

Supplementary Information

Green-solvent-processed formamidinium-based perovskite solar cells with uniform grain growth and strengthened interfacial contact via nanostructured tin oxide layer

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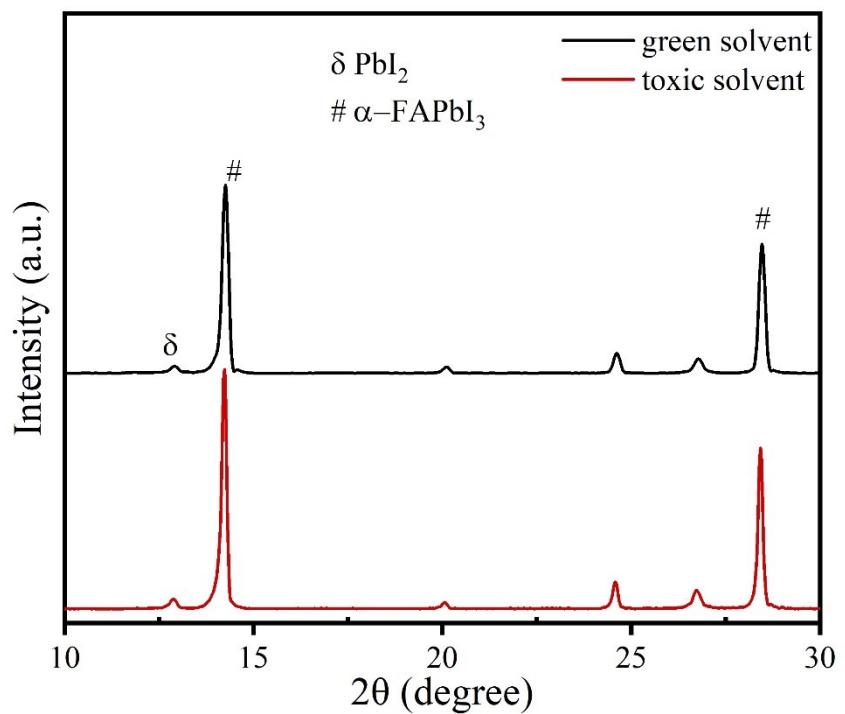


Fig. S1 XRD patterns from top-surfaces of perovskite films prepared via green solvent and toxic solvent, respectively.

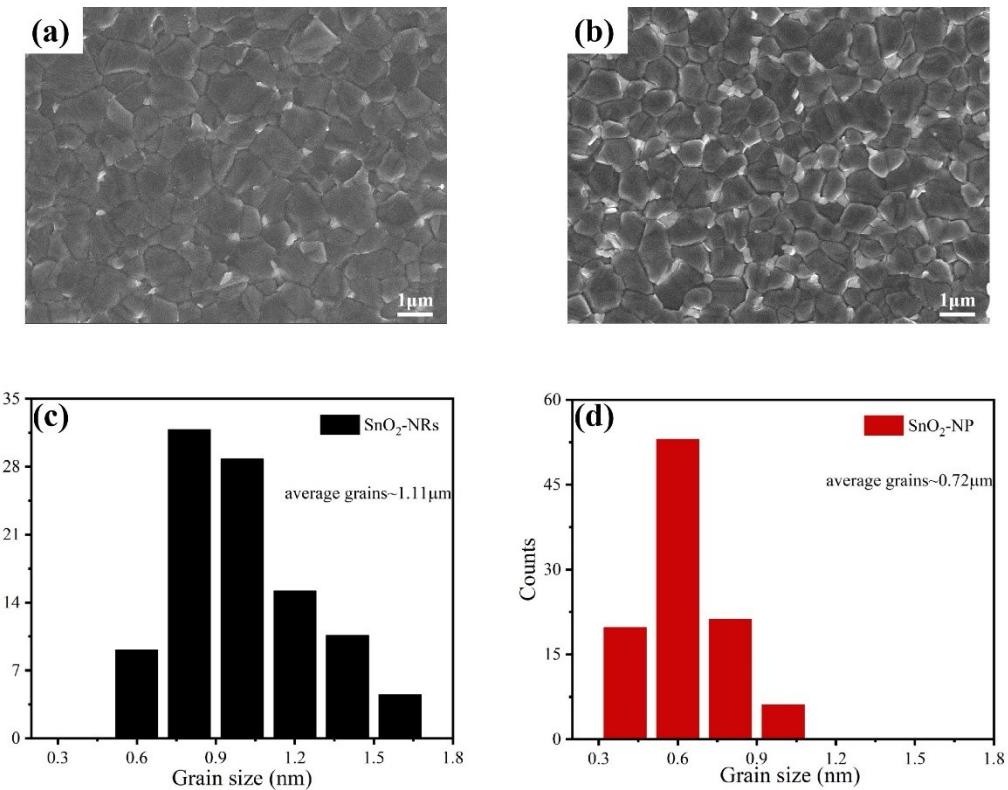


Fig. S2 **(a)** Top-view SEM image and **(c)** the size distribution of perovskite film based on SnO_2 -NRs, respectively. **(b)** Top-view SEM image and **(d)** the size distribution of perovskite film based on SnO_2 -NP, respectively.

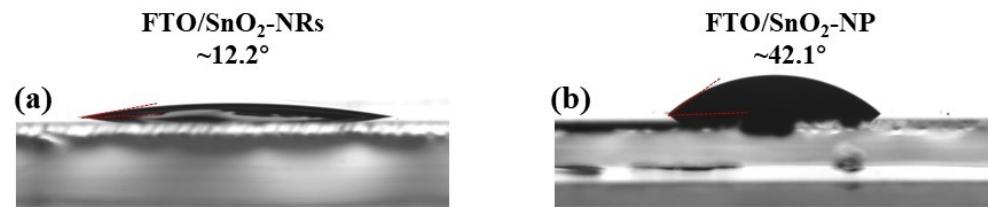


Fig. S3 The deionized water contact angles of (a) the SnO₂-NRs and (b) SnO₂-NP prepared on FTO substrates.

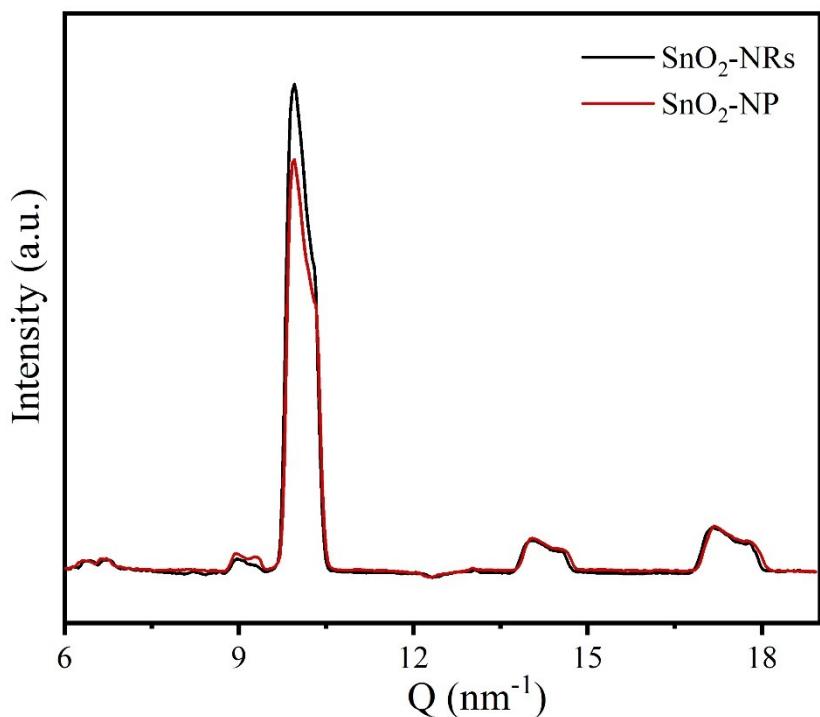


Fig. S4 Integrated intensity profiles along the q_z direction of GIWAXS pattern for SnO_2 -NRs and SnO_2 -NP substrates.

