

## Electronic Supplementary Information (ESI)

### **TiO<sub>2</sub>/ZIF-67 nanocomposites synthesized by the microwave-assisted solvothermal method: a correlation between the synthesis conditions and antimicrobial properties**

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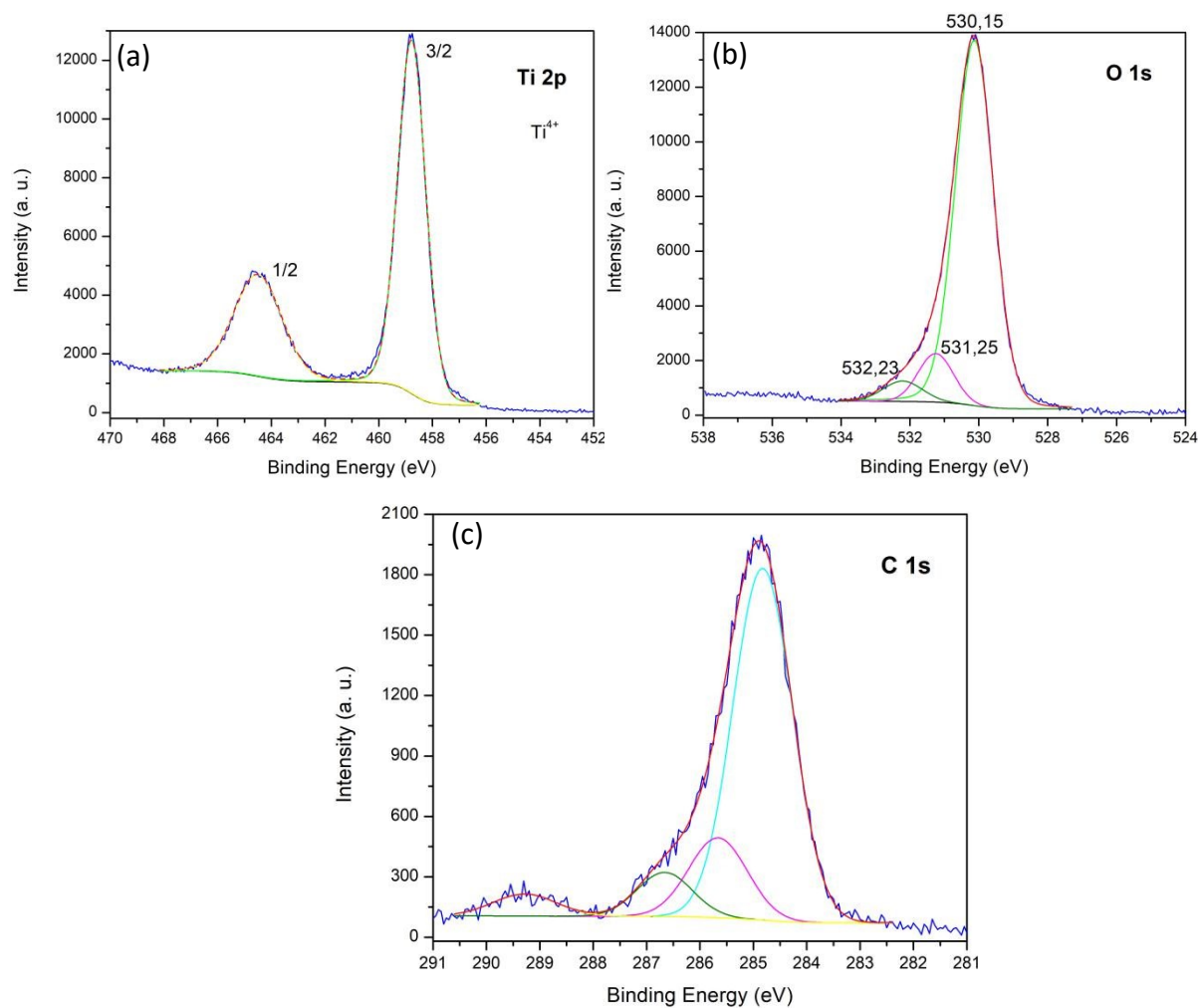
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IMGS: [ieda@quimica.ufpb.br](mailto:ieda@quimica.ufpb.br)

