

Supporting Information

Enhanced Photocatalytic Activity of Titanium Dioxide in Nitrogen Fixation by Photon Localization Effect of SiO₂ Opal Photonic Crystal

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1. The detailed preparation procedures of SiO₂ microspheres

One mixture containing methanol anhydrous, ammonia and ultra-pure water and another mixture containing methanol anhydrous and Tetraethoxysilane were concurrently dropped into the seed solution including methanol anhydrous, LUDOX AS-40 colloidal silica, ultra-pure water and ammonia under stirring condition at 198 rpm. After all the reagents were added dropwise, the SiO₂ suspension was continuously stirred for 2 hours, and finally monodisperse SiO₂ microspheres with suitable diameter were sorted out by the gravity settling method.

2. Calculation of UV-Vis-to-ammonia efficiency

The catalytic experiment for determining UV-Vis-to-ammonia efficiency(E) was performed in pure water (60 mL) without sacrificial reagent, and the irradiation area was 38 cm². TiO₂/SiO₂ opal composite membrane was used as the photocatalyst. The 500 W Xe lamp (Zolix LSH-X500) was employed as the light source. The E was determined by the following equation:

$$E = \frac{\Delta G_A \times n_A}{W \times S \times t} \times 100\%$$

where ΔG_A is the Gibbs free energy for ammonia generation from water and N₂ (339 kJ • mol⁻¹); n_A is the molar number of generated ammonia; W , S and t represent the incident light intensity, irradiation area and time, respectively.

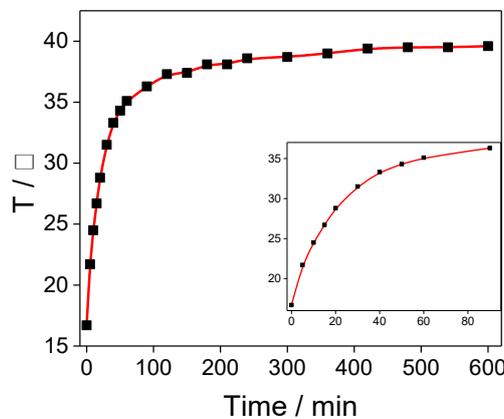


Figure S1. The change of reaction temperature with time.

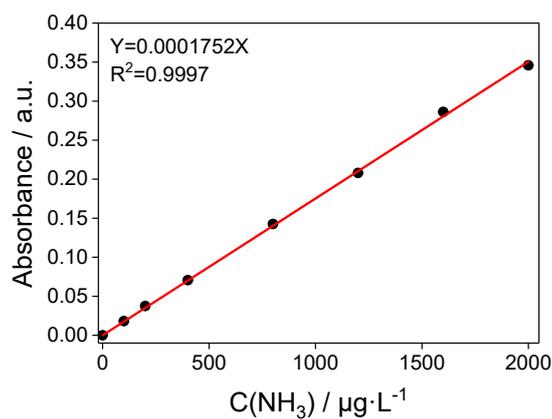


Figure S2. Standard curve of NH_4^+ with Nessler's reagent.

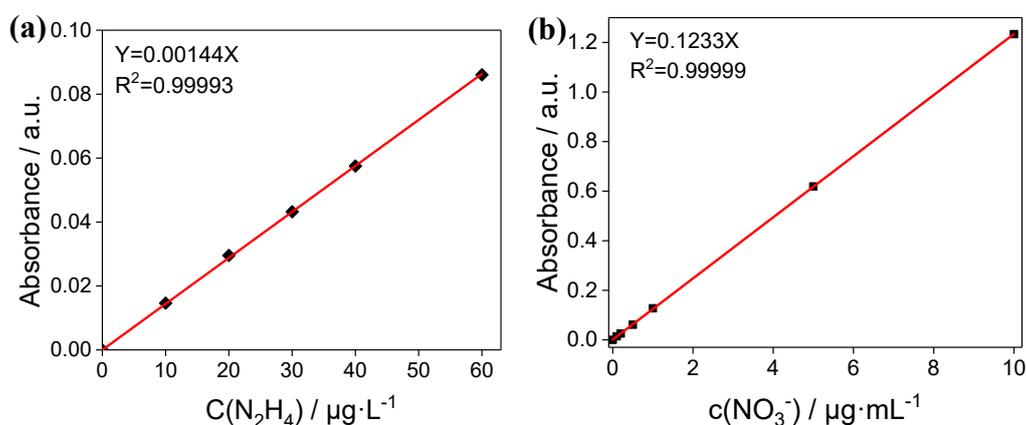


Figure S3. (a) Standard curve of N_2H_4 with para-(dimethylamino) benzaldehyde; (b) standard curve for NO_3^- and ammonium sulfamate.

