

Electronic Supplementary Information

Synthesis of green magnetic biopolymer derived from *Oak fruit* hull tannin for efficient simultaneous adsorption of a mixture of Malachite Green and Sunset Yellow dyes from aqueous solutions

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Figures

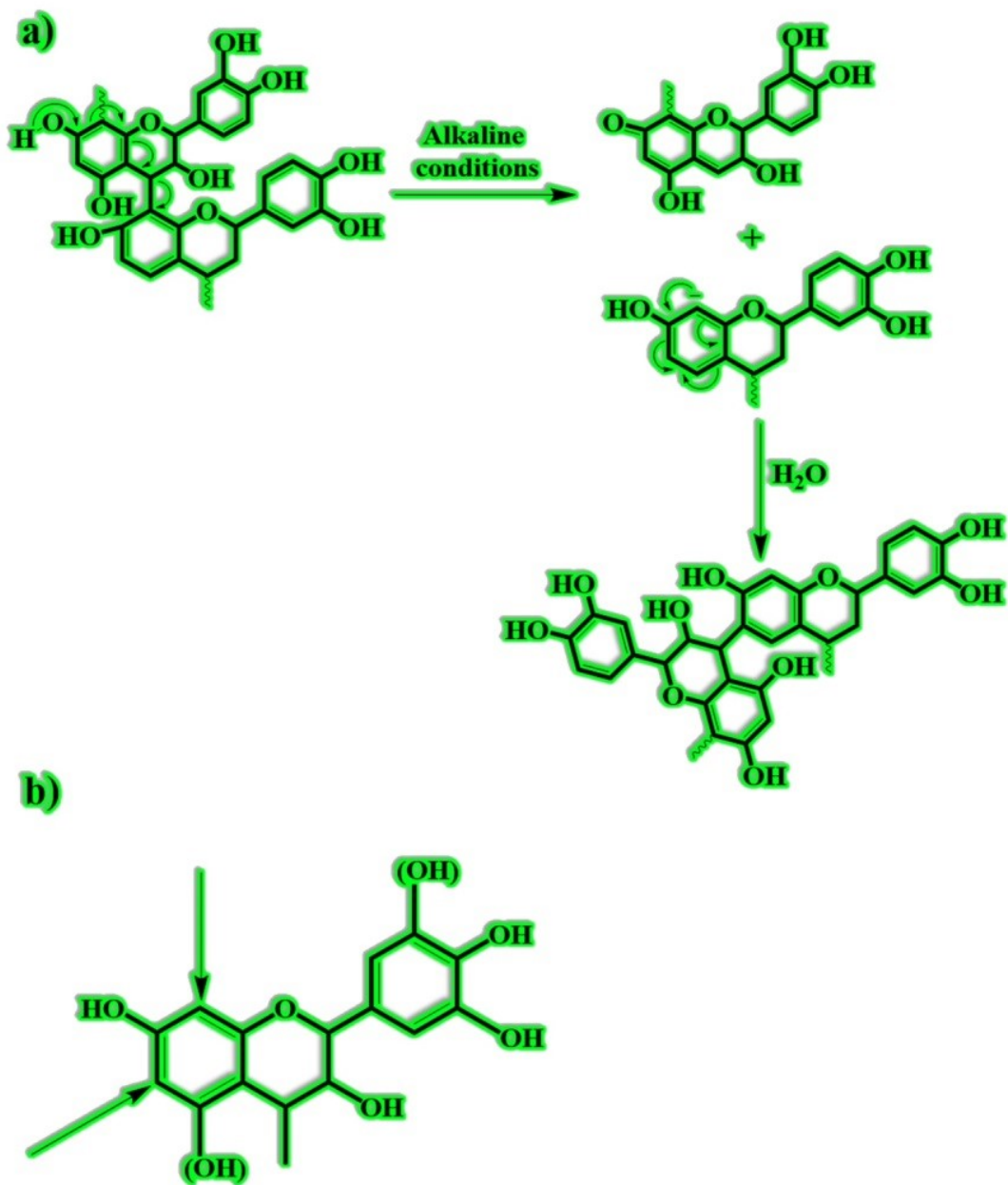
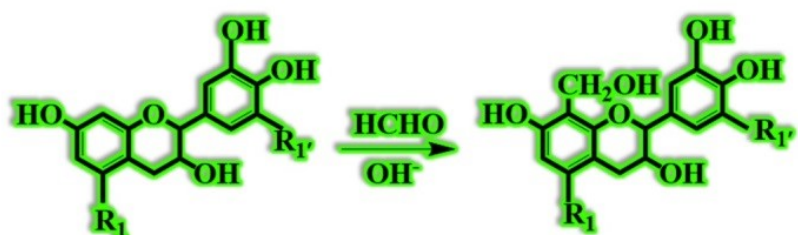
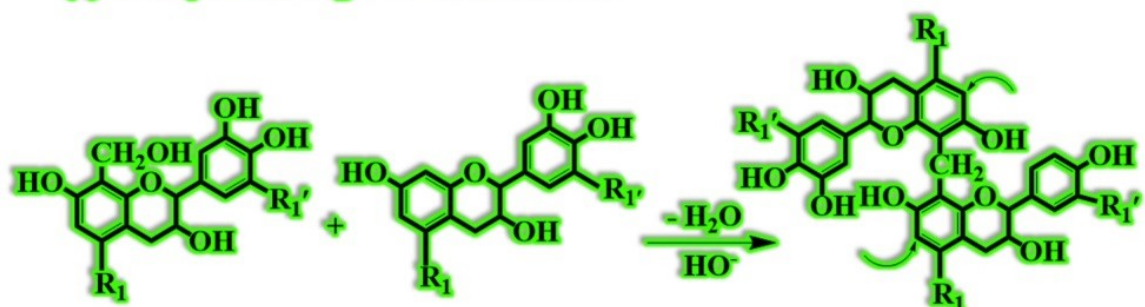