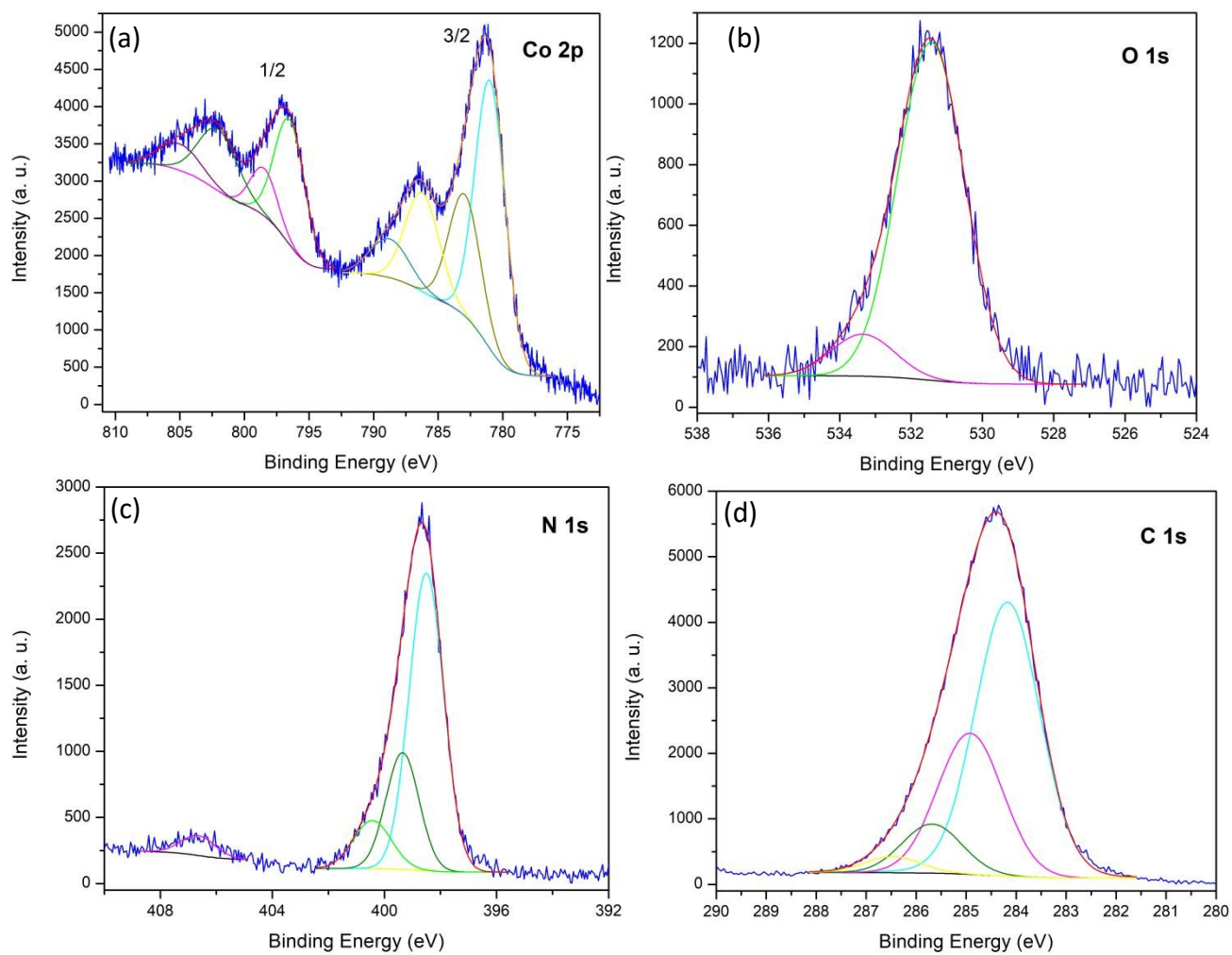
ACG: [ac\\_galca@infim.ro](mailto:ac_galca@infim.ro)

