

Supporting Information for

Carbon coated Fe₃O₄ nanoparticles in-situ grown on 3D cross-linked carbon nanosheets as anodic materials for high capacity lithium and sodium ion batteries

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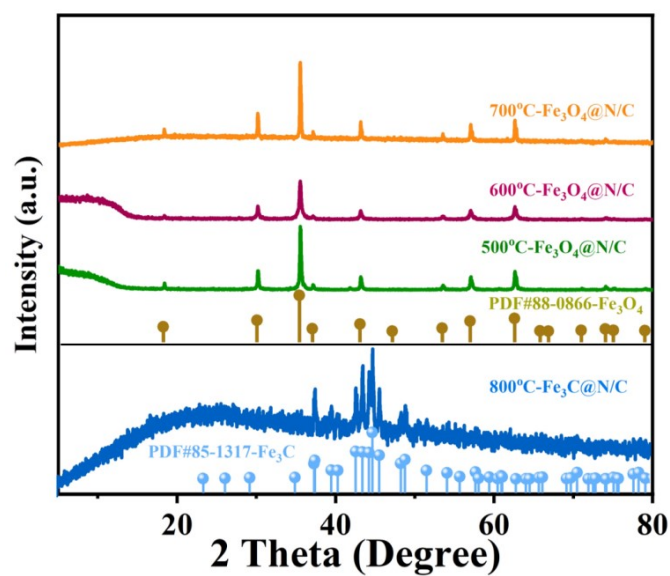


Fig. S1. XRD patterns of samples at different calcination temperatures.

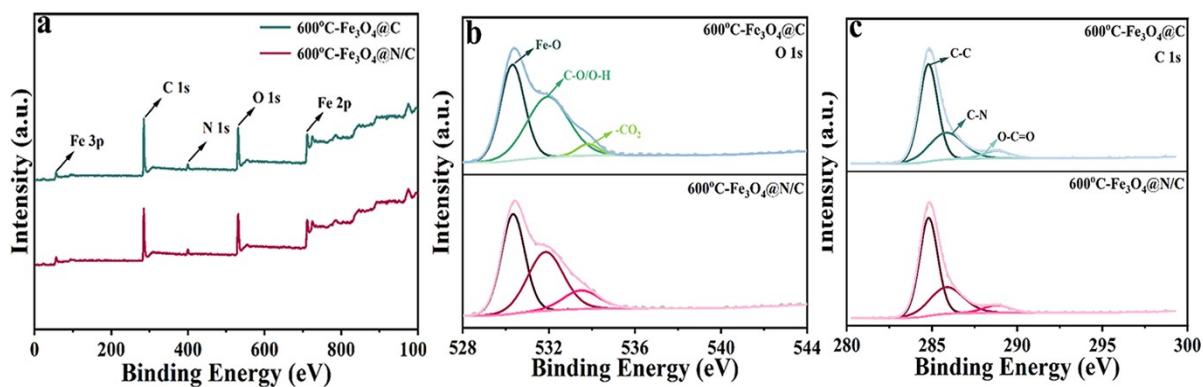


Fig. S2. (a) XPS survey and high-resolution XPS spectra of (b) C 1s and (c) O 1s of 600°C- $\text{Fe}_3\text{O}_4@\text{N/C}$ and 600°C- $\text{Fe}_3\text{O}_4@\text{C}$.

