

## Supporting information

### **The promoting effect of support pretreatment with sulfate acid on Ca resistance of CeO<sub>2</sub>/ZrO<sub>2</sub> catalyst for NH<sub>3</sub>-SCR of NO<sub>x</sub> with NH<sub>3</sub>**

Chenglong Li<sup>a</sup>, Zhitao Han<sup>a\*</sup>, Xinxin Wang<sup>a</sup>, Yu Gao<sup>a</sup>, Zhen Wang<sup>a</sup>, Xinxiang Pan<sup>a,b</sup>

a. Marine Engineering College, Dalian Maritime University, Dalian 116026, China;

b. School of Electronic and Information Technology, Guangdong Ocean University,

Zhanjiang 524088, China

Correspondence information: Dr. Zhitao Han, Dalian Maritime University, No.1,

Linghai Road, Dalian 116026, China, hanzt@dlnu.edu.cn

Table. S1 Summary of literature data on NH<sub>3</sub>-SCR catalysts over alkali metal poisoned

Catalysts	Preparation method	Reaction condition	Fresh catalysts T <sub>80%</sub> , °C	Alkali metals precursor and load	Poisoned catalysts		Ref
					T <sub>80%</sub> , °C	Max NO <sub>x</sub> Con, %	
CeO <sub>2</sub> /ZrO <sub>2</sub> -S	Precipitation (ZrO <sub>2</sub> )	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 5 vol. %, N <sub>2</sub> as balance GHSV = 60,000 h <sup>-1</sup>	245-500	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 1 wt.% CaO	255-490	100	This work
				(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 2 wt.% CaO	280-480	100	
	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 5 wt.% CaO			-	30		
CeO <sub>2</sub> /SO <sub>4</sub> <sup>2-</sup> -ZrO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3.5 vol. %, N <sub>2</sub> as balance GHSV = 180,000 h <sup>-1</sup>	360-545	(KNO <sub>3</sub> ) K/Ce = 0.4	325-530	100	1
				(KNO <sub>3</sub> ) K/Ce = 0.8	375-475	86	
				K/Ce was molar ratio	-	-	
V-Ce(SO <sub>4</sub> ) <sub>2</sub> /Ti	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3.5 vol. %, N <sub>2</sub> as balance GHSV = 150,000 h <sup>-1</sup>	275-450	(NaHCO <sub>3</sub> ) 0.5 wt.% Na	-	80	2
V <sub>2</sub> O <sub>5</sub> -WO <sub>3</sub> /TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 240,000 h <sup>-1</sup>	280-500	(CaO) 2 wt.% CaO	-	63	3
				(CaCO <sub>3</sub> ) 2 wt.% CaO	-	55	
				(CaSO <sub>4</sub> ) 2 wt.% CaO	325-480	90	
V <sub>2</sub> O <sub>5</sub> -WO <sub>3</sub> /TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 70,000 h <sup>-1</sup>	240-500	(Ca(OH) <sub>2</sub> ) 0.1 wt.% CaO	265-500	100	4
				(Ca(OH) <sub>2</sub> ) 0.5 wt.% CaO	280-500	100	
				(Ca(OH) <sub>2</sub> ) 1 wt.% CaO	290-500	100	
				(Ca(OH) <sub>2</sub> ) 2 wt.% CaO	-	56	

