## Supporting information

A Highly Stereoselectivity and Recyclable Microgel-Supported Bifunctional Sulfonamide Organocatalyst forAsymmetric Alcoholysis of meso-Cyclic Anhydrides: AThermo-responsive "Organic Nanoreactor"
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Table of Content

1. General Information. .....  2
2. General Procedure for the Synthesis of organocatalyst 5. .....  3
3. Synthesis and characterization of the microgel .....  6
3.1 Synthetic procedure .....  6
3.2 Elemental analysis .....  7
3.3 Sedimentation velocity of microgels at different temperatures. .....  7
4. General Procedure for the asymmetric alcoholysis of cyclic anhydrides .....  9
4.1 Synthetic procedure and characterization .....  9
4.2 Derivatization of the hemiesters ..... 13
4.3 Reaction optimization for the desymmetrization of meso-anhydrides ..... 13
5. ${ }^{1} \mathrm{H}$ spectra, ${ }^{13} \mathrm{C}$ spectra, HRMS spectra and the HPLC spectra ..... 15
6. Cartesian coordinates of the molecular complexes ..... 22

## 1. General Information

All alcohols were purified according to standard methods prior to use. Other reagents were obtained from commercial sources (Adamas-Beta) and used without further purification. All reactions were carried out in dried glassware with magnetic stirring.Reactions were performed in anhydrous solvents dried and distilled following standard procedures. Water for microgel synthesis was distilled water, for dialysis deionized water. Analytical thin-layer chromatography (TLC) was performed on GF254 silica gel plates, visualized by UV irradiation 254 nm or $\mathrm{KMnO}_{4}$ stain. Melting points were measured on $W R S$-1B digital melting-point apparatus. Optical rotations were measured by a Rudolph AUTOPOL I Automatic Polarimeter. Products were purified by flash column chromatography on silica gel purchased from Qingdao Haiyang Chemical Co. Ltd. HPLC analysis was performed using Daicel AD-H column ( $0.46 \mathrm{~cm} \times 25 \mathrm{~cm} \times 5 \mu \mathrm{~m}$ ) or Chiralcel OD-H column ( $0.46 \mathrm{~cm} \times 25 \mathrm{~cm} \times 5 \mu \mathrm{~m}$ ). ${ }^{1} \mathrm{H}(400 \mathrm{MHz})$ and ${ }^{13} \mathrm{C}(100 \mathrm{MHz})$ NMR were recorded on a Bruker Avance 400 spectrometer in $\mathrm{CDCl}_{3} / \mathrm{DMSO}$ using tetramethylsilane (TMS) as internal standards. Dialysis membranes MD34 with molecular weight cutoff 8000-14000 were purchased from Adamas-Beta. Hydrodynamic radii (Rh) were measured by dynamic light scattering (DLS) at a wavelength of 410 nm with a Brookhaven NanoBrook 90Plus Zeta. TEM pictures were taken at a Transmission Electron Microscope JEOL JEMF200 from Japan. The sedimentation velocities were determined with the stability analyzer LUMiSizer®651 at a rotation speed of 2000 rpm . Coupling constant ( $J$ ) values are given in Hz . Multiplicities are designated by the following abbreviations: s , singlet; d, doublet; t, triplet; q, quartet; br, broad; m, multiplet. Mass spectra (MS-ESI) were conducted on Thermo Scientific Q Exactive Combined Quadrupole Orbitrap Mass Spectrometer. IR spectra were recorded on a Thermo FT-IR spectrometer (Nicolet IS5) and are reported in terms of frequency of transmittance $\left(\mathrm{cm}^{-1}\right)$.

## 2. General Procedure for the Synthesis of organocatalyst 5



Conditions: a) $\mathrm{HCOOH}, \mathrm{HCHO}, \mathrm{H}_{2} \mathrm{O}$; b) $\mathrm{TrCl}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$; c) $\mathrm{Ph}_{3} \mathrm{P}$, DIAD, diphenyl phosphoryl azide, THF; d) $p$-styrenesulfonpl chloride, $\mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$.
Scheme S1 General synthesis procedure for chloramphenicol-based bifunctional sulfonamide catalyst 5

Compound 5 was prepared from commercially available ( $1 S, 2 S$ )-2-amino-1-( $p$ -nitrophenyl)-propane-1, 3-diol (1). The preparation procedure is referred to the following literatures: H. J. Yang, F. J. Xiong, J. Li, F. E. Chen, Chinese Chem. Lett. 2013, 24, 553-558; B. Vakulya, S. Varga, A. Csámpai, T. Soós, Org. Lett., 2005, 36, 1967-1969.
(1S, 2S)-2-( $N, N$-dimethylamino)-1-(p-nitrophenyl)-propane-1, 3-diol (2)


2
(1S, 2S)-2-amino-3-(p-nitrophenyl)-propane-1, 3-diol (10 $\mathrm{g}, ~ 47.2 \mathrm{mmol}$ ), formaldehyde $(37-40 \%, 12 \mathrm{~mL})$ and formic acid $(88 \%, 22 \mathrm{~mL})$ were added to a 25 mL flask and heated at reflux for 8 h . After the solvents were removed under reduced pressure, the residue was neutralized with 1 N sodium hydroxide ( 60 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The combined organic solution was concentrated, and purified by flash chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{CH}_{3} \mathrm{OH}=10: 1\right)$ to give a yellow solid product ( $1 S, 2 S$ )-2-( $N, N$-dimethylamino)-1-( $p$-nitrophenyl)-propane-1, 3-diol (2) (8.9 g, $79 \%$ yield). m.p. $88.2-89.3{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=+24.2\left(c 0.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz ,
$\left.\mathrm{CDCl}_{3}\right): \delta 8.20(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.59(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 4.58(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.60(\mathrm{~d}, J=4.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.60(\mathrm{~m}, 1 \mathrm{H}), 2.50(\mathrm{~s}, 6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 149.98, 147.53, 128.02, 123.64, 71.28, 69.87, 57.81, 41.61.

## (1S, 2S)-2-( $N, N$-dimethylamino)-3-trityl-1-(p-nitrophenyl)-propane-1-ol (3)



3
Trityl chloride ( $11.1 \mathrm{~g}, 40 \mathrm{mmol}$ ) and triethylamine ( $6.6 \mathrm{~g}, 66.6 \mathrm{mmol}$ ) were sequentially added to a solution of $2(8 \mathrm{~g}, 33.3 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$. The reaction was allowed to proceed at room temperature for 10 h and then washed by water $(3 \times 30 \mathrm{~mL})$. The organic phase was dried over anhydrous sodium sulfate, filtered and concentrated. The residues were recrystallized from ether to give $3(2.7 \mathrm{~g}$, $70 \%$ yield $)$ as a white solid. m.p. $155.6-156.9{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=-18.3\left(c 0.5, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.08(\mathrm{~d}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.40(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.21$ (s, $15 \mathrm{H}), 4.46(\mathrm{~d}, J=9.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.31-2.90(\mathrm{~m}, 3 \mathrm{H}), 2.55(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 149.09,147.56,142.94,128.43,128.30,127.88,127.34,123.63,87.71$, 70.33, 57.93, 41.15.
(1R, 2R)-2-( $N, N$-dimethylamino)-1-(4-nitrophenyl)-3-triphenylmethoxy-1, 2 propane-diamine (4)