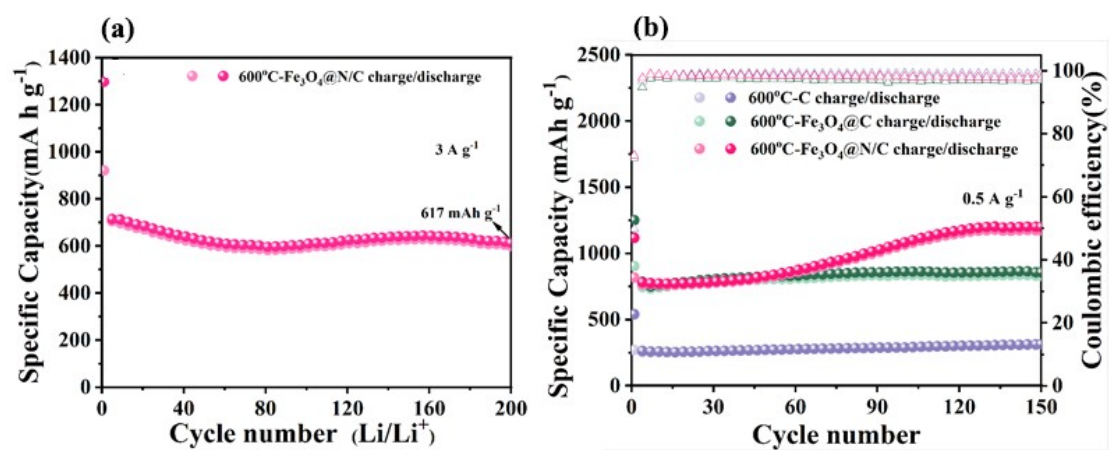


Fig. S3. (a) Cycling performance of the 600°C-Fe₃O₄@N/C at current densities of 3 A g⁻¹; (b) Cycling performance of the 600°C-C, 600°C-Fe₃O₄@C, and 600°C-Fe₃O₄@N/C at 0.5 A g⁻¹.

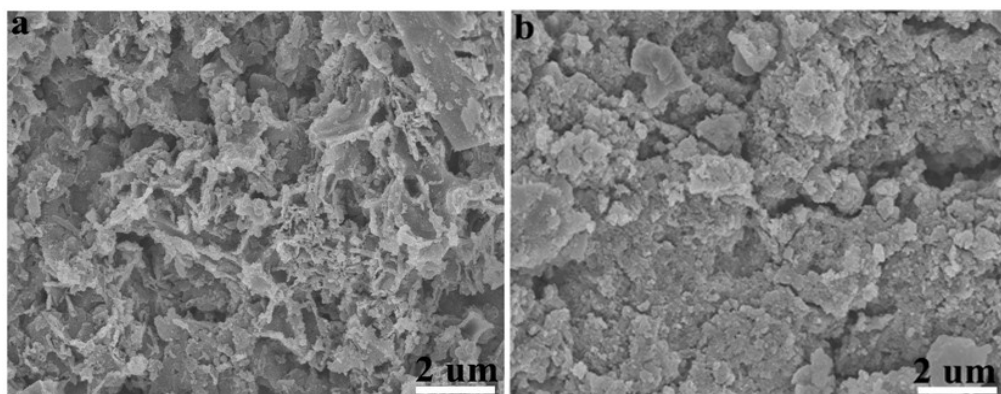


Fig. S4. SEM images of samples after cycles: (a) 600°C-Fe₃O₄@N/C; (b) 600°C-Fe₃O₄@C.

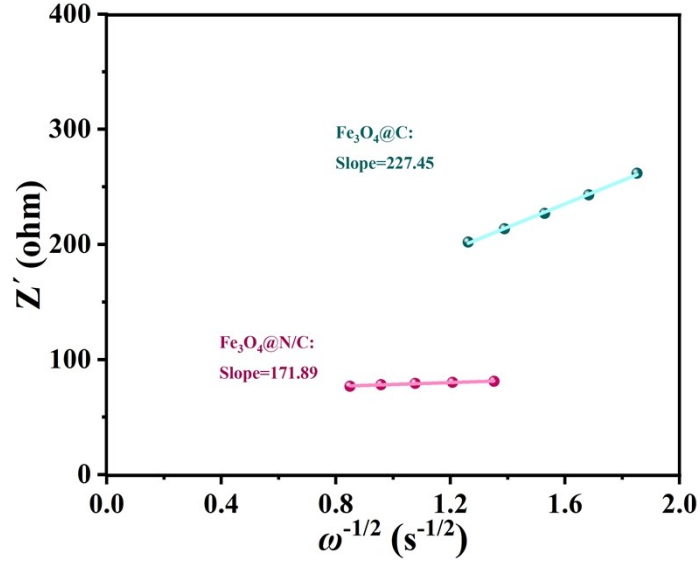


Fig. S5. Linear plots between Z' and $\omega^{-1/2}$ in the low-frequency region.

