

## **Supporting Information**

### **Oxidative C-C/C-X Coupling in Organometallic Nickel Complexes: Insights from DFT**

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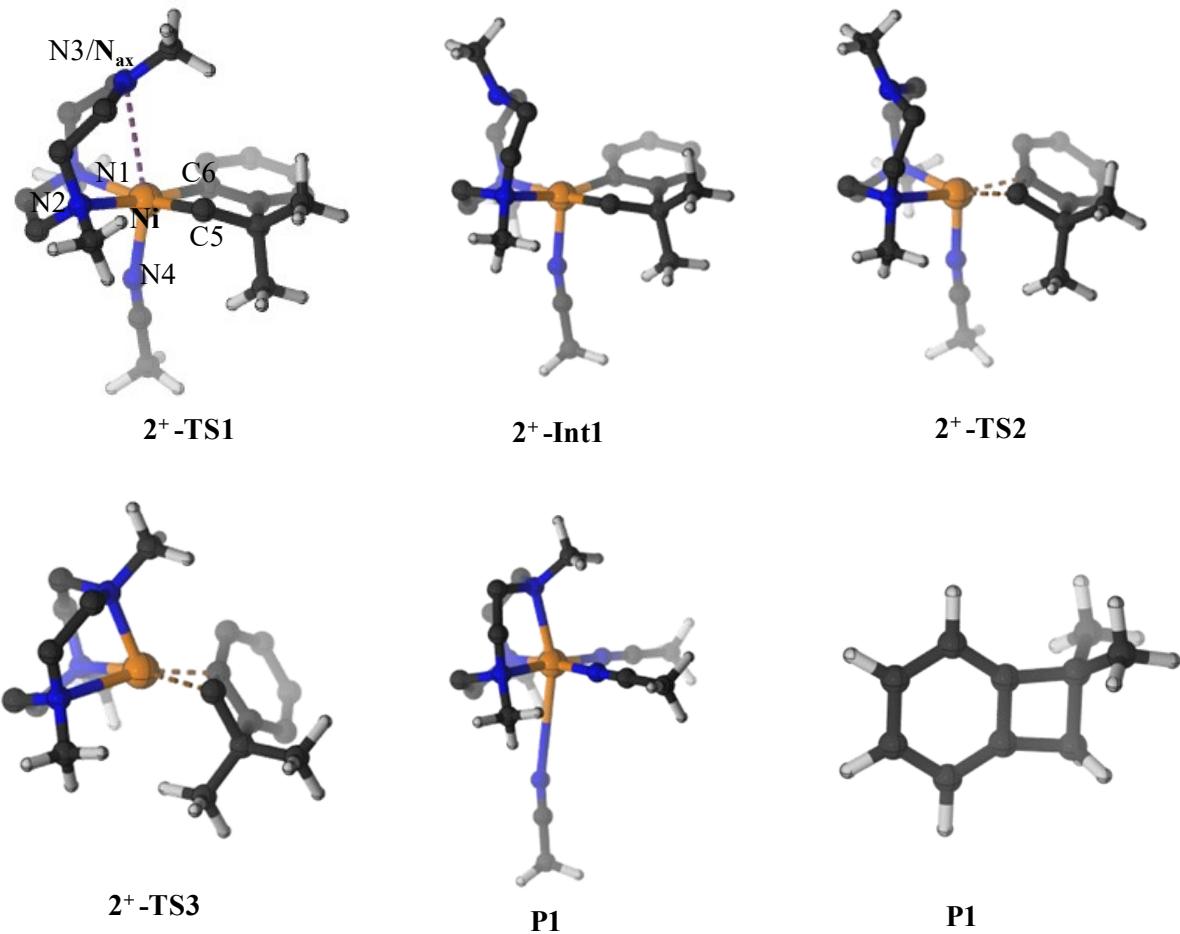
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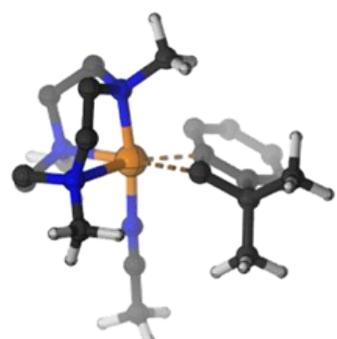
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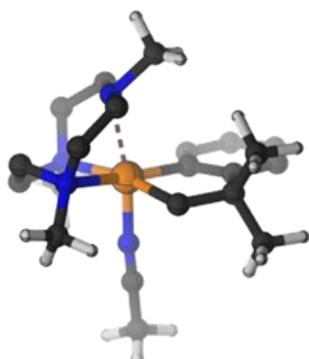
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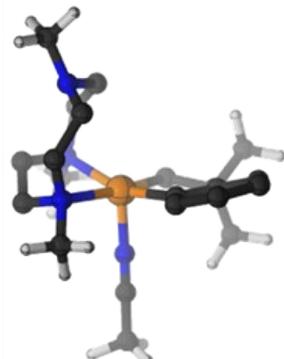
**Fig. S1.** Optimized geometries of  $2^+$  complex intermediates, and transition states.



