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Study on the Particle Characteristics and Stability of Ag-NPs Naturally Generated in Soil Matrix

- 3 Qiang Wang^{1,2*}, Ying-Rong Wang¹, Wen-Jing Wei¹, Hai-Lan Lin², Ping-Jian Yang^{3*}
- 4 ¹Key Laboratory of Hunan Province for Water Environment and Agriculture Product Safety, College of
- 5 Chemistry and Chemical Engineering, Central South University, Changsha 410083, Hunan, PR China
- 6 ² State Environmental Protection Key Laboratory of Monitoring for Heavy Metal Pollutants
- 7 ³ Chinese Research Academy of Environmental Sciences, Dayangfang 8, Anwai, Chaoyang District, Beijing
- 8 100012, PR China
- 9
- 10 *Corresponding author: Dr. Qiang Wang: qwangcsu@163.com; Dr. Ping-Jian Yang:
- 11 yang.pingjian@craes.org.cn
- 12 Postal address: No. 932 Lushan Nan Road, Yuelu District, Changsha 410083, PR China
- 13 **Phone number:** +86 731 88876961

15 This document contains: 8 pages, 1 text, 5 figures and 6 tables.

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17 Text S1. Steps of the iterative algorithm for data processing

The steps of iterative algorithm were outlined as follows: the standard deviation (σ) and 18 mean value (μ) of total raw dataset were calculated and the outliers which exceeding the value 19 of $(\mu + 5\sigma)$ were removed. Subsequently, μ and σ of the remaining fraction were recalculated 20 and the procedure was repeated until no dataset could be removed, that was, convergence was 21 22 achieved. Then iteration process end and the background signals were identified in this way. The particle events were identified and their intensities were recorded in succession. All target 23 24 particles detected were assumed to be spherical and zero-valent for each event. The nanoparticle number concentration N_{NP}, particle mass m_{NP} and nanoparticle diameter d_{NP} 25 26 could be calculated as described in equation (1), (3) and (4), respectively.

$$f_{NP} = N_{NP}Q_{sam}\eta_n \tag{1}$$

$$W = \eta_n Q_{sam} t_{dwell} C \tag{2}$$

$$m_{NP} = f_m^{-1} \left[\frac{\left(I_{NP} - I_{bgd} \right) * \eta_i}{m} \right]$$
(3)

$$d_{NP} = \sqrt[3]{\frac{6m_{NP}}{\pi\rho}} \tag{4}$$

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Where, f_{NP} is frequency of pulse events (pulse per event), Q_{sam} is the sample flow rate 31 (mL·min⁻¹), η_n is transport efficiency, W is the mass flux equation (mass per event), t_{dwell} is 32 the dwell time (ms per event), C is analytical element of ionic calibration curve equation, I_{NP} 33 is nanoparticle signal intensity (cps), Ibgd is background signal intensity (cps), fm is the mass 34 fraction of analytical element, η_i is the particle ionization efficiency, m is the slope of the 35 calibration curve for the dissolved ion (cps· μ g⁻¹·L), ϵ_{neb} is the nebulization efficiency, and ρ is 36 the particle density ($g \cdot cm^{-3}$). Background signal includes instrument noise and signal from 37 dissolved metal ions. 38





Figure S1. (a) Fulvic acid and (b) humus extracted from the environmental soil.





Figure S2. Time scans of Ag-NPs (a) with (b) without cloud point extraction.



Figure S3. Visual observation of the generation of Ag-NPs in (a) Greenbelt, (b) Farmland, (c)
Xiangjiang River, (d) Bottom Mud and (e) Control group.



Figure S4. Time scans of Ag-NPs generated in Greenbelt (a) with (b) without, Farmland (c) with, (d) without, Xiangjiang River (e) with, (f) without, Bottom Mud (g) with and (h) without Ag⁺ added, respectively (1.6 $*10^4$ times diluted).



Figure S5. TEM images of the Ag-NPs generated in the soil.

