

## Supporting information

# Interfacial engineering of a tri-phase CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> electrocatalyst for promoting oxygen evolution reaction

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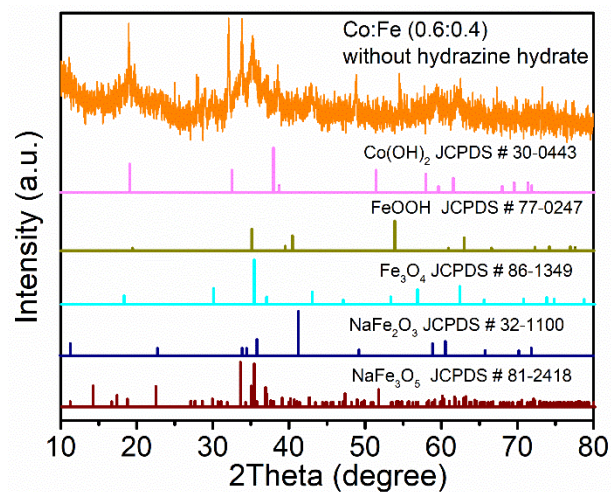


Fig. S1 The XRD pattern of the sample synthesized without hydrazine hydrate.

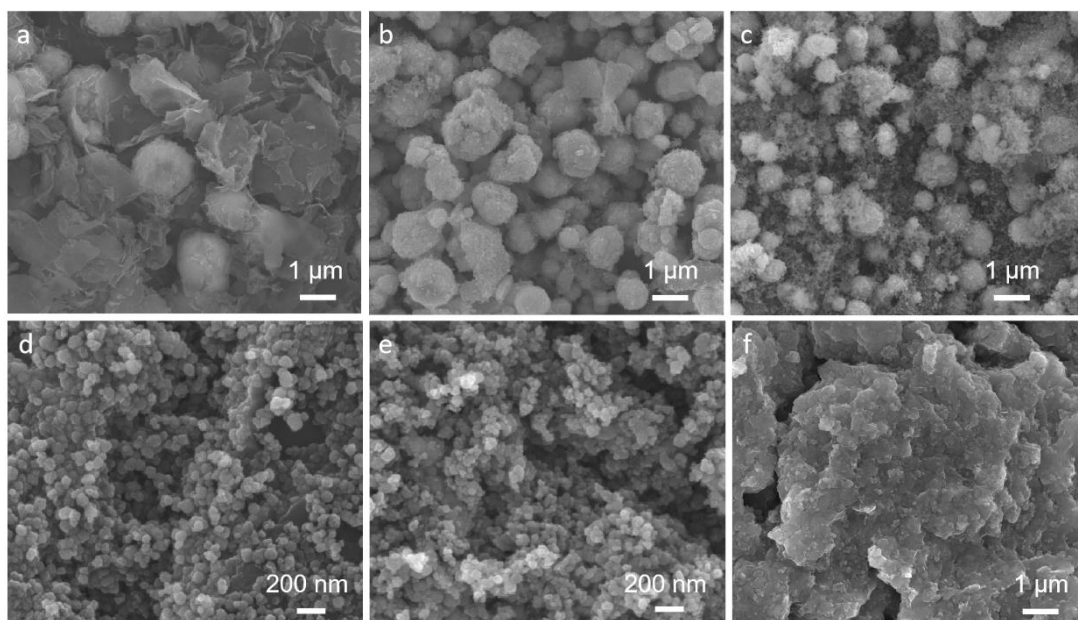


Fig. S2 FESEM images of the samples with different Co/Fe atom ratios: (a) 1:0, (b) 0.8:0.2, (c) 0.6:0.4, (d) 0.4:0.6, (e) 0.2:0.8, (f) 0:1.

