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Supporting Information

An efficient catalytic method for the borohydride reaction of esters

using diethylzinc as precatalyst

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General Information.

Unless otherwise stated, all reactions were performed using standard Schlenk techniques or inside an Etelux MB 200G glovebox. All solvents were refluxed over the appropriate drying agent and distilled prior to use. Commercially available chemicals are purchased from Minrell Chemical Reagent Company or Anaiji Reagent Company, and stored and used according to the instructions. The NMR spectrum was provided by the Bruker Ascend II 400 spectrometer. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, p = quintet, m = multiplet), integration and coupling constants (Hz). It should be noted that the boiling point of one of the reaction products, CH₃OBpin, is close to the boiling point of the catalyst solvent. Although we have tried various methods to remove the solvent, which resulted in the absence of CH₃OBpin to some extent. So in order to calculate the accuracy of the yield, our final spectrum retained solvent peaks (Figure S1). In addition, only the solvent peak at δ 0.8 ppm is marked in the ¹H NMR spectrum. Interestingly, in the subsequent experiments to explore the reaction mechanism, when zinc hydride was used to catalyze the hydroboration of dimethyl carbonate, satisfactory yield was obtained, which was the same as that of ZnEt₂. Furthermore, ¹H NMR showed that pure product was obtained (Figure S7.2, compared with Figure S7.1). This would also apply to other types of esters in this reaction.

In the NMR spectra we used the following markers:

* - MeOBPin: ¹H NMR (400 MHz, CDCl₃) δ 3.53 (s, 3 H, MeOBPin), 1.18 (s, 12 H, BOCMe₂). ¹³C NMR (101 MHz, CDCl₃) δ 81.7, 51.6, 23.6.

Hexane (ZnEt₂, 1.0 M in hexane): ¹H NMR (400 MHz, CDCl₃) δ 1.15(s, 4 H, CH₂CH₂CH₃), 1.17 (s, 4 H, CH₂CH₂CH₃), 0.82 (m, 6 H, CH₃). ¹³C NMR (101 MHz, CDCl₃) δ 30.6, 21.6, 13.1.

General Procedure for Hydroboration of Carbonates.

Carbonate (1.0 mmol, 1 equiv), HBpin (3.2 mmol, 3.2 equiv) and ZnEt₂ (0.07 mmol, 7 mol%, 1.0 M in hexane) were placed in a 10 mL Schlenk flask equipped with a magnetic stirring bar in a glove box. The reaction mixture was stirred at 80 °C for 12 h and then subjected to NMR analysis. The reaction process was monitored by ¹H NMR, ¹³C NMR and the results showed the disappearance of the carbonate protons and the appearance of the new product and CH₃OBpin, indicating the completion of the reaction.

BpinO OBpin 2a

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.87 (s, 4H, OCH₂), 1.18 (s, 24H, BOCMe₂).

¹³C NMR (101 MHz, CDCl₃) δ 81.7, 64.1, 23.6, 23.5.

2b

BpinO

NMR yield: 50%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.20 (q, J = 6.1 Hz, 1H, CH), 3.67 (dd, J = 5.6, 1.7 Hz, 2H, OCH₂), 1.18 (s, 24H, BOCMe₂), 1.10 (d, J = 6.4 Hz, 3H, Me).

¹³C NMR (101 MHz, CDCl₃) δ 81.7, 81.6, 69.3, 68.1, 23.6, 23.5, 17.5.



NMR yield: 87%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.05-3.93 (m, 1H, C*H*), 3.76-3.67 (m, 2H, OC*H*₂), 1.48-1.37 (m, 2H, C*H*₂), 1.17 (s, 24H, BOC*Me*₂), 0.86 (d, *J* = 7.4 Hz, 3H, *Me*).

¹³C NMR (101 MHz, CDCl₃) *δ* 81.7, 81.6, 74.4, 66.7, 24.3, 23.6, 23.5, 8.4.



NMR yield: 87%, colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 5.79-5.70 (m, 1H, CH=CH₂), 5.27 (d, J = 17.2 Hz, 1H, CH=CH₂), 5.09 (d, J = 10.6 Hz, 1H, CH=CH₂), 4.60-4.54 (m, 1H, CH), 3.79-3.67 (m, 2H, OCH₂), 1.18 (s, 24H, BOCMe₂). ¹³C NMR (101 MHz, CDCl₃) δ 134.4, 115.4, 81.7, 73.8, 66.8, 23.6, 23.5.



NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.23 (q, J = 4.1 Hz, 1H, CH), 3.87-3.77 (m, 2H, OCH₂), 3.39-3.34 (m, 2H, CH₂), 3.27 (s, 3H, CH₃), 1.17 (s, 24H, BOCMe₂).