Fig. S1. Alkaline auto condensation (a) ¹ and probable flavonoid unit in a condensed tannin (b). ²

(1) Hydroxy methylation:



(2) Methylene bridge immobilization:



(3) Methylene ether bridge immobilization:

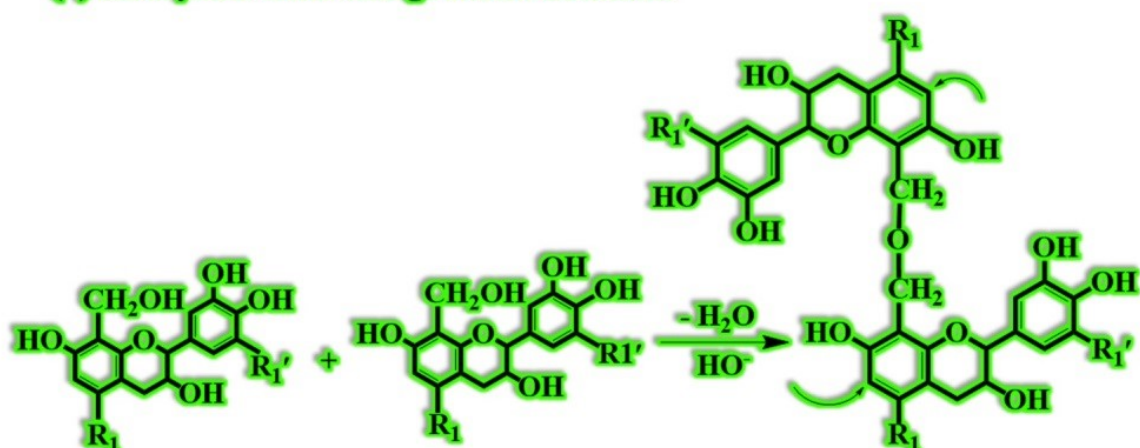


Fig. S2. Probable generic gelification mechanism between condensed tannin and formaldehyde.^{2,3}

