

Defect Engineering of Two-dimensional Nb-based Oxynitrides for Visible-light-driven Water Splitting to Produce H₂ and O₂

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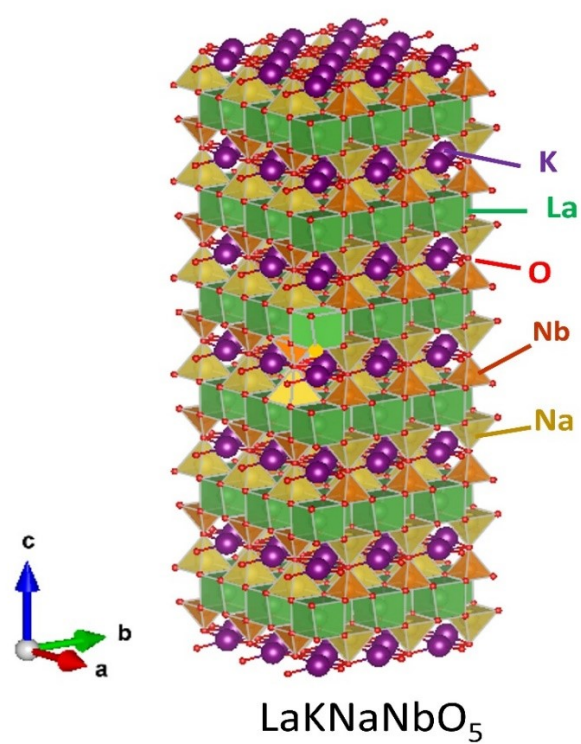


Figure S1. Crystal structure of LaKNaNbO₅.

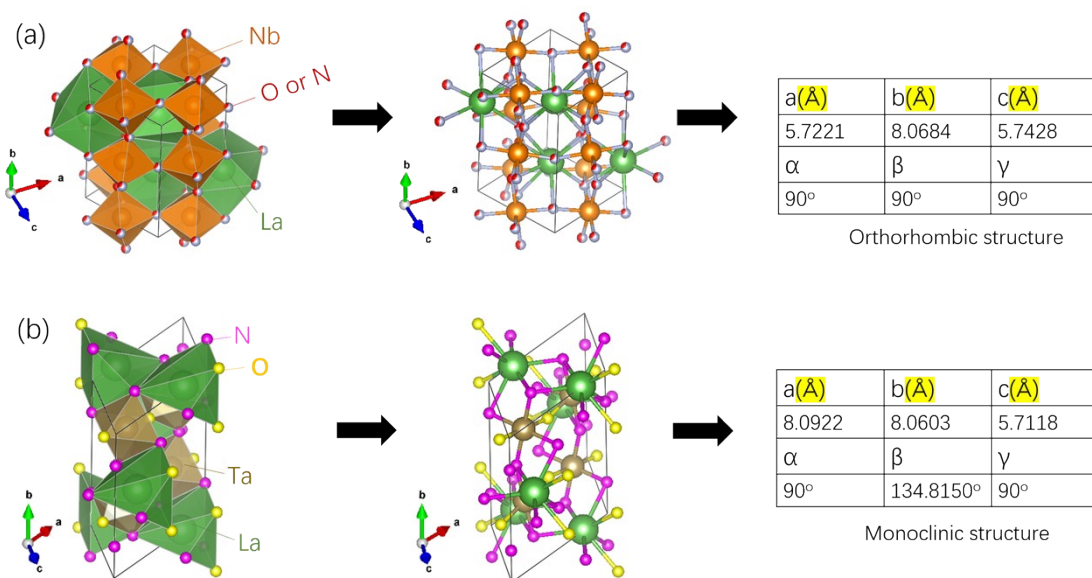


Figure S2. Crystal structure of LaNbON_2 and LaTaON_2 with polyhedral and ball and stick model and the corresponding unit cell parameters.