¹³C NMR (101 MHz, CDCl₃) *δ* 81.8, 81.7, 71.9, 71.8, 64.5, 57.6, 23.6, 23.5.

BpinO

OBpin 2f

NMR yield: 91%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.27-4.18 (m, 1H, CH),

3.93-3.87 (m, 2H, OCH₂), 3.86 -3.79 (m, 2H, CH₂), 1.22 (s, 36H, BOCMe₂).

¹³C NMR (101 MHz, CDCl₃) *δ* 81.8, 81.7, 72.6, 64.1, 23.6, 23.5.

BpinO OBpin 2g

NMR yield: 95%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.89-3.83 (m, 4H, OCH₂),

1.76 (dd, *J* = 8.7, 4.1 Hz, 2H, C*H*₂), 1.17 (s, 24H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) δ 81.8, 81.7, 60.5, 32.3, 23.7, 23.6.

BpinO OBpin 2h

NMR yield: 56%, colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 3.56 (s, 4H, OCH₂), 1.17

(s, 24H, BOC*Me*₂), 0.80 (s, 6H, *Me*)

¹³C NMR (101 MHz, CDCl₃) *δ* 81.7, 81.6, 69.3, 35.7, 23.6, 23.5, 19.8.

∠^{OBpin} 2i

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.51 (s, 3H, *Me*), 1.19 (s, 12H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) δ 81.8, 51.6, 23.6.

∽^{OBpin} 2i

NMR yield: 92%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.82 (q, *J* = 7.1, 5.7 Hz,

2H, OCH₂), 1.18 (s, 12H, BOCMe₂), 1.15-1.09 (m, 3H, Me).

¹³C NMR (101 MHz, CDCl₃) δ 81.6, 59.6, 23.6, 16.2.

OBpin 21

NMR yield: 98%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 5.84 (q, J = 8.7, 7.0 Hz, 1H, CH=CH₂), 5.21 (d, J = 17.1 Hz, 1H, CH=CH₂), 5.05 (d, J = 10.5 Hz, 1H, CH=CH₂), 4.29 (d, J = 4.9 Hz, 2H, CH₂), 3.53 (s, 3H, OCH₃), 1.18 (s, 12H, BOCMe₂) ¹³C NMR (101 MHz, CDCl₃) δ 134.4, 113.9, 81.8, 81.7, 64.5, 23.6, 23.5.

OBpin 2m

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.19 (t, *J* = 7.1 Hz, 2H, Ar–*H*), 7.05-6.92 (m, 3H, Ar–*H*), 1.21 (d, *J* = 24.9 Hz, 12H, BOC*Me*₂). ¹³C NMR (101 MHz, CDCl₃) δ 152.4, 128.3, 122.1, 118.5, 82.5, 81.8, 23.6.



NMR yield: 72%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.26-7.16 (m, 5H, *Ph*CH₂OBpin), 4.83 (s, 2H, OC*H*₂), 1.17 (s, 12H, BOC*Me*₂). ¹³C NMR (101 MHz, CDCl₃) δ 138.3, 127.3, 126.3, 125.7, 81.9, 81.7, 65.7, 23.6.

Optimization of Esters Hydroboration

	0 + 2.1HE	3pin	Cat.	2 OBpin
	3a			4a
Entry	Cat (mol %)	<i>T</i> (°C)	<i>t</i> (h)	Yield $(\%)^b$
1	None	60	8	0
2	5	25	8	32
3	5	60	8	86
4	5	60	10	92
5	5	60	12	99
6	3	60	12	90

Table S1. Optimization of ethyl acetate hydroboration^a

^{*a*}Ethyl Acetate (**3a**) (1 mmol, 1 equiv), HBpin (2.1 mmol, 2.1 equiv), catalysts (0.05 mmol, 5 mol %, 1.0 M in hexane). ^{*b*}The reaction was monitored by ¹H NMR spectroscopy.

General Procedure for Hydroboration of Esters.

In a glove box, the catalyst ZnEt₂ (0.05 mmol, 5 mol%, 1.0 M in hexane) was added to a solution of esters (1.0 mmol, 1 equiv) and HBpin (2.1 mmol, 2.1 equiv) in a 10 ml Schlenk flask. The reaction mixture was stirred at 60 °C for 12 h and then subjected to NMR analysis. The progress of the reaction was monitored by ¹H NMR and ¹³C NMR, which indicated the completion of the reaction by the disappearance of the ester proton and appearance of new products.

OBpin 4a

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.83 (q, *J* = 7.0 Hz, 2H, MeCH₂OBpin), 1.18 (d, *J* = 2.5 Hz, 12H, BOC*Me*₂), 1.14 (s, 3H, *Me*CH₂OBpin). ¹³C NMR (101 MHz, CDCl₃) δ 81.6, 59.6, 23.6, 16.2.