4
$3(5.0 \mathrm{~g}, 10 \mathrm{mmol})$ and $\mathrm{Ph}_{3} \mathrm{P}(3.28 \mathrm{~g}, 12.5 \mathrm{mmol})$ were placed under vacuum and purged with $\mathrm{N}_{2}$ three times, anhydrous THF ( 30 mL ) was added and the solution was cooled to $0{ }^{\circ} \mathrm{C}$. Diisopropyl azodicarboxylate ( $2.53 \mathrm{~g}, 12.5 \mathrm{mmol}$ ) was added followed by the dropwise addition of a solution of diphenyl phosphoryl azide ( 3.45 g $\mathrm{mL}, 12.5 \mathrm{mmol}$ ) in anhydrous THF ( 10 mL ). The mixture was then allowed to warm to room temperature and stir for 10 h , at which point the mixture was heated at $50^{\circ} \mathrm{C}$ for $2 \mathrm{~h} . \mathrm{Ph}_{3} \mathrm{P}(3.28 \mathrm{~g}, 12.5 \mathrm{mmol})$ was then added and the mixture was stirred for a further 2 h at $50^{\circ} \mathrm{C}$. The solution was allowed to cool to room temperature before
$\mathrm{H}_{2} \mathrm{O}(0.7 \mathrm{~mL})$ was added and then stirred for a further 2 h . The mixture was then concentrated in vacuo and the crude residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The mixture was washed three times with water. The combined organic phases were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated in vacuo. Purification by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}: \mathrm{Et}_{3} \mathrm{~N}=100: 1\right)$ afforded 4 as a brown oily. $(2.94 \mathrm{~g}, 59 \%)$. $[\alpha]_{\mathrm{D}}{ }^{25}=-11.3\left(c 1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 2.02(\mathrm{~s}, 2 \mathrm{H}), 2.41(\mathrm{~s}, 6 \mathrm{H})$, $2.76-2.81(\mathrm{~m}, 1 \mathrm{H}), 2.87(\mathrm{dd}, J=3.6 \mathrm{~Hz}, 10.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.12(\mathrm{dd}, J=6.0 \mathrm{~Hz}, 10.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.89(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.18$ ( $\mathrm{s}, 15 \mathrm{H}$ ), 7.37 (d, $J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 8.02(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 41.44,55.45,59.19,69.57,87.30$, 123.53, 126.89, 127.61, 128.58, 128.71, 143.52, 147.03, 151.93 ppm .

## $N$-((1R, 2R)-2-(dimethylamino)-1-(4-nitrophenyl)-3-(trityloxy)propyl)-4-vinyl benzenesulfonyl chloride (5)



5

To a solution of chloramphenicol base $4(1 \mathrm{~g}, 2 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added the solution of $\mathrm{Et}_{3} \mathrm{~N}(1.15 \mathrm{~mL}, 8 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ under $\mathrm{N}_{2}$ atmosphere. After cooling to $0{ }^{\circ} \mathrm{C}$, 4 -vinyl benzenesulfonyl chloride ( 3 mmol ) was added dropwise over 10 min . After addition, the reaction mixture was stirred for 3 h at room temperature and then quenched by water $(10 \mathrm{~mL})$. The organic phase was washed with $\mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL})$, brine ( 20 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated under reduced pressure to give a white solid. The crude product was purified by flash chromatography using PE: EA = 2: 1 to give product $5(2.7 \mathrm{~g}, 70 \%$ yield). m.p. $188.5-189.5{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{25}=+73.6\left(c 0.05, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.96-$ 7.76 (m, 2H), 7.53 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.24-6.97(\mathrm{~m}, 17 \mathrm{H})$, $6.68(\mathrm{dd}, J=17.6,10.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.83(\mathrm{~d}, J=17.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.42(\mathrm{~d}, J=10.9 \mathrm{~Hz}, 1 \mathrm{H})$, 3.93 (d, $J=10.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.08 (dd, $J=10.6,6.1 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.93 (dd, $J=10.6,3.4 \mathrm{~Hz}$, $1 \mathrm{H}), 2.70(\mathrm{ddd}, J=9.9,6.1,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.25(\mathrm{~s}, 6 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $143.14,135.25,129.15,128.53,127.82,127.26,126.39,123.44,117.73,87.66,68.03$,
58.05, 56.51, 40.95. HRMS (ESI) m/z: $[\mathrm{M}+\mathrm{H}]^{+}$calculated for $\mathrm{C}_{38} \mathrm{H}_{37} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{~S}$ 648.25138 , found 648.25140 .

## 3. Synthesis and characterization of the microgel

### 3.1 Synthetic procedure

First, we synthesized catalyst 5, a chloramphenicol-based sulfonamide containing a polymerizable styryl group. Polymeric catalyst 6a-c was prepared by reacting chiral monomer 5 with ethyl acrylate (EA) and $N, N$-methylenebis(acrylamide) (BIS) in the presence of 2, 2-azobis(2-methylpropionamidine) dihydrochloride (AMPA) as a radical initiator in water (Scheme S2). A series of microgels with $3 \mathrm{~mol} \%$ crosslinker and organocatalyst contents of 1,4 , and $20 \mathrm{~mol} \%$ was synthesized (polymeric catalysts $\mathbf{6 a}, \mathbf{6 b}$, and $\mathbf{6 c}$, respectively). The chemical structure of monomer $\mathbf{5}$ was confirmed by NMR and mass spectral analyses.

For the emulsion polymerization, a certain amount of organocatalyst monomer (1$20 \mathrm{~mol} \%$ ) was dissolved in ethyl acrylate ( $80-99 \mathrm{~mol} \%$ ) under continuous stirring. Then cetyltrimethylammonium bromide (CTAB, 1 mol\%), crosslinker $N, N^{\prime}$ methylenebisacrylamide (BIS, $3 \mathrm{~mol} \%$ ), and water were added. Dissolved oxygen was removed by purging nitrogen through the mixture within 60 min . In order to build an emulsion, the flask with the mixture was placed in an ultrasonic bath for 15 min . Then the flask was placed in oil bath and heated up to $70^{\circ} \mathrm{C}$ under constant stirring. Finally, a water solution of the initiator 2, 2'-azobis(2-methylpropionamidine) dihydrochloride (AMPA, $1 \mathrm{~mol} \%$ ) was added to the reaction mixture. After 6 hours reaction time, the mixture was cooled down to room temperature and dialyzed for 5 days against water to purify the microgel.


| $\mathbf{6 a}$ | 96 | 4 | 3 |
| :---: | :---: | :---: | :---: |
| $\mathbf{6 b}$ | 99 | 1 | 3 |
| $\mathbf{6 c}$ | 80 | 20 | 3 |

Scheme S2: Structures and composition of organocatalysts 6a-c.