	TI	Conductivity	Ionic	One of a method	Ag
Sample	рн (KCl)	$(\mu S cm^{-})$	strength (m	(%)	concentration (
		1)	M)		mg kg-1)
Greenbelt	7.13	315.0	4.0950	3.3791	0.0177
Farmland	6.43	158.2	2.0566	6.1909	0.0143
Xiangjiang River	7.41	107.3	1.3949	3.1963	0.0530
Bottom Mud	6.81	160.5	2.0865	12.7893	0.0128

Table S1. Basic physical and chemical properties of soil samples

Table S2. Operational Parameters for SP-ICP-MS Analysis

Instrumental parameters	Parameter values		
RF power	1550		
Plasma gas flow rate (L min ⁻¹)	15		
Sample flow rate (mL min ⁻¹)	0.36		
Spray chamber temperature (°C)	2		
Data acquisition mode	TRA		
Dwell time (ms)	3		
Acquisition time (s)	60		
Mass monitored	¹⁰⁷ Ag		

Table S3. Particle number concentration and size of Ag-NPs determined in the soil

	Spiked sample			Control sample		
Sample	Particle number concentration		Particle	Particle number concentration		Particle
	10 ⁶ particles	10 ⁶ particles g ⁻¹	size(nm)	10 ⁶ particles	10 ⁶ particles	size(nm)
	L-1	soil		L-1	g ⁻¹ soil	
Greenbelt	0.60	4.80	65	0.19	1.52	37
Farmland	0.09	0.72	50	0.22	1.76	30
Xiangjiang River	0.46	3.68	51	0.27	2.16	35
Bottom Mud	0.51	4.08	45	0.35	2.80	32

Table S4. Particle number concentration and average size of Ag-NPs

Sampla	Particle number concentration	Particle size (nm)	
Sample	$(10^6 \text{ particles } L^{-1})$	r article size (IIIII)	
Control	0.23 ± 0.17	37.94±3.01	
Ag^{+}	1.43 ± 0.13	41.55±3.77	
Ag-NPs	4.33 ±0.41	32.26±2.20	
Ag ⁺ and Ag-NPs	$4.90\pm\!\!0.14$	33.97±3.47	

Table S5. The influence of illumination on the generation of Ag-NPs

Illumination intensity	Illumination	Particle number concentration	Particle size				
$(\mu mol m^{-2} \cdot s^{-1})$	time (h) $(10^6 \text{ particles } L^{-1})$		(nm)				
Illumination intensity							
0		0.38±0.09	57.60±4.21				
100		0.85±0.13	63.63±4.63				
200	10	2.20±1.13	88.87±1.93				
300	10	2.46±0.63	78.93±4.40				
400		3.10±1.08	79.94±2.41				
500		4.50±1.27	79.63±1.70				
Illumination time							
	0	0.24 ± 0.03	61.55±1.08				
	5	0.86±0.12	72.29±0.35				
200	10	5.50±1.27	81.65±3.09				
300	24	5.83±0.78	80.95±3.14				
	48	6.63±0.55	78.44±1.10				
	72	6.50±0.76	82.14±1.36				

Fulvic Acid	Humus	Particle number concentration	Particle size				
(mg L ⁻¹)	(mg g ⁻¹)	(10 ⁷ particles L ⁻¹)	(nm)				
	Fulvic Acid						
0		0.43±0.09	33.38±2.09				
1		0.81±0.16	35.15±3.31				
2	/	1.02±0.13	45.96±1.29				
5	1	5.80±1.25	79.06±1.41				
10		7.43±0.64	80.16±1.19				
20		6.16±1.36	76.67±4.43				
Humus							
	0.00	0.24±0.05	67.58±2.78				
	0.01	1.61±0.19	45.02±1.81				
1	0.02	3.05±0.07	44.73±1.00				
/	0.05	3.10±0.28	60.09±5.26				
	0.10	2.54±0.21	46.73±4.61				
	0.15	2.33±0.21	42.08±6.54				

Table S6. 7	The influence	of humic	acid on th	e generation	of Ag-NPs