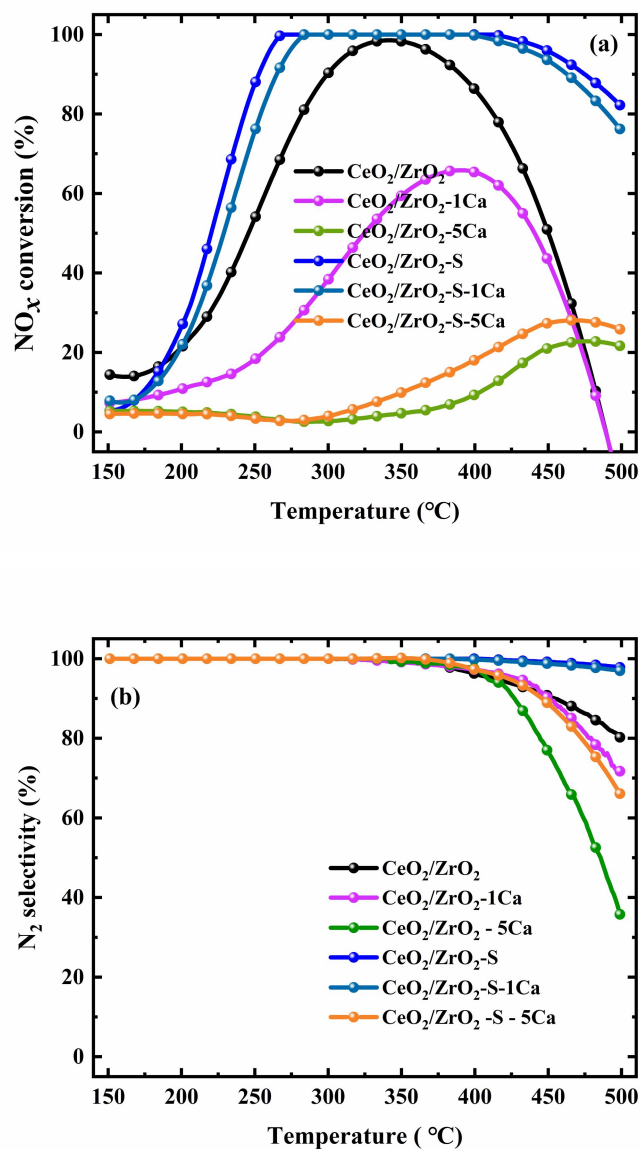
V <sub>2</sub> O <sub>5</sub> -WO <sub>3</sub> /TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 5 vol. %, N <sub>2</sub> as balance GHSV = 100,000 h <sup>-1</sup>	225-400	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) The molar ratio of Ca/V was 1:1	-	73	5
V <sub>2</sub> O <sub>5</sub> -WO <sub>3</sub> -CeO <sub>2</sub> / TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 5 vol. %, N <sub>2</sub> as balance GHSV = 100,000 h <sup>-1</sup>	180-400	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) The molar ratio of Ca/V was 1:1	240-400	96	6
CeO <sub>2</sub> -WO <sub>3</sub>	Impregnation Urea as precipitant	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 120,000 h <sup>-1</sup>	195-450	(Ca(OH) <sub>2</sub> ) 1 wt.% CaO (Ca(OH) <sub>2</sub> ) 3 wt.% CaO (Ca(OH) <sub>2</sub> ) 5 wt.% CaO	220-450 245-450 290-400	100 95 85	7
Ce/TiO <sub>2</sub>	Sol-gel	[NO] = [NH <sub>3</sub> ] = 600 ppm, [O <sub>2</sub> ] = 3 vol. %, Ar as balance GHSV = 108,000 h <sup>-1</sup>	210-400	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) The molar ratio of Ca/Ce was 0.05:1	-	64	8
Ce/TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 600 ppm, [O <sub>2</sub> ] = 3.5 vol. %, N <sub>2</sub> as balance GHSV = 100,000 h <sup>-1</sup>	-	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) The molar ratio of Ca/Ce was x:1(x=0.25, 0.5, 1 and 2)	-	40 16 10 18	9
CeO <sub>2</sub> -Nb <sub>2</sub> O <sub>5</sub> -TiO <sub>2</sub>	Sol-gel	[NO] = [NH <sub>3</sub> ] = 1000 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 90,000 h <sup>-1</sup>	230-490	(KNO <sub>3</sub> ) The molar ratio of K/Ce was 0.2:1	295-425	85	10
Ce/TNTs-eth	Impregnation	[NO] = [NH <sub>3</sub> ] = 600 ppm, [O <sub>2</sub> ] = 3.5 vol. %, N <sub>2</sub> as balance GHSV = 100,000 h <sup>-1</sup>	270-500	(KNO <sub>3</sub> ) K/Ce=0.25 (Ca(NO <sub>3</sub> ) <sub>2</sub> ) Ca/Ce=0.25 (NaNO <sub>3</sub> ) Na/Ce=0.25	330-440 310-470 295-470	88 96 98	11