### 3.2 Elemental analysis

To verify the incorporation of the organocatalyst 5 into the microgel, elemental analysis was performed by the external provider shiyanjia lab. The content of conventional organic elements in the sample was analyzed by using the principle of high temperature combustion method under high temperature and aerobic conditions. Reported values in Table S1 represent the average value of the C, H, N, S, element in the catalyst 6a-c. Among them, sulfur element as the characteristic element of monomer is used to accurately determine the true proportion of monomer contained in the polymer microgel 6a-c. As control experiments also the monomeric organocatalysts 5 were measured. All measured sulfur contents are in very good agreement with the calculated values. The only exception is catalyst 6c (poly-EA-5 (80:20)), catalyst $\mathbf{6 c}$ was thought to have a ratio of $0: 100$ (poly-EA-5).

Table S1 Elemental analysis

| Entry | Sample | $\mathrm{N}(\%)$ | $\mathrm{C}(\%)$ | $\mathrm{H}(\%)$ | $\mathrm{S}(\%)$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathbf{6 a}$ | 1.94 | 61.07 | 6.974 | 0.907 |
| 2 | $\mathbf{6 a}$ | 2.06 | 61.31 | 7.046 | 1.012 |
| 3 | $\mathbf{6 b}$ | 1.22 | 59.73 | 7.253 | 0.285 |
| 4 | $\mathbf{6 b}$ | 1.14 | 59.48 | 7.174 | 0.196 |
| 5 | $\mathbf{6 c}$ | 6.24 | 69.48 | 5.497 | 4.347 |
| 6 | $\mathbf{6 c}$ | 6.36 | 69.72 | 5.583 | 4.451 |
| 7 | $\mathbf{5}$ | 6.49 | 70.48 | 5.73 | 4.89 |
| 8 | $\mathbf{5}$ | 6.46 | 70.6 | 5.68 | 4.93 |
| $9^{a}$ | $\mathbf{5}$ | 6.48 | 70.4 | 5.7 | 4.94 |

[^0]
### 3.3 Sedimentation velocity of microgels at different temperatures

In order to further investigate the relationship between the swelling/collapse behavior of thermo-sensitive microgel colloidal particles and the external ambient temperature, we also conducted a sedimentation analysis experiment of the colloidal particles besides the DLS dynamic light scattering experiment.

Fig S1 (a) presents median particle migration velocity for microgels measured at different temperatures by sedimentation analysis experiment. For the microgel sample, we observed a rapid increase of the sedimentation velocity by heating the sample from $10^{\circ} \mathrm{C}$ to $50^{\circ} \mathrm{C}$. This can be attributed to that when the temperature is lower, the microgels undergo volume expansion, and the expanded particles exhibit lower sedimentation velocity due to greater buoyancy. In contrast, with the gradual increase of the temperature, the solvent existing in the highly cross-linked microgel network structure is discharged, the volume of the microgel particles shrinks. This also causes a rapid increase in the settling velocity of the particles. Reported data in Table S2 represented actual particle sedimentation velocity and distribution data. The integral distribution of particle migration velocity (Fig S1 (b)) shows that the particles in the system settle with different migration velocities in the sedimentation.

Table S2 Actual particle sedimentation velocity and distribution data

| Sample <br> Name | Median in nm | Harmonic Mean in nm | Std.Dev. in nm | $\begin{aligned} & \text { Span } \\ & (\mathrm{x} 90 \\ & \mathrm{x} 10) / \mathrm{x} 50 \end{aligned}$ | Mean <br> RCA <br> in $g$ | $\begin{aligned} & 10 \% \\ & \leq \quad \text { in } \\ & \mathrm{nm} \end{aligned}$ | $\begin{aligned} & 16 \% \\ & \leq \quad \text { in } \\ & \mathrm{nm} \end{aligned}$ | $\begin{aligned} & 50 \% \\ & \leq \quad \text { in } \\ & \mathrm{nm} \end{aligned}$ | $\begin{aligned} & 84 \% \\ & \leq \quad \text { in } \\ & \mathrm{nm} \end{aligned}$ | $\begin{aligned} & 90 \% \\ & \leq \quad \text { in } \\ & \mathrm{nm} \end{aligned}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $10^{\circ} \mathrm{C}-1$ | 1,406 | 1,036 | 842.3 | 1.594 | 532.5 | 530.2 | 677.9 | 1,406 | 2,500 | 2,772 |
| $10^{\circ} \mathrm{C}-2$ | 1,449 | 792.9 | 1,038 | 1.873 | 531.3 | 532.3 | 680.0 | 1,518 | 2,857 | 3,059 |
| $20^{\circ} \mathrm{C}-1$ | 1,518 | 1,052 | 1,022 | 1.679 | 531.3 | 532.3 | 680.0 | 1,518 | 2,857 | 3,082 |
| $20^{\circ} \mathrm{C}-2$ | 1,594 | 1,292 | 981.5 | 1.648 | 530.4 | 676.2 | 793.2 | 1,594 | 2,969 | 3,303 |
| $25^{\circ} \mathrm{C}-1$ | 1,705 | 1,264 | 1,039 | 1.597 | 528.7 | 771.3 | 910.3 | 1,705 | 3,000 | 3,496 |
| $25^{\circ} \mathrm{C}-2$ | 1,767 | 1,325 | 981.8 | 1.418 | 530.0 | 735.2 | 897.4 | 1,767 | 2,913 | 3,241 |
| $30^{\circ} \mathrm{C}-1$ | 1,910 | 1,439 | 1,225 | 1.717 | 529.2 | 765.5 | 926.6 | 1,910 | 3,644 | 4,044 |
| $30^{\circ} \mathrm{C}-2$ | 1,971 | 1,431 | 1,150 | 1.581 | 529.5 | 638.5 | 1,020 | 1,971 | 3,348 | 3,754 |
| $40^{\circ} \mathrm{C}-1$ | 2,123 | 1,415 | 951.1 | 1.200 | 531.9 | 748.7 | 1,066 | 2,123 | 3,030 | 3,296 |
| $40^{\circ} \mathrm{C}-2$ | 2,128 | 1,557 | 934.1 | 1.196 | 531.7 | 772.0 | 1,149 | 2,128 | 2,964 | 3,317 |
| $50^{\circ} \mathrm{C}-1$ | 2,245 | 1,672 | 1,167 | 1.391 | 531.1 | 835.6 | 1,091 | 2,245 | 3,649 | 3,958 |
| $50^{\circ} \mathrm{C}-2$ | 2,351 | 1,703 | 1,065 | 1.237 | 530.9 | 831.8 | 1,193 | 2,351 | 3,367 | 3,740 |




Fig S1. (a) Median particle migration velocity and (b) particle velocity distribution (2000 rpm) versus temperature plots for microgel-supported catalyst 6a in isopropyl alcohol.

