

Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
$\text{Li}_{10}\text{GeS}_2\text{P}_{12}$	$P4_2mc$	200	650, 900, 1150, 1400 [0.00216,0.00541,0.00922,0.012905]
$\text{Li}_2\text{B}_{12}\text{H}_{12}$	Pa_3	832	600 [0.00007]
$\text{Li}_{1.94}\text{B}_{12}\text{H}_{12}$	Pa_3	830	600 [0.00025]
Li_2SnS_3	$C2/c$	384	400, 600, 800 [0,0,0]
$\text{Li}_{1.97}\text{SnS}_3$	$C2/c$	382	400, 600, 800, 1000, 1200 [0,0,0.00001,0.00005,0.00006]
$\text{Li}_3\text{La}_4\text{Ti}_8\text{O}_{24}$	$Pmmm$	312	500, 700, 900, 1100 [0,0.00001,0.00033,0.00054]
Li_3N	$P6/mmm$	256	300, 500, 700, 900, 1100 [0,0.00010,0.00095,0.00234,0.00407]
$\text{Li}_{2.94}\text{N}_{0.98}$	$P6/mmm$	212	300, 500, 700, 900 [0,0.00012,0.00029,0.00547]
Li_3OCl	$P4/mmm$	320	1250 [0]
$\text{Li}_{2.95}\text{O}_{0.98}\text{Cl}_{0.98}$	$P4/mmm$	315	1000, 1250, 1500, 1650 [0.00007,0.00042,0.00671,0.01747]
$\text{Li}_6\text{PS}_5\text{Br}$	$F\bar{4}3m$	208	500 [0]
$\text{Li}_{5.94}\text{PS}_5\text{Br}$	$F\bar{4}3m$	207	500 [0.00001]
$\text{Li}_7\text{La}_3\text{Zr}_2\text{O}_{12}$	$I4_1/acd$	384	400, 600, 800 [0,0,0.00003]

(Continuation)			
Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
$\text{Li}_{6.88}\text{La}_3\text{Zr}_2\text{O}_{12}$	$I4_1/acd$	382	400, 600, 800 [0.00005,0.00017,0.00037]
$\text{Li}_7\text{P}_3\text{S}_{11}$	$P\bar{1}$	336	400, 600, 800 [0.00052,0.00270,0.00616]
$\text{Li}_{6.94}\text{P}_3\text{S}_{11}$	$P\bar{1}$	335	400, 600, 800 [0.00065,0.00304,0.00660]
LiF	$Fm\bar{3}m$	200	800, 1000, 1250 [0,0,0]
$\text{Li}_{0.98}\text{F}_{0.98}$	$Fm\bar{3}m$	196	400, 600, 800 [0,0,0.00002]
LiGaO_2	$Pna2_1$	128	300, 500, 700, 900, 1100, 1300 [0,0,0,0,0,0]
$\text{Li}_{0.97}\text{Ga}_{0.97}\text{O}_{1.94}$	$Pna2_1$	124	500, 700, 900, 1100, 1300, 1500 [0,0,0,0.00002,0.00010,0.00019]
LiIO_3	$P6_3$	270	300, 500, 700, 900, 1100 [0,0,0,0,0.00031]
$\text{LiI}_{0.94}\text{O}_3$	$P6_3$	267	300, 500, 700, 900 [0,0,0.00017,0.00022]
$\text{LiMn}(\text{HCO}_2)_3$	$P2_13$	224	500 [0]
$\text{Li}_{0.94}\text{Mn}(\text{HCO}_2)_3$	$P2_13$	223	500, 700, 900 [0,0,0.00001]
LiNbO_3	$R3c$	180	500, 700, 900 [0,0,0]
$\text{Li}_{0.97}\text{NbO}_3$	$R3c$	179	500, 700, 900 [0.00001,0.00003,0.00014]

(Continuation)			
Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
$\text{LiTi}_2(\text{PO}_4)_3$	$P\bar{3}c1$	432	400, 600, 800 [0,0,0]
$\text{Li}_{0.92}\text{Ti}_2(\text{PO}_4)_3$	$P\bar{3}c1$	430	400, 600, 800 [0,0,0.00001]

Supplementary Table I: Li-based SSE included in the DFT-AIMD database and details on the corresponding finite-temperature simulations. Ion diffusion coefficients, D , listed as “0” are $< 10^{-5} \text{ \AA}^2/\text{fs}$.

Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
$\text{Na}_2\text{B}_4\text{O}_7$	$P\bar{1}$	312	300, 450, 600, 750 [0,0,0,0]
Na_3SbS_4	$I\bar{4}3m$	288	500, 700, 900 [0,0,0.00003]
NaBH_4	$Pnma$	288	300, 450, 600, 750 [0,0,0,0]
$\text{Na}_{0.98}\text{BH}_4$	$Pnma$	287	300, 450, 600, 750 [0.00003,0.00006,0.00018,0.00027]
NaBO_2	$C2/c$	288	750 [0]
$\text{NaMn}(\text{HCO}_2)_3$	$P2_13$	224	500 [0]
$\text{Na}_{0.94}\text{Mn}(\text{HCO}_2)_3$	$P2_13$	223	500, 700, 900 [0,0,0.00001]
$\text{NaZr}_2(\text{PO}_4)_3$	$C2/c$	216	500 [0]
$\text{Na}_{0.92}\text{Zr}_2(\text{PO}_4)_3$	$C2/c$	215	500 [0]
AgCrSe_2	$P3m1$	192	500 [0]
$\text{Ag}_{0.98}\text{CrSe}_2$	$P3m1$	191	500 [0.00001]
AgI	$F\bar{4}3m$	250	200, 350, 500, 650 [0,0.00154,0.00408,0.00692]

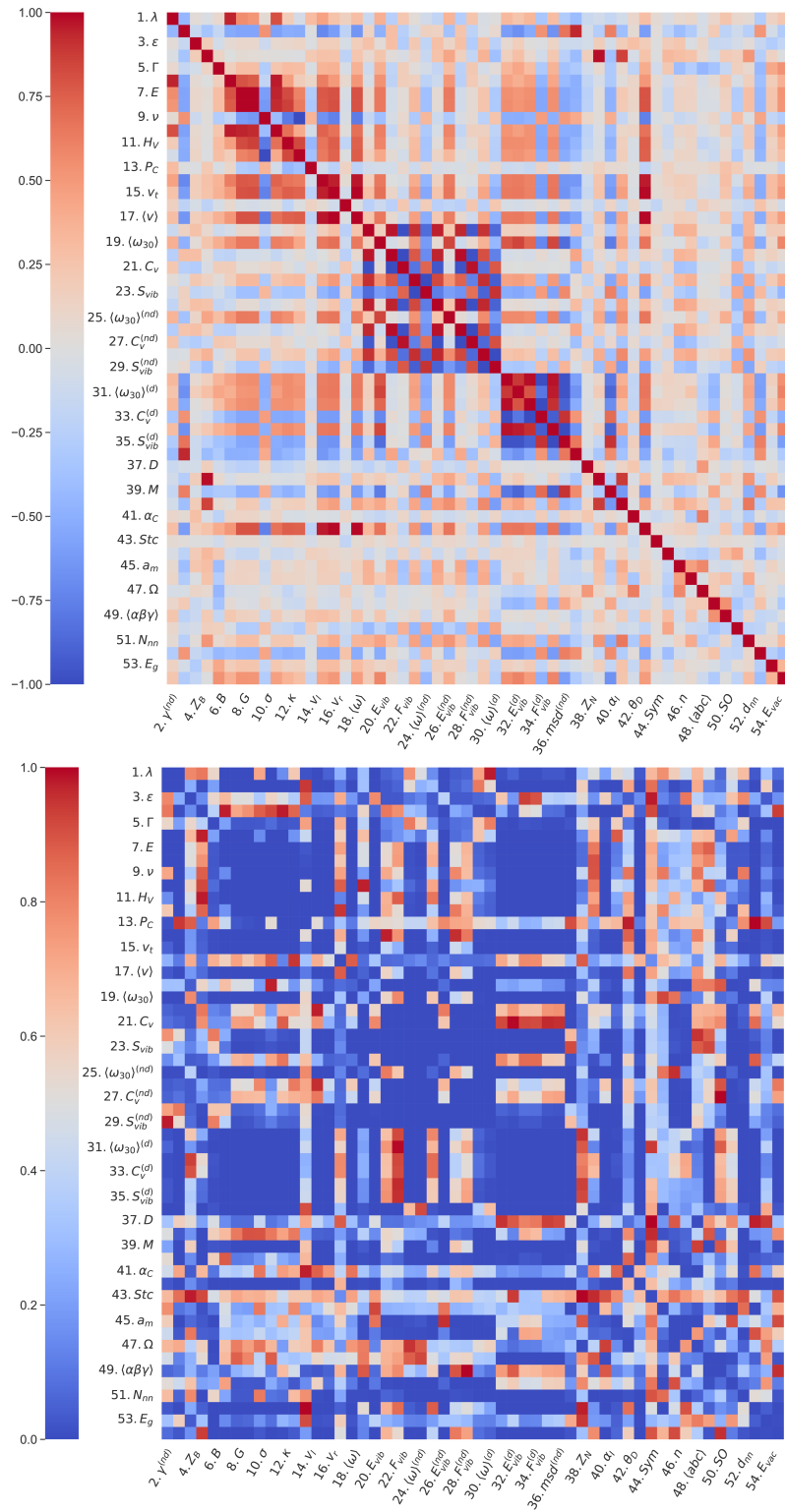
(Continuation)			
Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
$\text{Cu}_{1.99}\text{Se}$	$Fm\bar{3}m$	215	300, 500, 700, 900 [0.00002,0.00061,0.00281,0.00555]
CuI	$F\bar{4}3m$	216	500, 700, 900 [0.00072,0.00271,0.00746]