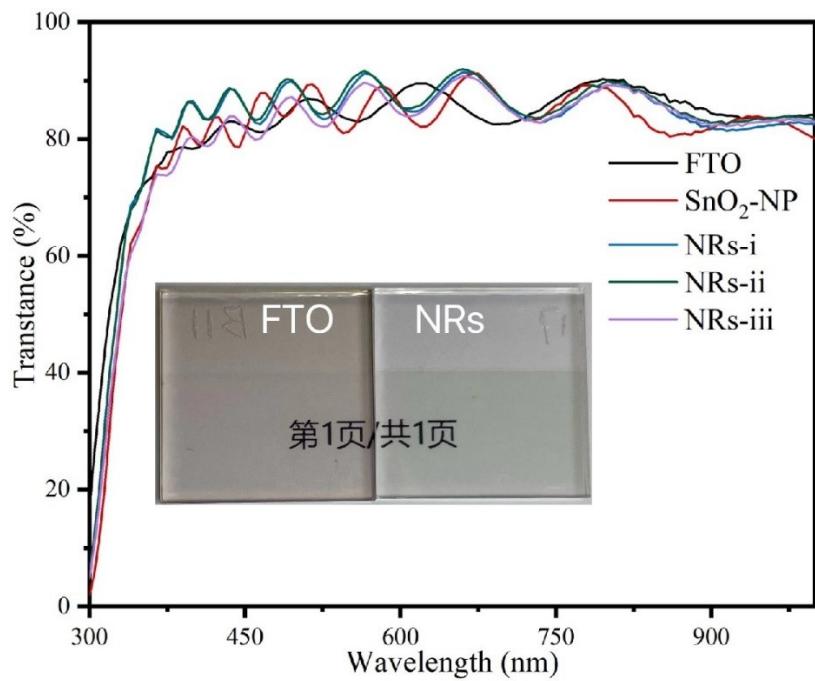


Fig. S5 UV-vis transmittance of bare FTO glass, SnO₂-NP, and SnO₂-NRs in different reaction temperature. The inset shows a typical optical image of the bare FTO glass and FTO/NRs-ii.

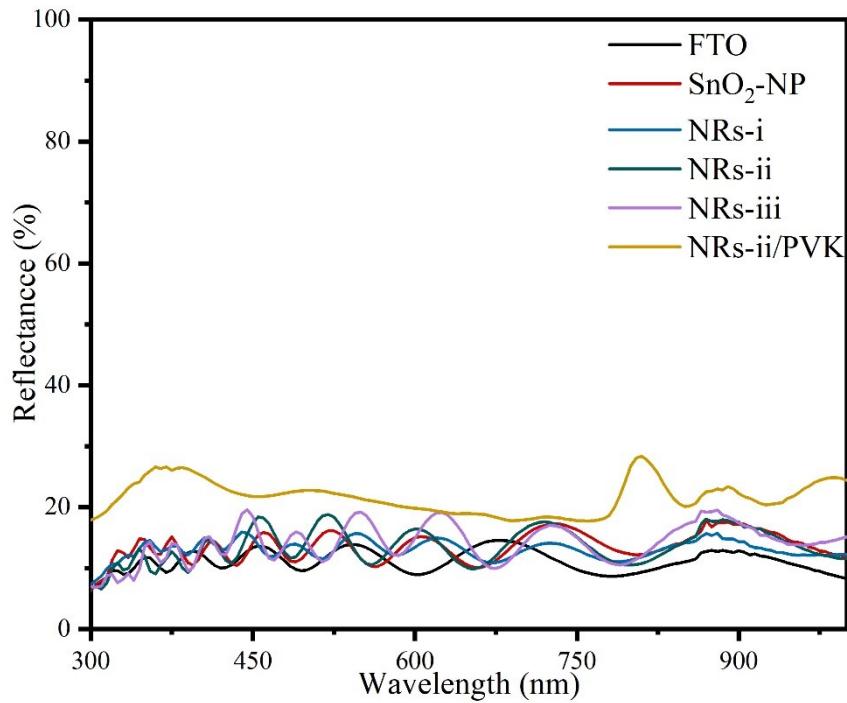


Fig. S6 Diffuse reflectance spectra of different films.

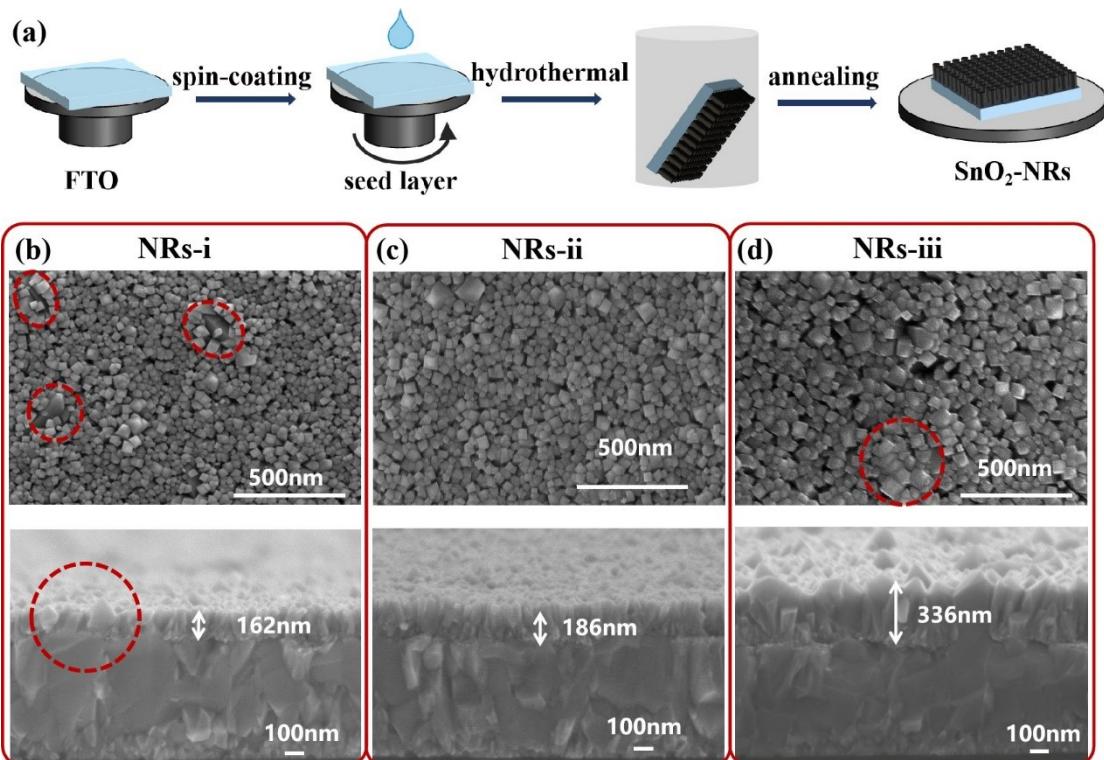


Fig. S7 (a) Schematic of the fabrication procedures of SnO_2 -NRs. Top-view and cross-sectional SEM images of SnO_2 -NRs in different reaction temperature: (b) NRs-i sample at 180°C , (c) NRs-ii sample at 200°C , and (d) NRs-iii sample at 220°C , respectively.

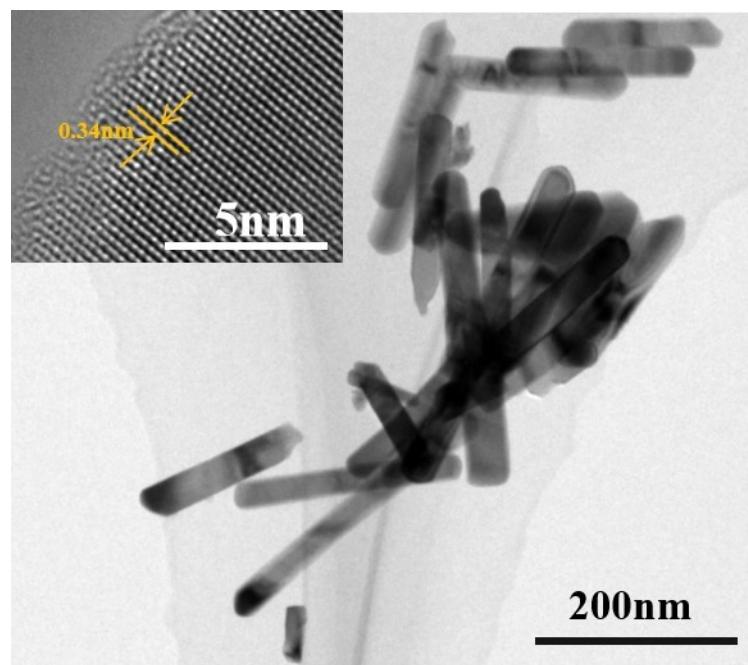


Fig. S8 TEM and HR-TEM images of SnO₂-NRs scraped from NR-ii sample.