### **XPS Analysis**

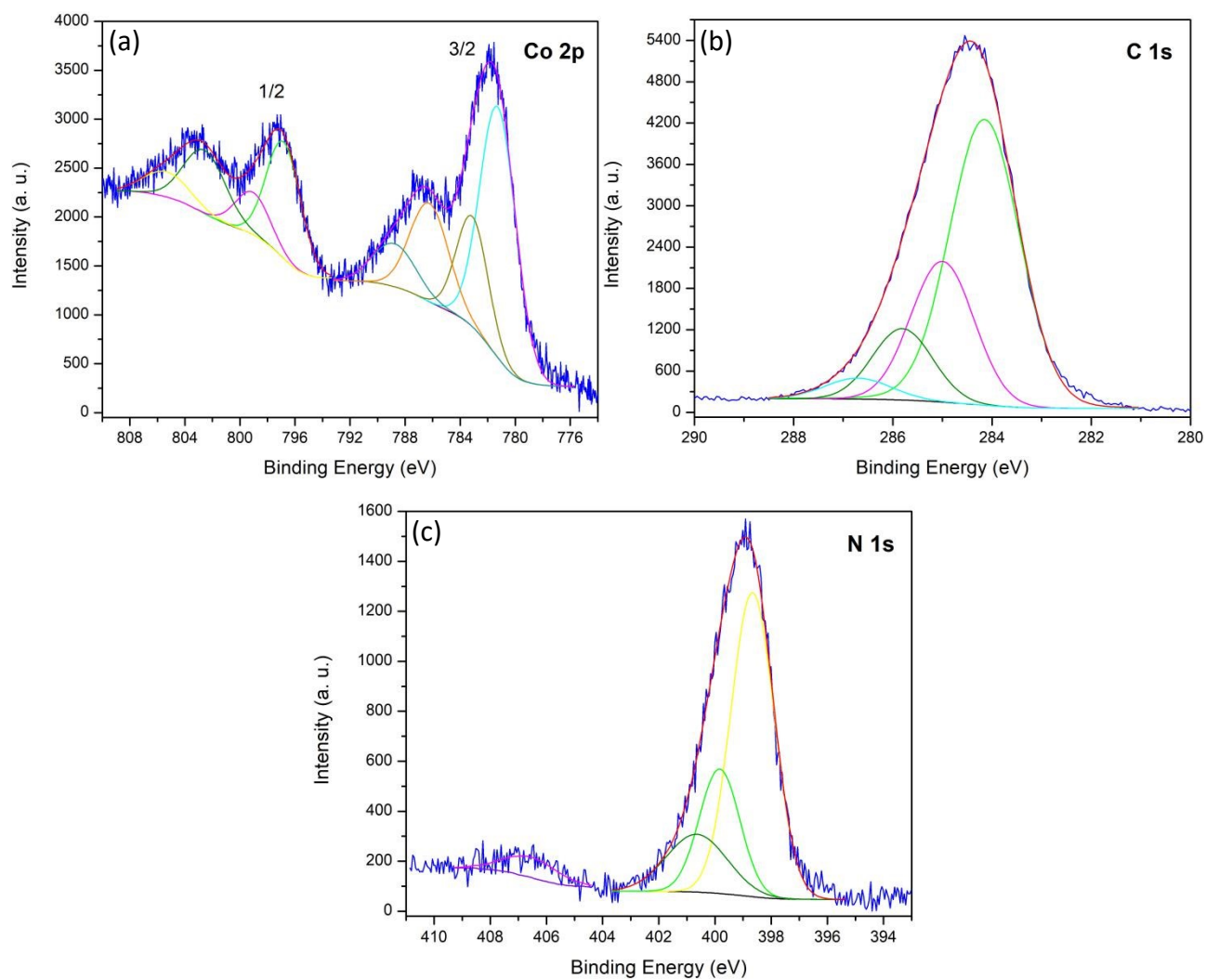
**Fig. S1** XPS deconvolutions for the synthesized TiO<sub>2</sub>: Ti 2p (a), O 1s (b) and C 1s (c).



