (c) shows no change in polydisperse microspheres on 24 h incubation.



Fig. S5. Schematic presentation showing the mechanism for fibril formation of peptide 1.

Experimental

Synthesis of NPG-Phe-OMe :

N-phenylglycine-OH (1.51 g, 10 mmol) was dissolved in 50 mL dry DCM in an ice-cold water bath. H₂N-Phe- OMe (1.81 g, 11 mmol) was dissolved in 10 mL DCM. It was then added to the reaction mixture, followed by immediate addition of 2.26 g (11 mmol) dicyclohexylcarbodiimide (DCC) and 1.48 g (11 mmol) of HOBt. The reaction mixture was allowed to come to room temperature and stir for 48 hrs. After that, DCM was evaporated, and the residue was dissolved in ethyl acetate (60 mL), and dicyclohexylurea (DCU) was filtered off. The organic layer was washed with 2 (M) HCl (3 ×50 mL), brine (2 × 50 mL), 1(M) sodium carbonate (3 × 50 mL) and brine (2 × 50 mL) and dried over anhydrous sodium sulfate.The products were purified by column chromatography using silica (60-120-mesh size) gel as stationary phase and n-hexane-ethyl acetate mixture as eluent.Yield: 2.05 g (6.5 mmol, 65 %).

¹H-NMR (400 MHz, CDCl₃, δ ppm):7.23-7.14 [5H aromatic & 1H NH proton], 7.16-7.15[b,1H NH proton] 6.93-6.91[2H aromatic], 6.83[1H aromatic], 6.57-6.53[2H aromatic], 4.95-4.93 [m, C^β1H], 3.77--3.75 [b, 2H], 3.69 [s, 3H, OCH₃], 3.06-3.05 [t, Cα 2H]. ¹³C-NMR (100 MHz, CDCl₃, δ ppm):172.03, 170.68, 147.81, 129.74, 129.48, 128.91, 127.40, 119.50, 113.61, 52.87, 52.76, 48.95, 38.23. Mass spectra, found m/z: 335.1495 [M+Na]+, calculated for C₁₈H₂₀N₂O₃Na 335.1407.



Fig. S3: ¹H NMR (400 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-OMe.



Fig. S4:¹³C NMR (100 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-OMe.



Fig. S5: Mass Spectra of NPG-Phe-OMe.

Synthesis of NPG-Phe-Phe-OMe 1 :

N-phenylglycine-Phe-OH (1.49 g, 5 mmol) was dissolved in 50 mL dry DCM in an icecold water bath. H₂N-Phe- OMe (1.15 g, 7 mmol) was dissolved in 10 mL DCM. It was then added to the reaction mixture, followed by immediate addition of 1.64 g (8 mmol) dicyclohexylcarbodiimide (DCC) and 1.08 g (8 mmol) of HOBt. The reaction mixture was allowed to come to room temperature and stir for 48 hrs. After that, DCM was evaporated, and the residue was dissolved in ethyl acetate (60 mL), and dicyclohexylurea (DCU) was filtered off. The organic layer was washed with 2 (M) HCl (3 ×50 mL), brine (2 × 50 mL), 1(M) sodium carbonate (3 × 50 mL) and brine (2 × 50 mL) and dried over anhydrous sodium sulphate.The products were purified by column chromatography using silica (100-200-mesh size) gel as stationary phase and n-hexane-ethyl acetate mixture as eluent. Yield: 1.6 g (3.5 mmol, 70 %).

¹H-NMR (400 MHz, CDCl₃, δ ppm): 7.25-7.13[m, 9H, aromatic proton], 7.13-7.06[1H, aromatic proton], 7.05-7.00[m, 2H, aromatic proton], 6.99-6.93[m, 2H, aromatic proton], 6.84-6.78[b, 1H, aromatic proton], 6.54-6.48[2H, b, NH] 6.46-6.40[1H, NH], 4.78-4.74[m, 1H, methine C^{\alpha} Phe], 4.73-4.67[m, 1H, Methine C^{\alpha} Phe], 3.65-3.69[5H, methylene protons of NPG & 3H-OMe], 3.08-3.00[4H, methylene protons of Phe]. ¹³C-

NMR (100 MHz, CDCl₃, δ ppm): 171.64, 171.56, 177.47, 147.05, 136.2, 135.96, 129.78, 129.54, 129.48, 128.85, 128.74, 127.23, 127.17, 119.34, 113.05, 53.96, 53.43, 52.46, 48.75, 38.05, 37.76. FT-IR (cm⁻¹): 3374, 2955, 1646, 1604. Mass spectra, found m/z: 482.8907 [M+Na]⁺, calculated for C₂₇H₂₉N₃O₄ 459.5369.