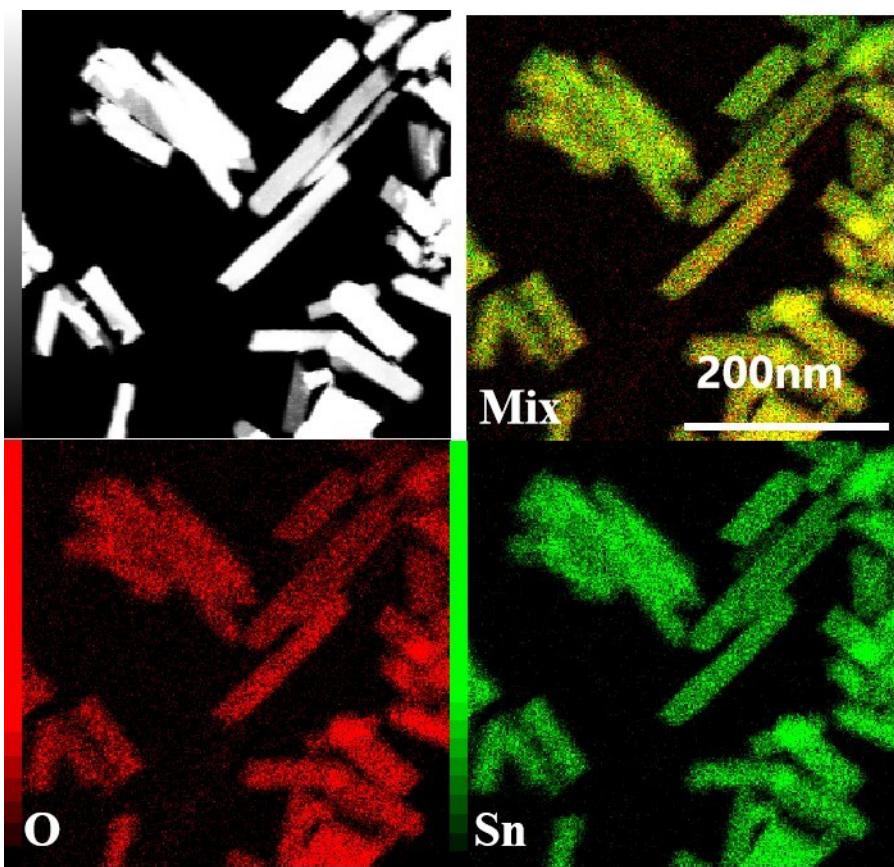


Fig. S9 High-angle annular dark-field (HADDF) STEM image of NRs-ii and energy dispersive X-ray (EDX) elemental mapping images of O, Sn elements.

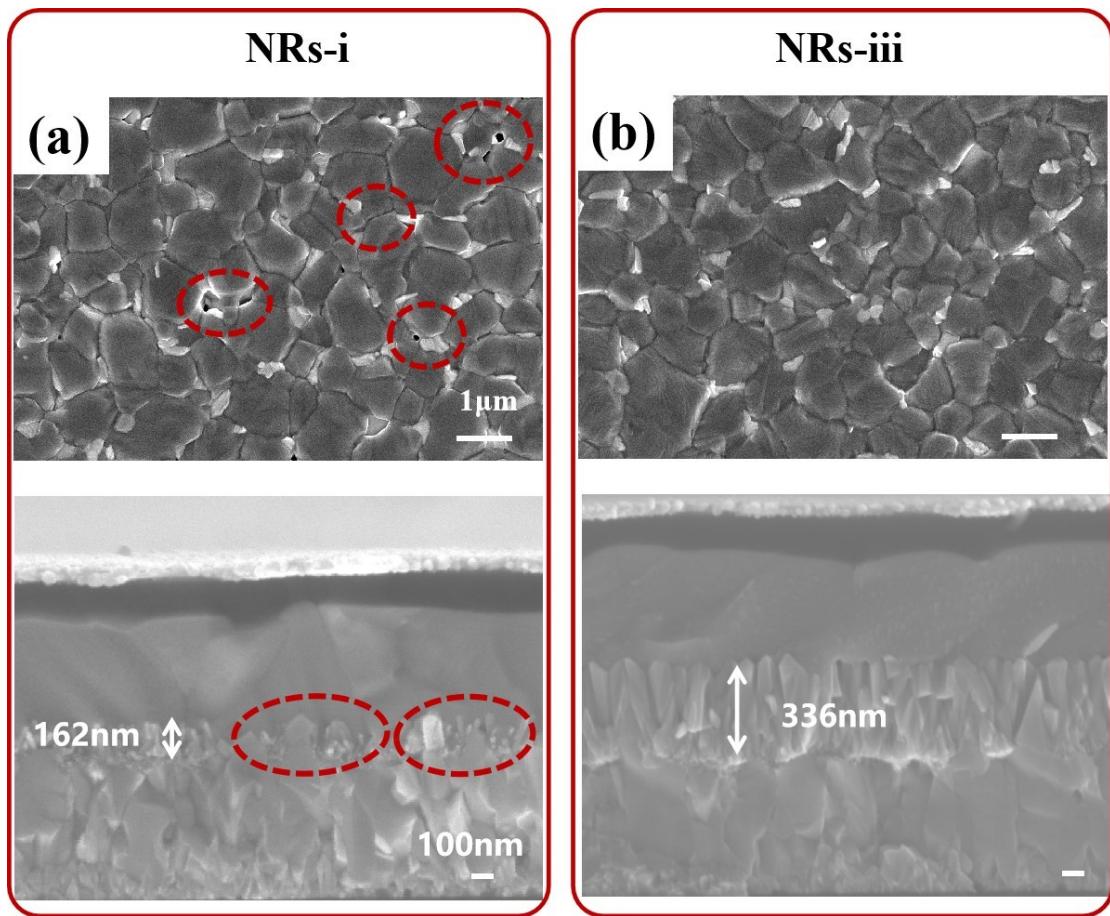


Fig. S10 Top-view and cross-sectional SEM images of perovskite film based on **(a)** NRs-i ETL substrate at 180°C, **(b)** NRs-iii ETL substrate at 220°C

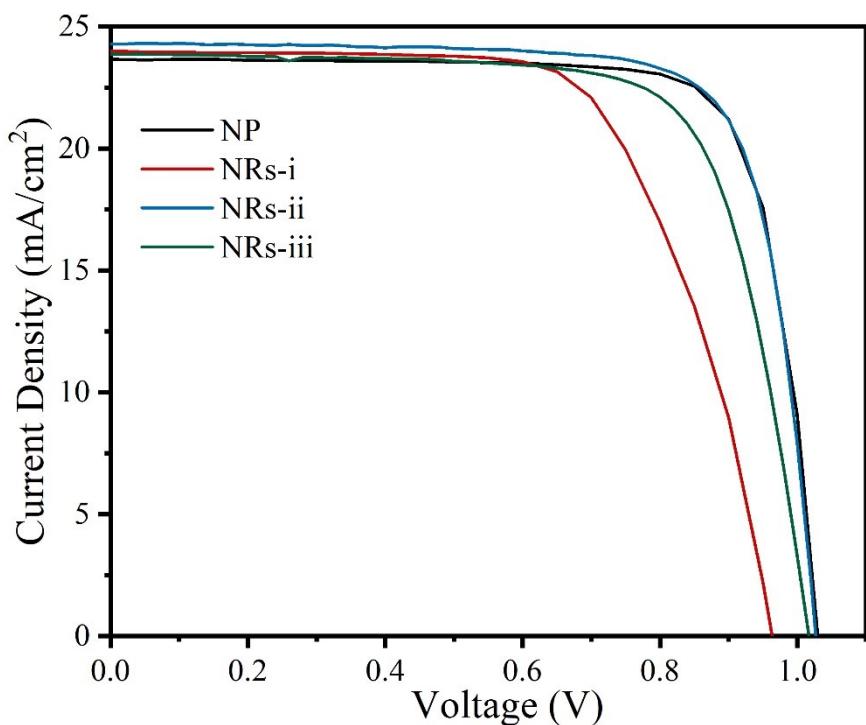


Fig. S11 Current-voltage curves of PSCs based on commercial SnO₂-NP and SnO₂-NRs in different reaction temperature: NRs-i sample at 180°C, NRs-ii sample at 200°C, and NRs-iii sample at 220°C, respectively.