Supplementary Table II: Na-, Ag- and Cu-based SSE included in the DFT-AIMD database and details on the corresponding finite-temperature simulations. Ion diffusion coefficients, D , listed as “0” are $< 10^{-5} \text{ \AA}^2/\text{fs}$.

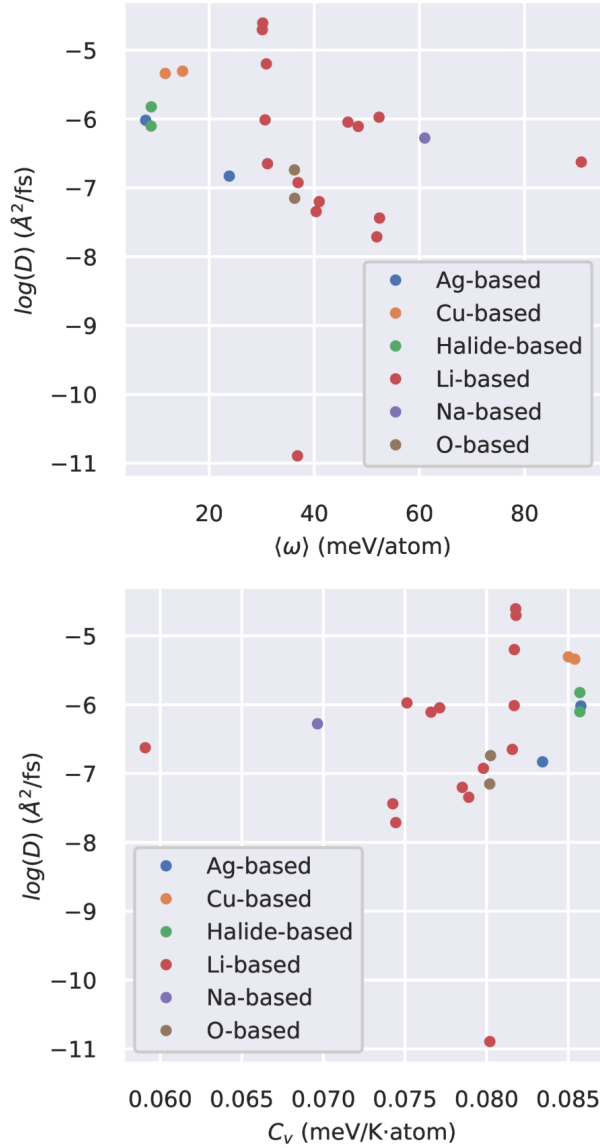
Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
BaF ₂	$Fm\bar{3}m$	192	1000, 1250, 1375, 1500 [0,0.00005,0.00406,0.00688]
CaF ₂	$Fm\bar{3}m$	192	500, 1000, 1250, 1500 [0,0,0.00006,0.00322]
CH ₃ NH ₃ PbBr ₃	Pm	324	250, 400 [0,0]
CH ₃ NH ₃ PbBr _{2.96}	Pm	323	400, 600, 800 [0,0.00001,0.00366]
CH ₃ NH ₃ PbIBr ₂	Pm	324	250, 400 [0,0]
CH ₃ NH ₃ PbI _{0.93} Br ₂	Pm	322	400 [0]
CH ₃ NH ₃ PbI ₂ Br ₁	Pm	324	100, 250, 400 [0]
CH ₃ NH ₃ PbI ₂ Br _{0.93}	Pm	322	400 [0]
CH ₃ NH ₃ PbI ₃	Pm	324	100, 250, 400 [0,0,0]
CH ₃ NH ₃ PbI _{2.96}	Pm	323	400, 600, 800 [0,0.00016,0.00407]
