

Ba₃Zr₂Cu₄S₉: the first quaternary phase of the Ba–Zr–Cu–S system

Sayani Barman^a, Subhendu Jana^a, Gopabandhu Panigrahi^a, Sweta Yadav^a, Manish K. Niranjan^b, and Jai Prakash^{a,*}

^aDepartment of Chemistry, Indian Institute of Technology Hyderabad, Kandi, Sangareddy, Telangana 502284, India

^bDepartment of Physics, Indian Institute of Technology Hyderabad, Kandi, Sangareddy, Telangana 502284, India

Electronic Supplementary Information (ESI)

The complex dielectric function $\varepsilon(\omega)$ and absorption coefficient $\alpha(\omega)$ are computed in the independent-particle approximation. The imaginary part of the dielectric function $\varepsilon''(\omega)$ is obtained from the joint density of states (JDOS) and the optical momentum matrix in the long wave length limit $\vec{q} = |\vec{k}' - \vec{k}| \rightarrow 0$. The expression for the $\varepsilon''(\omega)$ is given as¹:

$$\varepsilon_{ij}''(\omega) = \frac{4\pi^2 e^2}{V_c} \lim_{q \rightarrow 0} \sum_{c,v,\vec{k}} 2w_{\vec{k}} \delta(\varepsilon_{c\vec{k}} - \varepsilon_{v\vec{k}} - \omega) \times \langle u_{c\vec{k} + \hat{e}_i q} | u_{v\vec{k}} \rangle \langle u_{c\vec{k} + \hat{e}_j q} | u_{v\vec{k}} \rangle^* \quad (\text{S1-1})$$

Here V_c is the primitive cell volume; $w_{\vec{k}}$ is the weight of the k-points; \hat{e}_i and \hat{e}_j are the unit vectors for the three Cartesian directions; indices c and v refer to conduction band (CB) and valence band (VB) states respectively; $u_{c\vec{k}}$ is the cell periodic part of the orbitals at the k-point \vec{k} ; $\varepsilon_{c\vec{k}}$ and $\varepsilon_{v\vec{k}}$ are CB and VB single-electron energy at wave vector \vec{k} . The real part of the dielectric function $\varepsilon'(\omega)$ is obtained from the Kramers-Kronig transformation as follows:

$$\varepsilon'_{ij}(\omega) = 1 + \frac{2}{\pi} P \int_0^\infty \frac{\varepsilon_{ij}''(\omega') \omega'}{\omega'^2 - \omega^2 + i\eta} d\omega' \quad (\text{S1-2})$$

where P is the principal value and η is a small complex shift. The absorption coefficient $\alpha(\omega)$ is obtained from the dielectric function using the expression:

$$\alpha(\omega) = \frac{\sqrt{2}\omega}{c} [\sqrt{\varepsilon'(\omega)^2 + \varepsilon''(\omega)^2} - \varepsilon'(\omega)]^{1/2} \quad (\text{S1-3})$$

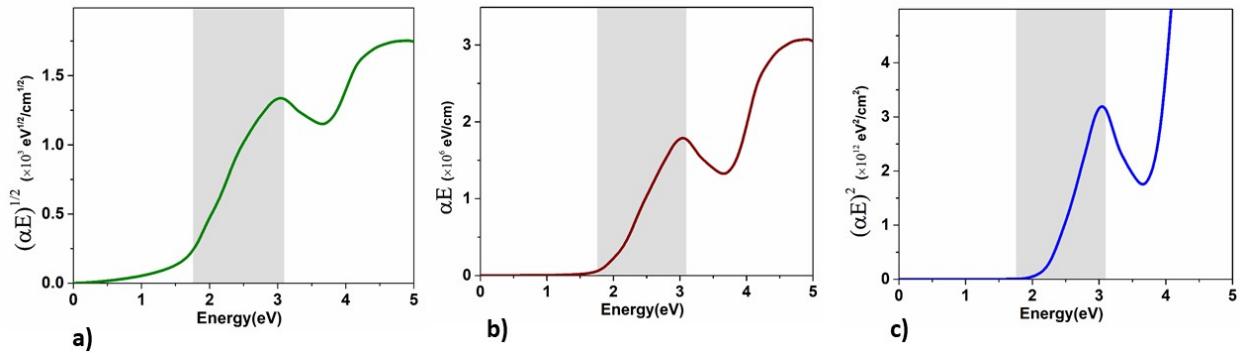


Fig. S1: (a) $(\alpha E)^{1/2}$, (b) (αE) , and (c) $(\alpha E)^2$ as function of energy. α and E are absorption and energy, respectively.

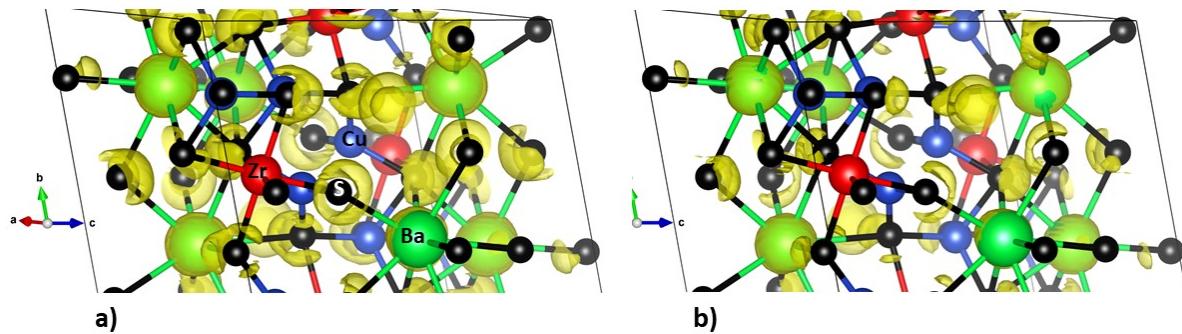


Fig. S2: 3D isosurfaces of the electron localization function (ELF) for $\text{Ba}_3\text{Zr}_2\text{Cu}_4\text{S}_9$ (a) $ELF = 0.85$ and (b) $ELF = 0.87$. The yellow cloud indicates the charge density due to charge transfer. Green, red, blue and black back ball indicate Ba, Zr, Cu and S atoms respectively.

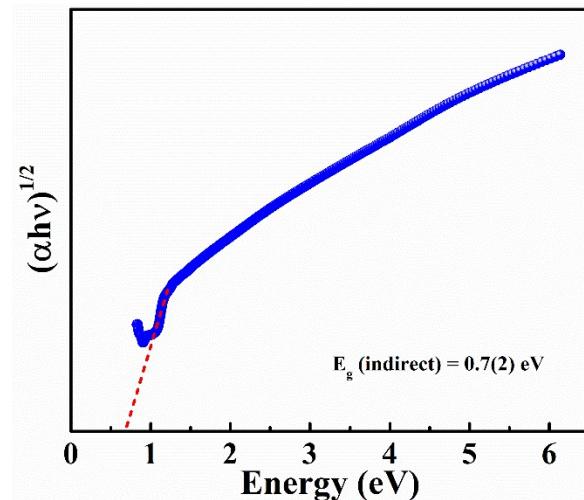


Fig. S3: The indirect bandgap Tauc plot of the $\text{Ba}_3\text{Zr}_2\text{Cu}_4\text{S}_9$ polycrystalline sample.