## 4. General Procedure for the asymmetric alcoholysis of cyclic anhydrides

### 4.1 Synthetic procedure and characterization

To perform the catalytic test reaction, the polymer catalyst (1-20 mol\%) was dissolved/suspended in the appropriate mixture of solvents, 1 equiv. meso-anhydride and the nucleophile (5 equiv.) were added. The reaction mixture was stirred at $20^{\circ} \mathrm{C}$ (TLC control, petroleum ether/ethyl acetate $2: 1, \mathrm{KMnO}_{4}$ stained). Once the anhydride consumption was complete, heating the reaction mixture to $40-50^{\circ} \mathrm{C}$. The microgel catalyst, which precipitated from the solvent, was immediately separated by centrifuging the mixture for 5 min . The supernatant was concentrated under reduced pressure to afford the corresponding hemiester without further purification by flash chromatography. The recovered catalyst was repeatedly washed with fresh solvent to remove residual contaminants.

All monoesters are known compounds and their NMR spectra data were identical to those reported in the literature: S. X. Wang, F. E. Chen, Adv. Synth. Catal. 2009, 351, 547-552; S. E. Park, E. H. Nam, H. B. Jang, J. S. Oh, S. Some, Y. S. Lee, C. E. Song, Adv. Synth. Catal. 2010, 352, 2211-2217; L. Xu, S. Han, L. Yan, H. Wang, H. Peng, F Chen, Beilstein. J. Org. Chem. 2018, 14, 309-317. The enantiomeric excess of the monoester was determined by chiral HPLC, analysis of the diastereoisomeric mixture of the corresponding amide ester derived from (S)-1-phenylethylamine according to the reported procedure.
(1S, 6R)-6-(methoxycarbonyl)cyclohex-3-enecarboxylic acid (8)

yield $93 \%$, ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=4.93(\mathrm{dd}, J=24.1,3.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.67(\mathrm{~s}$, $3 \mathrm{H}), 3.05-2.99(\mathrm{~m}, 2 \mathrm{H}), 1.84-1.82(\mathrm{~m}, 2 \mathrm{H}), 1.56-1.51(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=179.7,173.7,125.1,125.0,51.9,39.6,39.4,25.7,25.5 \mathrm{ppm}$.
(1S, 2R)-2-(Methoxycarbonyl)cyclohexanecarboxylic acid (10)

yield $94 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}$ ): $\delta=3.67$ (s, 3H), 2.84 (s, 2H), 122.00 (br, $2 \mathrm{H}), 1.78(\mathrm{br}, 2 \mathrm{H}), 1.54-1.47(\mathrm{~m}, 2 \mathrm{H}), 1.42-1.39(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO) $\delta=180.2,174.2,51.8,42.6,42.4,26.3,26.0,23.8,23.7 \mathrm{ppm}$.
(R)-5-Methoxy-3-methyl-5-oxopentanoic acid (12)

yield $85 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=3.67(\mathrm{~s}, 3 \mathrm{H}), 2.50-2.39(\mathrm{~m}, 3 \mathrm{H}), 2.31-$ $2.24(\mathrm{~m}, 2 \mathrm{H}), 1.05(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO) $\delta=173.8$, 172.8, 51.6, 40.6, 40.3, 27.3, 19.8 ppm .
(R)-5-Methoxy-3-isobutyl -5-oxopentanoic acid (14)

yield $88 \%$, ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=0.86-0.88(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 6 \mathrm{H}), 1.18-1.21$ $(\mathrm{m}, 2 \mathrm{H}), 1.53-1.67(\mathrm{~m}, 1 \mathrm{H}), 2.36-2.37(\mathrm{~m}, 5 \mathrm{H}), 3.65(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta=22.49,25.15,29.70,38.39,43.37,51.50,173.02,178.91 \mathrm{ppm}$.
(1S, 2R, 3S, 4R)-3-(Methoxycarbonyl)-7-oxabicyclo[2.2.1]hept-5-ene-2-carboxylic acid (16)

yield $90 \%$, ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.69(\mathrm{br}, 1 \mathrm{H}), 6.49-6.45(\mathrm{~m}, 2 \mathrm{H}), 5.29(\mathrm{~d}$, $J=16.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 2.88-2.83(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{DMSO}$ ) $\delta=173.0,172.5,137.1,137.0,80.4,80.1,51.9,47.0,46.3 \mathrm{ppm}$.

## Monomethyl 7-oxabicyclo[2.2.1]heptane-2,3-dicarboxylate (18)


yield $87 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=1.52-1.54(\mathrm{~m}, 2 \mathrm{H}), 2.99-3.05(\mathrm{~m}, 2 \mathrm{H}), 3.67$ (s, 3H), 4.89-4.96 (m, 2H) ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=28.95,29.02,52.16$, 52.27, 52.30, 78.34, 78.65, 171.31, 175.83 ppm .
(1R, 2S, 3R, 4S)-3-(Methoxycarbonyl)bicyclo[2.2.1]hept-5-ene-2-carboxylic acid (20)

yield $97 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=6.31(\mathrm{dd}, J=5.1,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.22(\mathrm{dd}, J$ $=5.3,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{~s}, 3 \mathrm{H}), 3.35-3.26(\mathrm{~m}, 2 \mathrm{H}), 3.18(\mathrm{~d}, J=12.9 \mathrm{~Hz}, 2 \mathrm{H}), 1.49(\mathrm{~d}$, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=$ $178.3,172.9,135.5,134.3,51.5,48.7,48.2,48.0,46.5,46.1 \mathrm{ppm}$.
(4R, 5R)-1,3-diphenyl-5-(methoxycarbonyl)-2-oxo-4-carboxylic acid imidazolidine (22)

yield $78 \%$, ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=3.62(\mathrm{~s}, 3 \mathrm{H}), 4.03-4.11(\mathrm{~m}, 4 \mathrm{H}), 4.98(\mathrm{~d}$, $J=15.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.08(\mathrm{~d}, J=14.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-7.32$, (m, 10H), 7.85 (br, 1H) ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=47.0,47.2,53.0,57.1,128.2,128.8,128.9,129.2$, 135.7,160.1, 168.9, 171.6 ppm .
(1S, 6R)-6-(Ethoxycarbonyl)cyclohex-3-enecarboxylic acid (23)

yield $88 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=11.14(\mathrm{br}, 1 \mathrm{H}), 5.68-5.62(\mathrm{~m}, 2 \mathrm{H}), 4.13$ (q, $J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.03-3.02(\mathrm{~m}, 2 \mathrm{H}), 2.58-2.51(\mathrm{~m}, 2 \mathrm{H}), 2.34(\mathrm{~d}, J=18.4 \mathrm{~Hz}, 2 \mathrm{H})$, $1.21(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=179.9,173.2,125.2,125.0$, $60.8,39.7,39.6,25.9,25.5,14.0 \mathrm{ppm}$.
(1S, 6R)-6-(Isopropoxycarbonyl)cyclohex-3-enecarboxylic acid (24)

yield $76 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=11.18(\mathrm{br}, 1 \mathrm{H}), 5.69-5.62(\mathrm{~m}, 2 \mathrm{H}), 5.00$ (dt, $J=12.4,6.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.02 (s, 2H), 2.58-2.50 (m, 2H), 2.33 (d, $J=16.5 \mathrm{~Hz}, 2 \mathrm{H}$ ), $1.19-1.18(\mathrm{~m}, 6 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=180.2,172.6,125.3,125.0$, 68.3, 39.7, 25.9, 25.4, 21.7, 21.6 ppm .
(1S, 6R)-6-((Allyloxy)carbonyl)cyclohex-3-enecarboxylic acid (25)