CsPbBr ₃	$Pm\bar{3}m$	180	500, 700, 850 [0,0.00018,0.00047]
CsPbBr _{2.97}	$Pm\bar{3}m$	179	500, 700, 900 [0.00009,0.00016,0.00028]
SrCl ₂	$C2/m$	192	1500, 2000, 2500 [0, 0.00001,0.00873]
SrF ₂	$Fm\bar{3}m$	192	1250, 1375, 1500, 1750 [0,0,0.00111,0.00856]

(Continuation)			
Composition	Space group	Number of atoms	Temperatures (K) [D ($\text{\AA}^2/\text{fs}$)]
Bi_2O_3	$Pn\bar{3}m$	180	500 [0]
$\text{LaGaO}_{2.98}$	$C2/c$	239	500, 700, 900 [0,0,0.00001]
SrCoO_3	$Pbam$	160	200, 400 [0,0]
$\text{SrCoO}_{2.75}$	$Pbam$	152	200, 400 [0.00001,0.00002]
$\text{SrTiO}_{2.95}$	Pc	317	1000, 1500, 2000 [0,0.00002,0.00003]

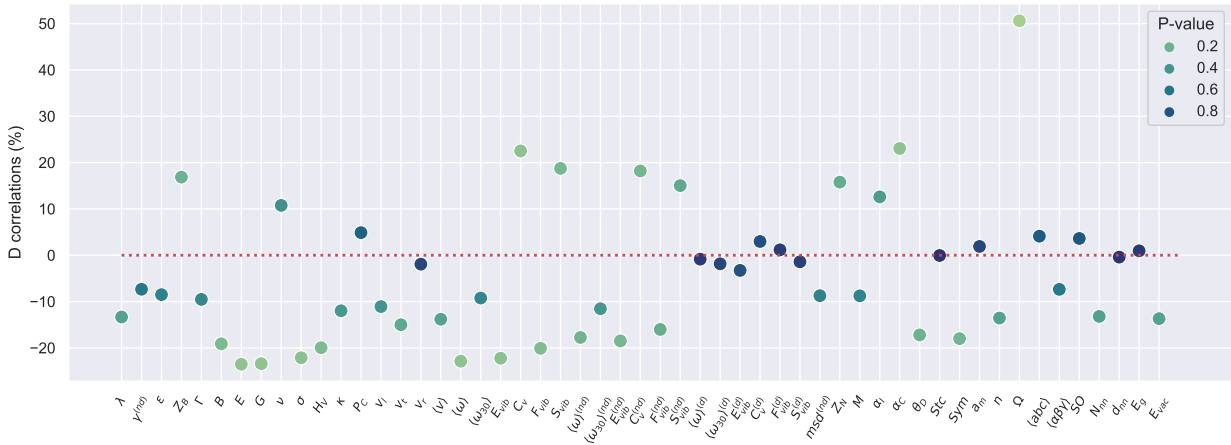
Supplementary Table III: Halide- and O-based SSE included in the DFT-AIMD database and details on the corresponding finite-temperature simulations. Ion diffusion coefficients, D , listed as “0” are $< 10^{-5} \text{ \AA}^2/\text{fs}$.