Table S1: The computed fractional atomic coordinates (x, y, z) in $\text{Ba}_3\text{Zr}_2\text{Cu}_4\text{S}_9$ unit cell.

	x		y		z	
	Cal.	Exp.	Cal.	Exp.	Cal.	Exp.
Ba(1)	0.01747	0.01802	0.27960	0.27871	0.51270	0.51290
Ba(2)	0.30659	0.30795	0.24541	0.24620	0.83308	0.83313
Ba(3)	0.65079	0.65117	0.23106	0.23065	0.15857	0.15715
Zr(1)	0.15555	0.15627	0.48505	0.48563	0.15889	0.15931
Zr(2)	0.33895	0.33931	0.01538	0.01462	0.34447	0.34427
Cu(1)	0.13244	0.13197	0.05723	0.05741	0.11966	0.12001
Cu(2)	0.14526	0.14696	0.75271	0.75455	0.08268	0.08244
Cu(3)	0.38731	0.38616	0.42916	0.42810	0.36928	0.36920
Cu(4)	0.53698	0.53636	0.25820	0.26012	0.60098	0.60063
S(1)	0.05490	0.05218	0.43269	0.43441	0.32647	0.32618
S(2)	0.14670	0.14581	0.74847	0.74816	0.27357	0.27349
S(3)	0.15722	0.15909	0.22901	0.22928	0.04666	0.04698
S(4)	0.30810	0.30684	0.05454	0.05320	0.56118	0.56037
S(5)	0.40068	0.40573	0.94857	0.95002	0.13206	0.13424
S(6)	0.54034	0.54174	0.26557	0.26485	0.41321	0.41189
S(7)	0.54183	0.54166	0.54198	0.54088	0.27358	0.27128
S(8)	0.76866	0.76900	0.44856	0.44921	0.02199	0.02232
S(9)	0.99470	0.99430	0.06239	0.06044	0.26597	0.26729

Table S2: The atomic displacement parameters (\AA^2) for $\text{Ba}_3\text{Zr}_2\text{Cu}_4\text{S}_9$ structure.

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ba(1)	0.01604(11)	0.01584(11)	0.00967(10)	0.00746(8)	0.00387(9)	0.00429(9)
Ba(2)	0.01126(11)	0.01336(10)	0.01206(11)	0.00422(8)	0.00428(9)	0.00623(9)
Ba(3)	0.01455(11)	0.01335(10)	0.01217(11)	-0.00128(8)	0.00156(9)	0.00537(9)
Zr(1)	0.00921(17)	0.00930(15)	0.00891(16)	0.00198(12)	0.00291(13)	0.00377(13)

Zr(2)	0.00941(17)	0.00924(15)	0.00847(16)	0.00265(12)	0.00276(13)	0.00364(13)
Cu(1)	0.0247(3)	0.0472(4)	0.0295(3)	0.0209(3)	0.0164(2)	0.0287(3)
Cu(2)	0.0156(2)	0.0132(2)	0.0176(3)	0.00174(18)	0.0069(2)	0.0039(2)
Cu(3)	0.0179(3)	0.0452(4)	0.0337(3)	0.0143(3)	0.0115(2)	0.0299(3)
Cu(4)	0.0147(2)	0.0161(2)	0.0170(3)	0.00248(18)	0.0054(2)	0.0040(2)
S(1)	0.0111(4)	0.0112(4)	0.0110(4)	0.0020(3)	0.0044(3)	0.0043(3)
S(2)	0.0106(4)	0.0080(4)	0.0091(4)	0.0019(3)	0.0027(3)	0.0031(3)
S(3)	0.0156(4)	0.0108(4)	0.0116(5)	0.0043(3)	0.0042(4)	0.0047(4)
S(4)	0.0099(4)	0.0122(4)	0.0092(4)	0.0034(3)	0.0033(3)	0.0050(3)
S(5)	0.0114(4)	0.0094(4)	0.0105(4)	0.0025(3)	0.0045(3)	0.0037(3)
S(6)	0.0131(4)	0.0113(4)	0.0108(5)	-0.0001(3)	0.0011(4)	0.0052(3)
S(7)	0.0096(4)	0.0104(4)	0.0121(4)	0.0025(3)	0.0034(3)	0.0042(3)
S(8)	0.0098(4)	0.0106(4)	0.0098(4)	0.0030(3)	0.0035(3)	0.0045(3)
S(9)	0.0105(4)	0.0117(4)	0.0104(4)	0.0034(3)	0.0033(3)	0.0039(3)

Table S3: The geometric parameters of Ba₃Zr₂Cu₄S₉ structure.

Ba1—S9 ⁱ	3.0504(10)	Zr1—S7	2.5524(10)
Ba1—S1 ⁱⁱ	3.1485(10)	Zr1—S3	2.5611(10)
Ba1—S6 ⁱ	3.1685(10)	Zr1—S2	2.6355(9)
Ba1—S2 ⁱⁱ	3.1873(10)	Zr1—S8 ⁱ	2.6679(10)
Ba1—S1	3.2492(10)	Zr1—S8 ^{ix}	2.6817(10)
Ba1—S7 ⁱⁱⁱ	3.3015(10)	Zr1—Cu3	2.9334(7)
Ba1—S4	3.3504(9)	Zr1—Cu2 ^x	3.1565(6)
Ba1—Cu4	3.5116(6)	Zr1—Cu4 ⁱⁱⁱ	3.2169(6)
Ba1—S4 ^{iv}	3.5749(9)	Zr1—Cu2	3.2340(6)

Ba1—Cu4 ⁱ	3.7599(5)	Zr2—S9 ⁱ	2.4904(10)
Ba1—Cu3	3.8999(6)	Zr2—S6	2.5483(10)
Ba1—Ba2	4.0550(4)	Zr2—S5 ^{viii}	2.6093(10)
Ba2—S5 ⁱⁱⁱ	3.0530(10)	Zr2—S4	2.6138(10)
Ba2—S2 ⁱⁱ	3.0837(10)	Zr2—S2 ^{viii}	2.6673(9)
Ba2—S3 ^v	3.1007(10)	Zr2—S4 ^{vi}	2.7043(10)
Ba2—S7 ⁱⁱⁱ	3.1054(10)	Zr2—Cu1	2.9499(7)
Ba2—S9 ^{vi}	3.2382(10)	Zr2—Cu4	3.1308(6)
Ba2—S8 ⁱⁱⁱ	3.2514(9)	Zr2—Cu2 ^{viii}	3.2364(6)
Ba2—S4	3.2675(10)	Zr2—Cu4 ^{vi}	3.3018(6)
Ba2—S8 ^v	3.3558(10)	Cu1—S9 ⁱ	2.2412(11)
Ba2—Cu2 ⁱⁱ	3.5581(5)	Cu1—S3	2.2575(11)
Ba2—Cu4	3.6311(6)	Cu1—S5 ^{viii}	2.3057(11)
Ba2—Cu2 ⁱⁱⁱ	3.6407(6)	Cu1—Cu1 ^{xi}	2.7801(13)
Ba3—S1 ^{vii}	3.0225(10)	Cu1—Cu2 ^{viii}	3.0418(8)
Ba3—S5 ^{viii}	3.0800(10)	Cu2—S8 ^{ix}	2.3082(10)
Ba3—S3	3.3132(10)	Cu2—S5	2.3263(11)
Ba3—S6	3.3134(11)	Cu2—S3 ^x	2.3513(12)
Ba3—S5 ^{ix}	3.3194(10)	Cu2—S2	2.3637(12)
Ba3—S7	3.3211(10)	Cu3—S6	2.2443(11)
Ba3—S8	3.3630(9)	Cu3—S1	2.2449(11)
Ba3—S9	3.3898(10)	Cu3—S7	2.2974(11)
Ba3—Cu2 ^{ix}	3.6005(6)	Cu3—Cu3 ⁱⁱⁱ	2.9289(13)
Ba3—Cu1	3.6610(7)	Cu4—S4	2.3356(10)