OBpin + OBpin 4b

NMR yield: 77%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.25-7.12 (m, 5H, Ar-*H*), 4.82 (s, 2H, PhCH₂OBpin), 3.79 (q, *J* = 7.0 Hz, 2H, MeCH₂OBpin), 1.14 (d, *J* = 4.4 Hz, 24H, BOC*Me*₂), 1.11 (s, 3H, *Me*CH₂OBpin).

¹³C NMR (101 MHz, CDCl₃) δ 138.3, 127.2, 126.3, 125.7, 81.8, 81.5, 65.6, 59.6, 23.7, 23.6, 16.2.



NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.91 (q, *J* = 4.8 Hz, 1H, *Cy*OBpin), 3.82 (q, *J* = 7.1 Hz, 2H, MeCH₂OBpin), 1.79-1.73 (m, 2H, *Cy*OBpin), 1.64 (dt, *J* = 8.1, 4.4 Hz, 2H, *Cy*OBpin), 1.43 (dt, *J* = 11.8, 4.5 Hz, 2H, *Cy*OBpin), 1.30 (d, *J* = 9.0 Hz, 1H, *Cy*OBpin), 1.24 (d, *J* = 11.2 Hz, 3H, *Cy*OBpin), 1.18 (s, 24H, BOC*Me*₂), 1.14 (s, 3H, *Me*CH₂OBpin).

¹³C NMR (101 MHz, CDCl₃) *δ* 81.6, 81.4, 71.6, 59.6, 33.2, 24.5, 23.6, 22.8, 16.2.

NMR yield: 95%, colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.28-7.14 (m, 5H, Ar-H), 4.84 (s, 2H, PhCH₂OBpin), 1.17 (s, 12H, BOCMe₂).

¹³C NMR (101 MHz, CDCl₃) δ 138.2, 127.2, 126.3, 125.7, 81.9, 65.6, 23.6.

4e

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.28-7.12 (m, 7H, Ar-*H*), 7.00 (d, J = 8.7 Hz, 2H, Ar-*H*), 6.95 (d, J = 7.4 Hz, 1H, Ar-*H*), 4.84 (s, 2H, PhCH₂OBpin), 1.20 (d, J = 20.9 Hz, 24H, BOCMe₂).

¹³C NMR (101 MHz, CDCl₃) *δ* 152.4, 138.2, 128.3, 127.3, 126.4, 125.7, 122.1, 118.5, 82.5, 81.9, 65.7, 23.6, 23.5.

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.29 (d, *J* = 7.8 Hz, 2H, Ar-*H*), 7.13-7.03 (m, 3H, Ar-*H*), 1.30 (d, *J* = 24.7 Hz, 12H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) δ 154.9, 128.3, 122.1, 118.5, 82.6, 81.9, 23.6, 23.5.

`OBpin 4g

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 3.54 (d, J = 6.5 Hz, 2H, CH₂OBpin), 1.74 (dt, J = 13.3, 6.7Hz, 1H, CHMe₂), 1.18 (d, J = 2.9 Hz, 12H, BOCMe₂), 0.82 (d, J = 6.7 Hz, 6H, CHMe₂).

¹³C NMR (101 MHz, CDCl₃) δ 81.6, 70.4, 28.8, 23.8, 23.6, 17.7.

OBpin **4h**

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.90 (s, 1H, C=CH₂), 4.75 (s, 1H, C=CH₂), 4.18 (s, 2H, CH₂OBpin), 1.65 (s, 3H, *Me*C=CH₂), 1.18 (s, 12H, BOC*Me*₂).

¹³C NMR (101MHz, CDCl₃) *δ* 141.7, 108.9, 81.8, 67.3, 23.6, 23.5, 17.9.

OBpin OBpin 4i

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 4.10 (q, J = 6.0 Hz, 1H, CH₃CHOBpin), 3.76 (td, J = 6.2, 3.4 Hz, 2H, CH₂), 1.56 (dd, J = 15.5, 8.7 Hz, 2H, CH₂CH₂OBpin), 1.48-1.42 (m, 2H, CH₂CH₂OBpin), 1.17 (d, J = 1.7 Hz, 24H, BOCMe₂), 1.11 (d, J = 6.3 Hz, 3H, CH₃CHOBpin).

¹³C NMR (101 MHz, CDCl₃) *δ* 81.6, 81.4, 69.6, 63.8, 33.1, 26.5, 23.6, 23.5, 21.6.

OBpin OBpin

4j

¹³C NMR (101 MHz, CDCl₃) δ 81.6, 63.9, 30.4, 24.3, 23.6.