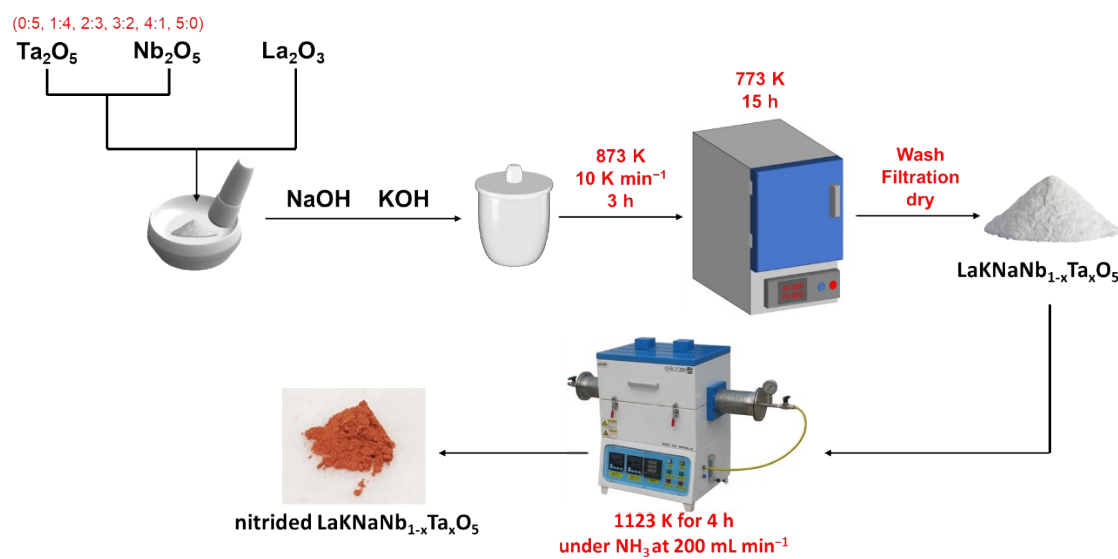


Figure S3. The synthesis procedure scheme of $\text{LaKNaNb}_{1-x}\text{Ta}_x\text{O}_5$ oxide precursors and nitrided $\text{LaKNaNb}_{1-x}\text{Ta}_x\text{O}_5$ ($x = 0, 0.2, 0.4, 0.6, 0.8, 1.0$).

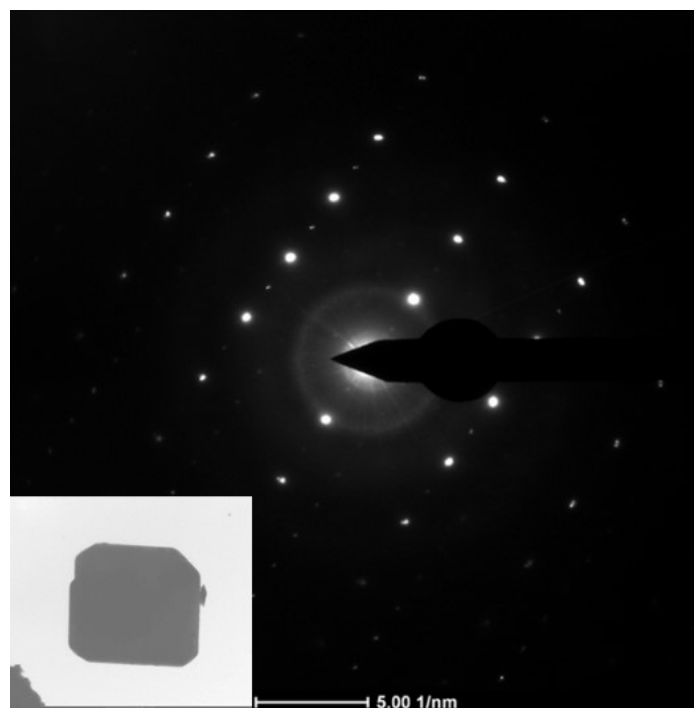


Figure S4. TEM image (inset) of the layered $\text{LaKNbNb}_{1-x}\text{Ta}_x\text{O}_5$ heated at 873 K for 3 h and maintained at 773 K for 15 and corresponding electron diffraction pattern.