yield $92 \%$, ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=11.11(\mathrm{br}, 1 \mathrm{H}), 5.92-5.82(\mathrm{~m}, 1 \mathrm{H}), 5.69-$ $5.64(\mathrm{~m}, 2 \mathrm{H}), 5.28(\mathrm{~d}, J=17.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.62-4.54(\mathrm{~m}, 2 \mathrm{H})$, 3.06 (d, $J=5.9 \mathrm{~Hz}, 2 \mathrm{H}$ ), 2.62-2.53 (m, 2H), 2.39-2.34 (m, 2H) ppm; ${ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta=179.7,172.9,132.1,125.2,118.1,65.4,39.7,39.6,25.9,25.6 \mathrm{ppm}$.
(1S, 6R)-6-((Benzyloxy)carbonyl)cyclohex-3-enecarboxylic acid (26)

yield $95 \%$, ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=10.51(\mathrm{br}, 1 \mathrm{H}), 7.37-7.29(\mathrm{~m}, 5 \mathrm{H}), 5.69(\mathrm{~s}$, $2 \mathrm{H}), 5.19-5.11(\mathrm{~m}, 2 \mathrm{H}), 3.11(\mathrm{t}, J=5.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.66-2.57(\mathrm{~m}, 2 \mathrm{H}), 2.42-2.35(\mathrm{~m}, 2 \mathrm{H})$ ppm; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=179.8,173.0,135.9,128.5,128.2,128.1,125.2$, 125.1, 66.6, 39.7, 39.6, 25.8, 25.6 ppm.
(1S, 6R)-6-((Cinnamyloxy)carbonyl)cyclohex-3-enecarboxylic acid (27)

yield $96 \%,{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=7.42-7.29(\mathrm{~m}, 5 \mathrm{H}), 6.64(\mathrm{~d}, J=3.2 \mathrm{~Hz}$, $2 \mathrm{H}), ~ 6.41-6.36(\mathrm{~m}, 1 \mathrm{H}), 5.73(\mathrm{~s}, 2 \mathrm{H}), 4.34-4.33(\mathrm{~m}, 2 \mathrm{H}), 3.14-3.09(\mathrm{~m}, 2 \mathrm{H}), 2.66-2.62$ $(\mathrm{m}, 2 \mathrm{H}), 2.43-2.39(\mathrm{~m}, 2 \mathrm{H}) \mathrm{ppm} ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta=177.7,173.2$, 136.7, 131.0, 128.5, 127.6, 126.4, 125.1, 123.1, 65.3, 63.4, 39.7, 39.6, 25.8, 25.6 ppm.

### 4.2 Derivatization of the hemiesters

To a solution of 1.0 equiv. hemiester in dry DCM was added 1.2 equiv. EDCI, 0.3 equiv. DMAP and 1.1 equiv. (S)-1-phenyl-ethylamine. The reaction mixture was stirred for 3 h at RT (TLC control, ethyl acetate : petroleum ether = 1:2), the solvent was concentrated under reduced pressure and the residue was washed 3 times with $10 \%$ aqueous hydrochloric acid. The enantiomeric excess of the hemiester was determined by HPLC analysis of the diastereoisomeric mixture of the corresponding amide-ester derived from ( $S$ )-1-phenyl-ethylamine according to the reported procedure (Scheme S3).


Scheme S3 General procedure for derivatization of hemiesters for HPLC analysis

### 4.3 Reaction optimization for the desymmetrization of meso-anhydrides

Table S3 Screening table for the desymmetrization of meso-anhydrides ${ }^{a}$

| Entry | Catalyst | Mol $[\text { [\%] }$ | Anhydride | Solvent | Nu | Conc. <br> [M] | Time <br> [h] | Yield <br> [\%] | $\begin{gathered} \mathrm{Ee}^{f} \\ {[\%]} \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 6 a | 5 | 7 | $\mathrm{Et}_{2} \mathrm{O}$ | MeOH | 0.02 | 120 | 94 | 31 |
| 2 | 6b | 5 | 7 | $\mathrm{Et}_{2} \mathrm{O}$ | MeOH | 0.02 | 120 | 87 | 25 |
| $3^{b}$ | 6c (5) | 5 | 7 | $\mathrm{Et}_{2} \mathrm{O}$ | MeOH | 0. 02 | 120 | 95 | 33 |
| 4 | 6 a | 5 | 7 | THF | MeOH | 0.02 | 60 | 89 | 15 |
| 5 | 6a | 5 | 7 | Toluene | MeOH | 0.02 | 36 | 96 | 60 |
| 6 | 6 a | 5 | 7 | $\mathrm{CHCl}_{3}$ | MeOH | 0.02 | 36 | 92 | 41 |
| 7 | 6 a | 5 | 7 | MTBE | MeOH | 0.02 | 70 | 91 | 61 |
| 8 | 6 a | 5 | 7 | Acetone | MeOH | 0.02 | 48 | 78 | 0 |
| 9 | 6 a | 5 | 7 | Dioxane | MeOH | 0.02 | 72 | 93 | 40 |
| 10 | 6 a | 5 | 7 | $\mathrm{CCl}_{4}$ | MeOH | 0.02 | 40 | 93 | 60 |
| 11 | 6a | 5 | 7 | MTBE | MeOH | 0.02 | 70 | 91 | 61 |
| 12 | 6 a | 5 | 7 | MTBE | MeOH | 0.05 | 36 | 90 | 76 |
| 13 | 6a | 5 | 7 | MTBE | MeOH | 0.1 | 20 | 93 | 71 |