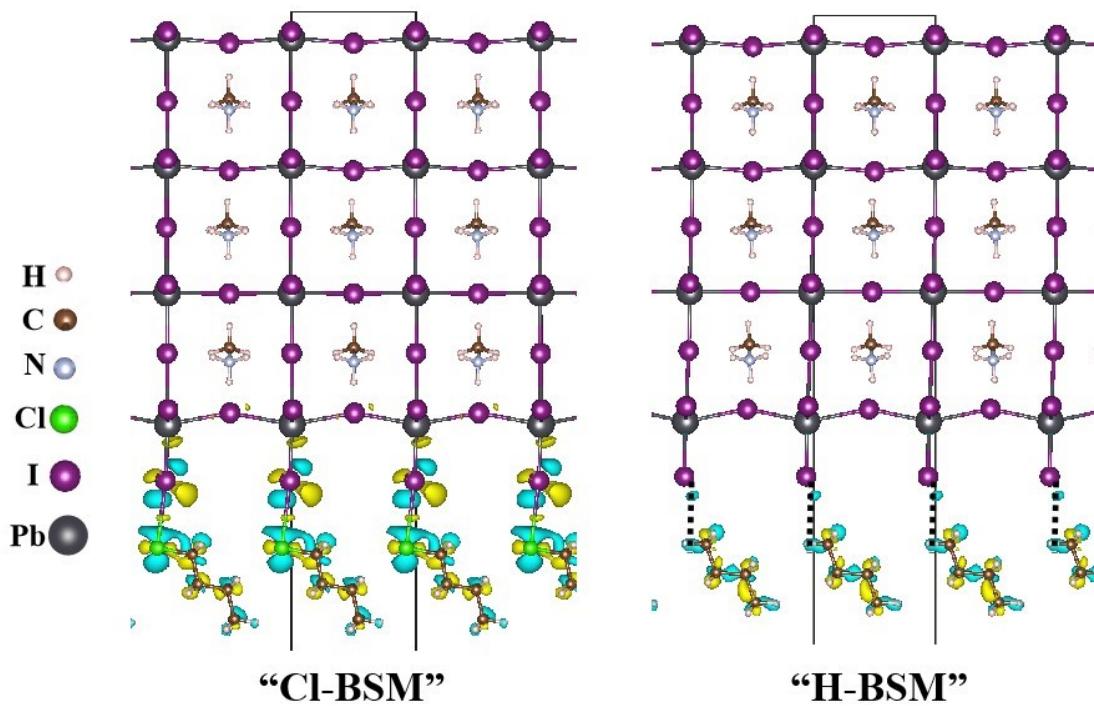


Fig. S12 Charge transfer density difference plots for (a) $\text{H}(\text{CH}_2)_4\text{Cl}$ (“Cl-BSM”) and (b) $\text{H}(\text{CH}_2)_4\text{H}$ (“H-BSM”) interfaced with I-terminated MAPbI_3 (001) surface. Dashed lines across the interface indicate no bonding.

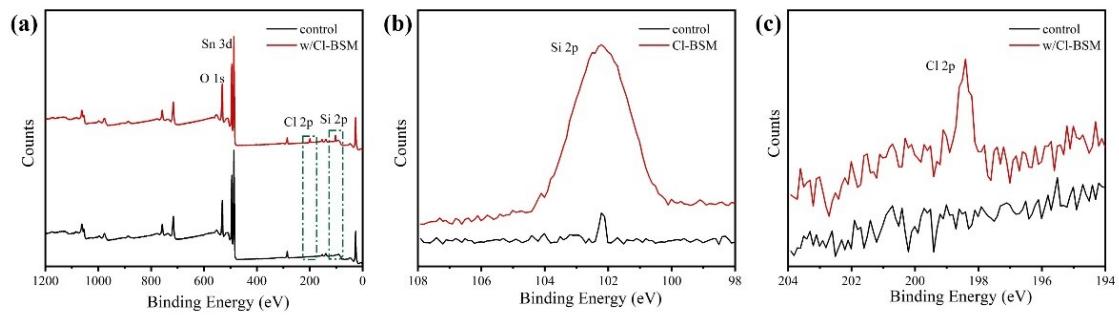


Fig. S13 XPS spectra from top-surface of SnO_2 -NRs (control) and Cl-BSM/ SnO_2 -NRs (w/Cl-BSM): **(a)** typical XPS survey, **(b)** Si 2p core level and **(c)** Cl 2p core level.

w/Cl-BSM

~7.6°

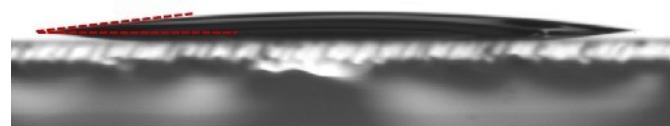


Fig. S14 The deionized water contact angles of SnO₂-NRs/Cl-BSM prepared on FTO substrates.

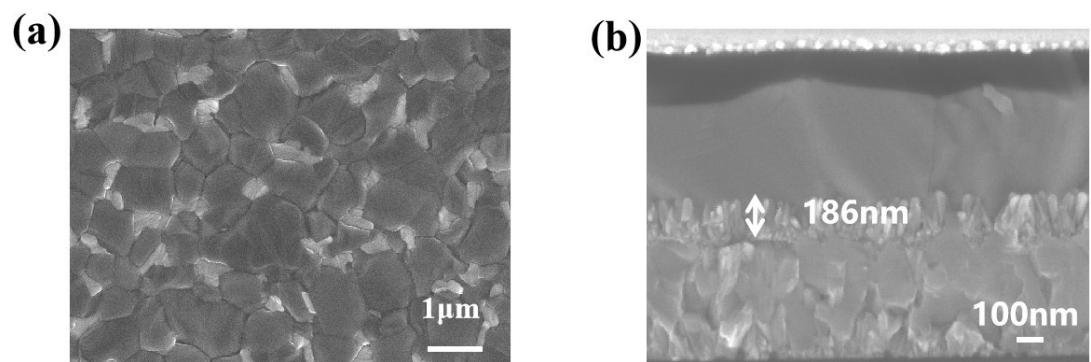


Fig. S15 (a) Top-view SEM image of the perovskite film based on Cl-BSM/SnO₂-NRs and (b) cross-sectional SEM image of PSCs device.

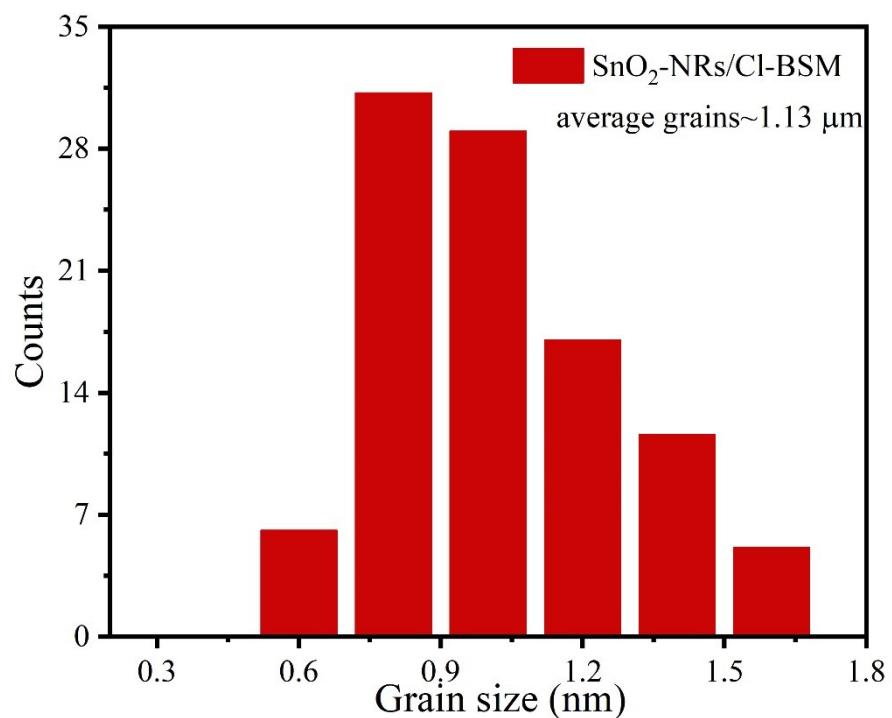


Fig. S16 The size distribution of perovskite film based on SnO_2 -NRs/Cl-BSM substrate.