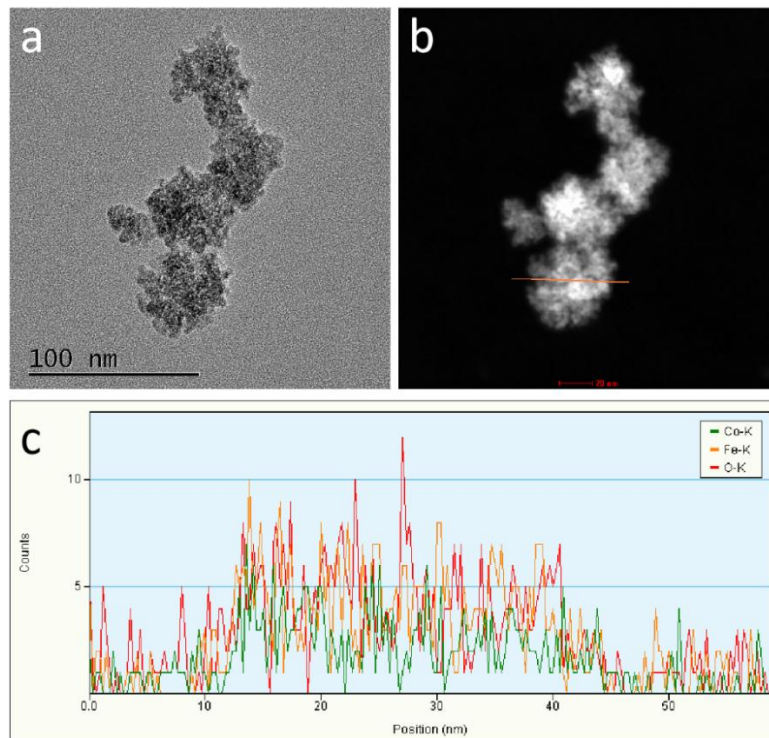


Fig. S3 (a) a typical TEM of the surrounded Co-Fe<sub>3</sub>O<sub>4</sub> nanoparticles in CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.4: 0.6), (b) HAADF-STEM image of the nanoparticles, and (c) the line-scan profile across the nanoparticles.

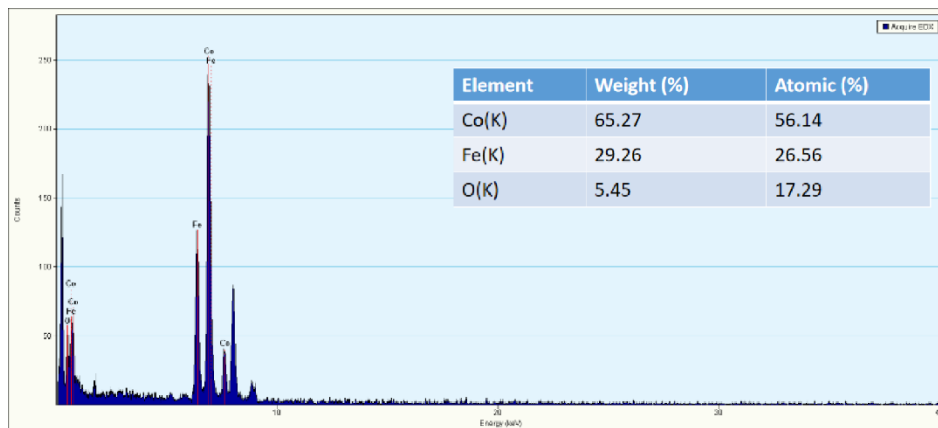


Fig. S4 EDS-TEM spectrum of the CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub>.

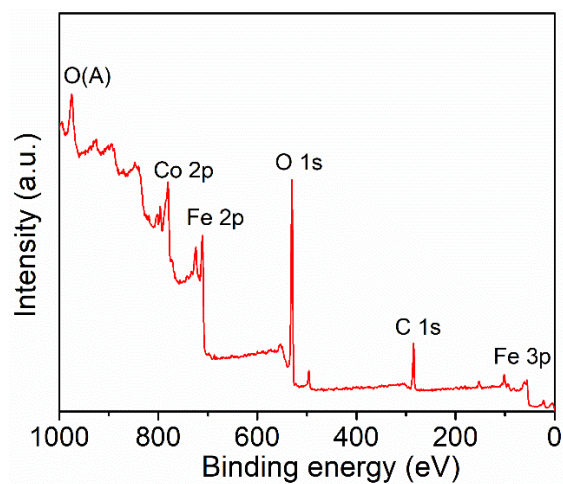


Fig. S5 XPS survey spectrum of the CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub>.

