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Dual Fluorometric Biosensor Based on Nanoceria Encapsulated Metal Organic Framework and Signal Amplification Strategy of Hybridization Chain Reaction for the Detection of Melamine and Pb²⁺ ions in Food Samples

Arunjegan Amalraj and Panneerselvam Perumal*

Department of Chemistry, SRM Institute of Science and Technology, Kattankulathur, 603 203, Tamil Nadu, India.

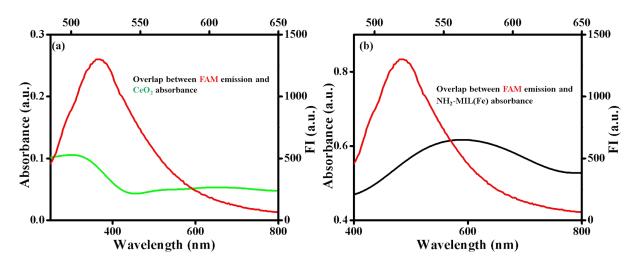


Figure S1. (a) UV-Vis absorption spectra for CeO₂ (green) (b) NH₂-MIL(Fe) (black) and FAM emission spectra (red).

The selectivity of the proposed Pb^{2+} sensor was studied with some other heavy metal ions. The effect of Zn^{2+} , Hg^{2+} , Cu^{2+} , and Fe^{3+} ions were investigated by the FAM fluorescence emission. As shown in **Figure. S2a**, the FAM emission of the reaction solution having each of the selected metal ions was constant except the Hg^{2+} ions shows a slight decreases in the FAM emission. The reason for the interference of Hg^{2+} ions might be that Hg^{2+} ions could bind to melamine. Since the real milk samples does not contains significant amount of Hg^{2+} ions [1], the interference of Hg^{2+} ions can be avoided in a fluorometric assay for Pb^{2+} ions used in milk samples. Furthermore, herein, we were able to avoid Hg^{2+} ion interference by using PDCA reagent also. PDCA was added to the detection solution prior to analysis to mask the Hg^{2+} ions significantly in order to achieve the unique detection. PDCA forms considerably more stable complexes with Hg^{2+} ions (log $\beta 2 = 20.28$). As shown in **Figure. S2b**, the FAM signal showed

a dramatic decrease with increasing Pb^{2+} ions concentrations in presence of PDCA while the introduction of Hg^{2+} and other ions had no influence on the FAM emission [2, 3].

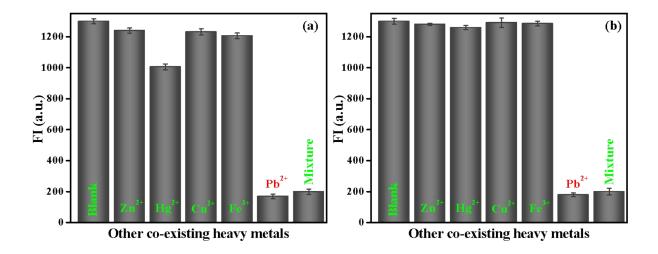


Figure S2. (a) Selectivity study for Pb²⁺ with coexisting other heavy metals (b) selectivity study for Pb²⁺ with coexisting other heavy metals and masking agent PDCA.

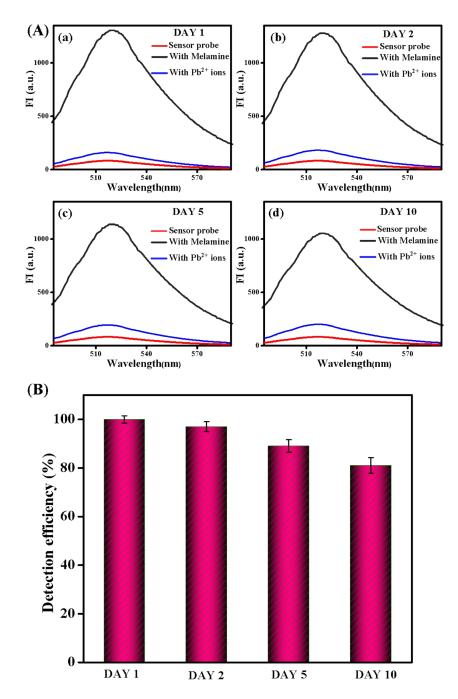


Figure S3. (A) a-d FAM fluorescence response for melamine and Pb²⁺ sensor (B) sensor detection efficiency at different days.

REFFERENCE

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