Ba3—Cu3	3.7574(7)	Cu4—S6	2.3372(12)
Ba3—Zr1 ^{vii}	4.1140(4)	Cu4—S7 ⁱⁱⁱ	2.3583(11)
Zr1—S1	2.5254(10)	Cu4—S2 ⁱⁱⁱ	2.3617(13)
S9 ⁱ —Ba1—S1 ⁱⁱ	152.01(3)	S5 ^{viii} —Zr2—Cu2 ^{viii}	45.34(2)
S9 ⁱ —Ba1—S6 ⁱ	90.13(3)	S4—Zr2—Cu2 ^{viii}	132.58(2)
S1 ⁱⁱ —Ba1—S6 ⁱ	73.99(2)	S2 ^{viii} —Zr2—Cu2 ^{viii}	45.96(3)
S9 ⁱ —Ba1—S2 ⁱⁱ	131.41(3)	S4 ^{vi} —Zr2—Cu2 ^{viii}	100.99(2)
S1 ⁱⁱ —Ba1—S2 ⁱⁱ	68.07(2)	Cu1—Zr2—Cu2 ^{viii}	58.686(18)
S6 ⁱ —Ba1—S2 ⁱⁱ	76.82(2)	Cu4—Zr2—Cu2 ^{viii}	177.962(18)
S9 ⁱ —Ba1—S1	70.90(3)	S9 ⁱ —Zr2—Cu4 ^{vi}	129.90(3)
S1 ⁱⁱ —Ba1—S1	84.33(3)	S6—Zr2—Cu4 ^{vi}	132.41(3)
S6 ⁱ —Ba1—S1	83.75(3)	S5 ^{viii} —Zr2—Cu4 ^{vi}	96.70(2)
S2 ⁱⁱ —Ba1—S1	149.62(2)	S4—Zr2—Cu4 ^{vi}	75.17(2)
S9 ⁱ —Ba1—S7 ⁱⁱⁱ	122.75(3)	S2 ^{viii} —Zr2—Cu4 ^{vi}	45.07(3)
S1 ⁱⁱ —Ba1—S7 ⁱⁱⁱ	71.80(2)	S4 ^{vi} —Zr2—Cu4 ^{vi}	44.40(2)
S6 ⁱ —Ba1—S7 ⁱⁱⁱ	145.54(2)	Cu1—Zr2—Cu4 ^{vi}	133.73(2)
S2 ⁱⁱ —Ba1—S7 ⁱⁱⁱ	86.73(3)	Cu4—Zr2—Cu4 ^{vi}	102.928(15)
S1—Ba1—S7 ⁱⁱⁱ	96.79(2)	Cu2 ^{viii} —Zr2—Cu4 ^{vi}	75.180(15)
S9 ⁱ —Ba1—S4	70.54(3)	S9 ⁱ —Zr2—Ba1	47.92(2)
S1 ⁱⁱ —Ba1—S4	136.36(3)	S6—Zr2—Ba1	72.54(2)
S6 ⁱ —Ba1—S4	135.99(2)	S5 ^{viii} —Zr2—Ba1	133.92(2)
S2 ⁱⁱ —Ba1—S4	86.93(2)	S4—Zr2—Ba1	54.67(2)
S1—Ba1—S4	122.87(2)	S2 ^{viii} —Zr2—Ba1	110.24(2)
S7 ⁱⁱⁱ —Ba1—S4	71.58(2)	S4 ^{vi} —Zr2—Ba1	130.31(2)

S9 ⁱ —Ba1—Cu4	84.32(2)	Cu1—Zr2—Ba1	86.626(15)
S1 ⁱⁱ —Ba1—Cu4	111.92(2)	Cu4—Zr2—Ba1	56.254(12)
S6 ⁱ —Ba1—Cu4	174.09(2)	Cu2 ^{viii} —Zr2—Ba1	125.459(14)
S2 ⁱⁱ —Ba1—Cu4	105.37(2)	Cu4 ^{vi} —Zr2—Ba1	126.781(14)
S1—Ba1—Cu4	96.33(2)	S9 ⁱ —Cu1—S3	125.11(4)
S7 ⁱⁱⁱ —Ba1—Cu4	40.356(19)	S9 ⁱ —Cu1—S5 ^{viii}	110.78(4)
S4—Ba1—Cu4	39.712(17)	S3—Cu1—S5 ^{viii}	114.66(4)
S9 ⁱ —Ba1—S4 ^{iv}	66.98(2)	S9 ⁱ —Cu1—Cu1 ^{xi}	117.19(4)
S1 ⁱⁱ —Ba1—S4 ^{iv}	123.11(2)	S3—Cu1—Cu1 ^{xi}	74.44(4)
S6 ⁱ —Ba1—S4 ^{iv}	65.73(2)	S5 ^{viii} —Cu1—Cu1 ^{xi}	109.14(4)
S2 ⁱⁱ —Ba1—S4 ^{iv}	64.97(2)	S9 ⁱ —Cu1—Zr2	55.32(3)
S1—Ba1—S4 ^{iv}	126.82(2)	S3—Cu1—Zr2	136.66(4)
S7 ⁱⁱⁱ —Ba1—S4 ^{iv}	133.06(2)	S5 ^{viii} —Cu1—Zr2	58.00(3)
S4—Ba1—S4 ^{iv}	70.31(2)	Cu1 ^{xi} —Cu1—Zr2	148.35(4)
Cu4—Ba1—S4 ^{iv}	109.970(18)	S9 ⁱ —Cu1—Cu2 ^{viii}	87.02(3)
S9 ⁱ —Ba1—Cu4 ⁱ	116.36(2)	S3—Cu1—Cu2 ^{viii}	146.95(4)
S1 ⁱⁱ —Ba1—Cu4 ⁱ	63.789(19)	S5 ^{viii} —Cu1—Cu2 ^{viii}	49.25(3)
S6 ⁱ —Ba1—Cu4 ⁱ	38.24(2)	Cu1 ^{xi} —Cu1—Cu2 ^{viii}	84.33(3)
S2 ⁱⁱ —Ba1—Cu4 ⁱ	38.65(2)	Zr2—Cu1—Cu2 ^{viii}	65.367(17)
S1—Ba1—Cu4 ⁱ	117.76(2)	S9 ⁱ —Cu1—Ba3	127.70(4)
S7 ⁱⁱⁱ —Ba1—Cu4 ⁱ	118.46(2)	S3—Cu1—Ba3	62.93(3)
S4—Ba1—Cu4 ⁱ	116.553(18)	S5 ^{viii} —Cu1—Ba3	56.85(3)
Cu4—Ba1—Cu4 ⁱ	143.95(2)	Cu1 ^{xi} —Cu1—Ba3	114.49(3)
S4 ^{iv} —Ba1—Cu4 ⁱ	59.604(16)	Zr2—Cu1—Ba3	83.872(17)