| 14 | 6 a | 5 | 7 | MTBE | MeOH | 0.15 | 20 | 90 | 65 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| -15 | 6a | 5 | 7 | MTBE | MeOH | 0.2 | 10 | - 87 | 59 |
| 16 | 6 a | 1 | 7 | MTBE | MeOH | 0.05 | 120 | 91 | 54 |
| 17 | 6 a | 5 | 7 | MTBE | MeOH | 0.05 | 36 | 90 | 76 |
| 18 | 6 a | 10 | 7 | MTBE | MeOH | 0.05 | 20 | 93 | 83 |
| 19 | 6a | -20 | 7 | MTBE | MeOH | 0.05 | 10 | - 92 | 74 |
| 20 | 6 | 10 | 7 | MTBE | MeOH | 0.05 | 20 | 93 | 83 |
| 21 | 6 a | 10 | 9 | MTBE | MeOH | 0.05 | 20 | 94 | 77 |
| 22 | 6 a | 10 | 11 | MTBE | MeOH | 0.05 | 36 | 85 | 67 |
| 23 | 6a | 10 | 13 | MTBE | MeOH | 0.05 | 36 | 88 | 37 |
| 24 | 6 a | 10 | 15 | MTBE | MeOH | 0.05 | 12 | 90 | 60 |
| 25 | 6 a | 10 | 17 | MTBE | MeOH | 0.05 | 20 | 87 | 35 |
| 26 | 6 a | 10 | 19 | MTBE | MeOH | 0.05 | 24 | 97 | 53 |
| 27 | 6 a | 10 | 21 | MTBE | MeOH | 0.05 | 72 | 78 | 15 |
| $28^{\text {c }}$ | 6a | - 10 | 7 | MTBE | MeOH | 0.05 | 20 | -95 | 60 |
| 29 | 6 | 10 | 7 | MTBE | MeOH | 0.05 | 20 | 93 | 83 |
| 30 | 6a | 10 | 7 | MTBE | EtOH | 0.05 | 96 | 88 | 53 |
| $31{ }^{\text {d }}$ | 6a | 10 | 7 | MTBE | $i-\mathrm{PrOH}$ | 0.05 | 120 | 76 | 14 |
| 32 | 6a | 10 | 7 | MTBE | allylalcohol | 0.05 | 48 | 92 | 77 |
| 33 | 6a | 10 | 7 | MTBE | BnOH | 0.05 | 12 | 95 | 86 |
| 34 | 6 a | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 12 | 96 | 90 |
| $35^{e}$ | 6 | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 12 | 95 | 88 |
| $36^{e}$ | 6 a | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 15 | 95 | 89 |
| $37^{e}$ | 6 a | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 12 | 97 | 85 |
| $38^{e}$ | 6a | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 16 | 93 | 89 |
| $39^{e}$ | 6a | 10 | 7 | MTBE | cinnamyl alcohol | 0.05 | 20 | 95 | 87 |

${ }^{a}$ unless otherwise noted, all reactions were carried out with anhydride and ROH in solvents at $20^{\circ} \mathrm{C} .{ }^{b}$ elemental analysis shows that catalyst $\mathbf{6} \mathbf{b}$ did not undergo polymerization reaction, thus, catalyst $\mathbf{6 b}$ is equivalent to monomer compound 5. ${ }^{c}$ reaction performed at $50^{\circ} \mathrm{C} .{ }^{d}$ incomplete conversion after 120 h . ${ }^{e}$ catalyst 6a recycling experiments. ${ }^{f}$ determined by chiral HPLC analysis.

## 5. ${ }^{1} \mathrm{H}$ spectra, ${ }^{13} \mathrm{C}$ spectra, HRMS spectra and the HPLC spectra

All compounds except 5 are known compounds and their NMR spectra data were identical to those reported in the literature: H. J. Yang, F. J. Xiong, J. Li, F. E. Chen, Chinese Chem. Lett. 2013, 24, 553-558.





## (1S,6R)-6-(methoxycarbonyl)cyclohex-3-enecarboxylic acid (8)

Chiralcel OD-H, hexane/IPA: 90/10, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=19.062 \mathrm{~min}, t($ minor $)=21.614 \mathrm{~min}$;

(1S, 2R)-2-(Methoxycarbonyl)cyclohexanecarboxylic acid (10)
Chiralcel OD-H, hexane/IPA: 90/10, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=23.326 \mathrm{~min}, t($ minor $)=25.834 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 20.362 | 1930.2 | 0.4943 | 50.450 |
| 2 | 23.069 | 1779.9 | 0.5183 | 49.550 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 23.326 | 543.9 | 0.5612 | 11.557 |
| 2 | 25.837 | 2634.1 | 0.8614 | 88.443 |

## (R)-5-Methoxy-3-methyl-5-oxopentanoic acid (12)

Chiralcel OD-H, hexane/IPA: 85/15, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=15.113 \mathrm{~min}, t($ minor $)=17.354 \mathrm{~min}$;


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 15.043 | 2638.9 | 0.5098 | 49.110 |
| 2 | 17.416 | 2498 | 0.5901 | 50.890 |



## (R)-5-Methoxy-3-isobutyl-5-oxopentanoic acid (14)

Chiralcel OD-H, hexane/IPA: $80 / 20$, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=9.326 \mathrm{~min}, t($ minor $)=10.7455 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 9.8566 | 271.1 | 0.3769 | 49.899 |
| 2 | 11.445 | 538.6 | 0.1905 | 50.101 |


| Entry | Time | Height | Widれ | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 9.326 | 1754.6 | 0.5366 | 31.774 |
| 2 | 10.745 | 2600.2 | 0.7068 | 68.226 |

(1S, 2R, 3S, 4R)-3-(Methoxycarbonyl)-7-oxabicyclo[2.2.1]hept-5-ene-2carboxylic acid (16)

Chiralcel AD-H, hexane/IPA: 90/10, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=17.157 \mathrm{~min}, t$ (minor) $=18.586 \mathrm{~min}$;


7-Oxa-bicyclo[2.2.1]heptane-2,3-dicarboxylic acid monomethyl ester (18)
Chiralcel AD-H, hexane/IPA: 80/20, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=21.422 \mathrm{~min}, t$ (minor) $=24.347 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 16.84 | 197.3 | 0.3767 | 49.933 |
| 2 | 19.036 | 129.6 | 0.5752 | 50.067 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 21.422 | 1387.1 | 0.5602 | 67.878 |
| 2 | 24.347 | 765.7 | 0.4832 | 32.122 |

(1R, $2 S, 3 R, 4 S$ )-3-(Methoxycarbonyl)bicyclo[2.2.1]hept-5-ene-2-carboxylic acid (20)

Chiralcel AD-H, hexane/IPA: 91/9, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=21.095 \mathrm{~min}, t$ (minor) $=24.155 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 21.544 | 2701 | 1.0607 | 50.626 |
| 2 | 24.487 | 2645.5 | 1.0562 | 49.374 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 21.095 | 2838.7 | 1.0358 | 76.816 |
| 2 | 24.155 | 1132.9 | 1.0938 | 23.814 |

(4R, 5R)-1,3-diphenyl-5-(methoxycarbonyl)-2-oxo-4-carboxylic acid

## imidazolidine (22)

Chiralcel OD-H, hexane/IPA: 80/20, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=23.143 \mathrm{~min}, t($ minor $)=32.168 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 25.033 | 81400.5 | 1.4608 | 48.933 |
| 2 | 33.329 | 80570.99 | 1.4750 | 51.067 |


| Entry | Time | Height | Widh | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 23.143 | 6.9 | 0.7045 | 42.065 |
| 2 | 32.168 | 6.2 | 0.9072 | 57.935 |

(1S, 6R)-6-(Ethoxycarbonyl)cyclohex-3-enecarboxylic acid (23)
Chiralcel OD-H, hexane/IPA: 90/10, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=17.667 \mathrm{~min}, t($ minor $)=22.35 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 18.656 | 2973.3 | 0.7193 | 50.281 |
| 2 | 21.638 | 2812.5 | 0.752 | 49.719 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 17.667 | 1057.6 | 1.283 | 23.637 |
| 2 | 22.35 | 2696.8 | 1.3796 | 76.363 |