OBpin OBpin 4k

NMR yield: 92%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.33 (dd, *J* = 9.3, 4.0 Hz, 2H, Ar-*H*), 7.20-7.14 (m, 2H, Ar-*H*), 4.88 (s, 4H, PhC*H*₂OBpin), 1.20-1.12 (m, 24H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) δ 136.5, 127.5, 127.3, 82.8, 64.1, 24.6.



NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.22 (dd, J = 9.8, 4.3 Hz, 2H, Ar-H), 6.91 (t, J = 8.7 Hz, 2H, Ar-H), 4.78 (s, 2H, PhC H_2 OBpin), 1.16 (s, 12H, BOC Me_2).

¹³C NMR (101 MHz, CDCl₃) δ 162.5, 160.1, 134.1, 114.2, 81.9, 81.7, 65.1, 23.8, 23.6.

NMR yield: 98%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 7.12 (d, J = 4.4 Hz, 2H, Ar-H), 7.01 (d, J = 6.3 Hz, 2H, Ar-H), 4.77 (s, 2H, MePhCH₂OBpin), 2.22 (s, 3H, *Me*PhCH₂OBpin), 1.19-1.13 (m, 12H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) δ 135.9, 135.3, 127.9, 125.9, 81.9, 81.7, 65.6, 23.8, 23.6, 20.1.

NMR yield: 99%, white gum. ¹H NMR (400 MHz, CDCl₃) δ 8.10 (dt, *J* = 8.7, 3.0 Hz, 2H, Ar-*H*), 7.42 (d, *J* = 8.8 Hz, 2H, Ar-*H*), 4.94 (s, 2H, NO₂PhC*H*₂OBpin), 1.23-1.15 (m, 12H, BOC*Me*₂).

¹³C NMR (101 MHz, CDCl₃) *δ* 146.3, 145.7, 125.9, 122.6, 82.3, 81.7, 64.6, 23.6, 23.5.

¹H, ¹³C NMR Spectra of Hydroboration Products of Carbonates.



Figure S1.1.1 1 H NMR (400 MHz, CDCl₃) of the compound without removing the solvent



Figure S1.1.2 ¹H NMR (400 MHz, CDCl₃) of the compound with the solvent removed



Figure S2.1.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2a**



Figure S2.2.2 $^{\rm 13}C$ NMR (101 MHz, CDCl₃) of compound 2b



Figure S2.3.2 $^{13}\mathrm{C}$ NMR (101 MHz, CDCl₃) of compound 2c



Figure S2.4.2 $^{13}\mathrm{C}$ NMR (101 MHz, CDCl₃) of compound 2d



Figure S2.5.2 ^{13}C NMR (101 MHz, CDCl₃) of compound 2e



Figure S2.6.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2f**



Figure S2.7.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2g**



Figure S2.8.2 $^{13}\mathrm{C}$ NMR (101 MHz, CDCl_3) of compound 2h



Figure S2.9.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2i**



Figure S2.10.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2**j



Figure S2.11.2 ^{13}C NMR (101 MHz, CDCl₃) of compound 2k



Figure S2.12.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2**I



Figure S2.13.2 ^{13}C NMR (101 MHz, CDCl₃) of compound 2m



Figure S2.14.2 ¹³C NMR (101 MHz, CDCl₃) of compound **2n**



¹H, ¹³C NMR Spectra of Hydroboration Products of Esters.

Figure S3.1.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4a



Figure S3.2.2 $^{\rm 13}{\rm C}$ NMR (101 MHz, CDCl₃) of compound 4b



Figure S3.3.2 $^{\rm 13}C$ NMR (101 MHz, CDCl₃) of compound 4c



Figure S3.4.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4d



Figure S3.5.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4e



Figure S3.6.2 ¹³C NMR (101 MHz, CDCl₃) of compound **4f**



Figure S3.7.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4g



Figure S3.8.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4h



Figure S3.9.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4i



Figure S3.10.2 $^{\rm 13}C$ NMR (101 MHz, CDCl₃) of compound 4j



Figure S3.11.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4k



Figure S3.12.2 ¹³C NMR (101 MHz, CDCl₃) of compound 4I



Figure S3.13.2 ¹³C NMR (101 MHz, CDCl₃) of compound **4m**



Figure S3.14.2 $^{\rm 13}C$ NMR (101 MHz, CDCl₃) of compound 4n



Figure S3.15.2 ¹³C NMR (101 MHz, CDCl₃) of compound 40

Mechanism Experiment

a) Formation of zinc hydride

When ZnEt₂ (1 mmol, 1 equiv) and HBpin (2 mmol, 2 equiv) were mixed in a Schlenk bottle. The solution was stirred at room temperature for 4 h, forming a white suspension. Take out an aliquot of the solution and analysed by NMR spectroscopy. However, no Zn-H signal was detected in ¹H NMR (Figure S5.1), the formation of Et-Bpin was indeed observed [¹¹B NMR 34.3 ppm in CDCl₃ (Figure S5.2)]. Insoluble material was also observed to form, presumably $(ZnH_2)_{\infty}$. ^[S1, S2]