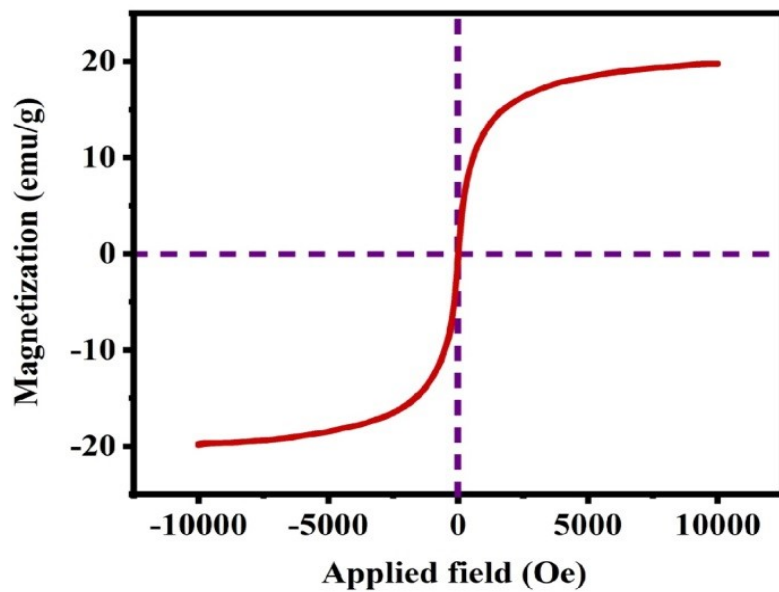


Fig. S3. Vibration sample magnetometry (VSM) data of $\text{Fe}_3\text{O}_4\text{-OT}$ recorded at room temperature.