## (1S, 6R)-6-(Isopropoxycarbonyl)cyclohex-3-enecarboxylic acid (24)

Chiralcel OD-H, hexane/IPA: 80/20, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=16.923 \mathrm{~min}, t($ minor $)=20.925 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 16.947 | 2552.7 | 0.6223 | 48.484 |
| 2 | 21.078 | 1587.5 | 1.128 | 51.516 |


| Entry | Time | Height | Widक | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 16.923 | 1589 | 0.4996 | 43.194 |
| 2 | 20.925 | 1013.3 | 0.8547 | 56.806 |

(1S, 6R)-6-((Allyloxy)carbonyl)cyclohex-3-enecarboxylic acid (25)
Chiralcel AD-H, hexane/IPA: 80/20, Flow rate: $0.50 \mathrm{~mL} / \mathrm{min}$, UV detection at 220 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=39.915 \mathrm{~min}, t$ (minor) $=41.099 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 39.392 | 2927.3 | 1.6025 | 48.809 |
| 2 | 41.947 | 2834.7 | 2.0384 | 51.191 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 39.915 | 743.6 | 0.9235 | 11.458 |
| 2 | 41.099 | 2875.8 | 1.3842 | 88.542 |

## (1S, 6R)-6-((Benzyloxy)carbonyl)cyclohex-3-enecarboxylic acid (26)

Chiralcel AD-H, hexane/IPA: 80/20, Flow rate: $0.80 \mathrm{~mL} / \mathrm{min}$, UV detection at 210 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=6.414 \mathrm{~min}, t$ (minor) $=9.376 \mathrm{~min}$;



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 7.727 | 2374.3 | 1.1061 | 49.211 |
| 2 | 11.055 | 2578.1 | 1.1512 | 50.789 |


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 6.414 | 459.3 | 0.6398 | 6.996 |
| 2 | 9.376 | 2813.1 | 1.0972 | 93.004 |

## (1S, 6R)-6-((Cinnamyloxy)carbonyl)cyclohex-3-enecarboxylic acid (27)

Chiralcel AD-H, hexane/IPA: 80/20, Flow rate: $0.80 \mathrm{~mL} / \mathrm{min}$, UV detection at 254 nm , $\mathrm{T}=25^{\circ} \mathrm{C}$, retention time: $t$ (major) $=12.357 \mathrm{~min}, t($ minor $)=16.463 \mathrm{~min}$;


| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 9.935 | 301.8 | 1.0309 | 49.375 |
| 2 | 13.623 | 300 | 1.0639 | 50.625 |



| Entry | Time | Height | Width | Area\% |
| :--- | :--- | :--- | :--- | :--- |
| 1 | 12.357 | 72.6 | 1.0796 | 4.951 |
| 2 | 16.463 | 1166.6 | 1.247 | 95.049 |

## 6. Cartesian coordinates of the molecular complexes

|  |  |
| :---: | :---: |
| C 5.41474400-1.80846500-1.62158400 | C 5.57110300-1.40399800-1.57508400 |
| С 4.09265400-1.47643000-1.89206400 | C 4.25150300-1.06460900-1.84857900 |
| C 3.09268900-1.64525800-0.92176500 | C 3.22269300-1.38242300-0.94832500 |
| C 3.44414100-2.14323900 0.33789600 | C 3.54205200-2.03874100 0.24562100 |
| C 4.76311000-2.47749900 0.62734500 | C 4.85792500-2.38295400 0.53748300 |
| C 5.73071200-2.31078400-0.36013100 | C 5.85505500-2.06555300-0.38129100 |
| H 6.19613600-1.68476700-2.36031100 | Н $6.37455700-1.16643100-2.26036900$ |
| H 3.83380300-1.05391200-2.85866200 | H 4.01627500-0.52075600-2.75892400 |
| Н 2.69845000-2.26653800 1.11149200 | H $2.77303800-2.279913000 .96684300$ |
| H 5.05005600-2.85618600 1.59972700 | H 5.11994200-2.88361200 1.46049800 |
| N 7.12588300-2.66734800-0.06174600 | N 7.24729700-2.43156800-0.08067900 |
| O 7.37773800-3.11774300 1.05672400 | O 7.47091400-3.02366000 0.97605400 |
| O $7.96456300-2.49637000-0.94793900$ | O 8.11197700-2.12614300-0.90347900 |
| C 1.68429300-1.19203600-1.27622200 | C 1.81681800-0.91335100-1.29200600 |
| H 1.47046400-1.54165900-2.29124800 | H 1.64726900-1.12701300-2.35211800 |
| C 0.54960600-1.72809600-0.36640000 | C 0.66456200-1.58946400-0.50598600 |
| Н $0.73815800-1.396176000 .65812700$ | H 0.80896100-1.39511900 0.56020200 |
| C -0.77145000-1.10598000-0.84196800 | C -0.65397700-0.93501900-0.94363900 |
| H -0.94883100-1.39976000-1.87700200 | H -0.78778600-1.08994300-2.01469400 |
| H -0.70587300-0.01598400-0.81331000 | H -0.61629100 $0.14222900-0.76663700$ |
| C - $3.18349600-1.05544600-0.30831600$ | C -3.08371400-1.00442400-0.49994700 |
| C - $3.234438000 .47399900-0.11450500$ | C -3.17903000 0.48360100-0.10490900 |
| C -4.03610900 1.30799800-0.90352600 | C -3.97291800 1.40045600-0.80484000 |
| C -2.55748700 1.039781000 .97694200 | C -2.55477400 0.910215001 .07743400 |
| C -4.14586500 $2.67289500-0.61888000$ | C -4.12611800 $2.71205900-0.34414500$ |
| H-4.59332300 $0.89326900-1.73519400$ | H -4.49034400 1.09080300-1.70514200 |
| C -2.65713800 2.401125001 .25624100 | C -2.69760500 2.219219001 .53274100 |
| H -1.97293300 0.401380001 .62800600 | H -1.97779700 0.20146600 1.65873900 |
| C -3.45426800 3.225119000 .45818000 | C -3.48639300 3.127471000 .82268800 |
| H-4.78060100 3.29792500-1.24055200 | H -4.75392900 $3.40261000-0.90004200$ |
| H -2.11486100 2.819718002 .09906400 | H -2.19558600 2.530901002 .44404000 |
| H -3.53847400 4.285050000 .68028000 | H -3.60437500 4.145894001 .18153000 |
| C -3.56999900-1.50984700-1.73124600 | C -3.40894300-1.27026800-1.98458400 |
| C -4.38408700-2.63016700-1.94536300 | C - $4.18722700-2.36727900-2.37770500$ |
| C -3.04403500-0.85415400-2.85965500 | C -2.86015800-0.45813900-2.99411900 |
| C -4.65471400-3.08636500-3.23681700 | C -4.40123700-2.65035500-3.72799300 |
| H -4.81727400-3.15280500-1.10174600 | H -4.63664100-3.00735500-1.62882500 |
| C -3.30540100-1.31387100-4.14895700 | C -3.06490400-0.74481000-4.34239600 |
| H-2.41243700 $0.01804500-2.74156900$ | H -2.25473300 0.40242600-2.73628300 |
| C -4.11386200-2.43428400-4.34385200 | C -3.83823500-1.84439400-4.71628100 |
| H -5.29179200-3.95542600-3.37258400 | H -5.01159600-3.50553500-4.00312500 |
| H -2.86867300-0.79446100-4.99633100 | H -2.61174200-0.10735500-5.09529500 |