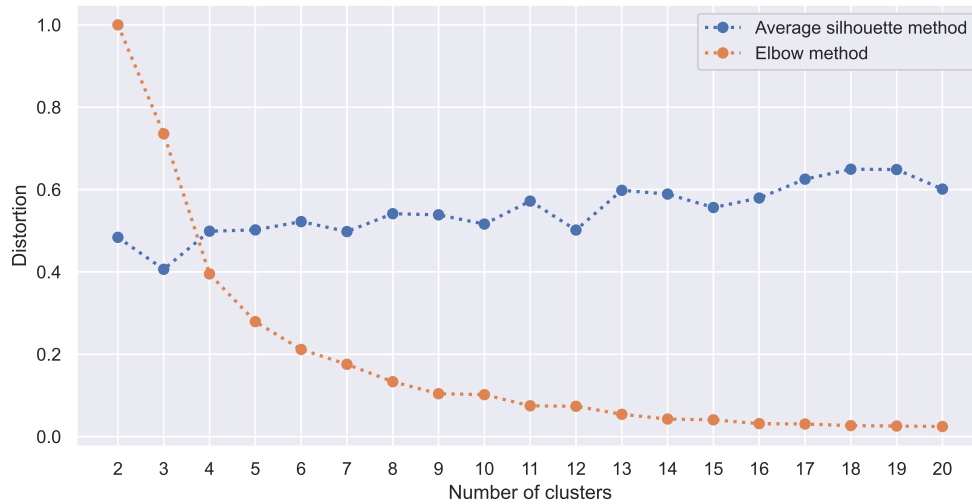
Supplementary Figure 1: Pearson correlogram calculated for all SSE (Top) and the corresponding p -value matrix (Bottom).



Supplementary Figure 2: Representation of the ion diffusion coefficient and other materials descriptors calculated at $T = 500 \pm 100$ K for all the materials contained in the SSE DFT-AIMD database. The diffusion coefficient is represented in logarithmic scale.

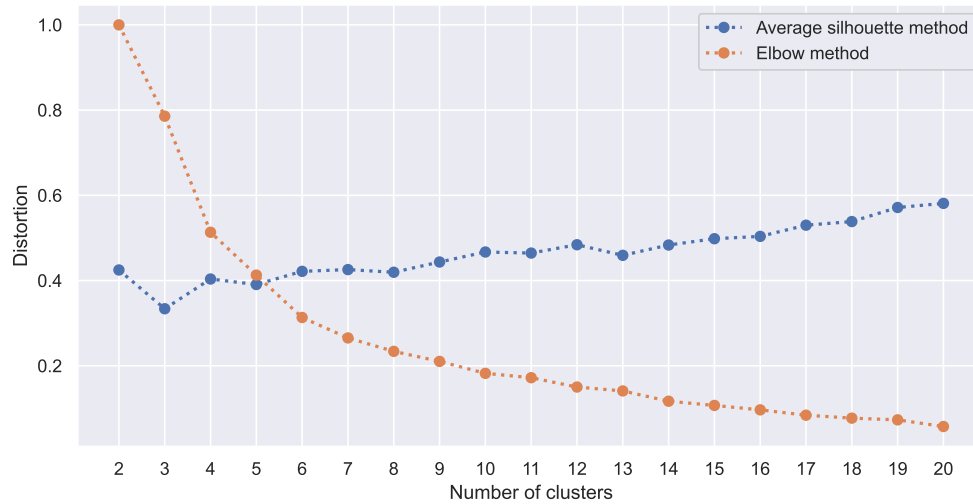


Supplementary Figure 3: Pearson D correlations estimated for all SSE.

