

Supplementary Information

Effect of introducing a cyclobutylmethyl group to an onium cation on the thermodynamic properties of ionic clathrate hydrate

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NMR characterization of tri-*n*-butyl(cyclobutylmethyl)phosphonium bromide (P444(1c4)-Br) synthesized in the present study.

¹H NMR (400.13 MHz in CDCl₃), δ 0.90–1.05 (t, 9H, P⁺CH₂(CH₂)₂CH₃); 1.41–1.67 (m, 12H, P⁺CH₂(CH₂)₂CH₃); 1.83–2.05 (m, 4H, P⁺CH₂CH(CH)₂CH₂); 2.15–2.31 (m, 2H, P⁺CH₂CH(CH)₂CH₂); 2.32–2.53 (m, 6H, P⁺CH₂(CH₂)₂CH₃); 2.55–2.75 (m, 3H, P⁺CH₂CH(CH)₂CH₂).

¹³C NMR (100.62 MHz in CDCl₃), δ 13.36 (s, P⁺CH₂CH₂CH₂CH₃); 18.70–19.03 (d, P⁺CH₂CH₂CH₂CH₃); 19.49 (s, P⁺CH₂CH(CH)₂CH₂); 23.67–23.88 (m, P⁺CH₂CH₂CH₂CH₃); 25.60 (s, P⁺CH₂CH₂CH₂CH₃); 26.04 (d, P⁺CH₂CH(CH)₂CH₂); 29.16–29.22 (d, P⁺CH₂CH(CH)₂CH₂), 30.17–30.27 (d, P⁺CH₂CH(CH)₂CH₂).

³¹P NMR (161.97 MHz in CDCl₃), δ 32.04 (s, P⁺(C₄H₉)₃(CH₂CH(CH)₂CH₂)).

Raman spectra of tri-*n*-butyl(cyclobutylmethyl)phosphonium bromide (P444(1c4)-Br) salt.

Figure S1 shows the Raman spectra of P444(1c4)-Br and P4444-Br salts. As noted in Figure 5, there are some differences derived from cyclobutylmethyl group, but no significant difference exists in the other peaks.

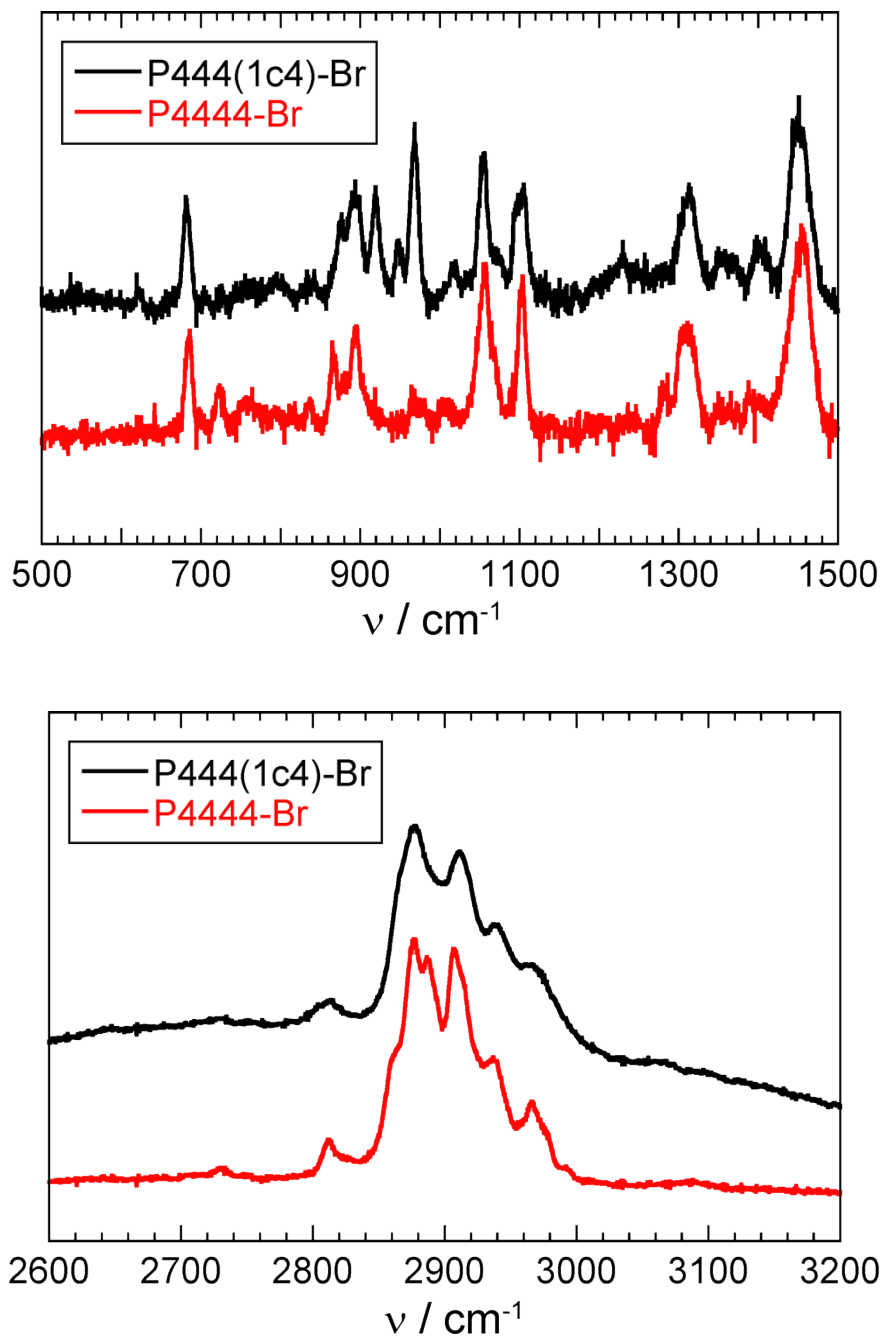


Figure S1. Raman spectra of P444(1c4)-Br and P4444-Br salts recorded at 77 K.