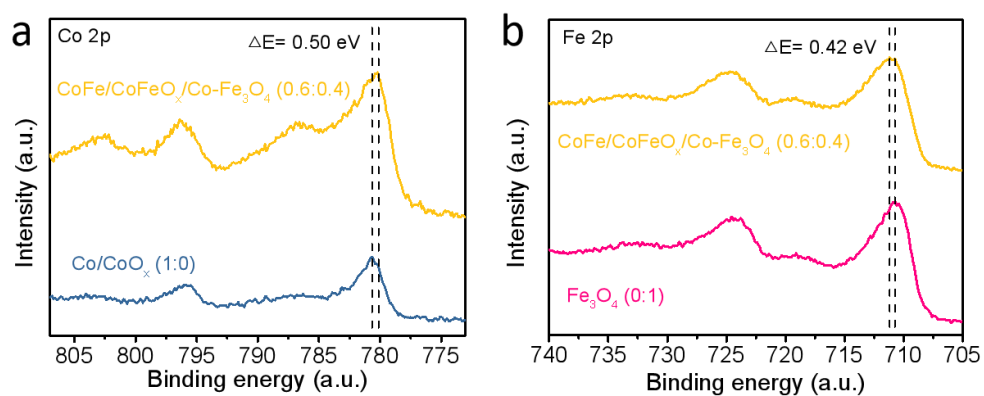


Fig. S6 XPS core level spectra of (a) the Co 2p and (b) the Fe 2p for different catalysts.

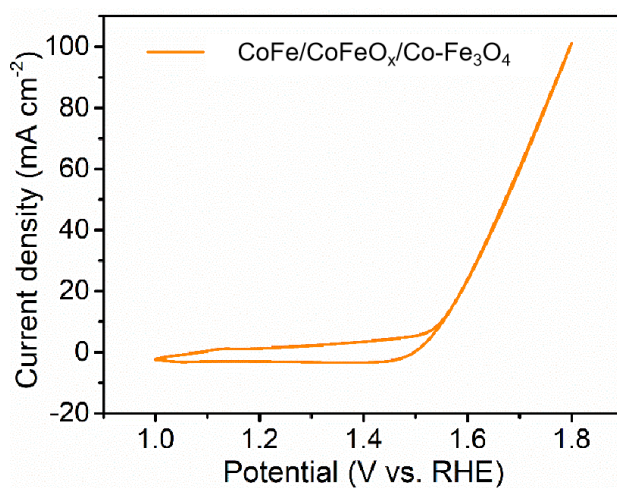


Fig. S7 CV curve of the CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.6:0.4) at a scan rate of 50 mV s<sup>-1</sup>.

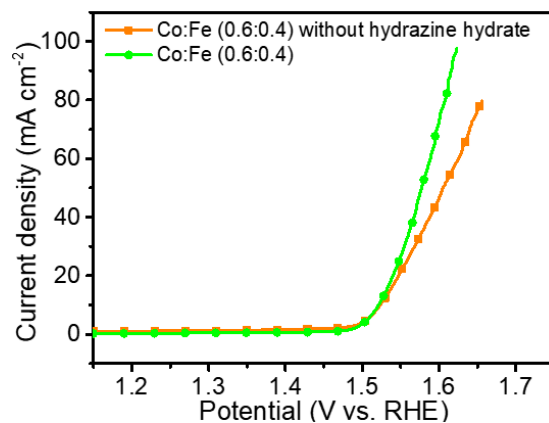


Fig. S8 The LSV curve of the sample synthesized without hydrazine hydrate.

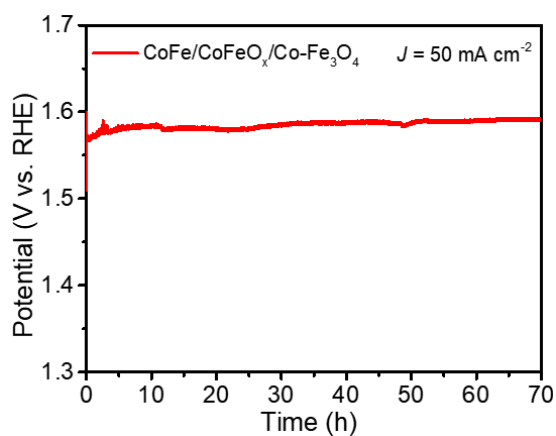


Fig. S9 The stability test of the CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.6:0.4) for a long-term electrolysis of 70 h at 50 mA cm<sup>-2</sup>.