Figure S5.1 ¹H NMR (400 MHz, CDCl₃) spectrum of the reaction mixture of $ZnEt_2$ and

HBpin



Figure S5.2 ¹¹B NMR (128 MHz, CDCl₃) spectrum of the reaction mixture of ZnEt₂ and HBpin

b) Reaction of ZnEt2 and HBpin under different conditions

When ZnEt₂ and HBpin were reacted at different temperatures and in different ratios, the formation of Et-Bpin was observed (¹¹B NMR 34.3 ppm in CDCl₃), but no Zn-H signal was detected by ¹H NMR. Insoluble white material could still be observed to form.



Figure S6. ¹¹B NMR (128 MHz, CDCl₃) of the reaction of ZnEt₂ and HBpin under different conditions

c) The reaction of carbonate esters catalyzed by zinc hydride

A Schlenk bottle was charged with 5 mL of THF followed by the addition of diethylzinc (1 mmol, 1 equiv) and HBpin (2 mmol, 2 equiv). The solution was stirred at room temperature for 4 h, forming a white suspension. The suspension was allowed to stand for a period of time, the supernatant was filtered off, and the solvent was removed under vacuum to obtain a white solid. Then added dimethyl carbonate (1 mmol, 1 equiv.) and HBpin (3.2 mmol 3.2 equiv.) into it, which reacted for 12 h at 80 °C. A portion of the sample was removed and analysed by NMR spectroscopy. The yield of ¹H NMR spectroscopy indicates that ZnH₂ is the real catalyst in the borohydride reaction.



Figure S7.1. ¹H NMR (400 MHz, CDCl₃), ¹³C NMR (101 MHz, CDCl₃) of dimethyl

carbonate catalyzed by $ZnEt_2$



Figure S7.2. ¹H NMR (400 MHz, CDCl₃), ¹³C NMR (101 MHz, CDCl₃) of dimethyl carbonate catalyzed by ZnH₂

Computational Details

In this study, the computations were performed by a DFT method using the Gaussian 16 suite of programs.^[S3] The BP86 functional^[S4,S5] and Ahlrichs' split valence plus polarization basis set (def2-SVP)^[S6,S7] were chosen to optimize the geometry structures. Vibrational frequency were calculated at the same level to confirm that the stable structures possess no imaginary frequency and every transition state possesses only one imaginary frequency. Intrinsic reaction coordinate (IRC) calculations were carried out to ensure that the transition state connects the corresponding reactant and product. Single-point calculations were further performed using R-BP86 functional and the triple- ζ valence plus polarization basis set (def2-TZVP)^[S6,S7] based on the optimized geometries at the BP86/def2-SVP level of theory to improve the accuracy of the energies. The Grimme's D3 empirical corrections were used to incorporate the London dispersion effects.^[S8] All the energy values reported in this article were obtained at the BP86-D3/def2-TZVP/BP86/def2-SVP level.

Table S2. Absolute energies (in Hartree) of all the studied structures. E_e = Electronic energy, G = gas-phase Gibbs free energy at 80 °C, and number of imaginary vibrational frequencies (Nimg) at the BP86-D3/def2-TZVP//BP86/def2-SVP level of theory.

Species	E_{e}	G	Nimg
1 a	-342.523443	-342.516974	0
ZnH ₂	-1780.842350	-1780.838092	0
HBpin	-411.897833	-411.889382	0
2a	-1051.812509	-1051.796494	0
CH ₃ OBpin	-526.504428	-526.494514	0
НСНО	-114.554723	-114.549966	0
INT-1	-2123.368276	-2123.360508	0
INT-2	-2123.368276	-2123.360508	0
INT-3	-2123.379528	-2123.371460	0
INT-4	-3789.661124	-3789.652169	0
INT-5	-4201.556730	-4201.543026	0
INT-6	-2420.732184	-2420.720289	0
INT-7	-2832.634118	-2832.616920	0
INT-8	-1895.424353	-1895.418303	0
INT-9	-2307.325635	-2307.314065	0

TS-1	-2123.328924	-2123.321168	1
TS-2	-2123.361644	-2123.354089	1
TS-3	-3904.172116	-3904.162093	1
TS-5	-4201.558013	-4201.544294	1
TS-7	-2832.636129	-2832.618916	1
TS-8	-1895.373344	-1895.367209	1
TS-10	-2307.325631	-2307.314394	1

Table S3. Cartesian coordinates (Å) of the reactants, intermediates and transition states at theBP86/def2-SVP level of theory.