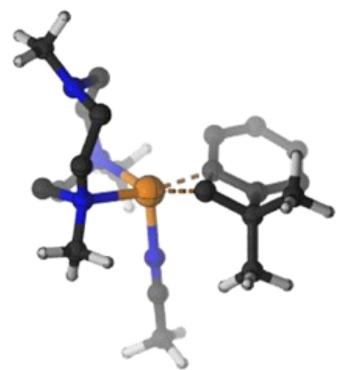
$3^{2+}$ -TS1



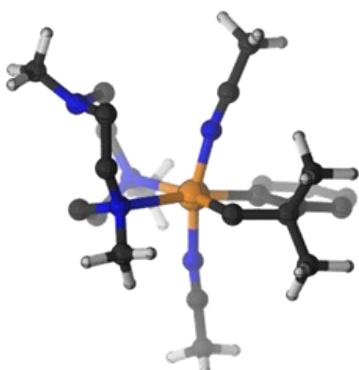
$3^{2+}$ -TS2



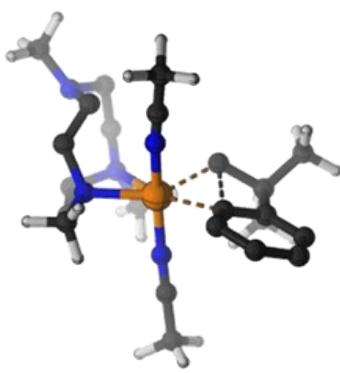
$3^{2+}$ -Int1



$3^{2+}$ -TS3

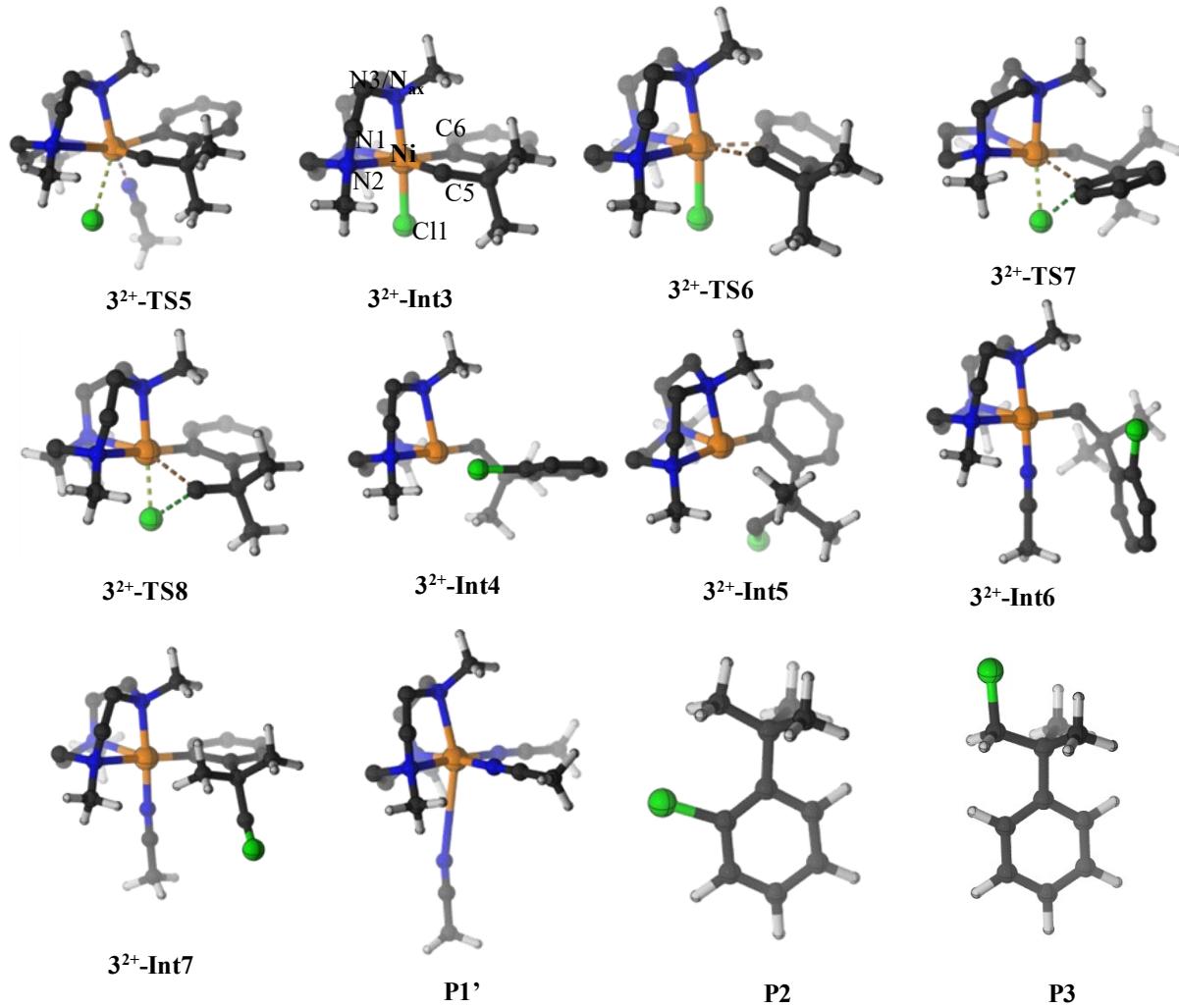


$3^{2+}$ -Int2



$3^{2+}$ -TS4

**Fig. S2.** Optimized geometries of  $3^{2+}$  complex intermediates, and transition states.



**Fig. S3.** Optimized geometries of  $\mathbf{3}^{2+}$  complex intermediates, and transition states.

**Table S1.** NPA charges on selected atoms in the complexes and intermediates.

	Ni	N1	N2	N3	N4/Cl	C5	C6
<b>1</b>	0.332	-0.549	-0.544	-0.585	-	-0.658	-0.199
<b>2<sup>+</sup></b>	0.718	-0.558	-0.554	-0.546	-	-0.565	-0.200
<b>2<sup>+</sup>-Int1</b>	0.686	-0.556	-0.558	-0.584	-0.382	-0.531	-0.264
<b>2<sup>+</sup>-TS2</b>	0.598	-0.580	-0.569	-0.581	-0.400	-0.521	-0.143
<b>2<sup>+</sup>-TS3</b>	0.789	-0.565	-0.568	-0.574	-	-0.588	-0.193
<b>3<sup>2+</sup></b>	0.481	-0.521	-0.524	-0.439	-0.351	-0.320	-0.200
<b>3<sup>2+</sup>-TS1</b>	0.579	-0.545	-0.510	-0.472	-0.371	-0.588	-0.193