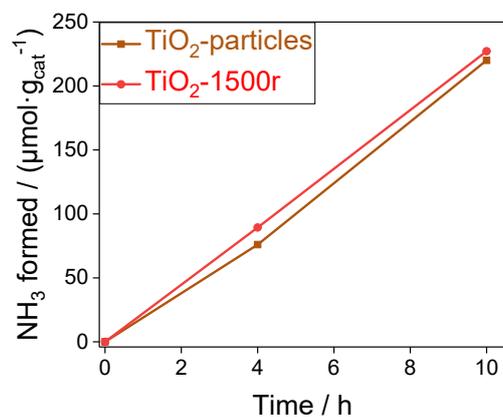


Figure S4. N_2 fixation catalytic performance of TiO_2 -1500r and TiO_2 -particles photocatalysts. Reaction conditions: pure water (60 mL), catalyst (38 cm²); light irradiation conditions: full-spectrum, 2.8mW·cm⁻².

(b)

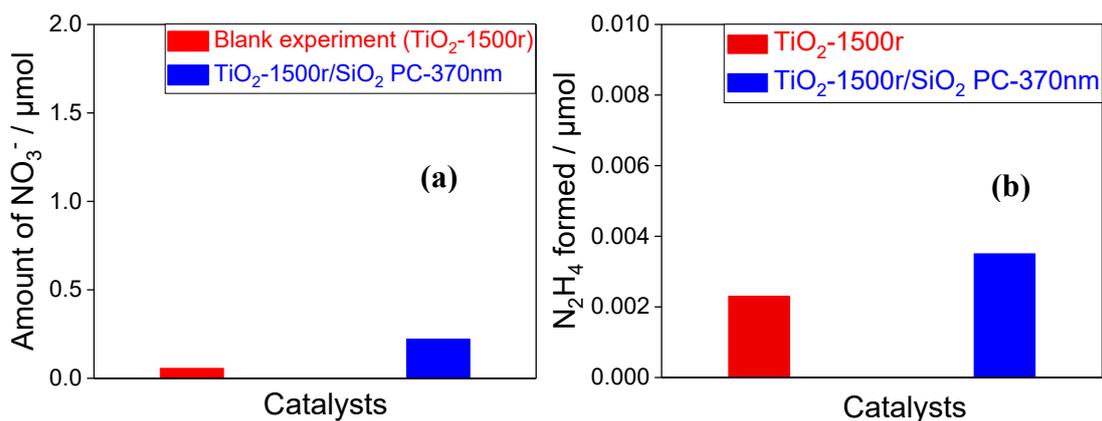


Figure S5. (a) Blank experiment for NO₃⁻ detection. Reaction conditions: pure water (60 mL), TiO₂-1500r catalyst (38 cm²), no light. And NO₃⁻ formed in phototatalytic reactions over TiO₂-1500r/SiO₂ PC-370nm catalysts. (b) Hydrazine formed in phototatalytic reactions over TiO₂-1500r, TiO₂-1500r/SiO₂ PC-370nm catalysts.

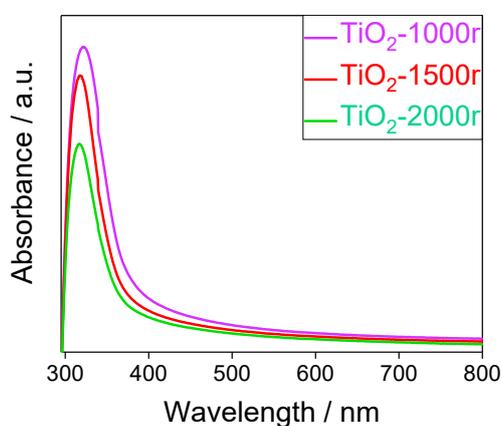


Figure S6. The UV-Vis absorption spectra of TiO₂ films with different thicknesses.

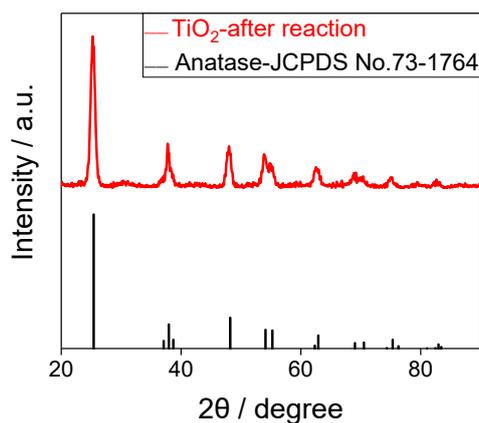


Figure S7. X-ray diffraction patterns of the reacted TiO₂.