S9 ⁱ —Ba1—Cu3	69.23(2)	Cu2 ^{viii} —Cu1—Ba3	105.665(18)
S1 ⁱⁱ —Ba1—Cu3	97.91(2)	S8 ^{ix} —Cu2—S5	111.17(4)
S6 ⁱ —Ba1—Cu3	118.61(2)	S8 ^{ix} —Cu2—S3 ^x	107.28(4)
S2 ⁱⁱ —Ba1—Cu3	156.54(2)	S5—Cu2—S3 ^x	109.74(4)
S1—Ba1—Cu3	35.129(19)	S8 ^{ix} —Cu2—S2	106.97(4)
S7 ⁱⁱⁱ —Ba1—Cu3	70.68(2)	S5—Cu2—S2	103.82(4)
S4—Ba1—Cu3	91.667(18)	S3 ^x —Cu2—S2	117.82(4)
Cu4—Ba1—Cu3	61.204(13)	S8 ^{ix} —Cu2—Cu1 ^{xii}	148.41(3)
S4 ^{iv} —Ba1—Cu3	136.025(19)	S5—Cu2—Cu1 ^{xii}	48.66(3)
Cu4 ⁱ —Ba1—Cu3	151.696(14)	S3 ^x —Cu2—Cu1 ^{xii}	68.16(3)
S9 ⁱ —Ba1—Ba2	121.406(19)	S2—Cu2—Cu1 ^{xii}	102.16(3)
S1 ⁱⁱ —Ba1—Ba2	86.409(18)	S8 ^{ix} —Cu2—Zr1 ^x	55.88(3)
S6 ⁱ —Ba1—Ba2	125.35(2)	S5—Cu2—Zr1 ^x	137.26(3)
S2 ⁱⁱ —Ba1—Ba2	48.60(2)	S3 ^x —Cu2—Zr1 ^x	53.00(3)
S1—Ba1—Ba2	145.283(19)	S2—Cu2—Zr1 ^x	118.85(3)
S7 ⁱⁱⁱ —Ba1—Ba2	48.621(17)	Cu1 ^{xii} —Cu2—Zr1 ^x	118.44(2)
S4—Ba1—Ba2	51.293(17)	S8 ^{ix} —Cu2—Zr1	54.85(3)
Cu4—Ba1—Ba2	56.812(11)	S5—Cu2—Zr1	130.42(3)
S4 ^{iv} —Ba1—Ba2	85.818(17)	S3 ^x —Cu2—Zr1	119.84(3)
Cu4 ⁱ —Ba1—Ba2	87.253(12)	S2—Cu2—Zr1	53.47(2)
Cu3—Ba1—Ba2	114.267(12)	Cu1 ^{xii} —Cu2—Zr1	155.63(2)
S5 ⁱⁱⁱ —Ba2—S2 ⁱⁱ	143.02(2)	Zr1 ^x —Cu2—Zr1	78.935(15)
S5 ⁱⁱⁱ —Ba2—S3 ^v	91.70(3)	S8 ^{ix} —Cu2—Zr2 ^{xii}	137.67(3)
S2 ⁱⁱ —Ba2—S3 ^v	81.52(2)	S5—Cu2—Zr2 ^{xii}	52.93(3)

S5 ⁱⁱⁱ —Ba2—S7 ⁱⁱⁱ	113.89(3)	S3 ^x —Cu2—Zr2 ^{xii}	115.01(3)
S2 ⁱⁱ —Ba2—S7 ⁱⁱⁱ	92.11(3)	S2—Cu2—Zr2 ^{xii}	54.21(2)
S3 ^v —Ba2—S7 ⁱⁱⁱ	141.90(3)	Cu1 ^{xii} —Cu2—Zr2 ^{xii}	55.948(16)
S5 ⁱⁱⁱ —Ba2—S9 ^{vi}	75.08(2)	Zr1 ^x —Cu2—Zr2 ^{xii}	164.051(19)
S2 ⁱⁱ —Ba2—S9 ^{vi}	68.12(2)	Zr1—Cu2—Zr2 ^{xii}	102.476(19)
S3 ^v —Ba2—S9 ^{vi}	73.38(3)	S8 ^{ix} —Cu2—Ba2 ⁱⁱ	122.46(3)
S7 ⁱⁱⁱ —Ba2—S9 ^{vi}	138.40(3)	S5—Cu2—Ba2 ⁱⁱ	126.23(3)
S5 ⁱⁱⁱ —Ba2—S8 ⁱⁱⁱ	141.25(3)	S3 ^x —Cu2—Ba2 ⁱⁱ	59.17(3)
S2 ⁱⁱ —Ba2—S8 ⁱⁱⁱ	70.54(2)	S2—Cu2—Ba2 ⁱⁱ	58.70(3)
S3 ^v —Ba2—S8 ⁱⁱⁱ	71.65(2)	Cu1 ^{xii} —Cu2—Ba2 ⁱⁱ	83.260(16)
S7 ⁱⁱⁱ —Ba2—S8 ⁱⁱⁱ	70.83(2)	Zr1 ^x —Cu2—Ba2 ⁱⁱ	81.739(14)
S9 ^{vi} —Ba2—S8 ⁱⁱⁱ	128.67(2)	Zr1—Cu2—Ba2 ⁱⁱ	82.649(13)
S5 ⁱⁱⁱ —Ba2—S4	72.95(2)	Zr2 ^{xii} —Cu2—Ba2 ⁱⁱ	82.693(13)
S2 ⁱⁱ —Ba2—S4	90.16(3)	S8 ^{ix} —Cu2—Ba3 ^{ix}	65.16(3)
S3 ^v —Ba2—S4	141.78(2)	S5—Cu2—Ba3 ^{ix}	63.95(3)
S7 ⁱⁱⁱ —Ba2—S4	75.20(2)	S3 ^x —Cu2—Ba3 ^{ix}	83.68(3)
S9 ^{vi} —Ba2—S4	68.93(2)	S2—Cu2—Ba3 ^{ix}	158.40(3)
S8 ⁱⁱⁱ —Ba2—S4	139.79(3)	Cu1 ^{xii} —Cu2—Ba3 ^{ix}	83.265(17)
S5 ⁱⁱⁱ —Ba2—S8 ^v	73.08(2)	Zr1 ^x —Cu2—Ba3 ^{ix}	74.690(14)
S2 ⁱⁱ —Ba2—S8 ^v	143.17(2)	Zr1—Cu2—Ba3 ^{ix}	119.412(16)
S3 ^v —Ba2—S8 ^v	92.92(3)	Zr2 ^{xii} —Cu2—Ba3 ^{ix}	116.872(16)
S7 ⁱⁱⁱ —Ba2—S8 ^v	70.10(2)	Ba2 ⁱⁱ —Cu2—Ba3 ^{ix}	142.848(19)
S9 ^{vi} —Ba2—S8 ^v	144.82(2)	S8 ^{ix} —Cu2—Ba2 ⁱⁱⁱ	64.18(3)
S8 ⁱⁱⁱ —Ba2—S8 ^v	73.15(3)	S5—Cu2—Ba2 ⁱⁱⁱ	56.51(3)