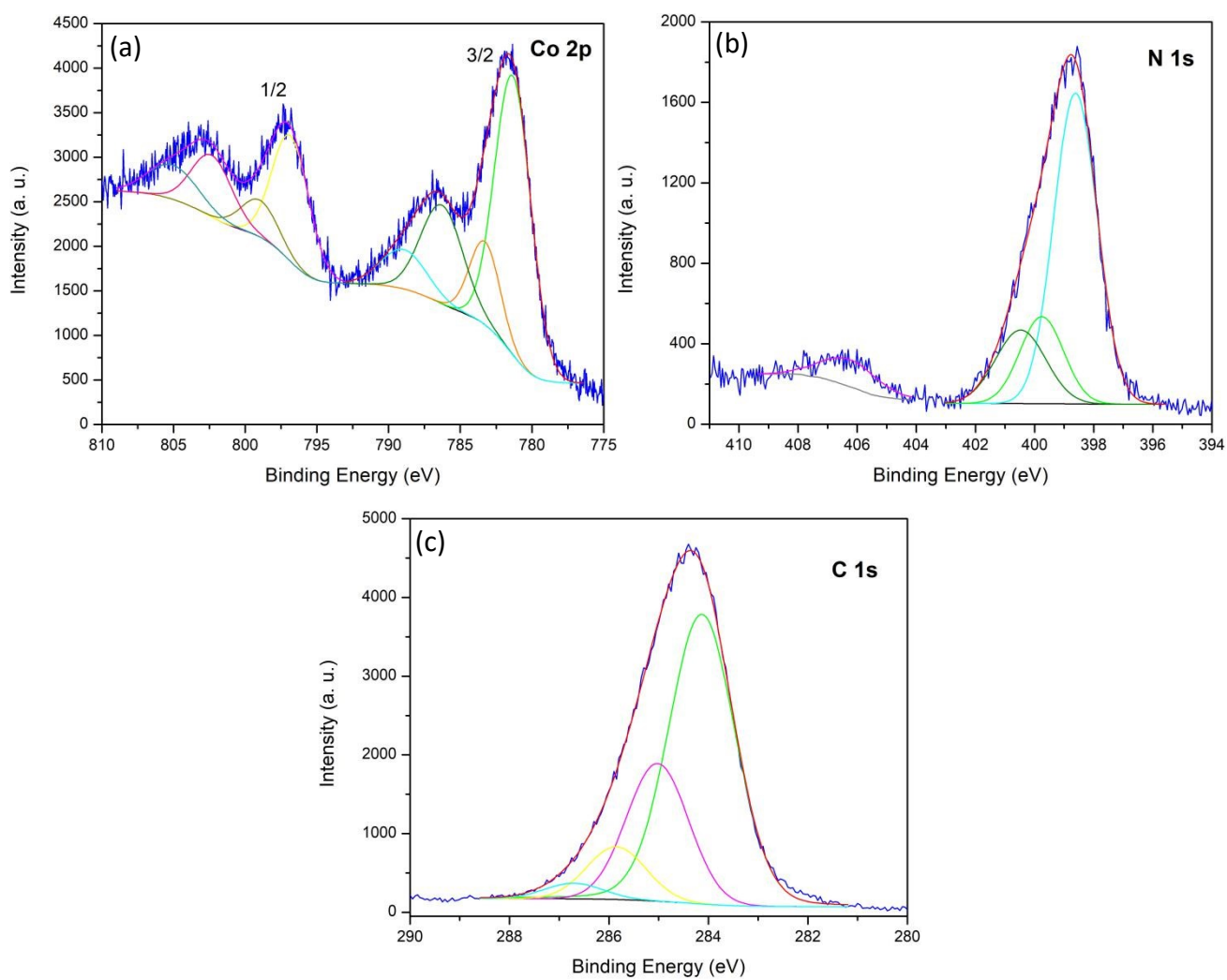
**Fig. S2** XPS deconvolutions for the synthesized ZIF-67: Co 2p (a), O 1s (b), N 1s (c) and C 1s (d).