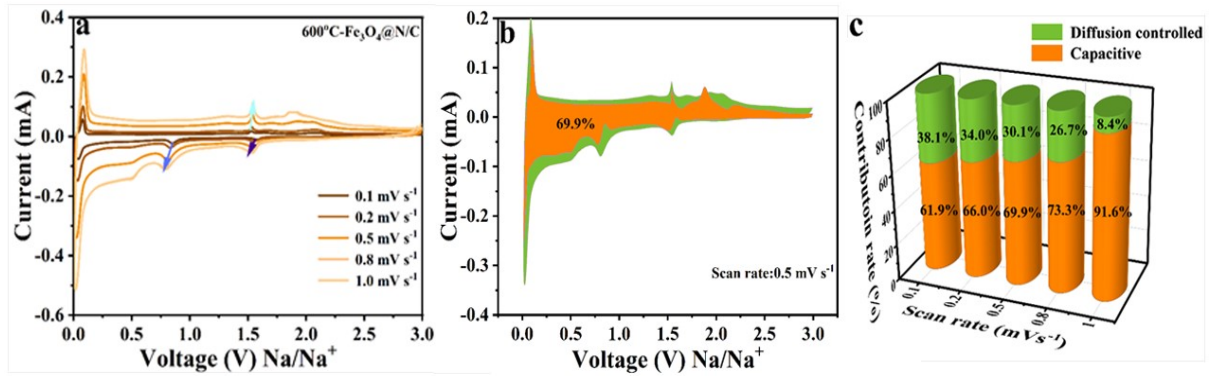


Fig. S6. (a) Cyclic voltammograms at different scan rates from 0.1 to 1 mV s^{-1} ; (b) Separation of the capacitive (yellow region) and diffusion (green region) currents in 600°C- $\text{Fe}_3\text{O}_4@\text{N/C}$ at a scan rate of 0.5 mV s^{-1} ; (c) Contribution ratio of the capacitive and diffusion-controlled charges at different scan rates.

Table S1. The fit resistance values and ionic diffusion coefficient of samples.

Sample	R_s	R_{ct}	D ($\text{cm}^2 \text{s}^{-1}$)
600°C- $\text{Fe}_3\text{O}_4@\text{N/C}$	4.782	8.625	9.39×10^{-14}
600°C- $\text{Fe}_3\text{O}_4@\text{C}$	2.655	92.600	3.62×10^{-14}

Table S2. Electrochemical performance of some transition metal oxides for LIBs.

Materials	Current density [A g^{-1}]	Capacity retention [mAh g^{-1}]	Ref.
$\text{Fe}_3\text{O}_4@\text{CTP QDs}$	0.1	810	1
$\text{Fe}_3\text{O}_4\text{-RGO}$	0.5	636	2
$\text{Fe}_3\text{O}_4@\text{NCG}$	0.5	895	3
$\text{Fe}_3\text{O}_4@\text{C}$	0.5	849	4
$\text{H-Fe}_3\text{O}_4@\text{C/GNSs}$	1	870.4	5
$\text{Fe}_3\text{O}_4@\text{N-HPCNs}$	1	581	6
$\text{OH-Fe}_3\text{O}_4@\text{NC}$	1	725	7
$\text{Fe}_3\text{O}_4@\text{C}$	1	514	8
$\text{IOC-Fe}_3\text{O}_4$	1	800	9
$\text{h-Fe}_3\text{O}_4$	1	714.6	10
$\text{GNSs/Fe}_3\text{O}_4 \text{ NPs}$	1	530	11
$\text{Fe}_3\text{O}_4@\text{ppy}$	1	796	12
$\text{GF-Fe}_3\text{O}_4$	1	1220	13
This work	1	1200	
$\text{H-Fe}_3\text{O}_4@\text{C/GNSs}$	10	285	5
$\text{Fe}_3\text{O}_4@\text{N-HPCNs}$	10	290	6
$\text{OH-Fe}_3\text{O}_4@\text{NC}$	10	361	7
This work	10	426	

Table S3. Electrochemical performance of some transition metal oxides for NIBs.

Materials	Current density [A g^{-1}]	Capacity retention [mA g^{-1}]	Ref.
RGO/ Fe_3O_4	0.04	208	11
Fe_3O_4 /graphene	0.1	216	14
$\text{Fe}_3\text{O}_4@\text{C}@\text{G}$	0.1	180	15
$\text{Fe}_3\text{O}_4@\text{CNT}$	0.1	377	16
$\text{Fe}_3\text{O}_4@\text{N-HG}$	0.1	440	17
H-CM $\text{Fe}_3\text{O}_4@\text{void}@\text{N-C}$	0.16	522	18
R- $\text{Fe}_3\text{O}_4@\text{C}$	0.2	160.3	19
Graphene- Fe_3O_4	0.5	220	9
$\text{Fe}_3\text{O}_4@\text{C}$	1	206	4
R- $\text{Fe}_3\text{O}_4@\text{C}$	2	160	20
h- $\text{Fe}_3\text{O}_4@\text{CNT}$	3.2	118	10
$\text{Fe}_3\text{O}_4@\text{C}$	4	142	21
Fe_3O_4 QD@C-GN	5	234	22
This work	5	237.5	
Fe_3O_4 QD@C-GN	10	149	22
This work	10	176	

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