

Electronic Supplementary Information

Three-dimensional flower-like Mn-Ni-Co-O microstructure as a high-performance electrocatalyst for methanol oxidation reaction

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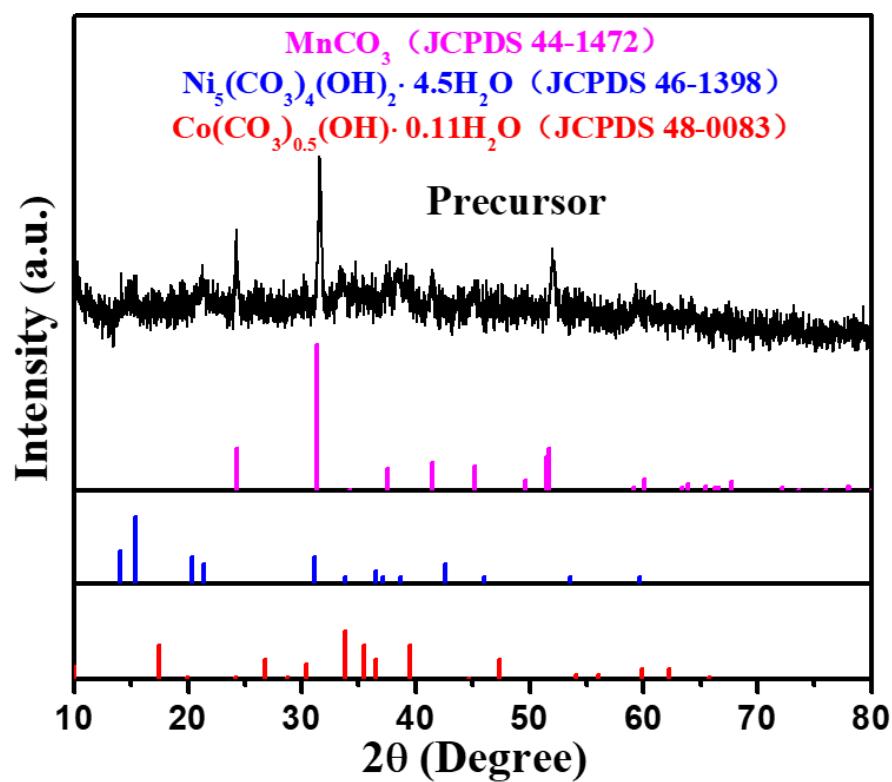


Fig. S1 XRD pattern of the Mn-Ni-Co precursor.

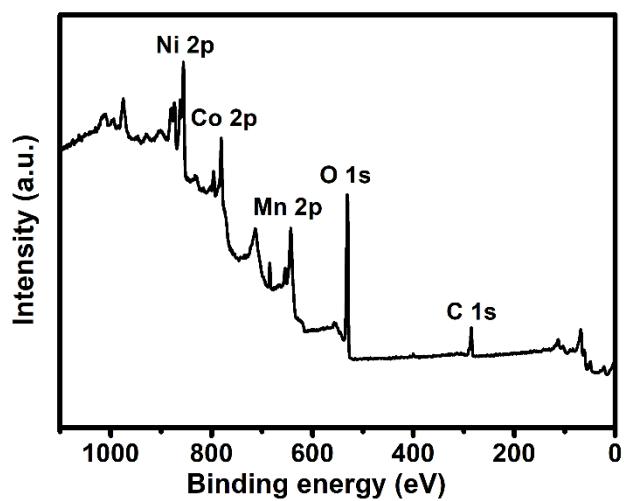


Fig. S2 XPS survey spectrum of the Mn-Ni-Co-O sample.

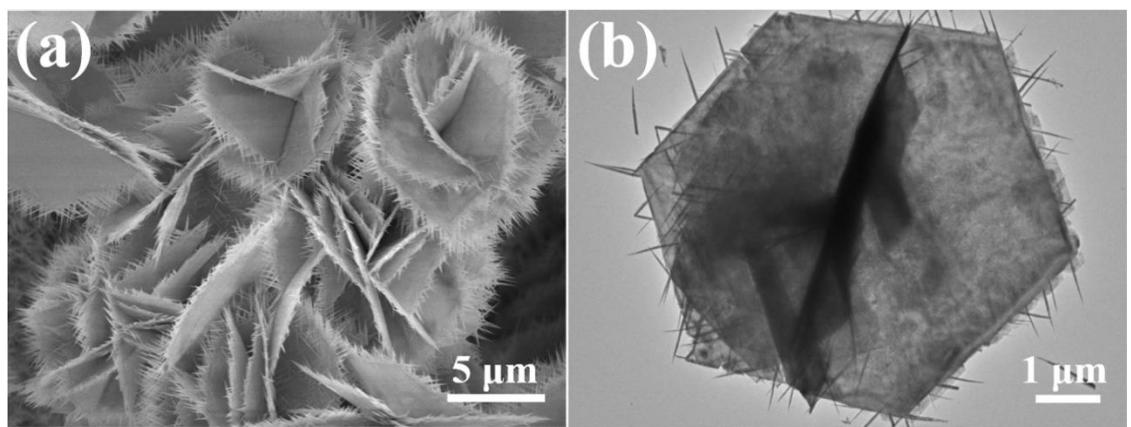


Fig. S3 (a) SEM image and (b) TEM image of the Mn-Ni-Co precursor.