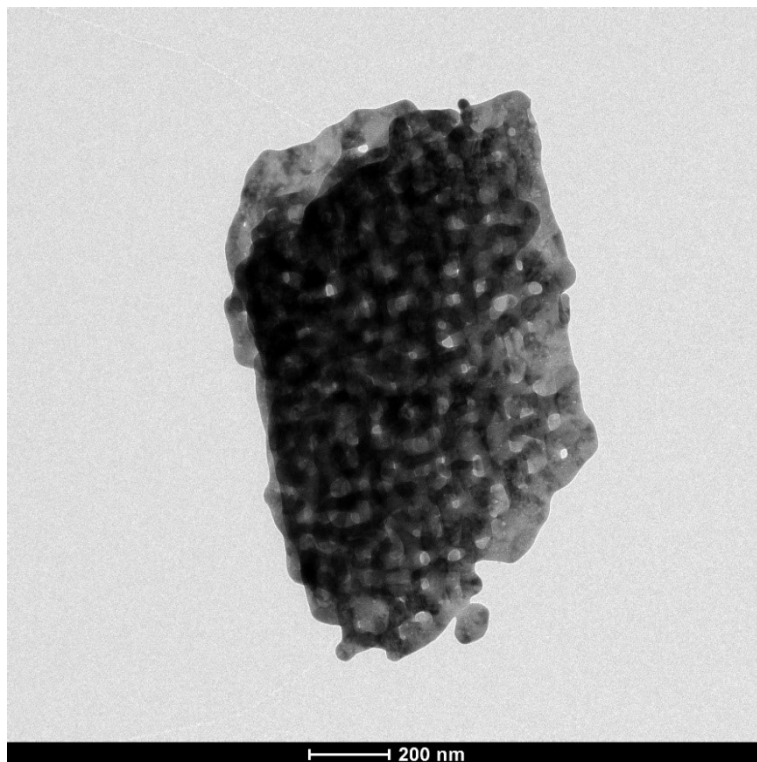


Figure S5. TEM image of the layered $\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ after nitridation at 1123 K for 4 h.

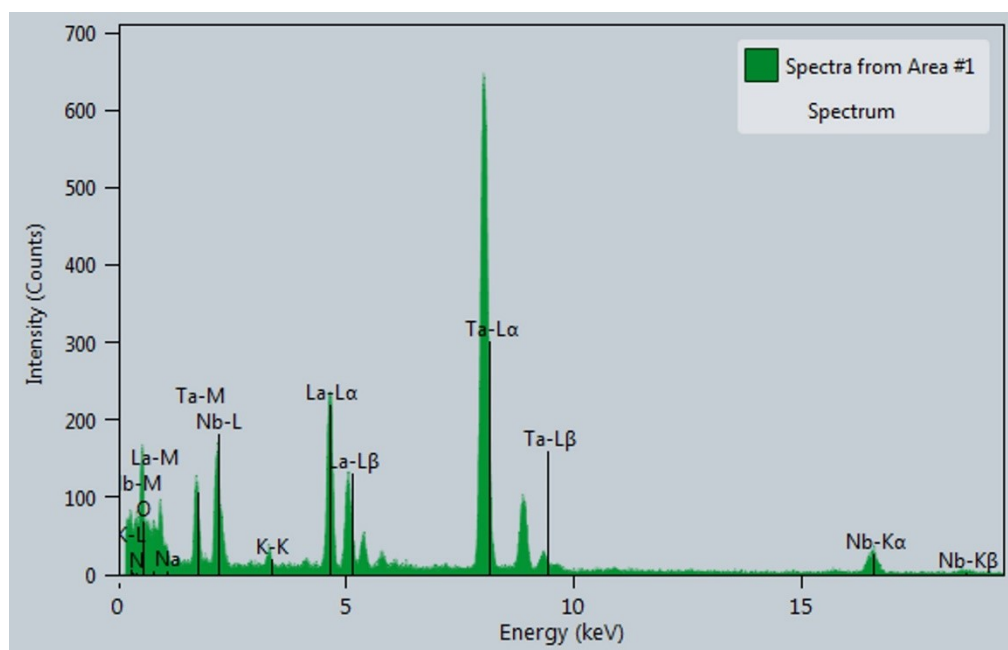


Figure S6. The EDS spectra of $\text{LaKNbNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ after nitridation at 1123 K for 4 h.