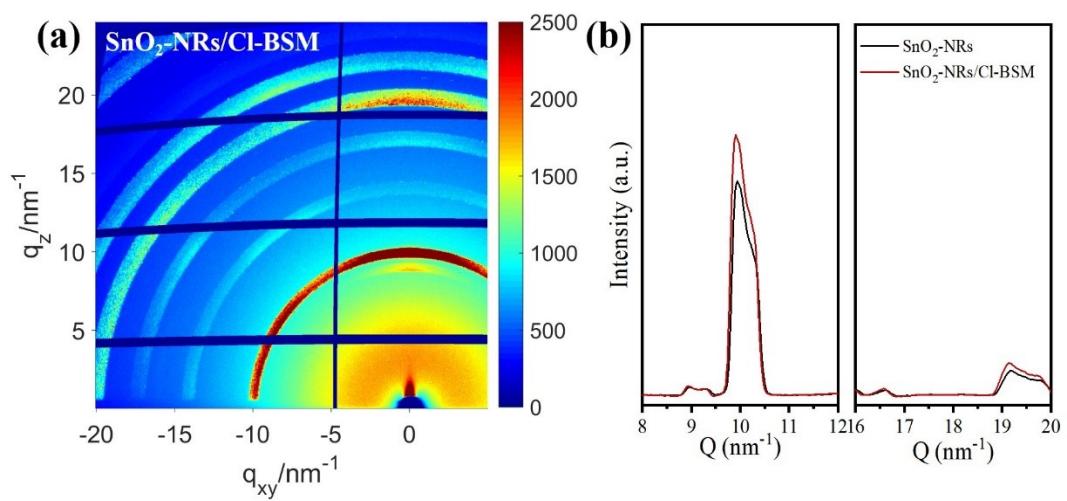


Fig. S17 (a) 2D GIWAXS patterns of perovskite film deposited on SnO_2 -NRs/Cl-BSM coated quartz substrates, (b) Integrated intensity profiles along the q_z direction of GIWAXS pattern for SnO_2 -NRs and SnO_2 -NRs/Cl-BSM substrates.

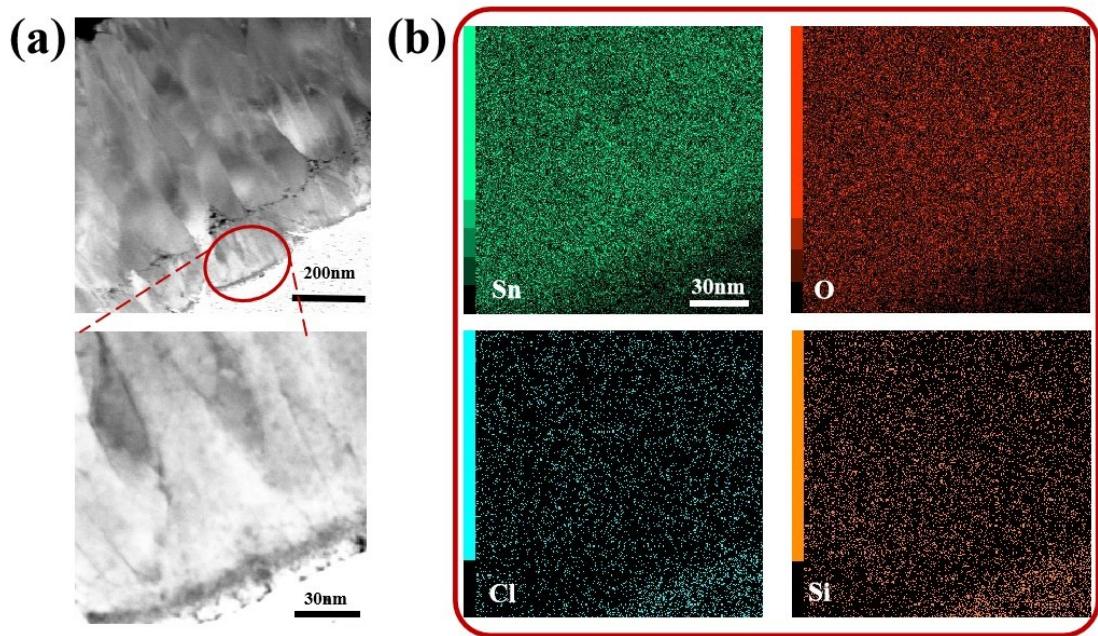


Fig. S18 **(a)** HAADF-STEM cross-sectional image (top) and zoom-in image of selected area (bottom) of Cl-BSM/SnO₂-NRs grown on FTO substrate. **(b)** EDX cross-sectional elemental mapping images of Sn, Si and Cl elements for Cl-BSM/SnO₂-NRs.

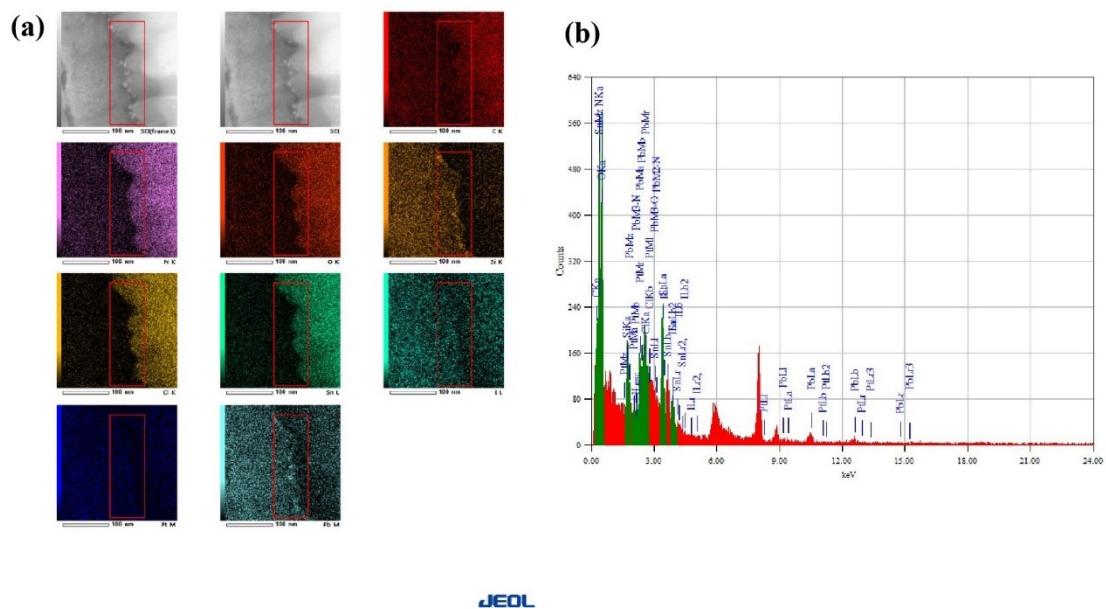


Fig. S19 **(a)** EDX elemental mapping and **(b)** EDX spectrum of PVK films based on Cl-BSM/SnO₂-NRs.

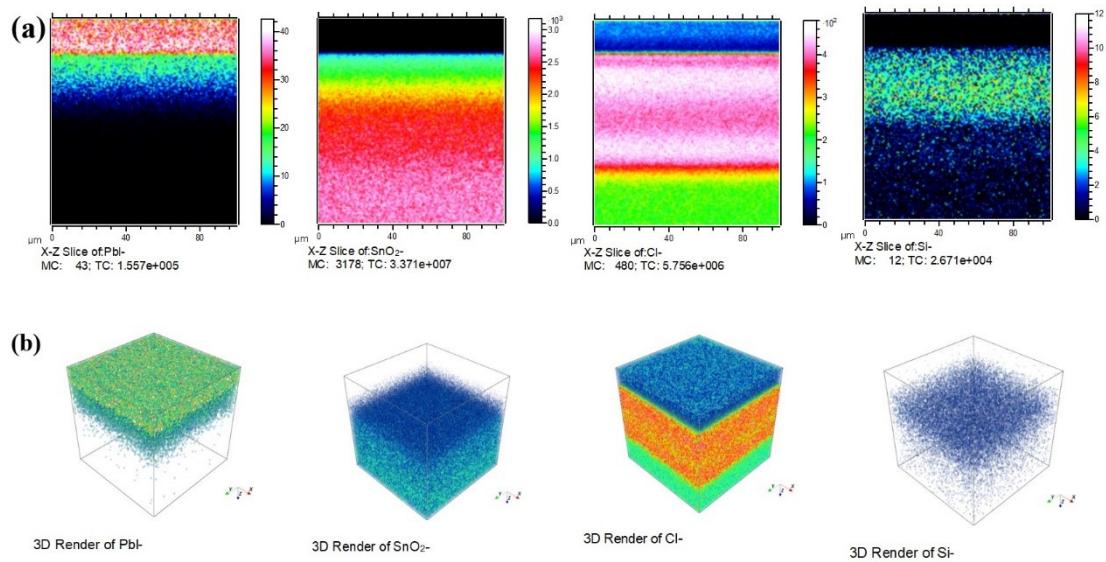


Fig. S20 **(a)** The X-Z Slice and **(b)** 3D render of various elements SIMS data.