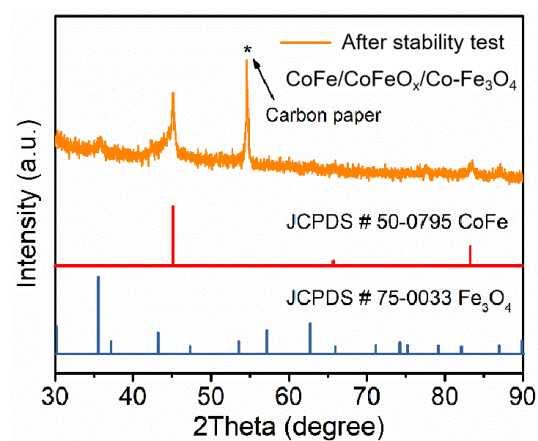


Fig. S10 XRD pattern of the spent CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.6:0.4).

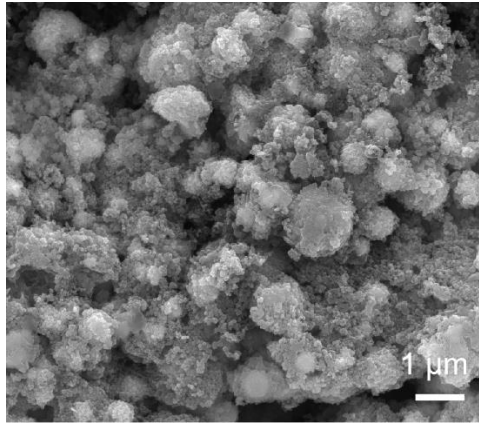


Fig. S11 FESEM image of the spent CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.6:0.4).

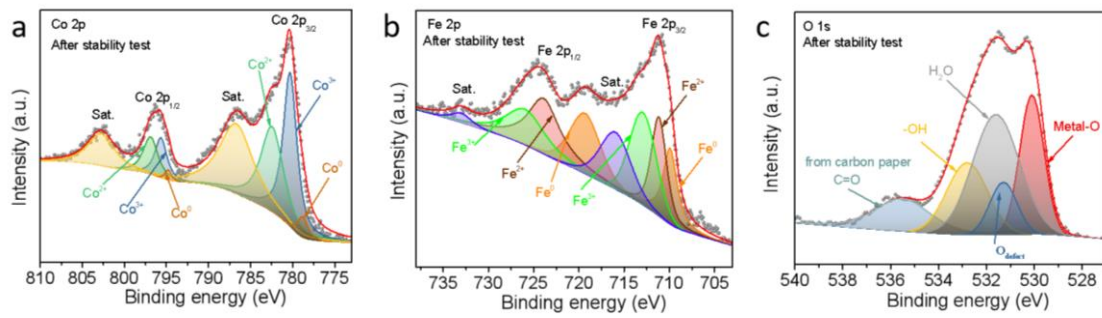


Fig. S12 XPS spectra of the spent CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub> (0.6:0.4).

Table S1 OER performance comparison of recently reported CoFe-based electrocatalysts.

Catalysts	Overpotential (mV) @ 10 mA cm <sup>-2</sup>	Tafel slope (mV dec <sup>-1</sup> )	Electrolyte	Substrate	Year published	Ref.
CoFeV hydroxide	376	26	1 M KOH	Glassy carbon	2021	1
CoFe alloy/N-doped carbon	340	77	1 M KOH	Glassy carbon	2022	2
La-doped CoFe layered double hydroxide	317	125	1 M KOH	Glassy carbon	2021	3
CoFe <sub>2</sub> O <sub>4</sub> /carbon nanotube	390	82	0.1 M KOH	Glassy carbon	2021	4
Se-doped Co-Fe sulfide	281	51.8	1 M KOH	Glassy carbon	2021	5
CoFe alloy/N-doped carbon nanosheets	285	39	1 M KOH	Glassy carbon	2021	6
CoFe/S, N co-doped carbon nanotubes	358	259	0.1 M KOH	Glassy carbon	2021	7
N-doped CoFe phosphate	313	58.92	1 M KOH	Glassy carbon	2021	8
Mesoporous Co-Fe oxides	280	68	1 M KOH	Glassy carbon	2021	9
Co <sub>3</sub> Fe <sub>1</sub> -layered double hydroxides	314	79.4	1 M KOH	Glassy carbon	2022	10
<b>Tri-phase CoFe/CoFeO<sub>x</sub>/Co-Fe<sub>3</sub>O<sub>4</sub></b>	<b>291</b>	<b>47.8</b>	<b>1 M KOH</b>	<b>Carbon paper</b>	<b>–</b>	<b>This work</b>

## References

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