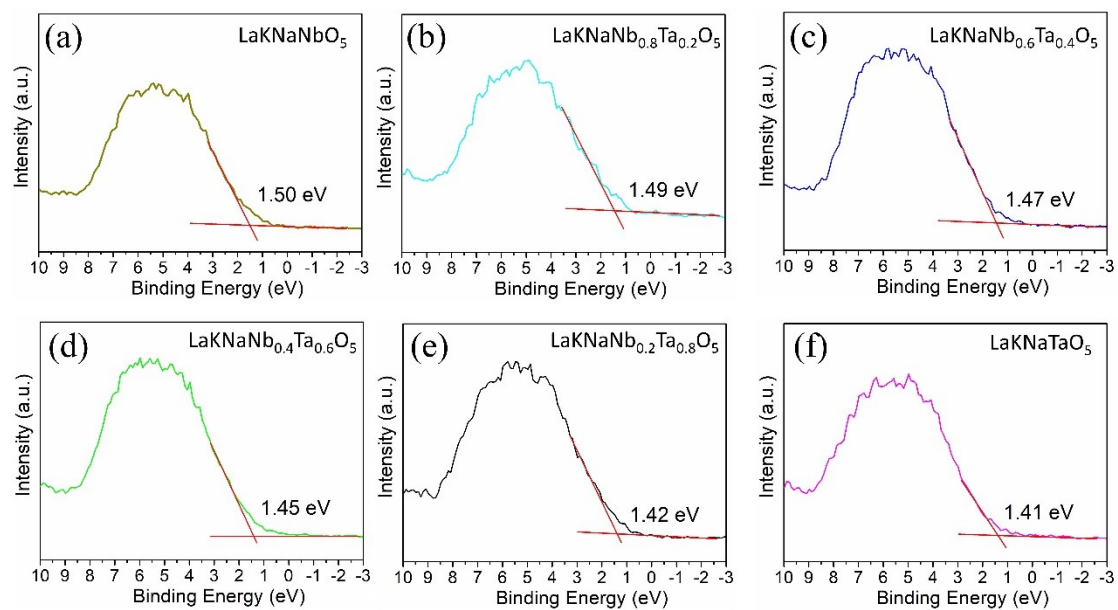


Figure S7. Valence band XPS spectra of (a) LaKNbNbO₅, (b) nitrified LaKNbNb_{0.8}Ta_{0.2}O₅, (c) nitrified LaKNbNb_{0.6}Ta_{0.4}O₅, (d) nitrified LaKNbNb_{0.4}Ta_{0.6}O₅, (e) nitrified LaKNbNb_{0.2}Ta_{0.8}O₅ and (f) nitrified LaKNbTaO₅.

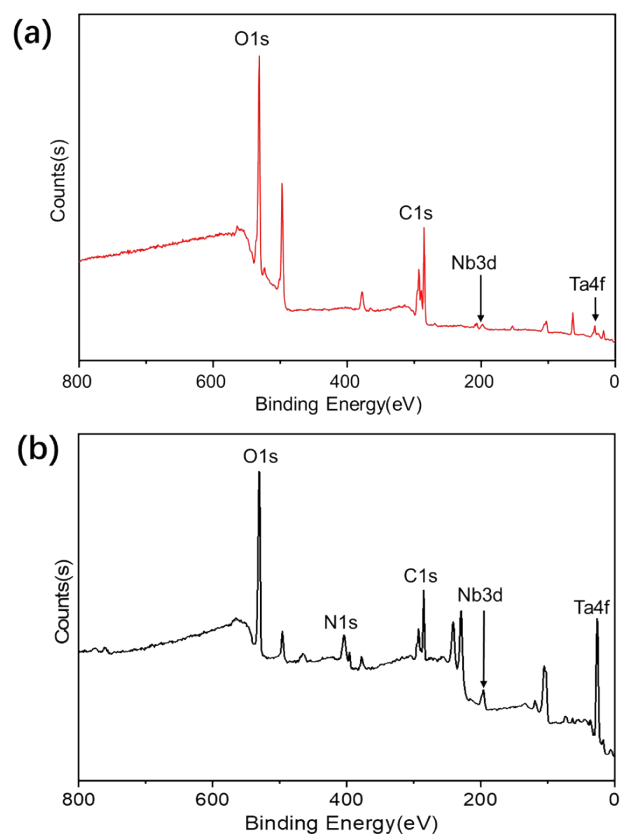


Figure S8. XPS survey spectra of (a) $\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ and $\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ nitrided at 1123 K for 4 h.