Fig. S6: ¹H NMR (400 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-Phe-OMe 1.



Fig. S7:¹³C NMR (100 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-Phe-OMe 1.



Fig. S8: Mass Spectra of NPG-Phe-Phe-OMe 1.

Synthesis of NPG-Phe-Tyr-OMe 2 :

N-phenylglycine-Phe-OH (1.49 g, 5 mmol) was dissolved in 50 mL dry DCM in an ice-cold water bath. H₂N-Tyr- OMe 1.36 g, 7 mmol) was dissolved in 10 mL DCM. It was then added to the reaction mixture, followed by immediate addition of 1.64 g (8 mmol) dicyclohexylcarbodiimide (DCC) and 1.08 g (8 mmol) of HOBt. The reaction mixture was allowed to come to room temperature and stir for 48 hrs. After that, DCM was evaporated, and the residue was dissolved in ethyl acetate (60 mL), and dicyclohexylurea (DCU) was filtered off. The organic layer was washed with 2 (M) HCl (3 ×50 mL), brine (2 × 50 mL), 1(M) sodium carbonate (3 × 50 mL) and brine (2 × 50 mL) and dried over anhydrous sodium sulphate.The products were purified by column chromatography using silica (100-200-mesh size) gel as stationary phase and n-hexane-ethyl acetate mixture as eluent. Yield: 1.5 g (3.2 mmol, 64 %).

¹H-NMR (400 MHz, CDCl₃, δ ppm): 7.95-7.85[b, 1H, NH], 7.17-7.09[m, 5H, aromatic proton], 7.00-6.90[m, 3H, aromatic proton], 6.85-6.74[m, 3H, aromatic proton], 6.70-6.64[m, 3H, aromatic proton], 6.48-6.41[b, 2H, NH], 4.88-4.81[m, 1H, methine C^{α} Phe], 4.80-4.74[m, 1H, Methine C^{α}-Tyr], 4.25[b, 1H, Tyr-OH], 3.68[s, 3H, OMe], 3.57[s, 2H, methylene protons of NPG] 3.04-2.84[m, 4H, methylene protons of Phe & Tyr]. ¹³C-

NMR (100 MHz, CDCl₃, δ ppm): 171.60, 170.72, 155.45, 146.74, 135.71, 130.41, 129.39, 129.21, 129.21, 128.50, 126.86, 119.06, 115.48, 113.13, 53.63, 53.46, 52.38, 48.35, 37.92, 37.01. FT-IR (cm⁻¹): 3374, 1740, 1654, 1606. Mass spectra, found m/z: 498.7579 [M+Na]⁺, calculated for C₂₇H₂₉N₃O₄Na 498.2005.



Fig. S9: ¹H NMR (400 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-Tyr-OMe 2.



Fig. S10:¹³C NMR (100 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Phe-Tyr-OMe 2.



Fig. S11: Mass Spectra of NPG-Phe-Tyr-OMe 2.

Synthesis of NPG-Tyr-OMe :

N-phenylglycine-OH (1.51 g, 10 mmol) was dissolved in 50 mL dry DCM in an ice-cold water bath. H₂N-Tyr- OMe (2.14 g, 11 mmol) was dissolved in 10 mL DCM. It was then added to the reaction mixture, followed by immediate addition of 2.26 g (11 mmol) dicyclohexyl carbodiimide (DCC) and 1.48 g (11 mmol) of HOBt. The reaction mixture was allowed to come to room temperature and stir for 48 hrs. After that, DCM was evaporated, and the residue was dissolved in ethyl acetate (60 mL), and dicyclohexylurea (DCU) was filtered off. The organic layer was washed with 2 (M) HCl (3 ×50 mL), brine (2 × 50 mL), 1(M) sodium carbonate (3 × 50 mL) and brine (2 × 50 mL) and dried over anhydrous sodium sulfate.The products were purified by column chromatography using silica (60-120-mesh size) gel as stationary phase and n-hexane-ethyl acetate mixture as eluent. Yield: 2.23 g (6.7 mmol, 67 %).