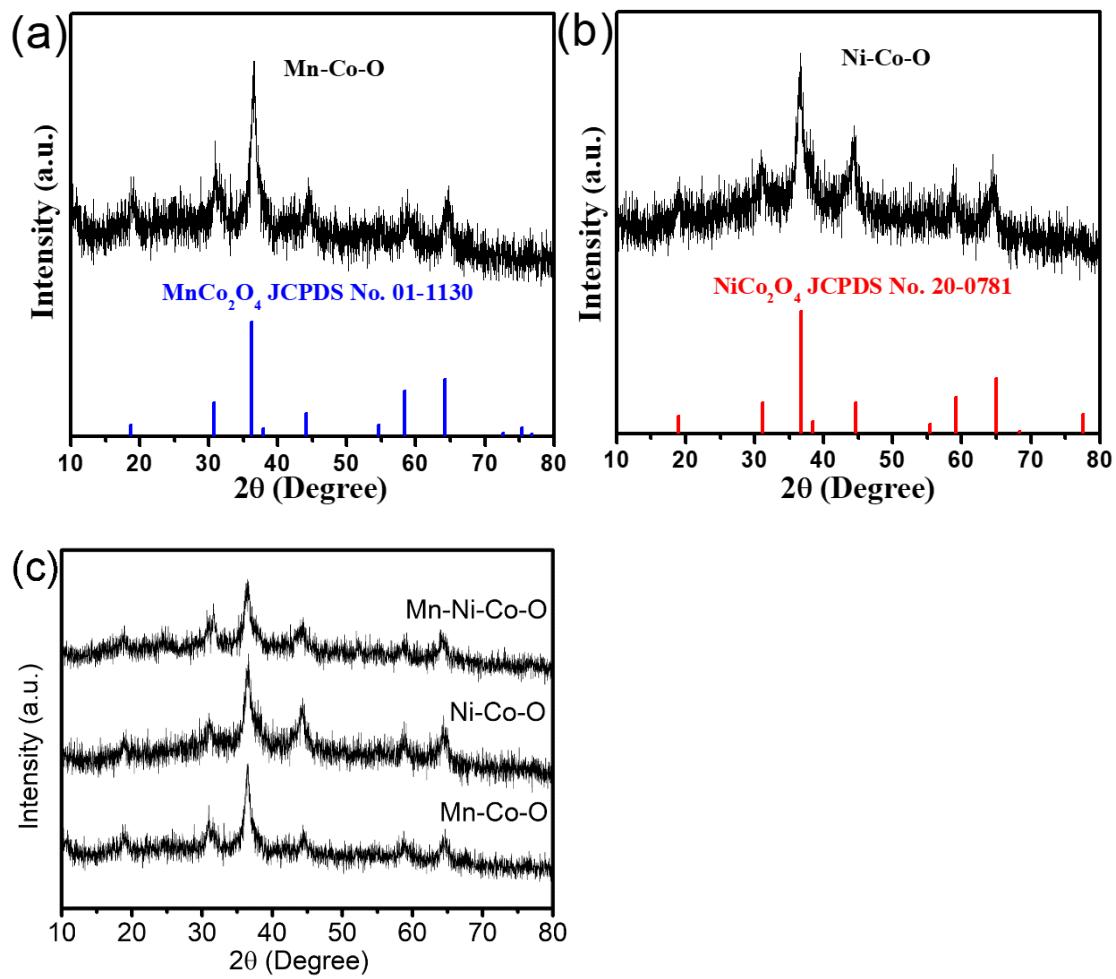


Fig. S4 (a) XRD pattern of the Mn-Co-O sample; (b) XRD pattern of the Ni-Co-O sample; (c) a comparison of the XRD patterns of Mn-Ni-Co-O, Ni-Co-O and Mn-Co-O samples.

