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Supplementary Material for

Fe-Ni-Co trimetallic oxides hierarchical nanosphere as highperformance bifunctional Electrocatalysts for Water electrolysis

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Fig. S2 The SEM image of NiCo precursor nanosphere.



Fig. S3 The half-quantitative SEM energy dispersive X-ray spectrum (SEM-EDX) of FeNiCo-15.



Fig. S4 a) Ni 2p, b) Co 2p XPS spectra of NiCo precursor.



Fig. S5 The TEM images of a) NiCo, b) FeNiCo-10, c) FeNiCo-15 and d) FeNiCo-20.



Fig. S6 The cyclic voltammograms curves of catalytic **a**) NiCo, **b**) FeNiCo-10, **c**) FeNiCo-15 and **d**) FeNiCo-20 toward to OER.



Fig. S7 Cdl for FeNiCo-5, FeNiCo-10, FeNiCo-15 and FeNiCo-20 toward to OER.



Fig. S8 a) The TOF curve of FeNiCo-15, FeNiCo-20, FeNiCo-10 and FeNiCo-5, b) TOF at 350 mV toward to OER.



Fig. S9 The cyclic voltammograms curves of catalytic a) NiCo, b) FeNiCo-10, c)

FeNiCo-15 and d) FeNiCo-20 toward to HER.



Fig. S10 Cdl for FeNiCo-5, FeNiCo-10, FeNiCo-15 and FeNiCo-20 toward to HER.



Fig. S11 a) The TOF curve of FeNiCo-15, FeNiCo-20, FeNiCo-10 and FeNiCo-5, **b)** TOF at 350 mV toward to HER.



Fig. S12 SEM images of FeNiCo-15 after electrochemical test.



Fig. S13 The XPS spectrum of Fe³⁺-NiCo₂O₄ nanospheres catalyst after OER

	NiCo	FeNiCo-5	FeNiCo-10	FeNiCo-15	FeNiCo-20	IrO ₂
Atom ratio	1/1	Fe:Co:Ni=	Fe:Co:Ni=	Fe:Co:Ni=	Fe:Co:Ni=	
Overpotential at 10 mA cm ⁻ 2	373 mV	303 mV	263 mV	204 mV	216 mV	271 mV
anodic peak position	1.41 V vs. RHE		-			
Rct	$1.5 \ \Omega \ cm^2$	1.4Ω cm ²	1.4Ω cm ²	1.2Ω cm ²	$1.3 \ \Omega \ cm^2$	
Cdl	3.62 mF cm ⁻²	4.39 mF cm ⁻²	7.21 mF cm ⁻²	21.71 mF cm ⁻²	16.42 mF cm ⁻²	
TOF(350 mV)		2.776 s ⁻¹	5.568 s ⁻¹	10.31 s ⁻¹	8.461 s ⁻¹	

Table S1. The detailed parameters of NiCo, FeNiCo-5, FeNiCo-10, FeNiCo-15, FeNiCo-20 and IrO₂ toward the OER.

Table S2. The detailed parameters of NiCo, FeNiCo-5, FeNiCo-10, FeNiCo-15, FeNiCo-20 and Pt/C toward the HER.

	NiCo	FeNiCo-5	FeNiCo-10	FeNiCo-15	FeNiCo-20	Pt/C
Overpotential						
at 10 mA cm ⁻	306 mV	289 mV	262 mV	178 mV	205 mV	16 mV
2						
anodic peak	1.41 V vs.					
position	RHE					
Rct	$1.7 \ \Omega \ cm^2$	$1.6 \ \Omega \ cm^2$	$1.4 \ \Omega \ cm^2$	$1.1 \ \Omega \ cm^2$	$1.2 \ \Omega \ cm^2$	
Cdl	4.04 mF	6.17 mF	8.85 mF	17.72 mF	12.63 mF	
	cm ⁻²					
TOF(350		1.7(21	6.246 -1	0.774	0.704	
mV)		1./62 s ⁻¹	6.246 s ⁻¹	9.//4 s ⁻¹	8./24 s ⁻¹	

Table S3. The comparisons of the catalytic activities of OER and HER on FeNiCo-15 with the recently reported catalysts in 1.0 M KOH media.

Catalysts	Overpotential at 10 mA cm ⁻² for OER(mV vs RHE)	Overpotential at 10 mA cm ⁻² for HER(mV vs RHE)	Electrolyte concentration (pH)	Ref.
FeNiCo-15	204	178	14	This
NiCo	373	306	14	work
Ni/NiO@G-SH	270		14	7
Ni/Ni(OH) ₂	310	168	14	8
1-D CoHCF /CFP	420		14	11
NiC _{0.2} NS/Ni/CF	228	121	14	12
NCO-0.1	297	143	14	14
NiFe-PBA	263	169	14	15
NiCo ₂ O ₄	420		14	16
FeP		69	0	22
Fe-NiMoO ₄	217	217	14	25
Ni-Fe-S	223	115	14	26
CoM-P-3DHFLMs	292		14	33
Ni ₄ Cu ₂ @C	280	137	14	34
ECT-S-Co _{0.37} Ni _{0.26} Fe _{0.37} O	232		14	37
FeNi ₃ N-Ni ₃ S	230	181	14	38
Co@NCNTAs	280		14	39
Co–Fe–P–Se	270		14	41