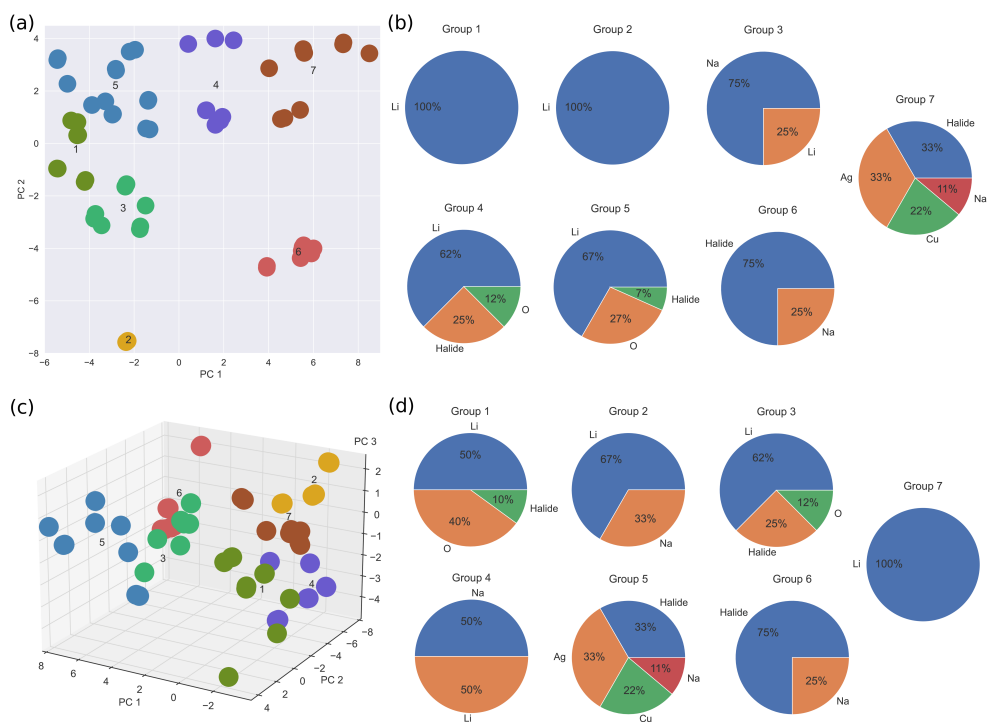


Supplementary Figure 4: K-means clustering analysis convergence for the two-dimensional PC1–PC2 representation. The elbow method is used to find the “elbow” point where adding additional data samples does not change cluster membership much.