1a		
2.055845	-0.000012	0.000010
0.857107	-0.000002	-0.000003
0.068221	1.118558	-0.108823
0.068199	-1.118557	0.108800
-1.301911	0.760379	0.120067
-1.301932	-0.760368	-0.120055
-1.580256	1.026739	1.163879
-1.948563	1.323470	-0.581068
-1.948583	-1.323439	0.581099
-1.580307	-1.026731	-1.163858
	1a 2.055845 0.857107 0.068221 0.068199 -1.301911 -1.301932 -1.580256 -1.948563 -1.948583 -1.580307	1a2.055845-0.0000120.857107-0.0000020.0682211.1185580.068199-1.118557-1.3019110.760379-1.301932-0.760368-1.5802561.026739-1.948563-1.323470-1.580307-1.026731

	ZnH ₂		
Zn	0.000000	0.000000	0.000000
Н	0.000000	0.000000	1.549086
Н	0.000000	0.000000	-1.549086
	HBpin		
С	-0.794660	-0.190744	-0.045745
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С	-5.576259	-0.668877	0.903450
Н	-6.345090	0.019269	1.309637
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Н	6.361180	-0.435053	-0.414577
Н	4.335006	2.948496	-0.601001
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Н	2.781727	2.577801	0.230326
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Н	1.385355	-2.395679	1.303975
Н	2.373223	-0.961742	1.750573
Н	0.612026	-0.999184	2.117627
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Н	2.317829	-2.192073	-0.983330
Н	1.885956	-0.820842	-2.047551
Н	2.544127	1.602067	0.992815
Н	1.131226	2.702324	1.094186
Н	1.200694	1.250728	2.138658
Н	1.978404	1.665804	-1.591366
Н	0.410289	0.975124	-2.138651
Н	0.445382	2.550620	-1.282689
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Н	-4.305012	0.186801	-0.654225
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С	0.670409	0.953103	0.545079
Ο	1.849245	0.922727	-0.240681
Ο	0.443866	-0.514729	0.910716
С	1.948305	-0.363775	-0.843351
С	1.490793	-1.283036	0.295060
Н	1.279767	-0.458984	-1.734521
Н	2.995525	-0.525203	-1.166831
Н	1.080428	-2.257794	-0.039912
Н	2.309906	-1.464563	1.027103
Zn	-1.511183	-0.186775	-0.134723
Н	0.917431	1.512691	1.478747
Н	-2.636534	-1.180089	-0.404914

INT-2

Ο	0.451605	1.359437	-0.057385
С	-0.670412	0.953100	0.545079
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0	-0.443861	-0.514729	0.910715
С	-1.948308	-0.363776	-0.843346
С	-1.490771	-1.283046	0.295052
Н	-2.995528	-0.525226	-1.166818
Н	-1.279779	-0.458967	-1.734525
Н	-2.309878	-1.464604	1.027093
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Н	-0.917434	1.512695	1.478742
Zn	1.511186	-0.186770	-0.134723
Н	2.636559	-1.180068	-0.404882

INT-3

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С	-1.343935	-1.450438	-0.324725
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С	-0.808773	1.702237	-0.136260
Н	-2.548249	1.181360	1.101368
Н	-1.029228	0.293996	1.531543
Н	-1.476747	2.094638	-0.947574
Н	-0.638411	2.578380	0.554408
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Zn	1.443402	-0.077011	0.084269
Н	2.745230	-0.734161	0.547277
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0	1.526831	1.034796	-0.139686
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С	0.199908	0.724582	0.166838
Н	0.047108	-0.906201	-1.247049
Н	0.482840	-1.399765	0.423228
Н	-0.047080	0.906235	1.247280
Н	-0.482838	1.399798	-0.422990
Н	-4.188461	0.910149	-0.075282
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INT-5

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С	-1.688365	-1.543883	0.823149
С	-0.852503	-1.530091	-0.466831
Н	-1.561889	-2.531478	1.329077
Н	-1.262687	-0.776517	1.514389
Н	-1.098761	-0.611672	-1.052101
Н	-1.094931	-2.405927	-1.105513
В	1.289247	-0.247655	-0.023195
0	1.280979	0.611082	-1.148363
0	1.051040	0.492894	1.165523
Н	2.607099	-0.839216	0.127886
С	1.487956	1.955517	-0.659119
С	0.864809	1.881968	0.807291
С	3.003487	2.235236	-0.650984
С	0.786042	2.932364	-1.608351
С	1.567631	2.762334	1.847808
С	-0.649340	2.174175	0.831226
Н	3.403351	2.057153	-1.669817