S4—Ba2—S8 ^v	114.44(2)	S3 ^x —Cu2—Ba2 ⁱⁱⁱ	152.97(3)
S5 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱ	123.90(2)	S2—Cu2—Ba2 ⁱⁱⁱ	89.02(3)
S2 ⁱⁱ —Ba2—Cu2 ⁱⁱ	40.92(2)	Cu1 ^{xii} —Cu2—Ba2 ⁱⁱⁱ	104.941(18)
S3 ^v —Ba2—Cu2 ⁱⁱ	40.63(2)	Zr1 ^x —Cu2—Ba2 ⁱⁱⁱ	118.584(16)
S7 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱ	122.16(2)	Zr1—Cu2—Ba2 ⁱⁱⁱ	77.697(13)
S9 ^{vi} —Ba2—Cu2 ⁱⁱ	65.435(19)	Zr2 ^{xii} —Cu2—Ba2 ⁱⁱⁱ	76.943(14)
S8 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱ	63.478(17)	Ba2 ⁱⁱ —Cu2—Ba2 ⁱⁱⁱ	147.717(19)
S4—Ba2—Cu2 ⁱⁱ	122.315(19)	Ba3 ^{ix} —Cu2—Ba2 ⁱⁱⁱ	69.405(10)
S8 ^v —Ba2—Cu2 ⁱⁱ	123.229(19)	S6—Cu3—S1	129.29(4)
S5 ⁱⁱⁱ —Ba2—Cu4	82.79(2)	S6—Cu3—S7	113.83(4)
S2 ⁱⁱ —Ba2—Cu4	104.82(2)	S1—Cu3—S7	110.49(4)
S3 ^v —Ba2—Cu4	173.62(2)	S6—Cu3—Cu3 ⁱⁱⁱ	77.69(3)
S7 ⁱⁱⁱ —Ba2—Cu4	40.036(19)	S1—Cu3—Cu3 ⁱⁱⁱ	112.23(4)
S9 ^{vi} —Ba2—Cu4	108.07(2)	S7—Cu3—Cu3 ⁱⁱⁱ	106.11(4)
S8 ⁱⁱⁱ —Ba2—Cu4	110.859(19)	S6—Cu3—Zr1	140.55(4)
S4—Ba2—Cu4	39.133(17)	S1—Cu3—Zr1	56.51(3)
S8 ^v —Ba2—Cu4	82.422(19)	S7—Cu3—Zr1	56.86(3)
Cu2 ⁱⁱ —Ba2—Cu4	145.726(17)	Cu3 ⁱⁱⁱ —Cu3—Zr1	140.62(4)
S5 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱⁱ	39.46(2)	S6—Cu3—Ba3	61.00(3)
S2 ⁱⁱ —Ba2—Cu2 ⁱⁱⁱ	171.33(2)	S1—Cu3—Ba3	128.00(4)
S3 ^v —Ba2—Cu2 ⁱⁱⁱ	107.14(2)	S7—Cu3—Ba3	61.01(3)
S7 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱⁱ	80.82(2)	Cu3 ⁱⁱⁱ —Cu3—Ba3	119.43(3)
S9 ^{vi} —Ba2—Cu2 ⁱⁱⁱ	114.198(19)	Zr1—Cu3—Ba3	85.192(17)
S8 ⁱⁱⁱ —Ba2—Cu2 ⁱⁱⁱ	111.307(19)	S6—Cu3—Ba1	79.34(3)

S4—Ba2—Cu2 ⁱⁱⁱ	83.227(19)	S1—Cu3—Ba1	56.39(3)
S8 ^v —Ba2—Cu2 ⁱⁱⁱ	38.254(17)	S7—Cu3—Ba1	166.53(3)
Cu2 ⁱⁱ —Ba2—Cu2 ⁱⁱⁱ	147.72(2)	Cu3 ⁱⁱⁱ —Cu3—Ba1	78.71(2)
Cu4—Ba2—Cu2 ⁱⁱⁱ	66.530(12)	Zr1—Cu3—Ba1	111.396(18)
S5 ⁱⁱⁱ —Ba2—Ba1	125.944(19)	Ba3—Cu3—Ba1	128.082(19)
S2 ⁱⁱ —Ba2—Ba1	50.84(2)	S4—Cu4—S6	107.17(4)
S3 ^v —Ba2—Ba1	132.33(2)	S4—Cu4—S7 ⁱⁱⁱ	111.96(4)
S7 ⁱⁱⁱ —Ba2—Ba1	52.914(18)	S6—Cu4—S7 ⁱⁱⁱ	114.87(4)
S9 ^{vi} —Ba2—Ba1	88.060(18)	S4—Cu4—S2 ⁱⁱⁱ	105.84(4)
S8 ⁱⁱⁱ —Ba2—Ba1	88.703(19)	S6—Cu4—S2 ⁱⁱⁱ	114.36(4)
S4—Ba2—Ba1	53.144(18)	S7 ⁱⁱⁱ —Cu4—S2 ⁱⁱⁱ	102.37(4)
S8 ^v —Ba2—Ba1	122.990(18)	S4—Cu4—Zr2	54.83(3)
Cu2 ⁱⁱ —Ba2—Ba1	91.701(12)	S6—Cu4—Zr2	53.18(3)
Cu4—Ba2—Ba1	54.029(10)	S7 ⁱⁱⁱ —Cu4—Zr2	140.11(3)
Cu2 ⁱⁱⁱ —Ba2—Ba1	120.497(11)	S2 ⁱⁱⁱ —Cu4—Zr2	117.29(3)
S1 ^{vii} —Ba3—S5 ^{viii}	136.36(3)	S4—Cu4—Zr1 ⁱⁱⁱ	136.19(3)
S1 ^{vii} —Ba3—S3	139.54(3)	S6—Cu4—Zr1 ⁱⁱⁱ	116.55(3)
S5 ^{viii} —Ba3—S3	73.76(2)	S7 ⁱⁱⁱ —Cu4—Zr1 ⁱⁱⁱ	51.74(3)
S1 ^{vii} —Ba3—S6	84.96(3)	S2 ⁱⁱⁱ —Cu4—Zr1 ⁱⁱⁱ	53.77(2)
S5 ^{viii} —Ba3—S6	69.46(2)	Zr2—Cu4—Zr1 ⁱⁱⁱ	164.875(19)
S3—Ba3—S6	83.81(2)	S4—Cu4—Zr2 ^{vi}	54.10(3)
S1 ^{vii} —Ba3—S5 ^{ix}	120.63(3)	S6—Cu4—Zr2 ^{vi}	115.82(3)
S5 ^{viii} —Ba3—S5 ^{ix}	82.33(3)	S7 ⁱⁱⁱ —Cu4—Zr2 ^{vi}	129.23(3)
S3—Ba3—S5 ^{ix}	83.48(2)	S2 ⁱⁱⁱ —Cu4—Zr2 ^{vi}	53.09(2)