<b>3<sup>2+</sup>-TS2</b>	0.511	-0.529	-0.522	-0.522	-0.354	-0.338	-0.041
<b>3<sup>2+</sup>-Int1</b>	0.525	-0.534	-0.543	-0.519	-0.364	-0.332	0.008
<b>3<sup>2+</sup>-TS3</b>	0.573	-0.526	-0.519	-0.519	-0.404	-0.291	-0.052
<b>3<sup>2+</sup>-Int2</b>	0.451	-0.507	-0.507	-0.582	-0.371	-0.280	-0.052
<b>3<sup>2+</sup>-TS4</b>	0.512	-0.471	-0.572	-0.571	-0.389	-0.238	-0.065
<b>3<sup>2+</sup>-TS5</b>	0.494	-0.546	-0.538	-0.470	-0.358/ -0.570	-0.298	-0.002
<b>3<sup>2+</sup>-Int3</b>	0.373	-0.528	-0.530	-0.479	-0.245	-0.359	-0.014
<b>3<sup>2+</sup>-TS6</b>	0.467	-0.549	-0.519	-0.508	-0.327	-0.400	-0.073
<b>3<sup>2+</sup>-TS7</b>	0.413	-0.545	-0.548	-0.524	-0.030	-0.497	-0.022
<b>3<sup>2+</sup>-TS8</b>	0.388	-0.548	-0.554	-0.513	-0.132	-0.249	-0.118
<b>3<sup>2+</sup>-Int4</b>	0.449	-0.569	-0.524	-0.565	-0.128	-0.624	-0.031
<b>3<sup>2+</sup>-Int5</b>	0.566	-0.552	-0.571	-0.524	-0.045	-0.400	-0.184
<b>3<sup>2+</sup>-Int6</b>	0.495	-0.557	-0.565	-0.529	-0.367	-0.594	-0.044
<b>3<sup>2+</sup>-Int7</b>	0.466	-0.554	-0.559	-0.517	-0.357	-0.420	-0.152