¹H-NMR (400MHz, CDCl₃, δ ppm):7.22-7.12[m,3H (aromatic)], 6.84-6.80[t, 1H aromatic], 6.76-6.74[d,2H aromatic], 6.59-6.57[d,2H aromatic], 6.54-6.52[d, 2H aromatic], 6.4-6.3[b,1H NH proton] 4.93-4.91[m,1H], 4.24-4.15[b,1H],3.74-3.73[b,2H, ^{\arrow}CH₂ NPG], 3.71[s,3H,OMe], 2.99-2.97[t,2H, ^{\arrow}CH Tyr]. ¹³C-NMR (100 MHz, CDCl₃, δ ppm): 172.32, 171.12, 155.57, 147.14, 130.59, 129.75, 127.21, 119.53,115.90, 113.61, 53.01, 52.77, 48.88, 37.47. Mass spectra, found m/z: 351.1856 [M+Na]⁺, calculated for C₁₈H₂₀N₂O₄Na 351.1768.



Fig. S13:¹³C NMR (100 MHz, CDCl₃, δ in ppm, 298K) spectra NPG-Tyr-OMe.



Fig. S14: Mass Spectra of NPG-Tyr-OMe.

Synthesis of NPG-Tyr-Phe-OMe 3 :

N-phenylglycine-Tyr-OH (1.57 g, 5 mmol) was dissolved in 50 mL dry DCM in an ice-cold water bath. H₂N-Phe- OMe (1.15 g, 7 mmol) was dissolved in 10 mL DCM. It was then added to the reaction mixture, followed by immediate addition of 1.64 g (8 mmol) dicyclohexylcarbodiimide (DCC) and 1.08 g (8 mmol) of HOBt. The reaction mixture was allowed to come to room temperature and stir for 48 hrs. After that, DCM was evaporated, and the residue was dissolved in ethyl acetate (60 mL), and dicyclohexylurea (DCU) was filtered off. The organic layer was washed with 2 (M) HCl (3 ×50 mL), brine (2 × 50 mL), 1(M) sodium carbonate (3 × 50 mL) and brine (2 × 50 mL) and dried over anhydrous sodium sulphate.The products were purified by column chromatography using silica (100-200-mesh size) gel as stationary phase and n-hexane-ethyl acetate mixture as eluent. Yield: 1.6 g (3.4 mmol, 68 %).

¹H-NMR (400 MHz, CDCl₃, δ ppm): 7.25-7.13[m, 6H, aromatic proton], 7.02-6.95[m, 2H, aromatic proton], 6.86-6.75[m, 3H, aromatic proton], 6.62-6.53[m, 3H, aromatic proton], 6.52-6.46[b, 2H, NH proton], 4.80-4.72[m, 1H, methine C^{\alpha} Phe], 4.71-4.63[m, 1H, Methine C^{\alpha}-Tyr], 3.70-3.65[m, 5H, OMe proton and methylene protons of NPG], 3.10-2.92[m, 4H, methylene protons of Phe], 2.90-2.80[m, 2H, methylene protons of Tyr]. ¹³C-NMR (100 MHz, CDCl₃, δ ppm): 171.66, 171.32, 170.74, 153.36, 146.78, 135.76, 130.53, 129.64, 129.36, 128.72, 127.39, 127.25, 119.45, 115.73, 113.36, 54.10, 53.56, 52.58, 48.65, 37.97, 37.15. FT-IR (cm⁻¹): 3382, 1650, 1602. Mass spectra, found m/z: 498.1987 [M+Na]⁺, calculated for C₂₇H₂₉N₃O₄Na 498.2005.



Fig. S15: ¹H NMR (400 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Tyr-Phe-OMe 3.



Fig. S16:¹³C NMR (100 MHz, CDCl₃, δ in ppm, 298K) spectra of NPG-Tyr-Phe-OMe 3.



Fig. S17: Mass Spectra of NPG-Tyr-Phe-OMe 3.