Mn-Ce-Ti	Precipitation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 18,000 h <sup>-1</sup>	150-330	(Ca(OH) <sub>2</sub> ) 2 wt.% Ca	-	80	12
				(CaCl <sub>2</sub> ) 2 wt.% Ca	-	77	
Mn/TiO <sub>2</sub>	Impregnation	[NO] = [NH <sub>3</sub> ] = 1000 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 75,000 h <sup>-1</sup>	100-200	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 0.5 wt.% Ca	125-200	100	13
				(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 1 wt.% Ca	147-200	100	
				(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 3 wt.% Ca	136-200	100	
				(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 5 wt.% Ca	147-200	97	
Fe-Ce/TiO <sub>2</sub>	Precipitation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 18,000 h <sup>-1</sup>	220-360	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 10 wt.% Ca	130-200	100	14
				(Ca(OH) <sub>2</sub> ) 1 wt.% Ca	-	76	
				(Ca(OH) <sub>2</sub> ) 2 wt.% Ca	-	74	
Fe-Ce/TiO <sub>2</sub>	Precipitation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 18,000 h <sup>-1</sup>	220-360	(Ca(OH) <sub>2</sub> ) 3 wt.% Ca	-	64	15
				(CaCO <sub>3</sub> ) 1 wt.% Ca	250-360	90	
				(CaCO <sub>3</sub> ) 2 wt.% Ca	265-360	89	
Zr-Cu/SZM-5	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 18,000 h <sup>-1</sup>	160-400	(CaCO <sub>3</sub> ) 3 wt.% Ca	270-360	88	16
				(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 1.26 wt.% CaO	170-400	100	
Cu-Ce/H-SPAO-3 4	Impregnation	[NO] = [NH <sub>3</sub> ] = 500 ppm, [O <sub>2</sub> ] = 3 vol. %, N <sub>2</sub> as balance GHSV = 30,000 h <sup>-1</sup>	180-360	(Ca(NO <sub>3</sub> ) <sub>2</sub> ) 1 wt.% Ca	210-360	100	17
				(KNO <sub>3</sub> ) 1 wt.% K	210-360	100	
				(Pb(NO <sub>3</sub> ) <sub>2</sub> ) 1 wt.% Pb	200-360	100	

## References

- 1 S. Gao, P. Wang, X. Chen, H. Wang, Z. Wu, Y. Liu and X. Weng, *Catal. Commun.*, 2014, **43**, 223-226.
- 2 W. Hu, Y. Zhang, S. Liu, C. Zheng, X. Gao, I. Nova and E. Tronconi, *Appl. Catal., B*, 2017, **206**, 449-460.
- 3 X. Li, X. Li, R. T. Yang, J. Mo, J. Li and J. Hao, *Mol. Catal.*, 2017, **434**, 16-24.
- 4 L. Chen, J. Li and M. Ge, *Chem. Eng. J.*, 2011, **170**, 531-537.
- 5 Y. Zheng, Y. Guo, J. Wang, L. Luo and T. Zhu, *J. Phys. Chem. C*, 2021, **125**, 6128-6136.
- 6 Y. Guo, X. Xu, H. Gao, Y. Zheng, L. Luo and T. Zhu, *Catal.*, 2021, **11**, 445-459.
- 7 X. Li, X. Li, J. Li and J. Hao, *Chem. Eng. J.*, 2017, **317**, 70-79.
- 8 S. Liu, R. Guo, S. Wang, W. Pan, P. Sun, M. Li and S. Liu, *J. Taiwan Inst. Chem. Eng.*, 2017, **78**, 290-298.
- 9 H. Wang, X. Chen, S. Gao, Z. Wu, Y. Liu and X. Weng, *Catal. Sci. Technol.*, 2013, **3**, 715-722.
- 10 Y. Jiang, D. Han, L. Yang, Z. Yang, H. Ge, R. Lin and X. Wang, *Process Saf. Environ. Protect.*, 2022, **160**, 876-886.
- 11 P. Wang, H. Wang, X. Chen, Y. Liu, X. Weng and Z. Wu, *J. Mater. Chem. A*, 2015, **3**, 680-690.
- 12 X. Wang, J. Zhou, J. Wang, A. Ding, K. Gui and H. R. Thomas, *React. Kinet. Mech. Catal.*, 2019, **129**, 153-164.
- 13 Y. Liu, T. Gu, Y. Wang, X. Weng and Z. Wu, *Catal. Commun.*, 2012, **18**, 106-109.