H -4.32089800 -2.79300700 -5.34773100 C - $4.06377000-1.721092000 .76532500$ C - $3.73513000-2.994104001 .25204400$ C - $5.22154900-1.099653001 .24783700$ C - $4.52977500-3.618931002 .21310700$ H - $2.85599800-3.495075000 .86785900$ C - $6.01984700-1.726611002 .20535300$ H - $5.49891800-0.116612000 .88502700$ C - $5.67497000-2.985832002 .69738900$ H - $4.25041900-4.601634002 .58177000$ H -6.91048400-1.22385300 2.57066000 Н -6.29295900 -3.46909000 3.44819900 O 2.66821700 1.16120900-3.55076300 O $0.264394000 .33710500-3.42313600$ C $0.898368002 .65780000-2.27677700$ C - 0.45872300 2.93087100-2.10878800 C $1.873603003 .62158200-2.00833600$ C - $0.836195004 .18385100-1.63655100$ H - 1.20210300 2.17466400 -2.33248600 C $1.481670004 .86107300-1.51616100$ H $2.918877003 .39795400-2.19280800$ C $0.122378005 .15517800-1.29818300$ H $-1.889669004 .39581800-1.49163300$ H 2.23686700 5.61286900-1.31305000 C - $0.331330006 .42905900-0.71672600$ H - $1.389358006 .64827000-0.85416900$ C $0.416174007 .29391000-0.01983400$ H - 0.011061008 .207819000 .37883100 H 1.468742007 .121832000 .18498800 N $1.634882000 .29863300-1.30400200$ H $2.399366000 .73685700-0.78770800$ N 0.58944400-3.19087100-0.36015500 C $0.23911200-3.837178000 .89523200$ H - $0.84710900-3.899739001 .07477700$ H $0.63609100-4.858864000 .89727600$ H $0.67622300-3.296020001 .73715100$ C - $0.02664800-3.84751900-1.50465400$ Н $0.26404800-4.90297100-1.50833400$ H-1.12934800 -3.79455600-1.49797300 H 0.32592300 - $3.41001400-2.44293900$ O -1.84053900-1.55574600-0.00436100 S $1.407101001 .02843500-2.80612900$ C - $1.40881200-2.032574003 .65858100$ H - $0.90007600-1.720675004 .57235200$ H - $2.48320900-1.828914003 .74792800$ H - $1.27222500-3.115904003 .53658600$ O -0.81992500-1.30632700 2.58253600 H - $1.27970500-1.540405001 .75299900$ C 0.473790003 .910014002 .33112600 C 0.506220003 .772214003 .65915300 H 0.198931004 .865584001 .88913700 H 0.265118004 .616444004 .30055700 C 0.916219002 .486889004 .33185900 H 0.457790002 .386211005 .32008800 H 2.002938002 .498916004 .50504200 C 0.813254002 .813881001 .35992500 H 1.758100003 .047945000 .85479100 H 0.061760002 .798204000 .56787800 C 0.914013001 .415679001 .99256300 H 0.289262000 .720316001 .42119100

H -4.00116400-2.06844100-5.76631800 C - $3.98432600-1.826001000 .44057400$ C - $3.64166500-3.146310000 .76434500$ C - $5.17320800-1.298238000 .95776100$ C - $4.45379600-3.910579001 .60210900$ H -2.73757700 -3.57347800 0.35010300 C - $5.98881300-2.064212001 .79158800$ H - $5.46183900-0.280898000 .71976000$ C - $5.63045500-3.371256002 .12308900$ H -4.16339300 -4.92833500 1.84602600 Н - $6.90377900-1.632954002 .18693900$ Н - $6.26222700-3.963440002 .77833400$ O 2.82142200 1.74409300-3.19152500 O $0.435620000 .86314000-3.26684800$ C $0.971824003 .02003400-1.79648000$ C - $0.396620003 .24117900-1.64501700$ C $1.913061003 .95781300-1.36443700$ C - 0.82109400 4.41133700-1.02372500 H - 1.11288800 2.50756800 -1.99612600 C $1.473774005 .11163400-0.72575900$ H $2.969332003 .78173400-1.53742700$ С $0.100867005 .34683100-0.52205800$ H $-1.883888004 .58103800-0.89177800$ H $2.202677005 .84388600-0.39517900$ C - 0.404106006 .521462000 .20729800 H - 1.461758006 .736335000 .06043700 C 0.297069007 .298961001 .04170800 H - 0.166206008 .142156001 .54279800 H 1.345751007 .121585001 .26140300 N $1.731732000 .56617900-1.12153700$ H $2.466684000 .94572600-0.52251800$ N $0.74015000-3.03864400-0.69440300$ C $0.36216700-3.855143000 .44879400$ H - $0.72779700-3.962723000 .57698100$ H $0.78386000-4.859753000 .32890600$ H $0.75613100-3.423989001 .37164000$ C $0.18078800-3.54692200-1.93915300$ H $0.49732700-4.58630300-2.07326900$ H - $0.92244100-3.51706400-1.96728000$ H $0.55506400-2.98000300-2.79625000$ O -1.74029500 -1.51461200-0.21507100 S $1.538700001 .48729200-2.51981100$ C - $1.42525700-2.472549003 .36459500$ H - 0.95660300 -2.27679500 4.33057300 H -2.50680800 -2.30399400 3.43960500 Н - $1.25786600-3.526624003 .10373200$ O -0.81717300-1.59634900 2.41876000 H - $1.24177600-1.725406001 .54848500$ C - 1.622500001 .304422003 .06815300 C-1.23547700 1.680478001 .84647700 H - 2.674175001 .348130003 .34398600 H-1.97136700 2.023931001 .12345600 C 0.199812001 .606612001 .39130400 H 0.413299002 .344999000 .61278100 H 0.384247000 .623935000 .93149500 C - 0.688939000 .792262004 .12934600 H - $0.86827700-0.277071004 .29379800$ H - 0.935067001 .272797005 .07846900 C 0.801841001 .010651003 .82074600 H 1.275873001 .497628004 .68008100

| C 0.534784001 .256971003 .47380600 | C 1.175796001 .825422002 .57199100 |
| :--- | :--- | :--- |
| H -0.519441001 .016410003 .60608700 | H 1.275231002 .888575002 .78780500 |
| O $2.43378400-0.086174002 .99596900$ | O $2.67636300-0.000382002 .80287100$ |
| C 1.368253000 .067101003 .90290900 | C 2.536663001 .268035002 .20873900 |
| C 2.291673000 .788328001 .94061700 | C $1.60867100-0.257718003 .63505300$ |
| O 3.16742200 0.93480900 1.12601400 | O $1.44606500-1.345374004 .12753100$ |
| O $1.28080500-0.632482004 .86768500$ | O 3.383032001 .700890001 .48475500 |


[^0]:    ${ }^{a}$ Theoretical calculation of organic elements (C, H, N, S) for monomeric compound 5.