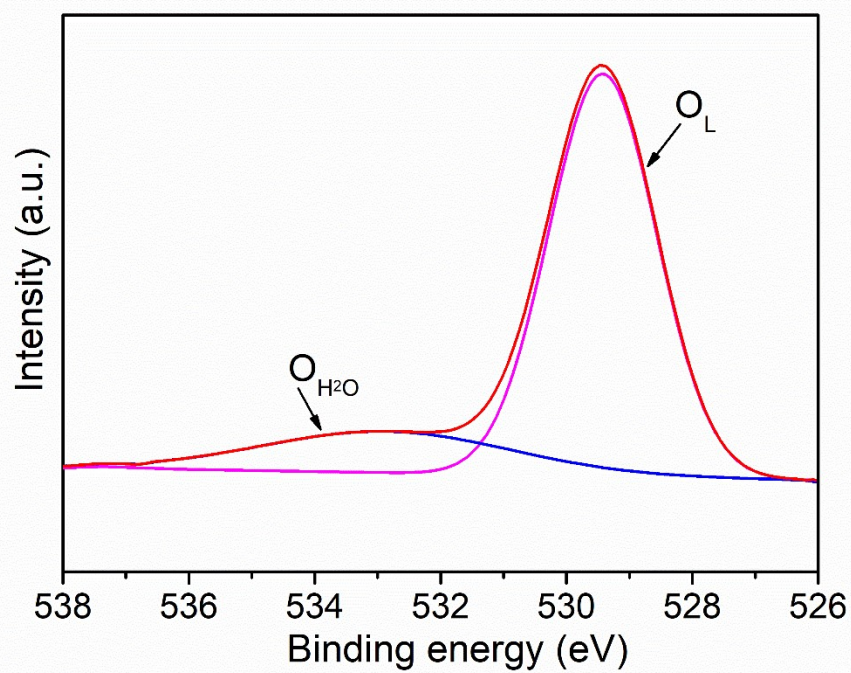


Figure S9. High-resolution XPS spectra of O 1s for LaKNaNbO₅

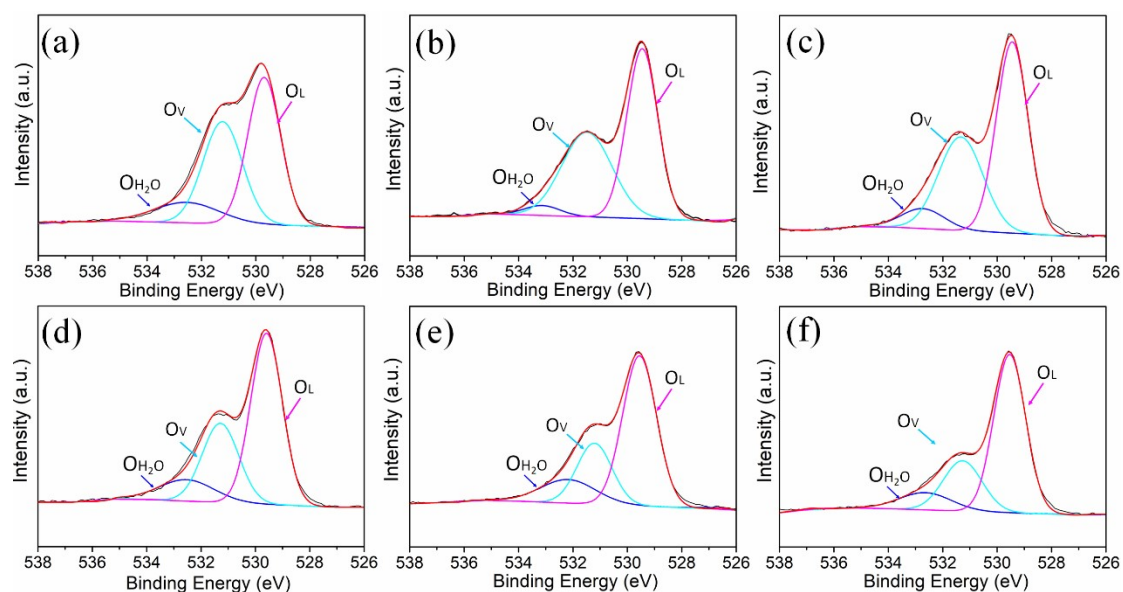


Figure S10. High-resolution XPS spectra of O 1s for the nitrided LaKNbO₅ (a), the nitrided LaKNb_{0.8}Ta_{0.2}O₅ (b), the nitrided LaKNb_{0.6}Ta_{0.4}O₅ (c), the nitrided LaKNb_{0.4}Ta_{0.6}O₅ (d), the nitrided LaKNb_{0.2}Ta_{0.8}O₅ (e) and the nitrided LaKNaTaO₅ (f) nitrided at 1123 K for 4 h.