S6—Ba3—S5 ^{ix}	151.33(2)	Zr2—Cu4—Zr2 ^{vi}	77.071(15)
S1 ^{vii} —Ba3—S7	73.07(2)	Zr1 ⁱⁱⁱ —Cu4—Zr2 ^{vi}	101.417(19)
S5 ^{viii} —Ba3—S7	125.12(3)	S4—Cu4—Ba1	66.42(3)
S3—Ba3—S7	66.55(2)	S6—Cu4—Ba1	87.26(3)
S6—Ba3—S7	70.00(2)	S7 ⁱⁱⁱ —Cu4—Ba1	65.03(3)
S5 ^{ix} —Ba3—S7	126.74(2)	S2 ⁱⁱⁱ —Cu4—Ba1	158.35(3)
S1 ^{vii} —Ba3—S8	72.54(2)	Zr2—Cu4—Ba1	75.900(14)
S5 ^{viii} —Ba3—S8	148.85(3)	Zr1 ⁱⁱⁱ —Cu4—Ba1	116.765(16)
S3—Ba3—S8	89.10(2)	Zr2 ^{vi} —Cu4—Ba1	119.930(16)
S6—Ba3—S8	135.50(2)	S4—Cu4—Ba2	62.00(3)
S5 ^{ix} —Ba3—S8	69.80(2)	S6—Cu4—Ba2	156.30(3)
S7—Ba3—S8	66.92(2)	S7 ⁱⁱⁱ —Cu4—Ba2	57.89(3)
S1 ^{vii} —Ba3—S9	69.28(2)	S2 ⁱⁱⁱ —Cu4—Ba2	89.28(3)
S5 ^{viii} —Ba3—S9	72.55(2)	Zr2—Cu4—Ba2	115.644(16)
S3—Ba3—S9	146.21(2)	Zr1 ⁱⁱⁱ —Cu4—Ba2	78.048(14)
S6—Ba3—S9	82.12(2)	Zr2 ^{vi} —Cu4—Ba2	76.293(13)
S5 ^{ix} —Ba3—S9	94.56(2)	Ba1—Cu4—Ba2	69.159(10)
S7—Ba3—S9	134.57(2)	S4—Cu4—Ba1 ^{vii}	125.05(3)
S8—Ba3—S9	121.97(2)	S6—Cu4—Ba1 ^{vii}	57.04(3)
S1 ^{vii} —Ba3—Cu2 ^{ix}	84.39(2)	S7 ⁱⁱⁱ —Cu4—Ba1 ^{vii}	122.46(3)
S5 ^{viii} —Ba3—Cu2 ^{ix}	119.86(2)	S2 ⁱⁱⁱ —Cu4—Ba1 ^{vii}	57.45(3)
S3—Ba3—Cu2 ^{ix}	103.53(2)	Zr2—Cu4—Ba1 ^{vii}	85.071(14)
S6—Ba3—Cu2 ^{ix}	169.26(2)	Zr1 ⁱⁱⁱ —Cu4—Ba1 ^{vii}	79.806(13)
S5 ^{ix} —Ba3—Cu2 ^{ix}	39.022(18)	Zr2 ^{vi} —Cu4—Ba1 ^{vii}	84.072(13)

S7—Ba3—Cu2 ^{ix}	105.428(19)	Ba1—Cu4—Ba1 ^{vii}	143.95(2)
S8—Ba3—Cu2 ^{ix}	38.527(17)	Ba2—Cu4—Ba1 ^{vii}	146.653(19)
S9—Ba3—Cu2 ^{ix}	95.367(19)	Cu3—S1—Zr1	75.64(3)
S1 ^{vii} —Ba3—Cu1	147.94(2)	Cu3—S1—Ba3 ⁱ	135.52(4)
S5 ^{viii} —Ba3—Cu1	38.81(2)	Zr1—S1—Ba3 ⁱ	95.31(3)
S3—Ba3—Cu1	37.352(19)	Cu3—S1—Ba1 ⁱⁱ	112.46(4)
S6—Ba3—Cu1	63.51(2)	Zr1—S1—Ba1 ⁱⁱ	104.25(3)
S5 ^{ix} —Ba3—Cu1	91.38(2)	Ba3 ⁱ —S1—Ba1 ⁱⁱ	111.97(3)
S7—Ba3—Cu1	89.79(2)	Cu3—S1—Ba1	88.48(3)
S8—Ba3—Cu1	125.89(2)	Zr1—S1—Ba1	158.09(4)
S9—Ba3—Cu1	109.34(2)	Ba3 ⁱ —S1—Ba1	85.43(2)
Cu2 ^{ix} —Ba3—Cu1	126.905(15)	Ba1 ⁱⁱ —S1—Ba1	95.67(3)
S1 ^{vii} —Ba3—Cu3	88.63(2)	Cu4 ⁱⁱⁱ —S2—Cu2	115.15(4)
S5 ^{viii} —Ba3—Cu3	90.39(2)	Cu4 ⁱⁱⁱ —S2—Zr1	79.94(3)
S3—Ba3—Cu3	60.52(2)	Cu2—S2—Zr1	80.42(3)
S6—Ba3—Cu3	36.329 (19)	Cu4 ⁱⁱⁱ —S2—Zr2 ^{xii}	81.83(3)
S5 ^{ix} —Ba3—Cu3	143.81(2)	Cu2—S2—Zr2 ^{xii}	79.83(3)
S7—Ba3—Cu3	37.235(19)	Zr1—S2—Zr2 ^{xii}	144.15(4)
S8—Ba3—Cu3	103.765(18)	Cu4 ⁱⁱⁱ —S2—Ba2 ⁱⁱ	164.46(4)
S9—Ba3—Cu3	117.03(2)	Cu2—S2—Ba2 ⁱⁱ	80.38(4)
Cu2 ^{ix} —Ba3—Cu3	141.819(14)	Zr1—S2—Ba2 ⁱⁱ	103.23(3)
Cu1—Ba3—Cu3	62.843(15)	Zr2 ^{xii} —S2—Ba2 ⁱⁱ	102.58(3)
S1 ^{vii} —Ba3—Zr1 ^{vii}	37.676(19)	Cu4 ⁱⁱⁱ —S2—Ba1 ⁱⁱ	83.90(4)
S5 ^{viii} —Ba3—Zr1 ^{vii}	153.84(2)	Cu2—S2—Ba1 ⁱⁱ	160.62(4)