Н	3.233219	3.282124	-0.366298
Н	3.538639	1.559337	0.045930
Н	0.841611	3.972964	-1.226549
Н	1.278770	2.907123	-2.601697
Н	-0.277996	2.665433	-1.755215
Н	1.507331	3.836276	1.573585
Н	1.080863	2.634470	2.836248
Н	2.632654	2.484834	1.960685
Н	-0.875950	3.240593	0.627702
Н	-1.192075	1.558735	0.082288
Н	-1.047942	1.921842	1.834966
Zn	-3.720845	0.101213	-0.278245
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Н	2.793364	-3.958345	-0.070936
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0	0.165759	-1.867174	-0.260003
С	-2.034100	-0.963242	0.317917
С	-1.096201	-1.457991	-0.791806
Н	-2.095077	-1.757350	1.104774
Н	-1.554603	-0.074437	0.801824
Н	-0.972400	-0.655504	-1.551961
Н	-1.557556	-2.338461	-1.286840
В	1.191246	-0.978488	-0.144968
0	2.431727	-1.359675	0.334100
Ο	1.114592	0.364881	-0.488938
С	3.196569	-0.138023	0.536590
С	2.477283	0.882935	-0.451268
С	3.043215	0.251932	2.017026
С	4.665750	-0.421193	0.213302
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С	3.025189	0.814956	-1.886822
Н	3.362088	-0.602880	2.646616
Н	3.664707	1.131244	2.282093
Н	1.987044	0.480603	2.265840
Н	5.273986	0.504557	0.280975
Н	5.073468	-1.151627	0.941208
Н	4.786190	-0.852108	-0.798592
Н	3.462820	2.741075	0.152412
Н	1.905208	2.962930	-0.708611
Н	1.905926	2.428156	0.999720
Н	4.035701	1.264441	-1.966438

Н	3.078001	-0.232174	-2.247730
Н	2.342824	1.373215	-2.559078
Zn	-4.426574	0.620284	0.176133
Н	-5.472169	1.688109	0.425285

INT-7

В	-2.499361	-0.147288	0.098746
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0	-2.023417	0.551667	-1.034249
С	-2.183256	1.963630	-0.753624
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С	-0.614107	2.152853	1.327379
Н	-0.219387	3.176034	1.158619
Н	-0.586323	1.949259	2.417372
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С	-2.948095	3.067311	1.512547
Н	-2.672748	4.090430	1.181400
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Н	-2.813690	3.017882	2.612518
С	-3.568860	2.398271	-1.269167
Н	-3.645956	2.139901	-2.344843
Н	-4.386552	1.873377	-0.735573
Н	-3.728670	3.490498	-1.162663
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Н	-0.081549	2.337394	-1.261230
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Н	-1.115887	3.813159	-1.245260
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В	2.557404	-1.286969	-0.231524
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Н	3.587304	-0.237125	2.299140
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Н	2.769104	-2.432689	-0.154432
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Н	3.625259	-0.609171	1.572825
Н	2.532664	0.162599	2.768352
Н	2.110772	-1.436425	2.087076
Н	3.417023	1.721464	0.374524
Н	1.812562	2.222915	-0.263692
Н	2.124287	2.254643	1.502401
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С	0.432474	-0.121012	0.376507
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0	1.206966	1.007628	0.570418
С	2.521123	-0.623534	-0.442383
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Н	3.288615	-1.021827	0.256324
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Н	2.079597	1.374274	-1.291960
Zn	-1.829062	0.066543	-0.251645
Н	-0.263436	0.271678	-0.958646
Н	-3.344688	0.123905	-0.465025
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Н	-1.195898	-0.256201	-1.702330
Н	-1.973674	-1.783396	0.902773
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Н	-0.909080	1.580336	1.516808
Zn	1.431607	-0.141537	-0.108222
Н	2.818978	-0.598404	-0.572414
	ТЅ-3		