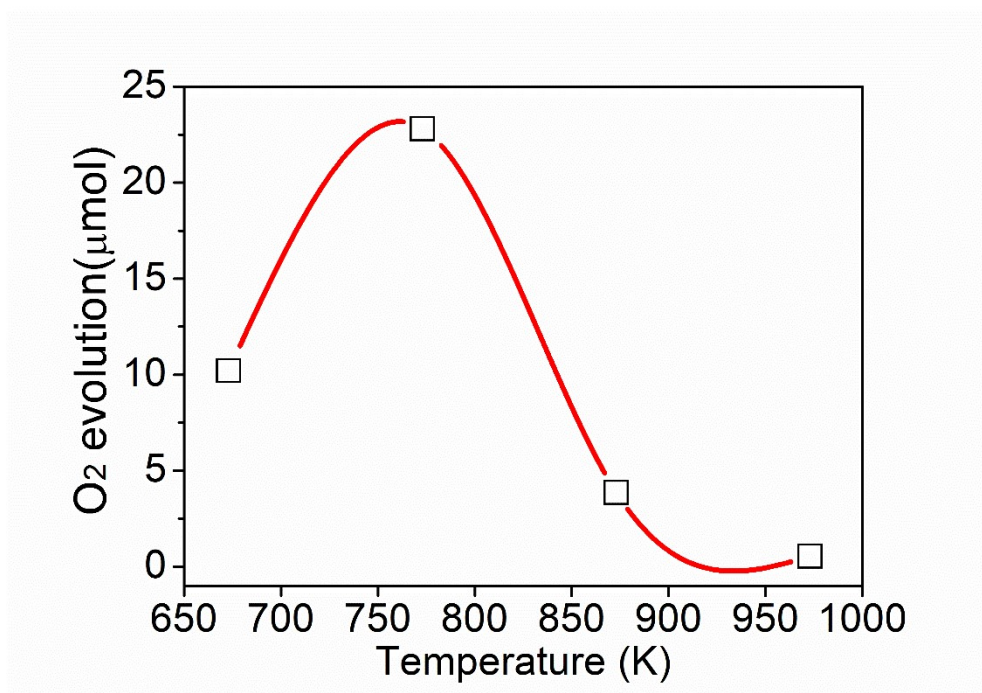


Figure S11. O₂ evolution rates for CoO_x modified the nitrided LaKNaNbO₅ treated at different temperature for 2 h.

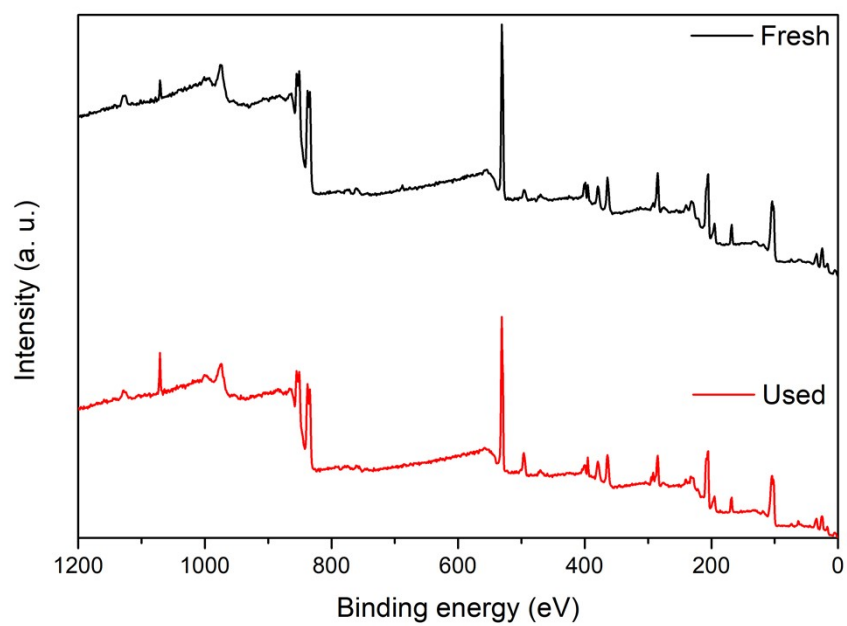


Figure S12. XPS survey of fresh and used nitrided $\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ after photoreaction.

