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High-frequency ultrasonic pyrolysis of 200 nm ultrafine Fe-doped NiO hollow

spheres for efficient oxygen evolution catalysis

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Supplementary Figure 1. (**a**, **b**) TEM images of NiO hollow nanospheres at 1.7 MHz frequency. (**c**, **d**) TEM images of NiO hollow nanospheres at 3.0 MHz frequency.



Supplementary Figure 2. TEM images of $Ni_{0.8}Fe_{0.2}O$, $Ni_{0.7}Fe_{0.3}O$, $Ni_{0.6}Fe_{0.4}O$ and $Ni_{0.5}Fe_{0.5}O$.



Supplementary Figure 3. (a) TEM image, (b) HRTEM image and (c) EDS mapping of $Ni_{0.9}Fe_{0.1}O$ hollow nanospheres after OER.

Catalyst	Electrode	Overpotential	Electrolyt	Reference
	substrate	(mV)	e	
Ni _{0.9} Fe _{0.1} O hollow	Glassy carbon	288	1 M KOH	This work
nanospheres				
Hollow cobalt nickel	Glassy carbon	310	1 M KOH	1
oxides microspheres				
Porous NiO hollow	Glassy carbon	323	1 M KOH	2
spheres				
Porous NiO nano flowers	Glassy carbon	346	1 M KOH	2
3D porous	Glassy carbon	353	1 М КОН	3
carbon@Ni/NiO				
Ni-Co mixed oxide	Glassy carbon	380	1 M KOH	4
nanocages				
Hollow NiCo ₂ O ₄ arrays	Glassy carbon	340	1 M KOH	5
CoNiO ₂ /SNC	Glassy carbon	280	1 M KOH	6
CoNiOx/NC	Glassy carbon	335	1 M KOH	6
Multi-shelled Co _{0.5} Ni _{0.5}	Glassy carbon	268	1 M KOH	7
oxide/phosphide				
Porous nanoscale	Glassy carbon	264	1 M KOH	8
NiO/NiCo ₂ O ₄				
U-NiO/NiCo ₂ O ₄	Glassy carbon	387	0.1M KOH	9
C-NiO/NiCo ₂ O ₄	Glassy carbon	430	0.1M KOH	9
Multilayer hollow	Glassy carbon	400	0.1M KOH	10
MnCo ₂ O ₄ microsphere				
3D NiFe ₂ O ₄ hollow	Glassy carbon	370	0.1M KOH	11
spheres				

Supplementary Table 1. Comparisons of the η_{10} of Ni_{0.9}Fe_{0.1}O hollow nanospheres with representative electrocatalysts reported previously.

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