Controllable synthesis and enhanced gas sensing performances of AuNP-modified ZnSnO₃ hollow nanocubes toward highly sensitive toluene detection

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Fig. S1. The full survey spectrum of the Au₂NPs@ZnSnO₃ nanostructure

Fig. S1 illustrated the full survey spectrum of Au₂NPs@ZnSnO₃ nanostructure ranging from 0~1300 eV, and the measured results exhibited the existence of Zn, Sn, O and Au elements. Meanwhile, there is no other impurity elements appearance in the full survey spectrum, which indicates high-purity of AuNP-modified ZnSnO₃ samples.

S2: Gas sensing performances of ZnSnO₃ sensors



Fig. S2 (a) Gas sensing performances as-prepared ZnSnO₃ sensors toward 100 ppm toluene under different operating temperatures. (b) Dynamic response-recovery curves of as-prepared pure ZnSnO₃ sensors toward 20-400 ppm toluene at 280°C.

Fig. S2 (a) shows the sensor radar plots of as-prepared ZnSnO₃-based gas sensors with different NaOH-assisted dissolution. The optimum operating temperature was investigated on all the as-prepared gas sensors, which selected the 280°C as the working temperature. Furthermore, gas sensor ZnSnO₃ with the volume NaOH-assisted dissolution of 60 mL shows the best response among all the as-prepared gas sensors. Fig. S2 (b) illustrates the dynamic cycle of ZnSnO₃ sensors toward 20-400 ppm toluene at 280°C, and the excellent gas sensing performances of ZnSnO₃ (NaOH-60 mL) sensor can be attributed the larger surface area of unique hollow nanocubes.

S3: Sensor response of Au_xNPs@ZnSnO₃ under different concentrations of toluene



Fig. S3 Response-concentration curves of as-fabricated Au_xNPs@ZnSnO₃ gas sensors toward different concentrations of toluene under their optimum working temperature.

Fig. S3 plots the response-concentration curves of as-fabricated $Au_xNPs@ZnSnO_3$ sensors toward different concentrations of toluene. Obviously, all the sensor responses were increasing with the increase concentration of toluene at the optimum working temperature. Moreover, the as-fabricated $Au_2NPs@ZnSnO_3$ sensor showed the highest response than other sensors under the same concentration of toluene.



Fig. S4 Dynamic response-recovery curves (a) and linear relationship fitting curves (b) of as-fabricated Au_xNPs@ZnSnO₃ sensors to 10-500 ppb toluene at their optimum working temperature.

Fig. S4 (a) displays the as-fabricated $Au_xNPs@ZnSnO_3$ sensors were exposed to a continuous concentration of toluene from 10 to 500 ppb under their optimal operating temperature. The measured results reveal that all the sensor responses of $Au_xNPs@ZnSnO_3$ were enhanced with the increasing concentration from 10 to 500 ppb.

Expectedly, the Au₂NPs@ZnSnO₃ sensor exhibited the highest response than other sensors, which was consistent well with the sensor responses under high concentration of toluene. Meanwhile, Fig. S4(b) displays the corresponding linear relationship fitting curves of Au_xNPs@ZnSnO₃ sensors, and the measured linear relationship of Au₂NPs@ZnSnO₃ sensor is $y_2 = 0.41x_2 + 0.72$, which showed good linear relationship toward ppb-level of toluene detection.

S5: Humidity effects of as-fabricated Au_xNPs@ZnSnO₃ sensors



Fig. S5 (a-b) Response curves of Au₀NPs@ZnSnO₃ and Au₂NPs@ZnSnO₃ sensors toward 50 ppm toluene under different humidity conditions.

Furthermore, the humidity effects were also conducted on the as-fabricated Au₀NPs@ZnSnO₃ and Au₂NPs@ZnSnO₃ sensors toward 50 ppm toluene under their optimum working temperatures. Obviously, the sensor responses gradually decreased with the increasing of humidity from 25% RH to 85% RH, and the results showed the humidity can significantly influence gas sensing properties. Meanwhile, the Au₂NPs@ZnSnO₃ sensor presented good anti-humidity property compared with Au₀NPs@ZnSnO₃ sensor. The possible reasons may be ascribed to the good chemical catalytic effect of AuNPs, which can greatly enhance absorption toluene and oxygen molecules.