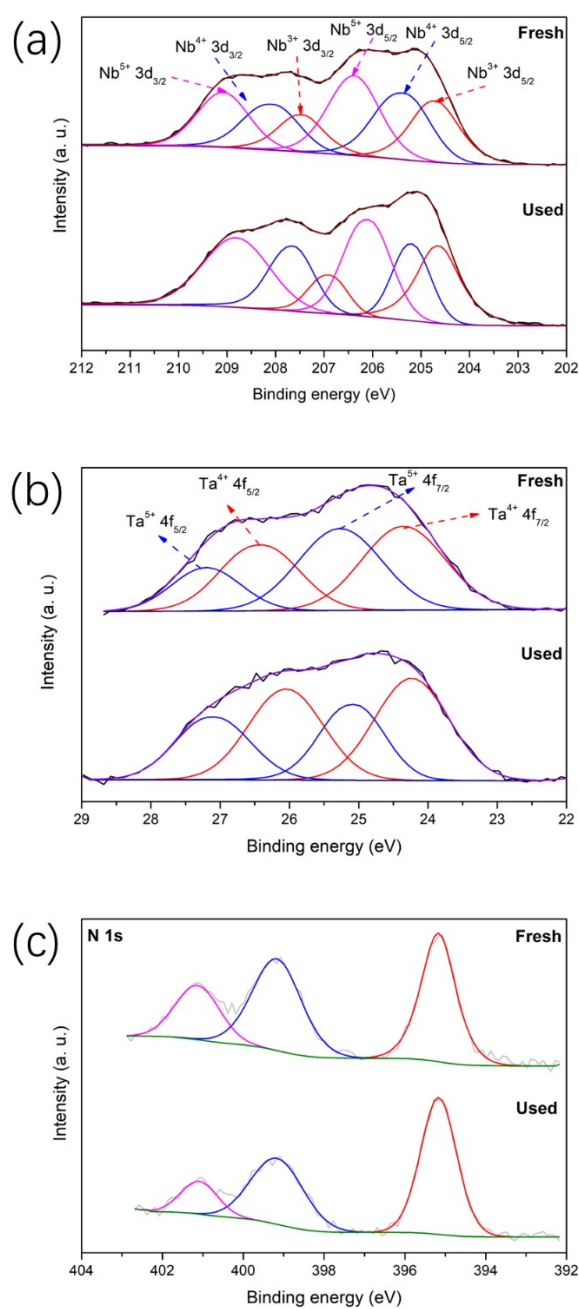


Figure S13. XPS Nb 3d (a), Ta 4f (b) and N 1s (c) of fresh and used nitrided $\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$ after photoreaction.

Table S1. The values of bandgap, VBE and CBE calculated from UV-vis DRS and VB-XPS spectra.

Samples	Bandgap (eV)	VBE (eV)	CBE (eV)
LaKNaNbO ₅	1.77	1.50	-0.27
LaKNaNb _{0.8} Ta _{0.2} O ₅	1.84	1.49	-0.35
LaKNaNb _{0.6} Ta _{0.4} O ₅	1.86	1.47	-0.39
LaKNaNb _{0.4} Ta _{0.6} O ₅	1.89	1.45	-0.44
LaKNaNb _{0.2} Ta _{0.8} O ₅	1.96	1.42	-0.54
LaKNaTaO ₅	2.12	1.45	-0.67

Table S2. Fractions of Nb species obtained from the XPS Nb 3d deconvolution regions for the nitrated $\text{LaKNaNb}_{1-x}\text{Ta}_x\text{O}_5$.

Samples	Nb 3d $_{3/2}$			Nb 3d $_{5/2}$		
	Nb ⁵⁺ 3d $_{3/2}$	Nb ⁴⁺ 3d $_{3/2}$	Nb ³⁺ 3d $_{3/2}$	Nb ⁵⁺ 3d $_{5/2}$	Nb ⁴⁺ 3d $_{5/2}$	Nb ³⁺ 3d $_{5/2}$
LaKNaNbO ₅	8.51%	9.80%	15.59%	9.43%	25.69%	30.98%
LaKNaNb _{0.8} Ta _{0.2} O ₅	4.37%	17.64%	12.62%	11.18%	24.67%	29.52%
LaKNaNb _{0.6} Ta _{0.4} O ₅	13.78%	11.23%	21.35%	11.13%	23.25%	19.25%
LaKNaNb _{0.4} Ta _{0.6} O ₅	15.41%	15.21%	11.37%	17.88%	20.92%	19.21%
LaKNaNb _{0.2} Ta _{0.8} O ₅	26.98%	12.48%	9.14%	24.41%	15.75%	11.24%
LaKNaTaO ₅	0	0	0	0	0	0

Table S3. Fractions of Ta species obtained from the XPS Ta 4f deconvolution regions for the nitrated $\text{LaKNaNb}_{1-x}\text{Ta}_x\text{O}_5$.