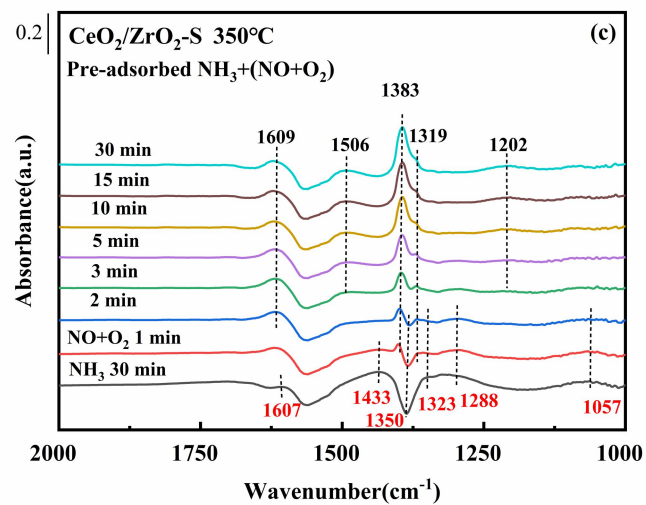
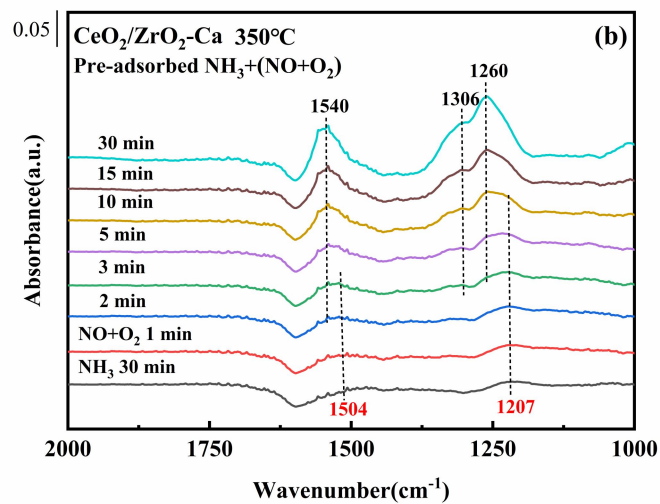
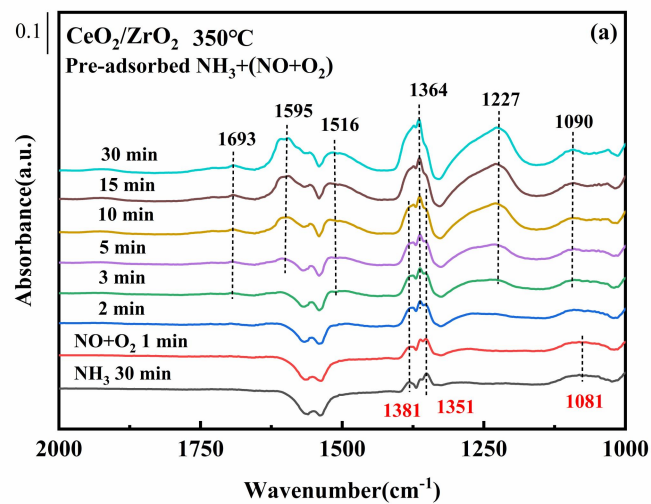
- 14 X. Wang, Q. Fang, J. Wang, K. Gui and H. R. Thomas, *React. Kinet. Mech. Catal.*, 2021, **133**, 245-258.
- 15 X. Wang, Q. Fang, J. Wang, K. Gui and H. R. Thomas, *RSC Adv*, 2020, **10**, 44876-44883.
- 16 H. Xue, T. Meng, F. Liu, X. Guo, S. Wang and D. Mao, *RSC Adv*, 2019, **9**, 38477-38485.
- 17 P. Wang, L. Yan, Y. Gu, S. Kuboon, H. Li, T. Yan, L. Shi and D. Zhang, *Environ. Sci. Technol.*, 2020, **54**, 6396-6405.