С	-1.646636	1.432744	0.562578

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С	1.051053	-0.408184	1.326048
С	0.033566	-0.456958	0.177966
Н	1.012059	0.611005	1.790284
Н	0.732912	-1.128082	2.118082
Н	0.005625	-1.476176	-0.263523
Н	0.343952	0.265099	-0.616718
Н	4.140443	0.550387	-1.558252
Н	-3.032894	1.167362	0.071482
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Н	-1.958600	1.669133	1.609291
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0	-0 034984	1 248343	-0 719480
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Н	1 577027	0.434916	1 320957
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0	-1 998825	-0 145970	-1 193292
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C	-2 748262	-1 177926	-0 497062
C	-1.726177	-1.628693	0.643147
C	-4.034019	-0.540249	0.060676
C	-3.108384	-2.273683	-1.505206
C	-2.377125	-2.015073	1.974480
C	-0.767263	-2.740633	0.182748
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Н	-4.694759	-1.293345	0.535850
Н	-3.809064	0.251669	0.802001
Н	-3.608680	-3.127861	-1.003614
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Н	-0.300506	-2.491606	-0.791882
Н	0.040761	-2.852648	0.934106
Zn	4.054460	-1.040038	-0.101283
Н	4.740437	-2.344633	-0.456120
Zn	-0.972198	2.936838	0.164960
Н	-0.324613	4.336087	0.236557
	TS-7		
В	-2.325657	-0.176953	0.033755
0	-2.297289	0.585624	1.197795
0	-2.192664	0.588097	-1.121161
С	-2.472187	1.966345	-0.745040
С	-2.110975	1.977636	0.811826
С	-0.637850	2.322476	1.090343
Н	-0.418502	3.390390	0.886200
Н	-0.423180	2.126404	2.160587
Н	0.052853	1.698920	0.487441
С	-3.026995	2.852832	1.672908
Н	-2.960078	3.918344	1.369828
Н	-4.083318	2.531350	1.604354
Н	-2.719831	2.781071	2.736128
С	-3.961225	2.234296	-1.027345
Н	-4.166630	2.015418	-2.094723
Н	-4.615389	1.579513	-0.418470
Н	-4.234434	3.290808	-0.830966
С	-1.603893	2.885647	-1.608941
Н	-0.531825	2.621419	-1.538389
Н	-1.907317	2.793583	-2.671608
Н	-1.729164	3.946774	-1.309251
0	-1.878102	-1.545281	0.053221
С	-0.506651	-1.845861	0.354487
С	0.358932	-1.729752	-0.905783
Н	-0.465711	-2.883148	0.746536
Н	-0.138125	-1.158131	1.146942
0	1.692501	-2.145028	-0.616594
Н	0.330246	-0.688888	-1.292784
Н	-0.047667	-2.402062	-1.691000
В	2.667259	-1.228691	-0.346602
0	3.973133	-1.612270	-0.117861
0	2.463184	0.143107	-0.277894
С	4.679933	-0.429519	0.356423
С	3.789775	0.755468	-0.224473
С	4.673938	-0.486098	1.893610

С	6.115835	-0.474526	-0.171024
С	3.723975	2.003513	0.657446
С	4.165637	1.137514	-1.665725
Н	5.113964	-1.450523	2.217478
Н	5.268369	0.335756	2.341536
Н	3.641915	-0.431810	2.295652
Н	6.668994	0.447707	0.102589
Н	6.649701	-1.337779	0.275273
Н	6.144628	-0.593072	-1.270649
Н	4.733541	2.443970	0.791847
Н	3.080650	2.769960	0.179580
Н	3.301173	1.779470	1.654931
Н	5.131081	1.680849	-1.710137
Н	4.238225	0.241514	-2.314890
Н	3.377401	1.797081	-2.081249
Н	-4.052871	-0.705718	-0.049860
Zn	-3.866539	-2.294787	0.122611
Н	-4.166132	-3.801183	0.290648
	ТS-8		
0	-1.202904	-0.818043	-0.000001
С	-1.575771	0.400008	0.000001
Н	-0.138121	1.472715	-0.000006
Н	-1.941429	0.899567	-0.937326
Н	-1.941424	0.899566	0.937330
Zn	0.704135	0.055533	0.000000
Н	1.974781	-0.793547	0.000001
	TS 1(h	
0	-1.551240	0.934517	-0.503930
С	-1.555702	2.364900	-0.534979
Н	-1.005911	2.735900	-1.426744
Н	-2.608195	2.704839	-0.598143
Н	-1.086748	2.782740	0.380994
В	-0.328860	0.230365	-0.197458
0	0.499701	-0.214211	-1.225789
0	0.377157	0.548242	0.964450
Н	-1.198194	-1.231194	0.240340
С	1.761230	-0.592724	-0.610103
С	1.789628	0.320906	0.700075
С	1.701374	-2.099090	-0.297607
С	2.887272	-0.314019	-1.610354
С	2.421876	-0.340256	1.928660
С	2.427281	1.700238	0.458340

Н	1.477984	-2.648443	-1.234439
Н	2.663750	-2.475473	0.104501
Н	0.898257	-2.333233	0.428764
Н	3.880754	-0.511118	-1.156853
Н	2.776089	-0.979502	-2.490423
Н	2.864109	0.730923	-1.973443
Н	3.487145	-0.590338	1.743350
Н	2.380359	0.355947	2.791068
Н	1.886358	-1.264785	2.215639
Н	3.526073	1.632442	0.324501
Н	1.996102	2.193196	-0.436200
Н	2.225460	2.347584	1.335796
Zn	-2.715728	-0.679498	0.209586
Н	-4.249942	-0.604495	0.368954

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