Samples	Ta 4f $_{5/2}$		Ta 4f $_{7/2}$	
	Ta ⁵⁺ 4f $_{5/2}$	Ta ⁴⁺ 4f $_{5/2}$	Ta ⁵⁺ 4f	Ta ⁴⁺ 4f $_{7/2}$
			$_{7/2}$	
LaKNaNbO ₅	0	0	0	0
LaKNaNb _{0.8} Ta _{0.2} O ₅	11.32%	32.58%	17.96%	38.14%
LaKNaNb _{0.6} Ta _{0.4} O ₅	12.39%	30.95%	24.33%	32.33%
LaKNaNb _{0.4} Ta _{0.6} O ₅	19.08%	27.76%	24.50%	28.66%
LaKNaNb _{0.2} Ta _{0.8} O ₅	23.56%	22.22%	28.71%	25.51%
LaKNaTaO ₅	33.39%	11.92%	40.76%	13.92%

Table S4. Fractions of O species obtained from the XPS O1s deconvolution regions for the nitrided $\text{LaKNaNb}_{1-x}\text{Ta}_x\text{O}_5$.

Samples	Area (O_L)	Area (O_V)	Area ($\text{O}_{\text{H}_2\text{O}}$)
LaKNaNbO_5	47.67%	43.27%	9.06%
$\text{LaKNaNb}_{0.8}\text{Ta}_{0.2}\text{O}_5$	53.45%	39.00%	7.55%
$\text{LaKNaNb}_{0.6}\text{Ta}_{0.4}\text{O}_5$	54.81%	37.48%	7.71%
$\text{LaKNaNb}_{0.4}\text{Ta}_{0.6}\text{O}_5$	57.07%	30.51%	12.42%
$\text{LaKNaNb}_{0.2}\text{Ta}_{0.8}\text{O}_5$	59.72%	25.03%	15.25%
LaKNaTaO_5	65.88%	20.75%	13.37%

Table S5. Photocatalytic H₂ and O₂ evolution rates over the nitrated LaKNaNb_{1-x}Ta_xO₅ at 1123 K for 4 h.

Samples	H ₂ evolution rate	O ₂ evolution rate
LaKNaNbO ₅	0.2635 umol/h	18.6485 umol/h
LaKNaNb _{0.8} Ta _{0.2} O ₅	1.4836 umol/h	22.8087 umol/h
LaKNaNb _{0.6} Ta _{0.4} O ₅	3.3684 umol/h	14.9002 umol/h
LaKNaNb _{0.4} Ta _{0.6} O ₅	4.7403 umol/h	5.9927 umol/h
LaKNaNb _{0.2} Ta _{0.8} O ₅	6.9226 umol/h	2.6307 umol/h
LaKNaTaO ₅	19.3713 umol/h	0.4457 umol/h

Table S6. The XPS peak contents of nitride LaKNaTa_{0.8}Nb_{0.2}O₅ before and

	Name	Peak BE	Height CPS	FWHM eV	Area (P) CPS.eV	Atomic %	Peak Type
Fresh	C 1s	284.8	25911.8	2.66	100441	61.41	Standard
	N 1s	394.91	16502.7	1.01	60060	23.58	Standard
	Nb 3d	205.59	39381.6	3.03	177463	10.97	Standard
	Ta 4f	24.83	12381.2	3.49	69693.1	4.03	Standard
	Name	Peak BE	Height CPS	FWHM eV	Area (P) CPS.eV	Atomic %	Peak Type
Used	C 1s	284.8	19785.4	2.56	89630	63.49	Standard
	N 1s	395.2	15287.4	1.07	46031.9	20.94	Standard
	Nb 3d	205.72	36628.6	2.71	163111	11.69	Standard
	Ta 4f	25.12	8925.04	3.61	58018.3	3.88	Standard

after 15 hours reaction.