SUPPLEMENTARY INFORMATION FOR

Low temperature curable titanium-based sols for visible light photocatalytic coatings for glass and polymeric substrates

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Photocatalyst	Catalyst quantity	Light source	Reaction media	Rto (%) / Rate constant (min ⁻¹)	Ref.	
P25 (TiO ₂)	20 mg			≈40%	- - - 1 -	
	5 mg	-		>20%		
	15 mg	- Xenon lamp	100 mL MB 10 mg/L	≈50%		
TiO ₂ (NaOH 10M)	20 mg	(300W)	120 min	≈90%		
-2(30 mg	-		<90%		
	50 mg	-		<90%		
MnTiO ₃ -TiO ₂ MnTiO ₃	- 5 mg	UV lamp (10W & 100W)	FO me NAD 10-5 NA	>80%	2	
			250 min	0.0059 min ⁻¹		
			nH 2-9	>80%		
			pi 2 3	0.0052 min ⁻¹		
Fe ₃ O ₄ /AC/TiO ₂	100 mg		100 mL MB 100 mg/L			
		UV lamp	60 min		3	
		(1000W)	pH 10	≈91%		
			pH 12	≈98%		
			pH 13	≈84%		
BCTi (TiO ₂)	0.5 wt%	UVA & visible light (8W x 4)	20 mL MB 10 ppm	0.1194 h ⁻¹	4	
			180 min	0.0019 min ⁻¹		
ZnO	4 . 4	Mercury light (150W)	100 mL MB 10-30 mg/L	0.0108 min ⁻¹	_ 5	
2%Fe-7nO	– 1g/L		180 min	0 0106 min ⁻¹		
2/01 € 2110			pH 1-10	0.0100 mm		
dye-TiO2	Unknown	Halogen lamp (150W)	200 mL MB 120 min	40%	6	
Cu-TiSi	Lavered	Visible light Bulbs (5700K. 2mW/cm ² .		≈90%	7	
	Unknown		20 ml MB 10⁻⁵ M	0.16h ⁻¹		
	– number of		13 h	0.0026min ⁻¹		
St-TiSi	substrates	400-750nm)	pH 6.5	≈99%		
	20x20x4mm		1	0.22h ⁻¹		
				0.0036min ⁻¹		
TiO2-NTU-9	1 g/L	(1000W/m ² , 450-	20 mL MB 10-40 mg/L	1000/	8	
			250 min	100%		
		600nm)	pH 2-10			
	Layered	Visible light	15 mL MB 15ppm 500 min	84.3%	9	
	number of	(Philing T5-6W/A				
N-1102-3102	substrates	max. peak 610nm)				
	76 2x25 4x1 1mm					
	70.2223.421.111111			23 41%		
ZnO	Layered			0.001 min ⁻¹	10	
	 Unknown number of substrates 10x60x1.5mm 	Visible light	200 mL MB 5 mg/L	36.62%		
N-ZnO N-ZnO/CNT		(200W tungsten lamp)	250 min	0.0017 min ⁻¹		
				63.69%		
				0.0040 min ⁻¹		
N-TiO ₂ @NH ₂ -MIL-88				0.0305min ⁻¹		
NH ₂ -MIL-88	-	Xe lamp	100 mL MB 50 mg/L	0.0133min ⁻¹	-	
N-TiO ₂	– 20 mg	(300W, 420-	80 min	0.0012min ⁻¹	_ 11	
TiO ₂	_	760nm)		0.0005min ⁻¹	-	
Membrane	Membrane		100 mL MB 10 ppm	49.26%		
NH ₂ -MIL-125	8x8cm	LED (6W, >420nm)	300 min	0.0023 min ⁻¹	12	
2			42, 80 mL 16.5 μM MB 0.22 120 min 36. pH 2 0.22	42.78%	13	
Porous TiO2				0.284h ⁻¹		
	Layered	UV fluorescence		0.0047min ⁻¹		
Dense TiO2	4 substrates	lamp (365 nm, 6W, 800uW/cm²)		36.17%		
	25x/5x1mm			0.222h ⁻¹		
				0.0037min ⁻¹		
Porous TiO ₂	Layered	UV fluorescence	80 mL MB 16.5 μM	42,78%		
	4 substrates	lamp (365 nm, 6W, 800uW/cm ²)	120 min	0.284h ⁻¹	14	
	25x75x1mm		pH 2	0.0047min ⁻¹		
TiO ₂ /NH ₂ BDC	Layered		80 mL MB 16.5 μM	>60%	This work	
	2 substrates	LED lamp (6W,	120 min	0.226 h ⁻¹		
	25x75x1mm	405nm)	pH 2-9	0.0037 min ⁻¹		

Table S1. Data on	photocatalysts	in the	photodegradation	of meth	vlene blue i	(MB)
Tuble St. Dutu on	photoculurysts		photoachiaaation	ormeen	yichic blue	11107

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Fig. S1 Infrared spectra of xerogel samples



Fig. S2 Coatings appearance of coated glass supports: (A) Ti-SG@450, (B) Ti-SG@150 and (C) Ti/L-SG@150.



Fig. S3 Tauc plots of coatings cured at 150 and 450 °C. Dashed lines depict the linear fitting of (A) $(\alpha h\nu)^{0.5}$ for the indirect band gap and (B) $(\alpha h\nu)^2$ for the direct band gap.



Fig. S4 Evolution of the relative dye concentration under visible and UV light (A) in absence of substrate and photocatalyst and (B) in presence of the substrate without photocatalyst.



Fig. S5 (A) N_2 adsorption isotherms and (B) pore size distributions of xerogel samples. Total surface area values and subtracted micropore and mesopore contributions are shown in the table below. These values were subtracted from the t-plot method. Pore size distribution were achieve using DFT methods as detailed in the experimental section.



Fig. S6 Evolution of the relative dye concentration using UV light illumination prior to visible irradiation for Ti/L-SG@150.



Fig. S7 Comparison of reflectance spectra of Ti/L-SG@150 and Ti/L-G@150.