For atom numbering see Fig. S1and S3

**Table S2.** Wiberg bond indices for selected bonds in nickel complexes at B3LYP/BS1

	Ni-N1	Ni-N2	Ni-N3	Ni-N4/C1	Ni-C5	Ni-C6	C5/C1-C6/C1
<b>1</b>	0.221	0.233	0.015	-	0.675	0.662	0.159
<b>2<sup>+</sup></b>	0.218	0.224	0.227	-	0.565	0.552	0.270
<b>2<sup>+</sup>-TS1</b>	0.229	0.233	0.011	0.289	0.557	0.559	0.266
<b>2<sup>+</sup>-Int1</b>	0.231	0.233	0.014	0.300	0.569	0.544	0.270
<b>2<sup>+</sup>-TS2</b>	0.194	0.172	0.016	0.305	0.288	0.286	0.702
<b>2<sup>+</sup>-TS3</b>	0.190	0.148	0.181	-	0.268	0.291	0.733
<b>3<sup>2+</sup></b>	0.266	0.254	0.412	0.449	0.659	0.651	0.104
<b>3<sup>2+</sup>-TS1</b>	0.213	0.320	0.453	0.467	0.332	0.339	0.587
<b>3<sup>2+</sup>-TS2</b>	0.266	0.250	0.396	0.433	0.663	0.650	0.010
<b>3<sup>2+</sup>-Int1</b>	0.257	0.272	0.089	0.422	0.668	0.752	0.055
<b>3<sup>2+</sup>-TS3</b>	0.266	0.365	0.096	0.402	0.320	0.634	0.451
<b>3<sup>2+</sup>-Int2</b>	0.248	0.255	0.006	0.439	0.674	0.661	0.064
<b>3<sup>2+</sup>-TS4</b>	0.408	0.045	0.009	0.465	0.349	0.457	0.447
<b>3<sup>2+</sup>-TS5</b>	0.087	0.247	0.325	0.367	0.572	0.680	0.011
<b>3<sup>2+</sup>-Int3</b>	0.244	0.232	0.328	0.794	0.666	0.654	0.114
<b>3<sup>2+</sup>-TS6</b>	0.185	0.294	0.360	0.842	0.317	0.305	0.638
<b>3<sup>2+</sup>-TS7</b>	0.179	0.245	0.253	0.556	0.341	0.785	0.658
<b>3<sup>2+</sup>-TS8</b>	0.255	0.151	0.300	0.632	0.741	0.309	0.521
<b>3<sup>2+</sup>-Int4</b>	0.233	0.290	0.138	-	0.779	-	0.965
<b>3<sup>2+</sup>-Int5</b>	0.252	0.276	0.187	-	-	0.771	1.011
<b>3<sup>2+</sup>-Int6</b>	0.219	0.116	0.293	0.429	0.777	-	1.036
<b>3<sup>2+</sup>-Int7</b>	0.246	0.112	0.324	0.420	-	0.743	0.990

For atom numbering see Fig. S1 and S3