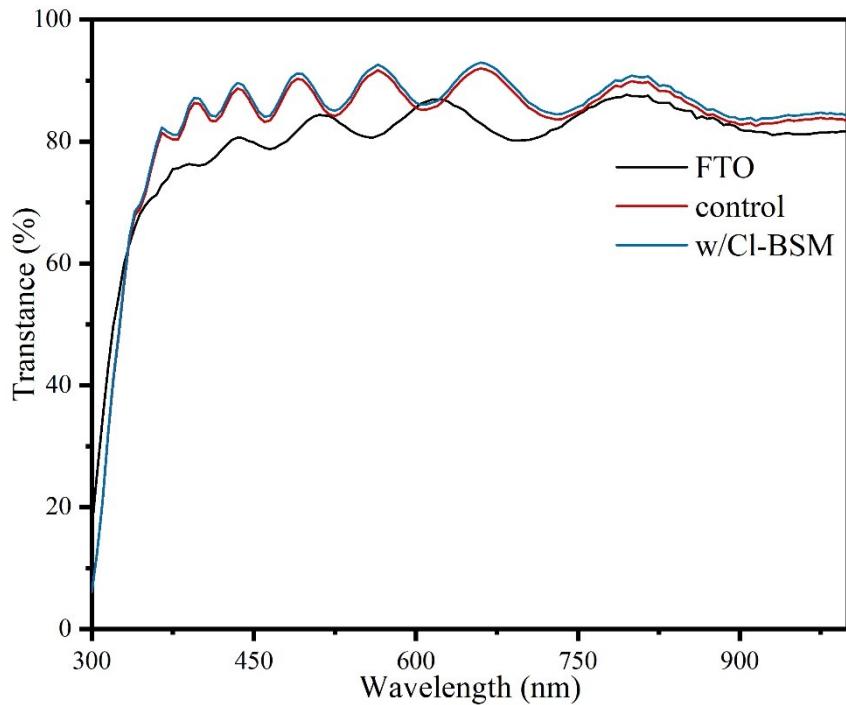


Fig. S21 Transmittance spectra based on bare FTO, SnO₂-NRs, and the Cl-BSM/SnO₂-NRs substrates, respectively.

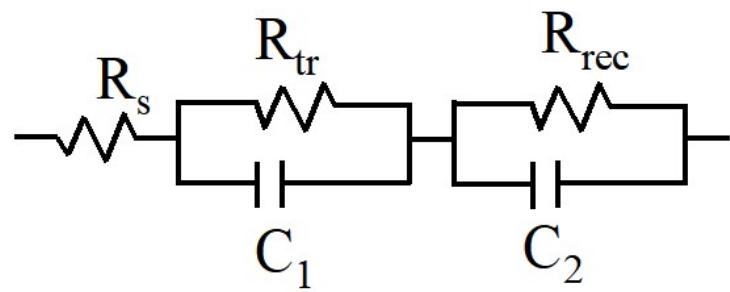


Fig. S22 Equivalent circuit of PSCs based on SnO_2 -NRs and SnO_2 -NRs/Cl-BSM used for fitting impedance data.

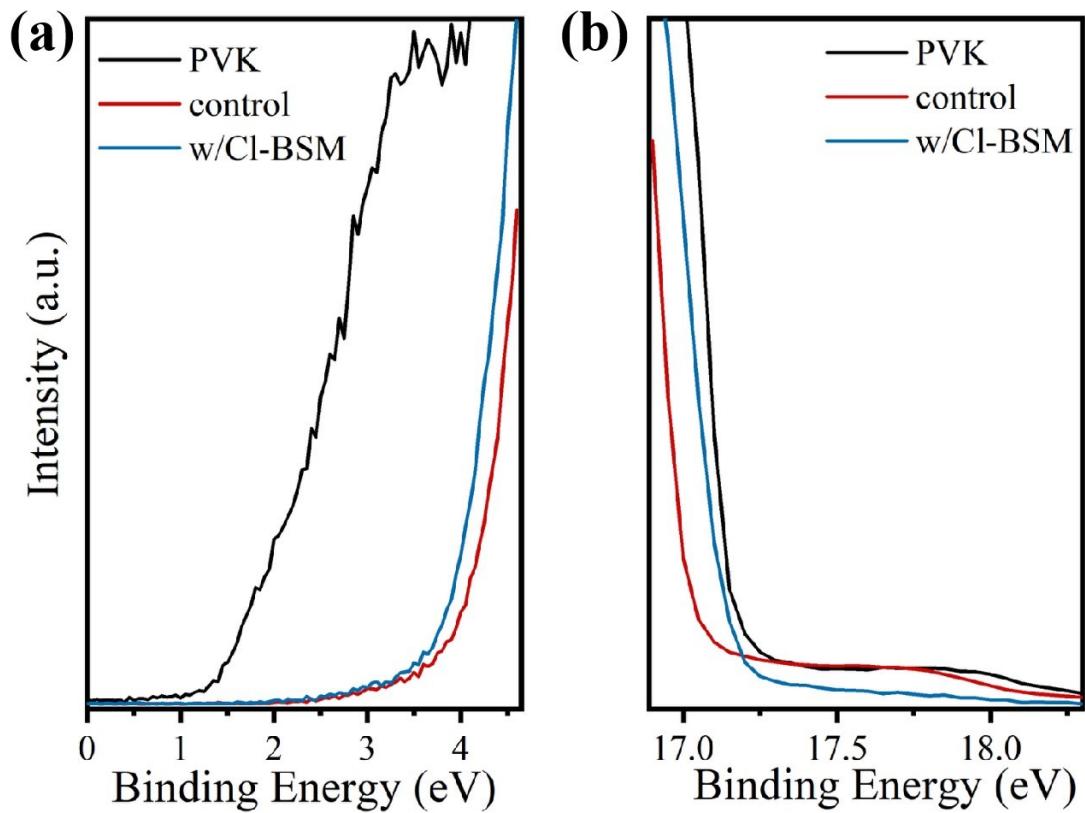


Fig. S23 The UPS spectra on perovskite, SnO_2 -NRs and Cl-BSM/ SnO_2 -NRs films: **(a)** Valance regions (E_{onset}) and **(b)** secondary-electron cutoff (E_{cutoff}). The values of E_{onset} are 1.45 eV, 4.05 eV and 4.03 eV for perovskite, SnO_2 -NRs and Cl-BSM/ SnO_2 -NRs, respectively. The conduction band minimum (E_{CB}) can be calculated from the formular $E_{\text{CB}} = \text{WF} + E_{\text{onset}} - E_g$, which are 3.99 eV, 4.16 eV and 4.03 eV for perovskite, SnO_2 -NRs and Cl-BSM/ SnO_2 -NRs, respectively.

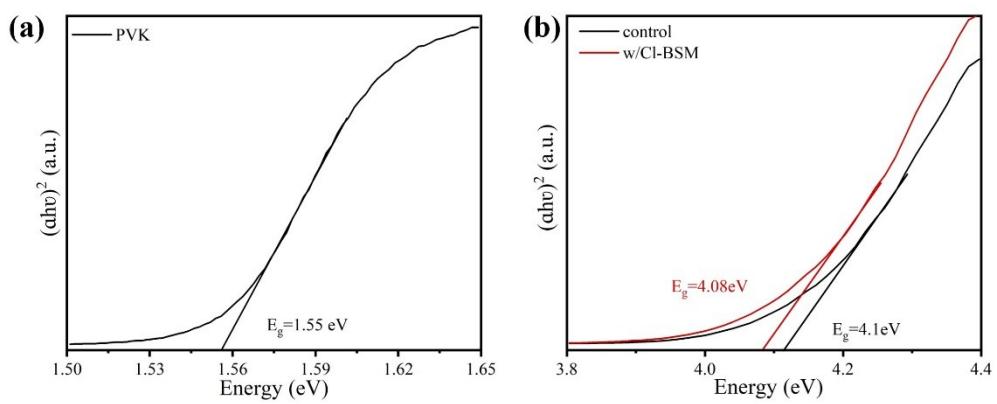


Fig. S24 Tauc plot of **(a)** perovskite and **(b)** SnO_2 -NRs and SnO_2 -NRs/Cl-BSM films.