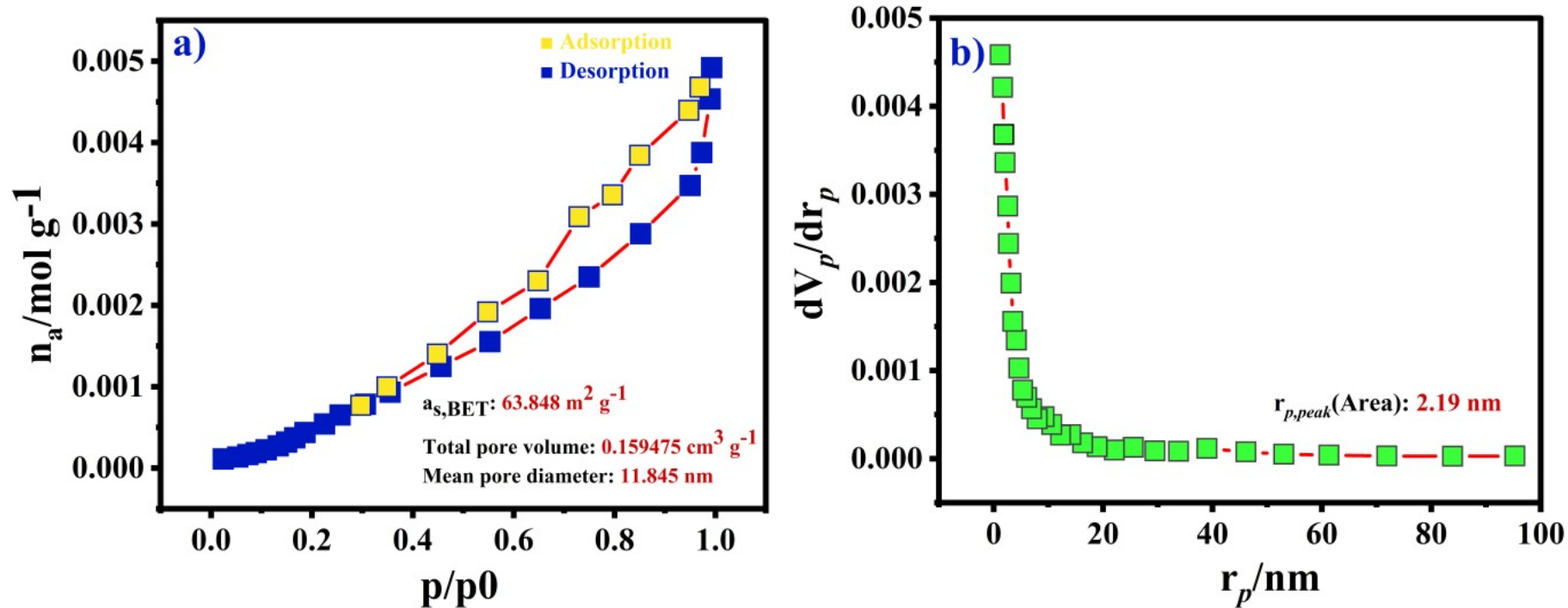


Fig. S4. Nitrogen adsorption and desorption diagram (a) and BJH diagram (b) for Fe₃O₄-OT adsorbent.

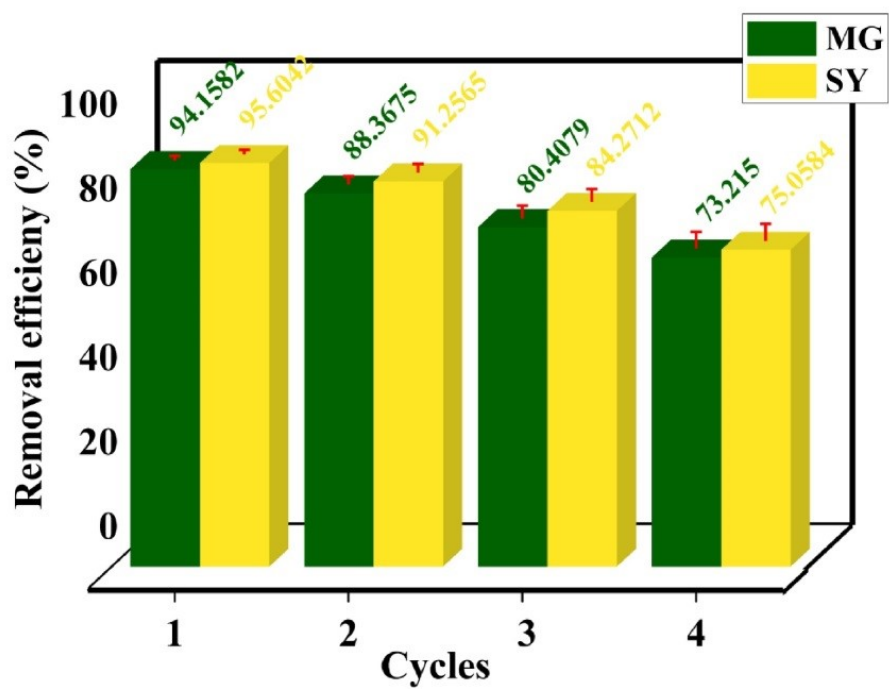


Fig. S5. The removal efficiency of MG and SY at 16 mg L⁻¹ initial concentration on Fe₃O₄-OT in different cycles.

Tables

Table S1. Physical properties and molecular structure of MG and SY.

Name of the commercial dye	Malachite Green (MG)	Sunset Yellow (SY)
Colour Index Name	Aniline green	Orange-yellow
λ_{max} (nm)	617	480
Molecular Weight (g mol ⁻¹)	364.911	452.400
Charge	(+)	(-)
Chemical Formula	C ₂₃ H ₂₅ N ₂	C ₁₆ H ₁₀ N ₂ Na ₂ O ₇ S ₂

Table S2. Assignment of groups in the FTIR spectrum of the Fe₃O₄-OT.