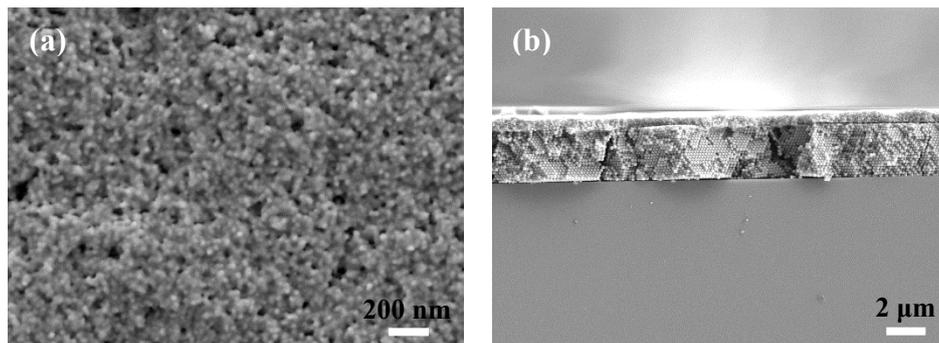


Figure S8. SEM images of top view (a) and cross section (b) of the reacted TiO_2 -1500r/ SiO_2 PC-370nm composite membranes.

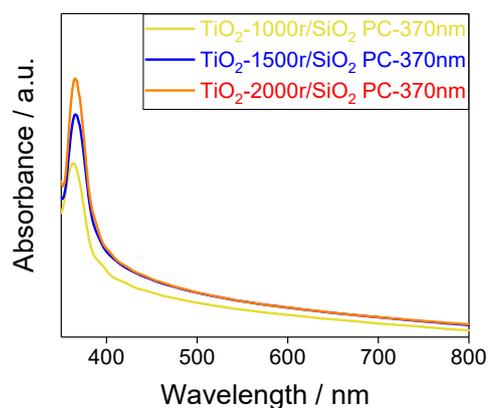


Figure S9. The UV-Vis absorption spectra of SiO_2 PC-370nm in TiO_2 -1000r/ SiO_2 PC-370nm, TiO_2 -1500r/ SiO_2 PC-370nm and TiO_2 -2000r/ SiO_2 PC-370nm photocatalysts, respectively.

Table S1. Nitrogen Fixation Rate of Various Photocatalysts for Photocatalytic N_2 Fixation

Catalyst	Reaction solvent	Light Source	Hole scavenger	Ammonia production rate	Ref.
0.2 wt% Fe-doped TiO_2	Water	360-W Hg-Arc Lamp (UV)	No	$10\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	11
0.4 wt% Co-doped TiO_2	Water	360-W Hg-Arc Lamp (UV)	No	$6.3\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	11
0.4 wt% Mo-doped TiO_2	Water	360-W Hg-Arc Lamp (UV)	No	$6.7\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	11
Defect TiO_2	Pure water	Hg lamp ($\lambda > 280\text{nm}$)	No	$0.73\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	12
Defective TiO_2 nanobamboo	Pure water	300 W Xe lamp (Full spectrum)	No	$48.3\text{ mg}\cdot\text{m}^{-2}\cdot\text{h}^{-1}$	13
$\text{TiO}_2/\text{SrTiO}_3/\text{g-C}_3\text{N}_4$	Water	300 W Xe lamp (AM 1.5G)	10 vol% methanol	$219.2\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$ 1	14
Fe-TiO_2/Au	Deionized water	300 W Xe lamp ($\lambda > 420\text{nm}$)	No	$22.4\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	15
Fe-doped TiO_2-SiO_2	Water	300 W Xe lamp (Full spectrum)	No	$32\mu\text{mol}\cdot\text{g}_{\text{cat}}^{-1}\cdot\text{h}^{-1}$	16

F-Vo-TiO₂	Deionized water	300 W Xe lamp (Full spectrum)	No	206 $\mu\text{mol} \cdot \text{g}_{\text{cat}}^{-1} \cdot \text{h}^{-1}$	18
TiO₂/Au/a-TiO₂	Deionized water	300 W Xe lamp (AM 1.5G)	No	13.4 $\text{nmol} \cdot \text{cm}^{-2} \cdot \text{h}^{-1}$	42
TiO₂-1500r/SiO₂ PC-370nm	Ultra-pure water	500 W Xe lamp (Full spectrum)	No	31 $\mu\text{mol} \cdot \text{g}_{\text{cat}}^{-1} \cdot \text{h}^{-1}$	this work
TiO₂-2000r/SiO₂ PC-370nm	Ultra-pure water	500 W Xe lamp (Full spectrum)	No	42.6 $\mu\text{mol} \cdot \text{g}_{\text{cat}}^{-1} \cdot \text{h}^{-1}$	this work