S3—Ba3—Zr1 ^{vii}	128.239(19)	Zr1—S2—Ba1 ⁱⁱ	100.63(3)
S6—Ba3—Zr1 ^{vii}	121.540(18)	Zr2 ^{xii} —S2—Ba1 ⁱⁱ	107.82(3)
S5 ^{ix} —Ba3—Zr1 ^{vii}	86.257(18)	Ba2 ⁱⁱ —S2—Ba1 ⁱⁱ	80.56(2)
S7—Ba3—Zr1 ^{vii}	80.329(18)	Cu1—S3—Cu2 ^x	116.19(4)
S8—Ba3—Zr1 ^{vii}	40.257(17)	Cu1—S3—Zr1	125.63(5)
S9—Ba3—Zr1 ^{vii}	85.043(18)	Cu2 ^x —S3—Zr1	79.84(3)
Cu2 ^{ix} —Ba3—Zr1 ^{vii}	47.732(10)	Cu1—S3—Ba2 ^{xiii}	131.17(4)
Cu1—Ba3—Zr1 ^{vii}	165.577(12)	Cu2 ^x —S3—Ba2 ^{xiii}	80.20(3)
Cu3—Ba3—Zr1 ^{vii}	112.214(13)	Zr1—S3—Ba2 ^{xiii}	101.71(3)
S1—Zr1—S7	94.61(3)	Cu1—S3—Ba3	79.72(3)
S1—Zr1—S3	92.91(3)	Cu2 ^x —S3—Ba3	159.89(4)
S7—Zr1—S3	90.77(3)	Zr1—S3—Ba3	101.35(3)
S1—Zr1—S2	86.79(3)	Ba2 ^{xiii} —S3—Ba3	79.92(2)
S7—Zr1—S2	90.27(3)	Cu4—S4—Zr2	78.25(3)
S3—Zr1—S2	178.93(3)	Cu4—S4—Zr2 ^{vi}	81.50(3)
S1—Zr1—S8 ⁱ	93.62(3)	Zr2—S4—Zr2 ^{vi}	97.86(3)
S7—Zr1—S8 ⁱ	171.28(3)	Cu4—S4—Ba2	78.87(3)
S3—Zr1—S8 ⁱ	91.70(3)	Zr2—S4—Ba2	153.73(3)
S2—Zr1—S8 ⁱ	87.30(3)	Zr2 ^{vi} —S4—Ba2	91.35(3)
S1—Zr1—S8 ^{ix}	173.94(3)	Cu4—S4—Ba1	73.86(3)
S7—Zr1—S8 ^{ix}	90.44(3)	Zr2—S4—Ba1	85.80(3)
S3—Zr1—S8 ^{ix}	90.36(3)	Zr2 ^{vi} —S4—Ba1	153.84(3)
S2—Zr1—S8 ^{ix}	89.85(3)	Ba2—S4—Ba1	75.563(19)
S8 ⁱ —Zr1—S8 ^{ix}	81.18(3)	Cu4—S4—Ba1 ^{iv}	175.35(4)

S1—Zr1—Cu3	47.85(3)	Zr2—S4—Ba1 ^{iv}	98.81(3)
S7—Zr1—Cu3	48.91(2)	Zr2 ^{vi} —S4—Ba1 ^{iv}	95.41(3)
S3—Zr1—Cu3	81.11(3)	Ba2—S4—Ba1 ^{iv}	104.77(2)
S2—Zr1—Cu3	99.42(3)	Ba1—S4—Ba1 ^{iv}	109.69(2)
S8 ⁱ —Zr1—Cu3	139.79(3)	Cu1 ^{xii} —S5—Cu2	82.09(4)
S8 ^{ix} —Zr1—Cu3	137.89(3)	Cu1 ^{xii} —S5—Zr2 ^{xii}	73.47(3)
S1—Zr1—Cu2 ^x	103.17(3)	Cu2—S5—Zr2 ^{xii}	81.73(3)
S7—Zr1—Cu2 ^x	134.29(3)	Cu1 ^{xii} —S5—Ba2 ⁱⁱⁱ	164.71(4)
S3—Zr1—Cu2 ^x	47.16(3)	Cu2—S5—Ba2 ⁱⁱⁱ	84.03(3)
S2—Zr1—Cu2 ^x	131.93(3)	Zr2 ^{xii} —S5—Ba2 ⁱⁱⁱ	98.21(3)
S8 ⁱ —Zr1—Cu2 ^x	45.75(2)	Cu1 ^{xii} —S5—Ba3 ^{xii}	84.34(3)
S8 ^{ix} —Zr1—Cu2 ^x	75.44(2)	Cu2—S5—Ba3 ^{xii}	163.94(4)
Cu3—Zr1—Cu2 ^x	122.24(2)	Zr2 ^{xii} —S5—Ba3 ^{xii}	102.59(3)
S1—Zr1—Cu4 ⁱⁱⁱ	78.88(3)	Ba2 ⁱⁱⁱ —S5—Ba3 ^{xii}	110.29(3)
S7—Zr1—Cu4 ⁱⁱⁱ	46.51(2)	Cu1 ^{xii} —S5—Ba3 ^{ix}	102.50(3)
S3—Zr1—Cu4 ⁱⁱⁱ	134.64(3)	Cu2—S5—Ba3 ^{ix}	77.03(3)
S2—Zr1—Cu4 ⁱⁱⁱ	46.29(3)	Zr2 ^{xii} —S5—Ba3 ^{ix}	158.74(4)
S8 ⁱ —Zr1—Cu4 ⁱⁱⁱ	132.96(2)	Ba2 ⁱⁱⁱ —S5—Ba3 ^{ix}	80.51(2)
S8 ^{ix} —Zr1—Cu4 ⁱⁱⁱ	102.36(2)	Ba3 ^{xii} —S5—Ba3 ^{ix}	97.67(3)
Cu3—Zr1—Cu4 ⁱⁱⁱ	60.275(18)	Cu3—S6—Cu4	111.52(4)
Cu2 ^x —Zr1—Cu4 ⁱⁱⁱ	177.437(18)	Cu3—S6—Zr2	120.78(4)
S1—Zr1—Cu2	130.87(3)	Cu4—S6—Zr2	79.58(3)
S7—Zr1—Cu2	98.06(3)	Cu3—S6—Ba1 ^{vii}	128.93(4)
S3—Zr1—Cu2	133.89(3)	Cu4—S6—Ba1 ^{vii}	84.72(3)