Band (cm⁻¹)	Functional group
3400	–OH stretching
2900	C–H stretching vibrations in methylene (-CH ₂ -)
2800	C–H stretching vibrations in methyl and methylene groups
1720	–C=O stretching vibrations
1627	C=C stretching in the aromatic ring
1450, 1500 and 1616	C–C skeletal vibrations
1100 and 1380	C–O stretching vibrations
1205	C–O–C symmetric stretch
1035	C–O ethers stretching
620 to 823	C–H deformation vibrations in the aromatic ring
455 and 764	=C–H bending vibrations
524	CCO and CCC ring deformation
578	Fe-O stretching vibrations of Fe ²⁺ and Fe ³⁺

Table S3. Variance analysis for response surface quadratic model for adsorption efficiency (%) of MG and SY onto Fe₃O₄-OT.

Source of variation	DF	R% MG				DF	R% SY			
		Sum of squares	Mean of squares	F-value	P-value		Sum of squares	Mean of squares	F-value	P-value
Model	20	370.3	18.52	23.22	< 0.0001	20	3321.8	221.5	41.79	< 0.0001
X ₁	1	0.442	0.442	0.554	0.4723	1	953.0	953	179.9	< 0.0001
X ₂	1	128.4	128.4	161.11	< 0.0001	1	91.31	91.31	17.23	0.0022
X ₃	1	16.21	16.21	20.33	0.0009	1	142.5	142.5	26.88	0.0004
X ₄	1	30.46	30.46	38.21	< 0.0001	1	235.4	235.4	44.42	< 0.0001
X ₅	1	4.00	4.00	5.02	0.0467	1	243.4	243.4	45.92	< 0.0001
X ₁ X ₂	1	15.79	15.79	19.81	0.0010	1	41.89	41.9	7.91	0.0211
X ₁ X ₃	1	3.73	3.73	4.68	0.0535	1	190.9	190.9	36.02	< 0.0001
X ₁ X ₄	1	20.18	20.18	25.31	0.0004	1	20.14	20.14	3.80	0.0893
X ₁ X ₅	1	7.50	7.50	9.41	0.0107	1	307.8	307.8	58.09	< 0.0001
X ₂ X ₃	1	0.505	0.505	0.633	0.4431	1	16.38	16.38	3.09	0.01211
X ₂ X ₄	1	0.105	0.105	0.132	0.7231	1	275.8	275.8	52.04	< 0.0001
X ₂ X ₅	1	15.37	15.37	19.28	0.0011	1	119.5	119.5	22.56	0.0008
X ₃ X ₄	1	0.393	0.393	0.493	0.4972	1	181.3	181.3	34.22	0.0002
X ₃ X ₅	1	2.25	2.25	2.82	0.1214	1	381.0	381.0	71.90	< 0.0001
X ₄ X ₅	1	17.57	17.57	22.03	0.0007	1	121.6	121.6	22.95	0.0008
X ₁ ²	1	1.67	1.67	2.09	0.1758	1	12.98	12.98	2.24	0.1629
X ₂ ²	1	81.8	81.8	102.6	< 0.0001	1	7.08	7.08	1.22	0.2928
X ₃ ²	1	24.4	24.4	30.58	0.0002	1	0.245	0.245	0.042	0.8410
X ₄ ²	1	4.34	4.34	5.44	0.0397	1	0.205	0.205	0.035	0.8545
X ₅ ²	1	3.68	3.68	4.62	0.0547	1	1.34	1.34	0.231	0.6403
Residual	11	8.77	0.797			11	63.81	5.80		
Lack of Fit	6	5.91	0.985	1.72	0.284	6	26.63	4.44	0.597	0.727
Pure Error	5	2.86	0.573			5	37.17	7.43		
Corr. Total	31	379.1				31	3406.6			

Table S4. Adsorption parameter of isotherm for the adsorption of MG and SY onto Fe₃O₄-OT (volume, 50 mL; adsorbent dose, 20 mg; contact time, 20 min; temperature, 25 °C; pH, 6.5).

