

## Supporting Information

# **Bi<sub>2</sub>S<sub>3</sub>-decorated three-dimensional BiOCl as a Z-scheme heterojunction with highly exposed {001} facets of BiOCl for enhanced visible-light photocatalytic performance**

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### **1.1. Characterizations of photocatalysts**

The patterns of X-ray diffraction (XRD) were collected on a X'Pert Pro (Panalytical, Netherlands) using a Cu-K $\alpha$  X-ray source in the range of 5°-70° (5°·min<sup>-1</sup>). SEM images were recorded on an S-4800 microscope (Hitachi, Japan). Transmission electron

microscope (TEM) and high-resolution transmission electron microscopy (HRTEM) were carried out on a JEM-200 (JEOL, Japan) at an accelerating voltage of 200 kV. X-ray photoelectron spectroscopy (XPS) was recorded on an ESCALAB-250Xi (Thermo Fisher, UK). UV-vis diffuse reflectance spectra (DRS) were collected on a Lambda 750 spectrophotometer (Perkin Elmer, USA) by using BaSO<sub>4</sub> as a reflectance standard. Photoluminescence (PL) spectra were measured using a Fluorolog3 spectrophotometer (HORIBA JOBIN YVON, USA) with excitation wavelength of 320 nm. The electron spin resonance (ESR) measurements were carried out by using a JES-FA200 spectrometer (JEOL, Japan).

## **1.2. Photoelectrochemical measurements**

Electrochemical impedance spectra (EIS) were recorded on an electrochemical system (CHI-660e, Shanghai) in three-electrode quartz cells with 0.5 M Na<sub>2</sub>SO<sub>4</sub> electrolyte solution. Platinum wire and Ag/AgCl electrode were used as the counter electrode and the reference electrode, respectively. Work electrode was prepared by depositing photocatalyst onto the glassy carbon electrode.

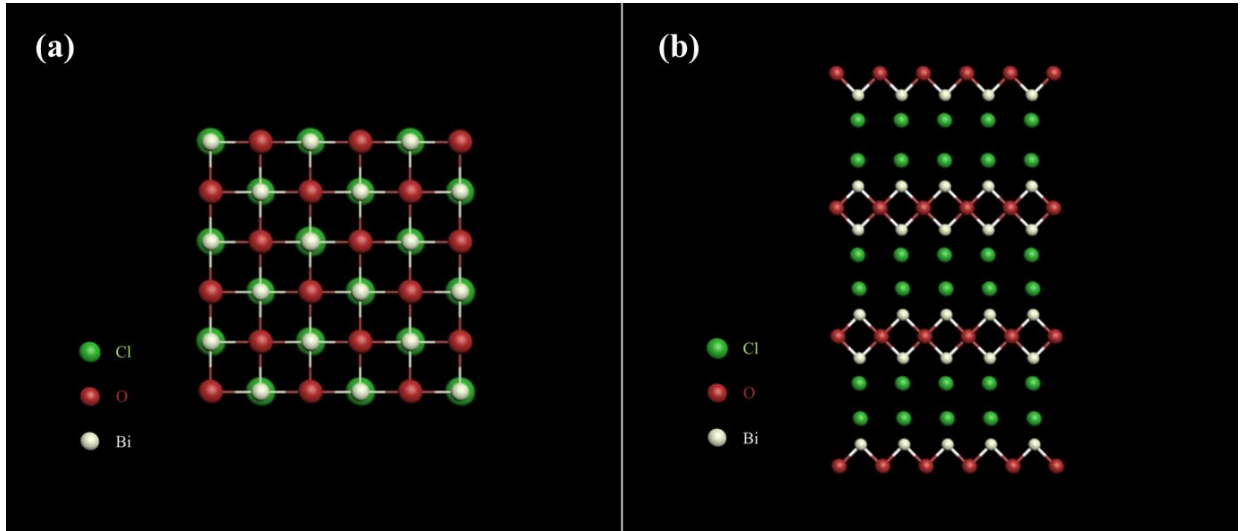


Fig. S1. The structure model of BiOCl crystals: (a)  $\{001\}$  facets and (b)  $\{110\}$  facets