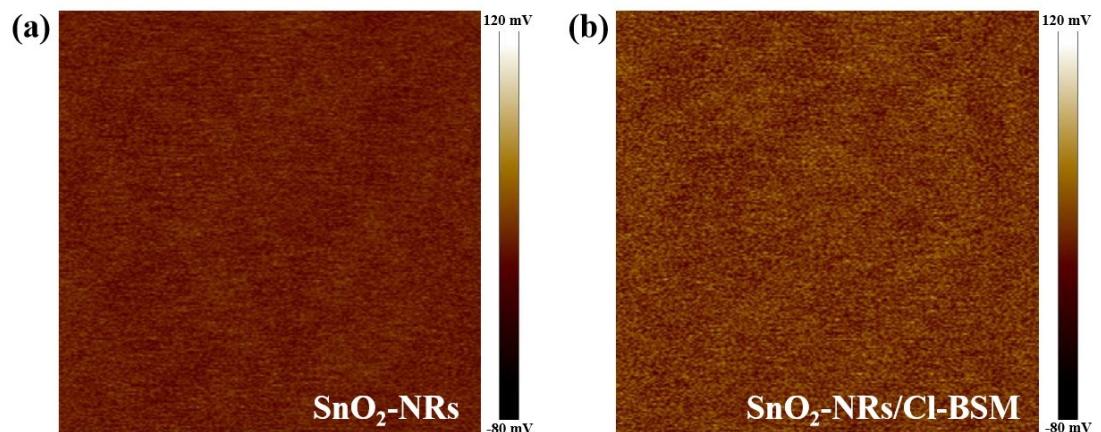


Fig. S25 KPFM images of (a) $\text{SnO}_2\text{-NRs}$ and (b) $\text{SnO}_2\text{-NRs/Cl-BSM}$ films.

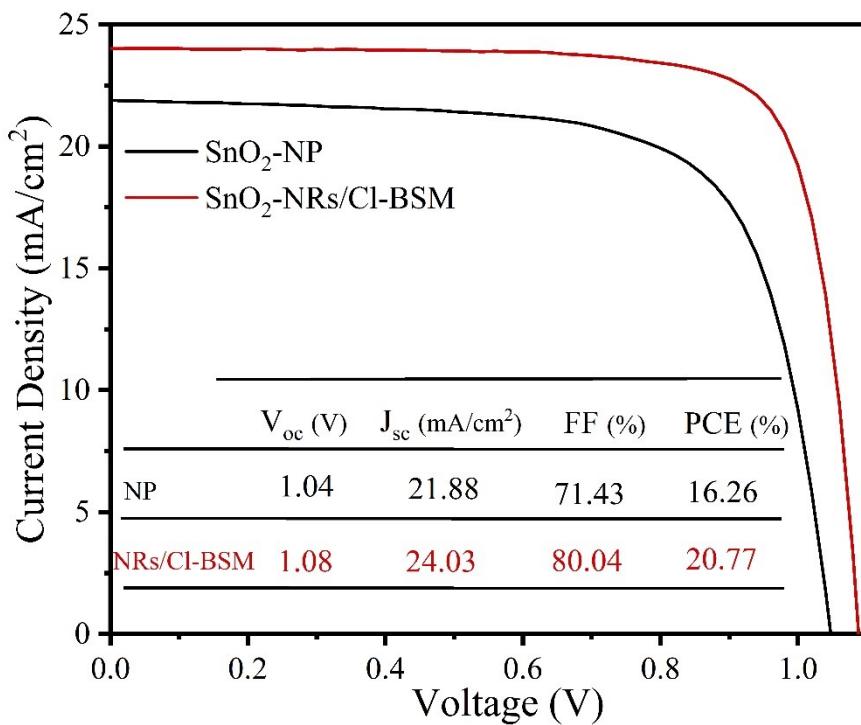


Fig. S26 Current-voltage curves of PSCs based on commercial SnO₂-NP and SnO₂-NRs/Cl-BSM, respectively. Inset shows the relevant paraments of PSCs.

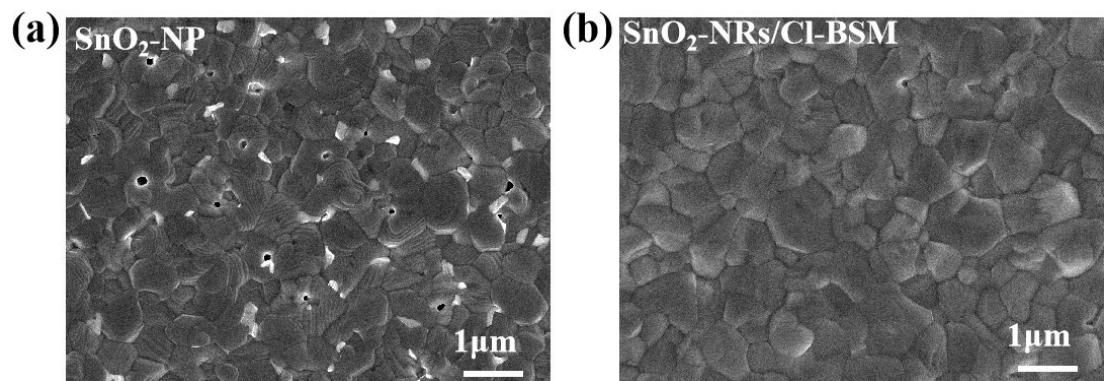


Fig. S27 Top-view SEM images of perovskite film based on SnO₂-NPs and SnO₂-NRs/Cl-BSM, respectively.

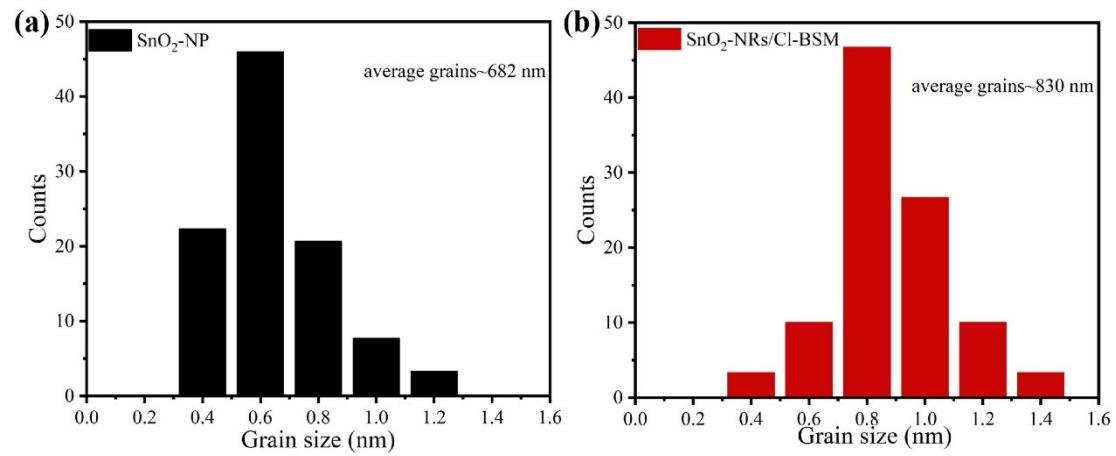


Fig. S28 The size distribution of perovskite film based on $\text{SnO}_2\text{-NPs}$ and $\text{SnO}_2\text{-NRs/Cl-BSM}$, respectively.