Isotherm	Plot	Parameters	MG	SY
Langmuir $\frac{C_e}{q_e} = \frac{1}{Q_m k_L} + \frac{C_e}{Q_m}$	<i>C_e/q_e vs. C_e</i>	Q _m (mg g ⁻¹)	49.00	53.96
		K _L (L mg ⁻¹)	3.496	1.903
		R ²	0.997	0.993
		R _L =1/(1+(K _L ×C ₀))	0.009-0.054	0.017-0.095
Freundlich $\ln q_e = \ln K_F + \frac{1}{n} \ln C_e$	<i>ln q_e vs. ln C_e</i>	1/n	0.162	0.311
		K _F (L mg ⁻¹)	4.597	4.345
		R ²	0.967	0.929
Temkin $q_e = B_1 \ln K_T + B_1 \ln C_e$	<i>q_e vs. ln C_e</i>	B ₁	4.240	8.903
		K _T (L mg ⁻¹)	1.00	1.00
		R ²	0.892	0.968
Dubinin-Radushkevich $\ln q_e = \ln Q_S - k\varepsilon^2$	<i>ln q_e vs. ε²</i>	Q _s (mg g ⁻¹)	38.88	44.36
		β	-4.3E ⁻⁰⁹	-3.7E ⁻⁰⁸
		E (kJ mol ⁻¹)	10.763	3.682
		R ²	0.785	0.930

Table S5. kinetic parameters for MG and SY dyes adsorption onto Fe₃O₄-OT (volume, 50 mL; initial concentration for each dye, 16 mg L⁻¹; adsorbent dose, 20 mg; temperature, 25 °C; pH, 6.5).

Model	Plot	Parameters	MG	SY
First-order- kinetic $\ln(q_e - q_t) = \ln q_e - k_1 t$	$\ln (q_e - q_t)$ vs. t	k_1 (min ⁻¹)	0.103	0.116
		q_e (calc) (mg g ⁻¹)	33.64	28.89
		R ²	0.948	0.977
Pseudo-second-order- kinetic $\frac{t}{q_t} = \frac{1}{k_2 q_e^2} + \frac{t}{q_e}$	t/q_t vs. t	k_2 (min ⁻¹)	0.002	0.004
		q_e (calc) (mg g ⁻¹)	48.51	45.44
		R ²	0.991	0.996
		h (mg g ⁻¹ min ⁻¹)	6.503	9.734
Intraparticle diffusion $q_t = k_{diff} t^{1/2} + C$	q_t vs. $t^{1/2}$	K_{diff} (mg g ⁻¹ min ^{-1/2})	5.435	4.274
		C (mg g ⁻¹)	9.643	16.44
		R ²	0.943	0.933
Elovich $q_t = \frac{1}{\beta} \ln(t) + \frac{1}{\beta} \ln(\alpha\beta)$	q_t vs. $\ln t$	β (g mg ⁻¹)	0.094	0.120
		α (mg g ⁻¹ min ⁻¹)	14.37	32.69
		R ²	0.971	0.966
Experimental data		q_e (exp) (mg g ⁻¹)	39.21	39.53

Table S6. Thermodynamic parameters for the adsorption of MG and SY onto magnetic Fe₃O₄-OT (volume: 50 mL; initial dye concentration for each dye: 16 mg L⁻¹; adsorbent dose: 20 mg; contact time: 20 min; pH: 6.5).

	MG		SY	
T(k)	k _C	ΔG°(kJ mol ⁻¹)	k _C	ΔG°(kJ mol ⁻¹)
288	6.98	-4.65	10.53	-5.64
298	15.24	-6.75	22.91	-7.76
308	16.88	-7.24	26.54	-8.40
318	36.78	-9.53	53.32	-10.51
R ²	0.933		0.948	
ΔS°(j mol ⁻¹ k ⁻¹)	151.0		152.7	
ΔH°(kj mol ⁻¹)	38.72		38.19	

References

1. A. Arbenz and L. Avérous, *Green Chem.* 2015, **17**, 2626-2646.
2. J. Beltrán-Heredia, P. Palo, J. Sánchez-Martín, J. R. Domínguez and T. González, *Ind. Eng. Chem. Res.*, 2012, **51**, 50-57.
3. J. Sánchez-Martín, J. Beltrán-Heredia and C. Carmona-Murillo, *Ind. Crops Prod.*, 2011, **33**, 409-417.