S2—Zr1—Cu2	46.11(3)	Zr2—S6—Ba1 ^{vii}	109.47(3)
S8 ⁱ —Zr1—Cu2	74.28(2)	Cu3—S6—Ba3	82.67(3)
S8 ^{ix} —Zr1—Cu2	44.73(2)	Cu4—S6—Ba3	164.90(4)
Cu3—Zr1—Cu2	136.42(2)	Zr2—S6—Ba3	97.92(3)
Cu2 ^x —Zr1—Cu2	101.065(15)	Ba1 ^{vii} —S6—Ba3	82.09(2)
Cu4 ⁱⁱⁱ —Zr1—Cu2	76.386(15)	Cu3—S7—Cu4 ⁱⁱⁱ	83.41(4)
S1—Zr1—Ba3 ⁱ	47.02(2)	Cu3—S7—Zr1	74.23(3)
S7—Zr1—Ba3 ⁱ	134.07(2)	Cu4 ⁱⁱⁱ —S7—Zr1	81.75(3)
S3—Zr1—Ba3 ⁱ	71.01(2)	Cu3—S7—Ba2 ⁱⁱⁱ	164.80(4)
S2—Zr1—Ba3 ⁱ	108.08(2)	Cu4 ⁱⁱⁱ —S7—Ba2 ⁱⁱⁱ	82.07(3)
S8 ⁱ —Zr1—Ba3 ⁱ	54.54(2)	Zr1—S7—Ba2 ⁱⁱⁱ	99.21(3)
S8 ^{ix} —Zr1—Ba3 ⁱ	129.93(2)	Cu3—S7—Ba1 ⁱⁱⁱ	101.93(4)
Cu3—Zr1—Ba3 ⁱ	86.028(15)	Cu4 ⁱⁱⁱ —S7—Ba1 ⁱⁱⁱ	74.62(3)
Cu2 ^x —Zr1—Ba3 ⁱ	57.576(12)	Zr1—S7—Ba1 ⁱⁱⁱ	156.35(4)
Cu4 ⁱⁱⁱ —Zr1—Ba3 ⁱ	124.078(14)	Ba2 ⁱⁱⁱ —S7—Ba1 ⁱⁱⁱ	78.46(2)
Cu2—Zr1—Ba3 ⁱ	125.243(14)	Cu3—S7—Ba3	81.75(3)
S9 ⁱ —Zr2—S6	96.15(3)	Cu4 ⁱⁱⁱ —S7—Ba3	163.35(4)
S9 ⁱ —Zr2—S5 ^{viii}	94.40(3)	Zr1—S7—Ba3	101.33(3)
S6—Zr2—S5 ^{viii}	90.00(3)	Ba2 ⁱⁱⁱ —S7—Ba3	113.22(3)
S9 ⁱ —Zr2—S4	93.02(3)	Ba1 ⁱⁱⁱ —S7—Ba3	101.15(3)
S6—Zr2—S4	93.50(3)	Cu2 ^{ix} —S8—Zr1 ^{vii}	78.37(3)
S5 ^{viii} —Zr2—S4	171.41(3)	Cu2 ^{ix} —S8—Zr1 ^{ix}	80.42(3)
S9 ⁱ —Zr2—S2 ^{viii}	86.71(3)	Zr1 ^{vii} —S8—Zr1 ^{ix}	98.82(3)
S6—Zr2—S2 ^{viii}	176.98(3)	Cu2 ^{ix} —S8—Ba2 ⁱⁱⁱ	174.04(4)

S5 ^{viii} —Zr2—S2 ^{viii}	88.78(3)	Zr1 ^{vii} —S8—Ba2 ⁱⁱⁱ	98.24(3)
S4—Zr2—S2 ^{viii}	87.33(3)	Zr1 ^{ix} —S8—Ba2 ⁱⁱⁱ	95.37(3)
S9 ⁱ —Zr2—S4 ^{vi}	173.33(3)	Cu2 ^{ix} —S8—Ba2 ^{xiii}	77.57(3)
S6—Zr2—S4 ^{vi}	88.76(3)	Zr1 ^{vii} —S8—Ba2 ^{xiii}	152.17(3)
S5 ^{viii} —Zr2—S4 ^{vi}	90.11(3)	Zr1 ^{ix} —S8—Ba2 ^{xiii}	90.77(3)
S4—Zr2—S4 ^{vi}	82.14(3)	Ba2 ⁱⁱⁱ —S8—Ba2 ^{xiii}	106.85(2)
S2 ^{viii} —Zr2—S4 ^{vi}	88.48(3)	Cu2 ^{ix} —S8—Ba3	76.31(3)
S9 ⁱ —Zr2—Cu1	47.74(3)	Zr1 ^{vii} —S8—Ba3	85.20(3)
S6—Zr2—Cu1	83.75(3)	Zr1 ^{ix} —S8—Ba3	155.08(3)
S5 ^{viii} —Zr2—Cu1	48.53(2)	Ba2 ⁱⁱⁱ —S8—Ba3	108.46(3)
S4—Zr2—Cu1	139.68(3)	Ba2 ^{xiii} —S8—Ba3	75.700(19)
S2 ^{viii} —Zr2—Cu1	97.50(3)	Cu1 ^{vii} —S9—Zr2 ^{vii}	76.94(3)
S4 ^{vi} —Zr2—Cu1	137.74(3)	Cu1 ^{vii} —S9—Ba1 ^{vii}	135.35(4)
S9 ⁱ —Zr2—Cu4	103.02(3)	Zr2 ^{vii} —S9—Ba1 ^{vii}	94.79(3)
S6—Zr2—Cu4	47.24(3)	Cu1 ^{vii} —S9—Ba2 ^{vi}	105.43(4)
S5 ^{viii} —Zr2—Cu4	134.75(3)	Zr2 ^{vii} —S9—Ba2 ^{vi}	102.60(3)
S4—Zr2—Cu4	46.92(2)	Ba1 ^{vii} —S9—Ba2 ^{vi}	119.15(3)
S2 ^{viii} —Zr2—Cu4	133.07(3)	Cu1 ^{vii} —S9—Ba3	88.66(3)
S4 ^{vi} —Zr2—Cu4	77.03(2)	Zr2 ^{vii} —S9—Ba3	156.72(4)
Cu1—Zr2—Cu4	123.15(2)	Ba1 ^{vii} —S9—Ba3	82.59(2)
S9 ⁱ —Zr2—Cu2 ^{viii}	78.87(3)	Ba2 ^{vi} —S9—Ba3	98.79(3)
S6—Zr2—Cu2 ^{viii}	133.60 (3)		

Symmetry codes: (i) $x-1, y, z$; (ii) $-x, -y+1, -z+1$; (iii) $-x+1, -y+1, -z+1$; (iv) $-x, -y, -z+1$; (v) $x, y, z+1$; (vi) $-x+1, -y, -z+1$; (vii) $x+1, y, z$; (viii) $x, y-1, z$; (ix) $-x+1, -y+1, -z$; (x) $-x, -y+1, -z$; (xi) $-x, -y, -z+1$.

$y, -z$; (xii) $x, y+1, z$; (xiii) $x, y, z-1$.

Reference:

- 1 M. Gajdoš, K. Hummer, G. Kresse, J. Furthmüller, and F. Bechstedt, *Phys. Rev. B* 2006, **73**, 045112.