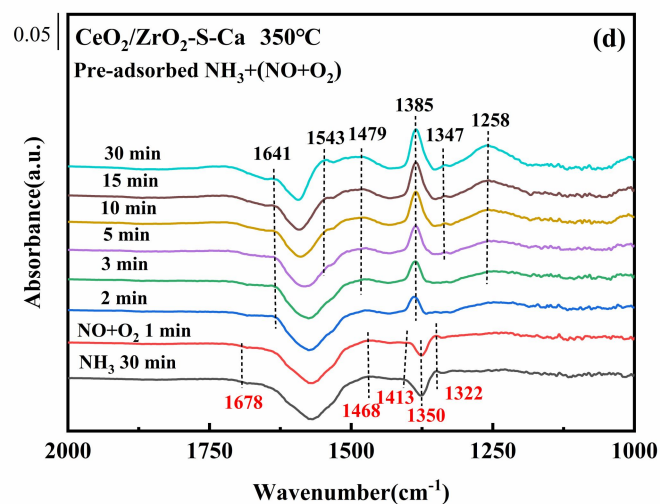
**Fig. S1.** SCR performance test results of CeO<sub>2</sub>/ZrO<sub>2</sub>, CeO<sub>2</sub>/ZrO<sub>2</sub>-1Ca, CeO<sub>2</sub>/ZrO<sub>2</sub>-5Ca, CeO<sub>2</sub>/ZrO<sub>2</sub>-S, CeO<sub>2</sub>/ZrO<sub>2</sub>-S-1Ca, CeO<sub>2</sub>/ZrO<sub>2</sub>-S-5Ca : (a), NO<sub>x</sub> conversion; (b), N<sub>2</sub> selectivity.

Reaction conditions: 0.5 mL catalyst, [NO] = [NH<sub>3</sub>] = 500 ppm, [O<sub>2</sub>] = 5 vol.%,

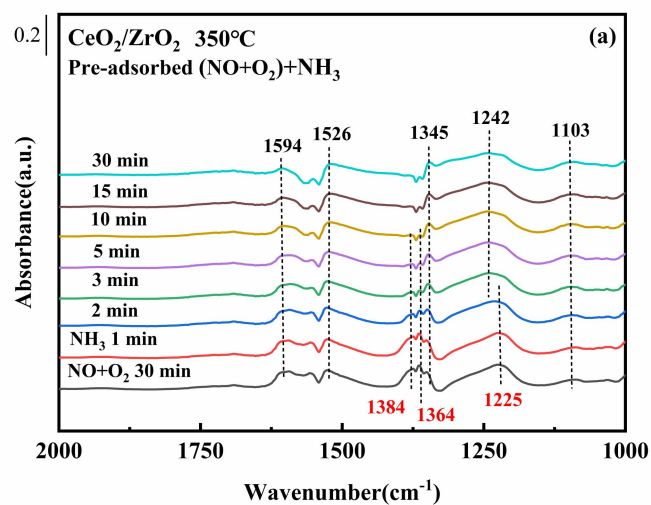
balance with N<sub>2</sub>, total flow rate = 500 mL min<sup>-1</sup> and GHSV = 60,000 h<sup>-1</sup>