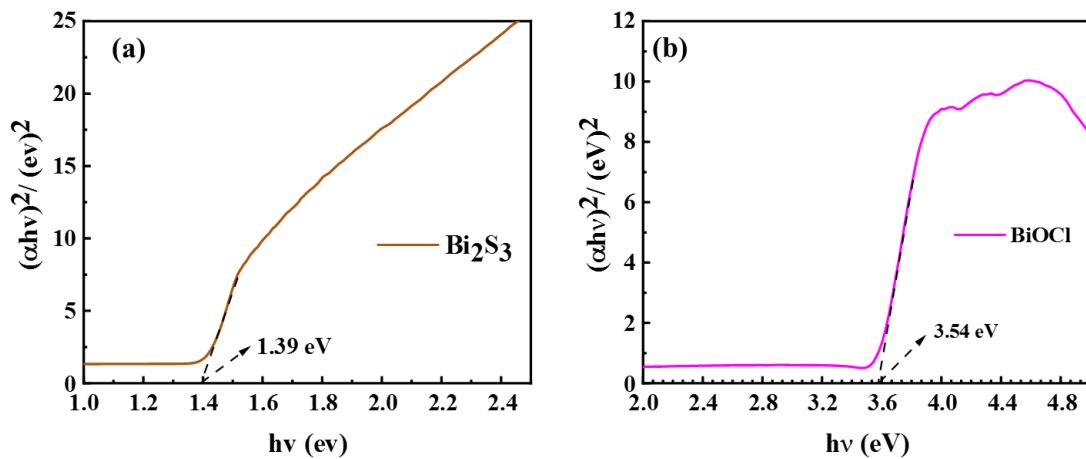


Fig. S2. plots of  $(\alpha h\nu)^2$  versus the photon energy ( $h\nu$ ) for (a) BiOCl and (b) Bi<sub>2</sub>S<sub>3</sub>.

The band gap energies ( $E_g$ ) of the samples were estimated by Kubelka-Munk expression:  $\alpha h\nu = A(h\nu - E_g)^{n/2}$ , where  $E_g$ ,  $\alpha$ ,  $h$ ,  $\nu$ , and  $A$  are band gap energy, light absorption coefficient, Planck's constant, photon frequency, and constant, respectively. As shown in Fig. S2, the band gaps of BiOCl and Bi<sub>2</sub>S<sub>3</sub> are 3.54 and 1.39 eV, respectively.

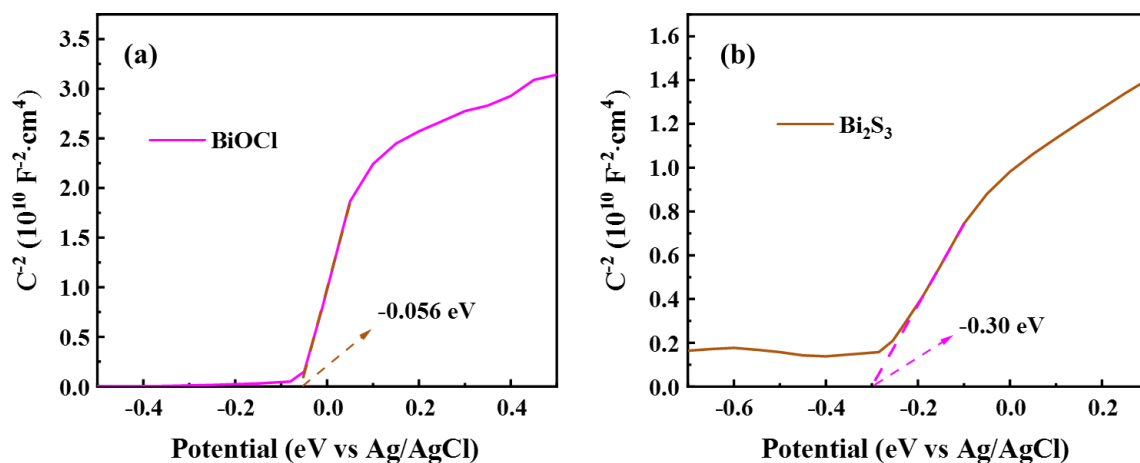


Fig. S3. Mott-Schottky curves of (a) BiOCl and (b) Bi<sub>2</sub>S<sub>3</sub>.

Furthermore, the potential of valence band (VB) and conduction band (CB) of BiOCl and Bi<sub>2</sub>S<sub>3</sub> were extrapolated from the Mott-Schottky curves. As shown in Fig. S3, the value of the  $E_{CB}$  relative to Ag/AgCl electrode can be obtained via extrapolating the linear part of the curve to  $C^{-2} = 0$ . Moreover, the values of  $E_g$ ,  $E_{CB}$  and  $E_{VB}$  of BiOCl and Bi<sub>2</sub>S<sub>3</sub> are displayed in Table S1.

Table S1. The values of  $E_g$ ,  $E_{CB}$  and  $E_{VB}$  of BiOCl and Bi<sub>2</sub>S<sub>3</sub>.