**Fig. S3** XPS deconvolutions for the synthesized TSC: Co 2p (a), N 1s (b) and C 1s (c).

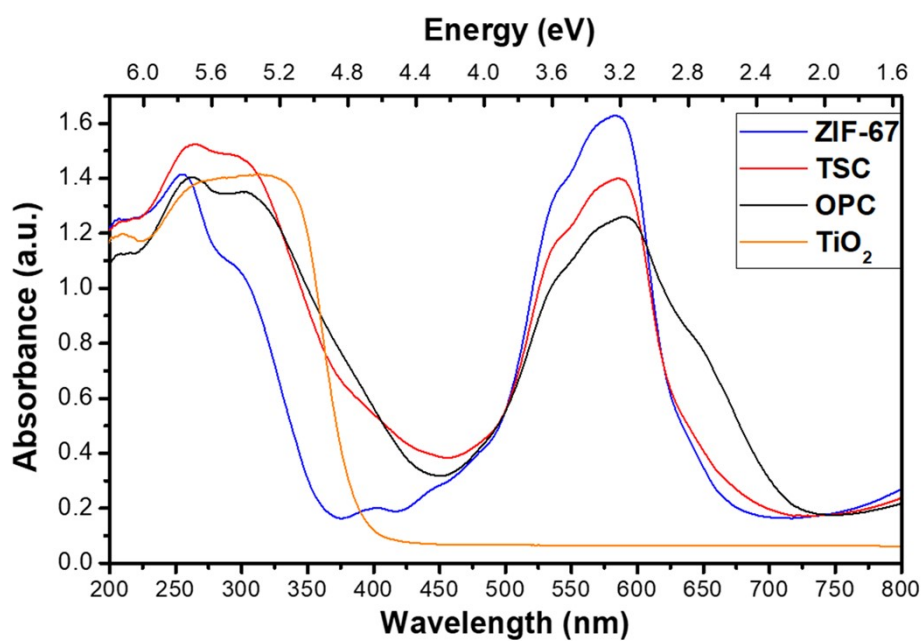


**Fig. S4** XPS deconvolutions for the synthesized OPC: Co 2p (a), N 1s (b) and C 1s (c).



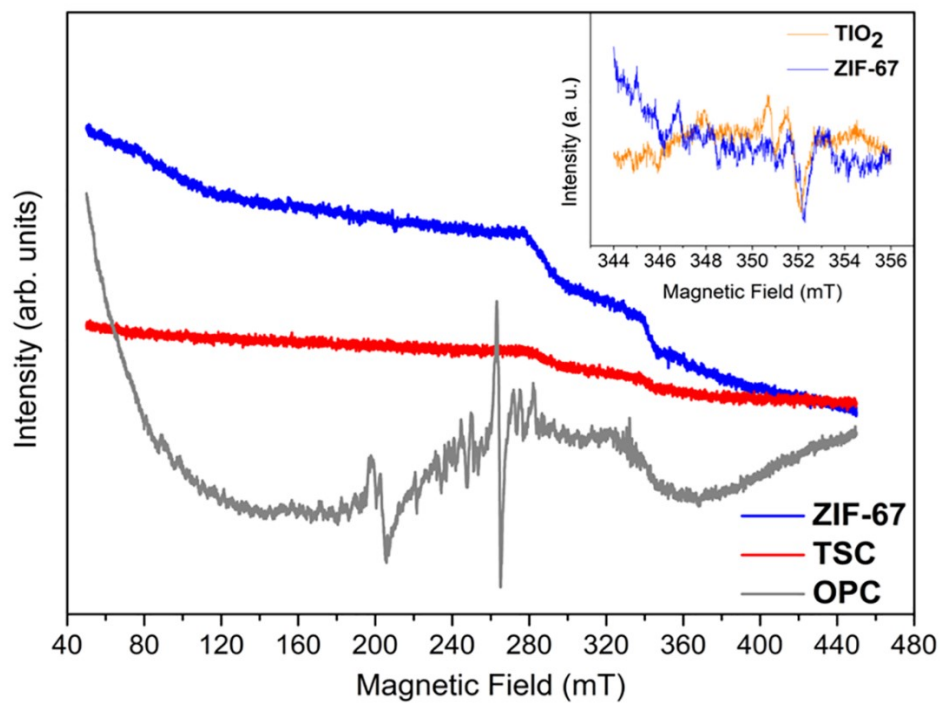
## UV-vis analysis

**Fig. S5** UV-vis spectra of