Table S1 Performance comparison of recent works about green-solvent engineering

host-solvent	anti-solvent/ air knife	PSCs structure	PCE/ %	year
non-green	Green			
DMF/DMSO	diisopropyl ether	FTO/cTiO ₂ /mTiO ₂ /Li-TiO ₂ /perovskite/Spiro/Au	21.26	2022 ¹
DMF/DMSO	ethyl acetate /hexane	ITO/SnO ₂ /perovskite/Spiro/Ag	20.06	2019 ²
DMF/DMSO	anisole	FTO/cTiO ₂ /mTiO ₂ /Li-TiO ₂ /perovskite/Spiro/Au	20.2	2018 ³
DMF/DMSO	methyl benzoate	FTO/SnO ₂ /perovskite/Spiro/Au	22.37	2020 ⁴
DMF/DMSO	tetraethyl orthosilicate	ITO/NiO _x /perovskite/PCBM/BCP/Ag	18.15	2020 ⁵
DMF/DMSO	methylamine bromide/ethane	ITO/PTAA/perovskite/PC ₆₁ BM/Phen-NADPO/Ag	21.53	2020 ⁶
DMF/DMPU	air knife	FTO/SnO ₂ /perovskite/Spiro/Au	20.56	2021 ⁷
DMF/NMP	antisolvent-free	FTO/SnO ₂ /perovskite/Spiro/Au	24.02	2022 ⁸
Green	Green			
DMSO	isopropanol	ITO/NiO _x /PTAA/perovskite/PC ₆₁ BM/BCP/Ag	19.5	2021 ⁹
DMSO	antisolvent-free	ITO/PTAA /SiO ₂ -NP/perovskite/PCBM/BCP/Ag	16.7	2021 ¹⁰
DMSO/GBL	antisolvent-free	FTO/NiO _x /perovskite/PCBM/PEI/BC P/Ag	17.02	2020 ¹¹
2-Me/ACN	air knife	ITO/PTAA/perovskite/C ₆₀ /BCP/metal cathode	21.3	2019 ¹²
2-Me/NMP	antisolvent-free	ITO/MeO-2PACz/perovskite/PC ₆₀ BM/BCP/Ag	20.39	2022 ¹³
2-Me/DMSO /MA ethanol	antisolvent-free	FTO/SnO ₂ /perovskite/Spiro/Au	18.27	2022 ¹⁴
TMP/DMF	dibutyl ether	FTO/SnO ₂ /perovskite/Spiro/Au	20.02	2021 ¹⁵
TEP	dibutyl ether	FTO/SnO ₂ /KCl/perovskite/PEAI/Spiro/Au	20.1	2022 ¹⁶
TEP	dibutyl ether	FTO/SnO ₂ -NRs/Cl-BSM/perovskite/Spiro/Ag	22.42	This work
2-Me	antisolvent-free	FTO/SnO ₂ -NRs/Cl-BSM/perovskite/Spiro/Ag	20.77	This work

Table S2. Photovoltaic parameters of the PSCs based on commercial SnO₂-NP and SnO₂-NRs in different reaction temperature: NRs-i sample at 180°C, NRs-ii sample at 200°C, and NRs-iii sample at 220°C, respectively.

ETL substrates	J _{sc} (mA/cm ²)	V _{oc} (V)	FF (%)	PCE (%)
SnO ₂ -NP	23.66	1.00	81.04	19.17
NRs-i	24.01	0.96	67.82	15.63
NRs-ii	24.30	1.02	77.86	19.30
NRs-iii	23.87	1.02	74.31	18.09

Table S3. DFT results for E_{int} for “BSM” on different surface terminations of MAPbI_3 (010). Here, “—” indicates bond formation with charge transfer, and “---” indicates no chemical bond formation

“BSM”/Termination per cell	Cl-	H-	Cl-	H-
	SMB/MAI-	SMB/MAI-	SMB/PbI ₂ -	SMB/PbI ₂ -
E_{int}/eV	-0.176944	-0.147035	-0.465694	-0.226885
Bond Type (Length, Å)	Cl—I (3.42567)	H---I (3.44440)	Cl—Pb (3.19359)	H---Pb (3.18806)

Table S4. PL decay lifetimes of PVK films on SnO₂-NRs (control) and SnO₂-NRs/Cl-BSM (w/Cl-BSM) coated FTO substrates.

Substrate	τ_1 (ns)	A ₁	τ_2 (ns)	A ₂	τ_t (ns)
Control	1.063	6588.92	78.103	543.47	67.19
w/Cl-BSM	1.797	1171.92	39.714	694.61	37.02

Table S5. The detailed photovoltaic parameters of the champion devices with or without Cl-BSM measured in reverse scan and forward scan at a scan rate of 100 mV/s under simulated AM 1.5G one sun illumination of 100mW/cm²

Device	Scan direction	J_{sc} (mA/cm ²)	J_{sc} from EQE			h (%)
			V_{oc} (V)	FF (%)	PCE (%)	
control	reverse scan	24.3	1.02	77.86	19.3	23.4
	forward scan	24.17	23.3	0.92	70.31	15.64
w/Cl-BSM	reverse scan	24.6	1.12	81.36	22.42	2.6
	forward scan	24.3	24.2	1.12	80.24	21.84

References

- [1] L. Wang, X. Wang, L.-L. Deng, S. Leng, X. Guo, C.-H. Tan, W. C. H. Choy, C.-C. Chen, *Materials Horizons*, 2020, **7**, 934-42.
- [2] H. B. Lee, M. K. Jeon, N. Kumar, B. Tyagi, J. W. Kang, *Advanced Functional Materials*, 2019, **29**, 1903213.
- [3] M. Yavari, M. Mazloum-Ardakani, S. Gholipour, M. M. Tavakoli, S.-H. Turren-Cruz, N. Taghavinia, M. Grätzel, A. Hagfeldt, M. Saliba, *Advanced Energy Materials*, 2018, **8**, 1800177.
- [4] Y. Yun, F. Wang, H. Huang, Y. Fang, S. Liu, W. Huang, Z. Cheng, Y. Liu, Y. Cao, M. Gao, L. Zhu, L. Wang, T. Qin, W. Huang, *Adv Mater*, 2020, **32**, e1907123.
- [5] M. Wang, Q. Fu, L. Yan, J. Huang, Q. Ma, M. Humayun, W. Pi, X. Chen, Z. Zheng, W. Luo, *Chemical Engineering Journal*, 2020, **387**, 123966.
- [6] W. Xu, Y. Gao, W. Ming, F. He, J. Li, X. H. Zhu, F. Kang, J. Li, G. Wei, *Adv Mater*, 2020, **32**, e2003965.
- [7] D. Lee, K. Lim, J. Lee, N. Park, *Journal of Materials Chemistry A*, 2021, **9**, 3018-3028.
- [8] T. Bu, L. Ono, J. Li, J. Su, G. Tong, W. Zhang, Y. Liu, J. Zhang, J. Chang, S. Kazaoui, F. Huang, Y.-B. Cheng, Y. Qi, *Nature Energy*, 2022, **7**, 528-536.
- [9] S. Shan, Y. Li, H. Wu, T. Chen, B. Niu, Y. Zhang, D. Wang, C. Kan, X. Yu, L. Zuo, H. Chen, *SusMat*, 2021, **1**, 537-44.
- [10] J. Küffner, J. Hanisch, T. Wahl, J. Zillner, E. Ahlswede, M. Powalla, *ACS Applied Energy Materials*, 2021, **4**, 11700-10.
- [11] H. Huang, Y. Tian, C. Huang, F. Li, C. Chu, M. Lee, C. Huang, F. Su, *ACS Appl Mater Interfaces*, 2020, **12**, 26041-9.
- [12] Y. Deng, C. Van Brackle, X. Dai, J. Zhao, B. Chen, J. Huang, *Sci Adv*, 2019, **5**, eaax7537.
- [13] S. Lee, S. Hong, H. Kim, *ACS Appl Mater Interfaces*, 2022, **14**, 39132-39140.
- [14] Q. Zhang, G. Ma, A. Green, K. Gollinger, J. Moore, T. Demeritte, C. Ray, A.

- Hill, X. Gu, E. Morgan, M. Feng, S. Banerjee, Q. Dai, *ACS Applied Energy Materials*, 2022, **5**, 1487-1495.
- [15] Yun. Y, Vidyasagar. D, Lee. M, Gong. O. Y, Jung. J, Jung. H. S, Kim. D. H, Lee. S, *Adv Sci*, 2021, **8**, e2102492-501.
- [16] Cao. X, Hao. L, Liu. Z, Su. G, He. X, Zeng. Q, Wei. J, *Chemical Engineering Journal*, 2022, **437**, 135458-66.