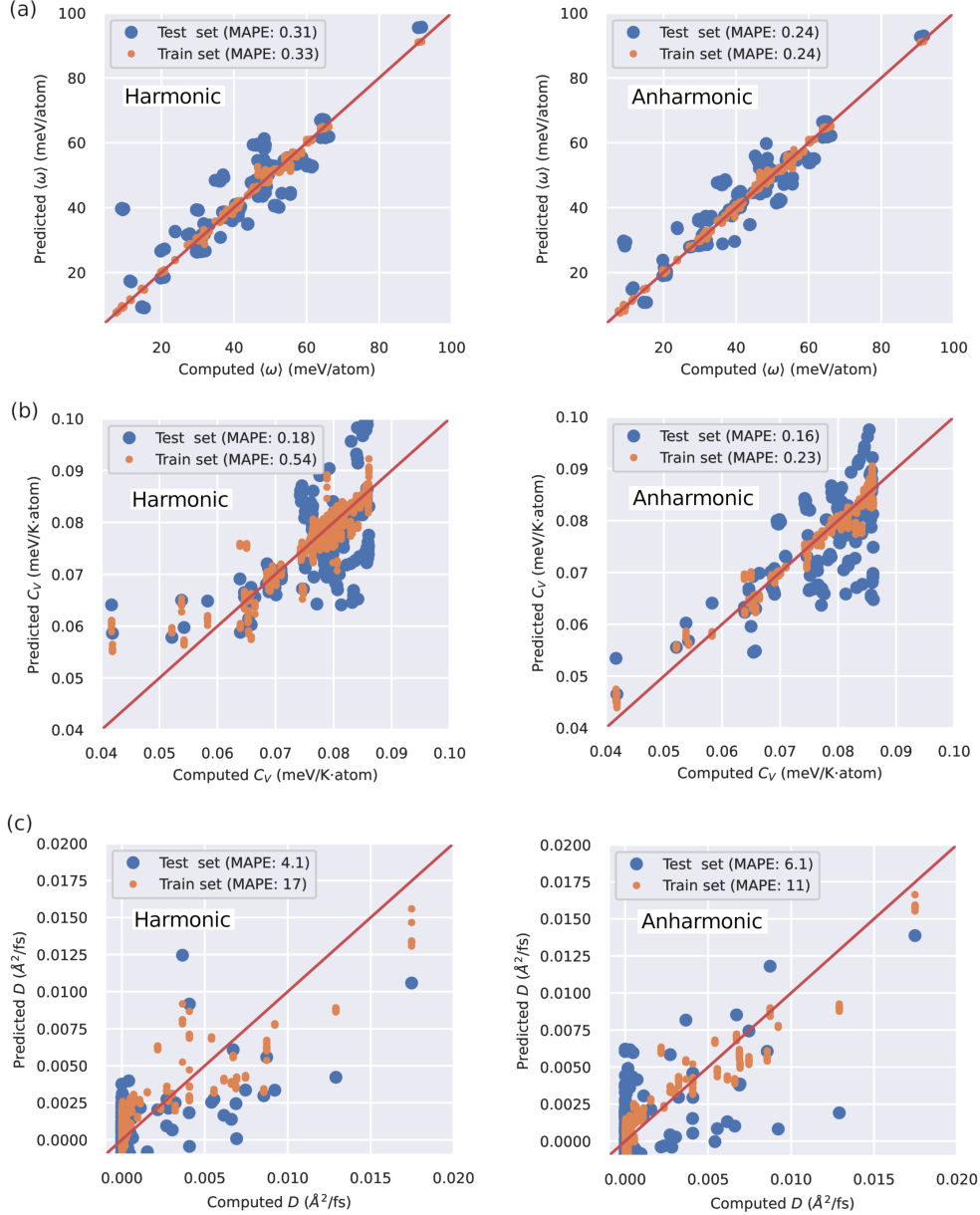
Silhouette score determines whether there are large gaps between each sample and all other samples within the same cluster or across different clusters. A subminimal number of 7 clusters was selected to account for the SSE database variance.



Supplementary Figure 5: K-means clustering analysis convergence for the three-dimensional PC1–PC2–PC3 representation. The elbow method is used to find the “elbow” point where adding additional data samples does not change cluster membership much. Silhouette score determines whether there are large gaps between each sample and all other samples within the same cluster or across different clusters. A subminimal number of 7 clusters was selected to account for the SSE database variance.



Supplementary Figure 6: K-means clustering analysis results obtained for the SSE DFT-AIMD database considering 7 clusters. (a)-(b) Classification of the analyzed materials in the orthogonal bidimensional space PC1-PC2. (c)-(d) Classification of the analyzed materials in the orthogonal tridimensional space PC1-PC2-PC3. The position of the cluster numbers in the PC plots coincide with the clusters centroids.



Supplementary Figure 7: Machine learning (ML) models trained in our DFT-AIMD database for prediction of different SSE T -dependent quantities, considering a kernel-ridge regression algorithm. The ML models were trained by considering and neglecting AIMD-based vibrational descriptors that explicitly incorporate anharmonic effects, labelled as “anharmonic” and “harmonic”, respectively. K-fold validation results obtained for (a) $\langle \omega \rangle$, (b) constant volume heat capacity and (c) ionic diffusion coefficient obtained from AIMD simulations. “MAPE” stands for the mean absolute percentage error of the ML predictions (Methods).