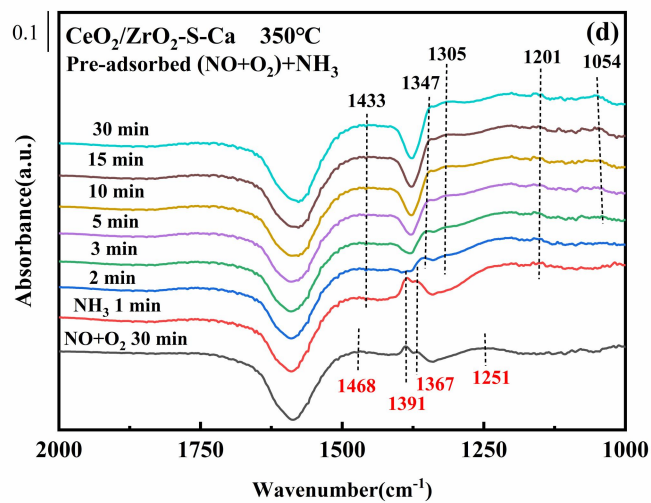
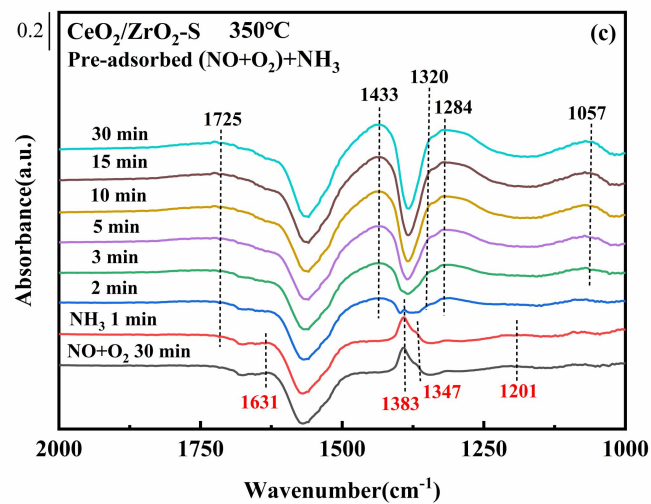
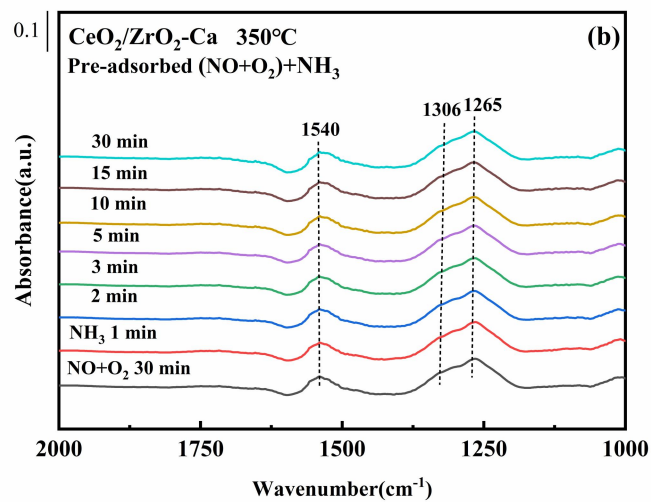






**Fig. S2.** In situ DRIFTS of reaction between  $\text{NO} + \text{O}_2$  and adsorbed  $\text{NH}_3$  species over (a)  $\text{CeO}_2/\text{ZrO}_2$ , (b)  $\text{CeO}_2/\text{ZrO}_2\text{-Ca}$ , (c)  $\text{CeO}_2/\text{ZrO}_2\text{-S}$  and (d)  $\text{CeO}_2/\text{ZrO}_2\text{-S-Ca}$  catalysts at 350 °C.





**Fig. S3.** In situ DRIFTS of reaction between  $\text{NH}_3$  and adsorbed  $\text{NO}+\text{O}_2$  species over (a)  $\text{CeO}_2/\text{ZrO}_2$ , (b)  $\text{CeO}_2/\text{ZrO}_2\text{-Ca}$ , (c)  $\text{CeO}_2/\text{ZrO}_2\text{-S}$  and (d)  $\text{CeO}_2/\text{ZrO}_2\text{-S-Ca}$  catalysts at 350 °C.