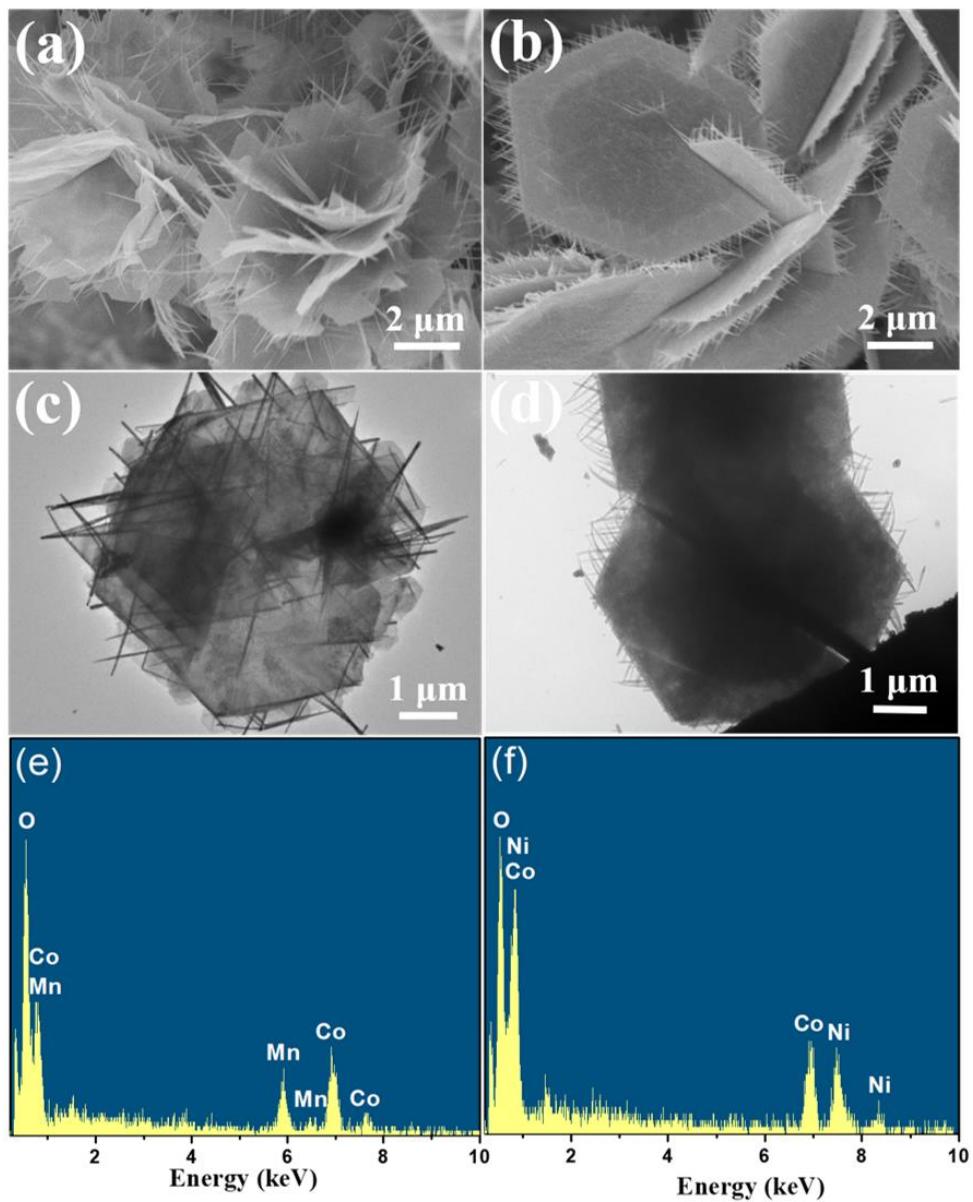


Fig. S5 (a, b) SEM images; (c, d) TEM images and (e, f) EDX spectra of Mn-Co-O and Ni-Co-O samples.

Table S1 Comparison of the electrocatalysts in this work with other reports in literature.

Catalyst	Potential window (V)	Current density	Scan rate (mV s ⁻¹)	Ref.
Mn-Co-O	0–0.6 (vs. Ag/AgCl)	33 mA mg ⁻¹	10	This work
Ni-Co-O	0–0.6 (vs. Ag/AgCl)	73 mA mg ⁻¹	10	This work
Mn-Ni-Co-O	0–0.6 (vs. Ag/AgCl)	113 mA mg ⁻¹	10	This work
MnCo ₂ O ₄	0–0.7 (vs. Ag/AgCl)	79.9 A g ⁻¹	10	[1]
MnCo ₂ O ₄	1.0–1.62 (vs. RHE)	96 A g ⁻¹	10	[2]
NiCo ₂ O ₄	0–0.6 (vs. Hg/HgO)	50 A g ⁻¹	10	[3]
Co ₃ O ₄ /NiCo ₂ O ₄	0–0.6 (vs. Hg/HgO)	140 mA cm ⁻²	10	[4]
NiCo ₂ O ₄ /Ni(OH) ₂	0–0.6 (vs. Ag/AgCl)	92.3 A g ⁻¹	10	[5]

References:

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- [4] L. Qian, S. Luo, L. Wu, X. Hu, W. Chen, W. Chen, X. Wang, *Appl. Surf. Sci.*, 2020, **503**, 144306.
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