Sample	$E_g$ (eV)	$E_{CB}$ (eV vs. Ag/AgCl)	$E_{CB}$ (eV vs. NHE)	$E_{VB}$ (eV) <sup>a</sup>
BiOCl	3.54	-0.056	0.164	3.704
Bi <sub>2</sub> S <sub>3</sub>	1.39	-0.30	-0.08	1.31

a:  $E_{VB} = E_{CB} + E_g$

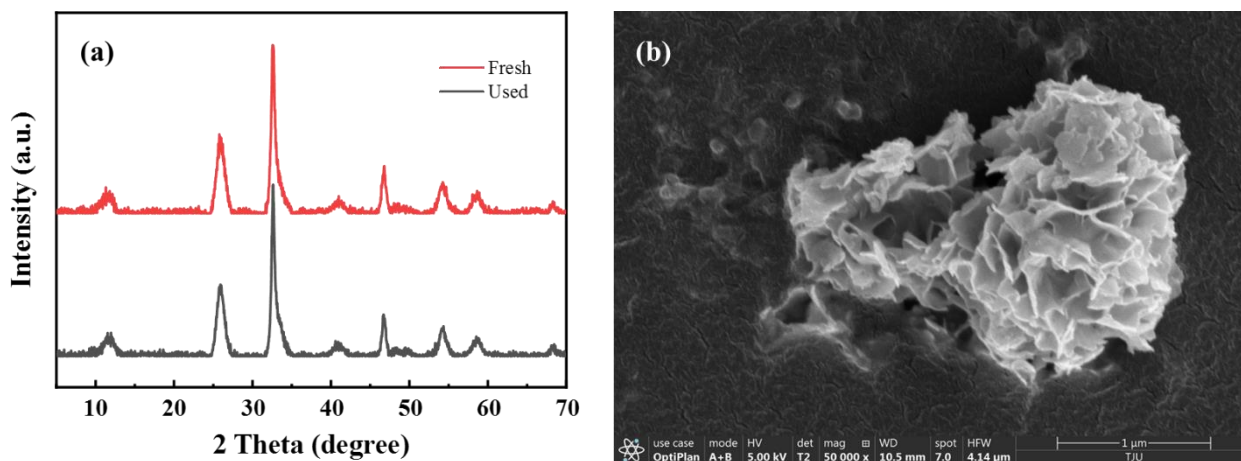


Fig. S4. The XRD patterns and SEM image of BOC-BS-5 after four cyclic experiments.

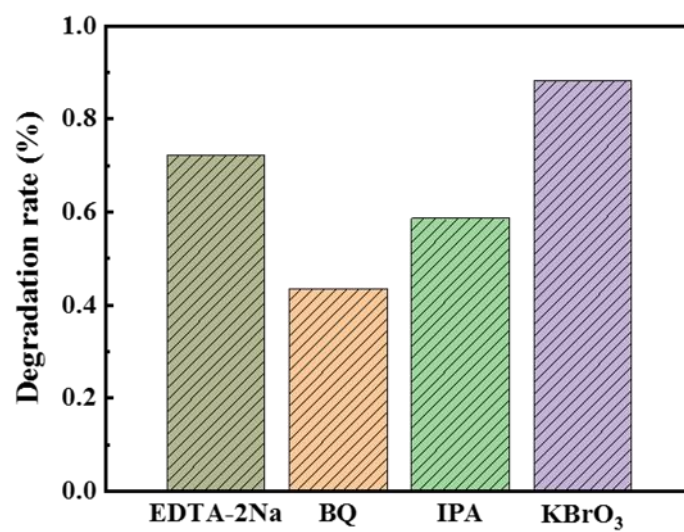
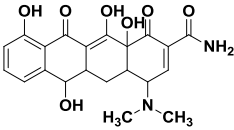
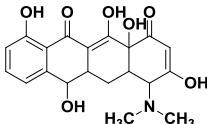
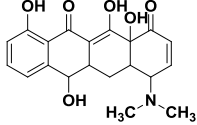
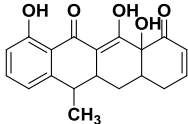
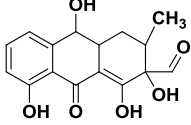
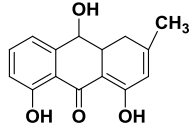
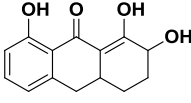
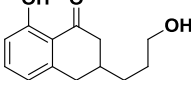


Fig. S5. Effects of different scavengers to photocatalytic degradation.

Table S2. Major intermediates identified by LC-MS.

Compounds	m/z	formula	Proposed structure
TC	445	C <sub>22</sub> H <sub>24</sub> N <sub>2</sub> O <sub>8</sub>	
Product 1	415	C <sub>21</sub> H <sub>24</sub> N <sub>2</sub> O <sub>7</sub>	
Product 2	388	C <sub>20</sub> H <sub>21</sub> NO <sub>7</sub>	
Product 3	372	C <sub>20</sub> H <sub>21</sub> NO <sub>6</sub>	
Product 4	327	C <sub>19</sub> H <sub>18</sub> O <sub>5</sub>	
Product 5	305	C <sub>16</sub> H <sub>16</sub> O <sub>6</sub>	
Product 6	259	C <sub>15</sub> H <sub>14</sub> O <sub>4</sub>	
Product 7	247	C <sub>14</sub> H <sub>14</sub> O <sub>4</sub>	
Product 8	221	C <sub>13</sub> H <sub>16</sub> O <sub>3</sub>	
Product 9	177	C <sub>10</sub> H <sub>8</sub> O <sub>3</sub>	