the composites and pristine compounds



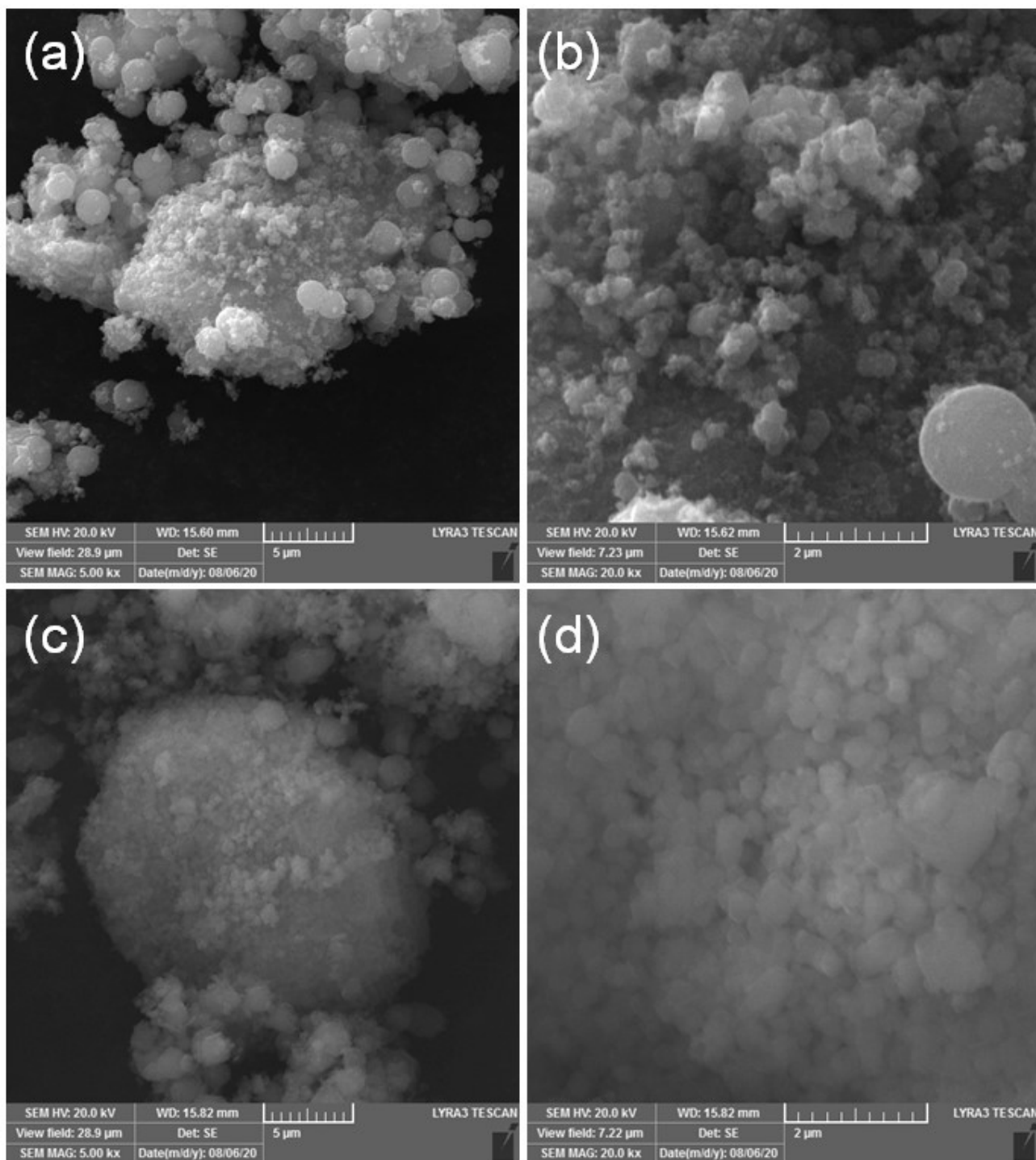
## EPR Measurements

**Fig. S6** EPR spectra of the pure TiO<sub>2</sub> and ZIF-67 compounds and their corresponding OPC and TSC composites. The inset indicates the low EPR signal observed for the pristine compounds.



## SEM analysis

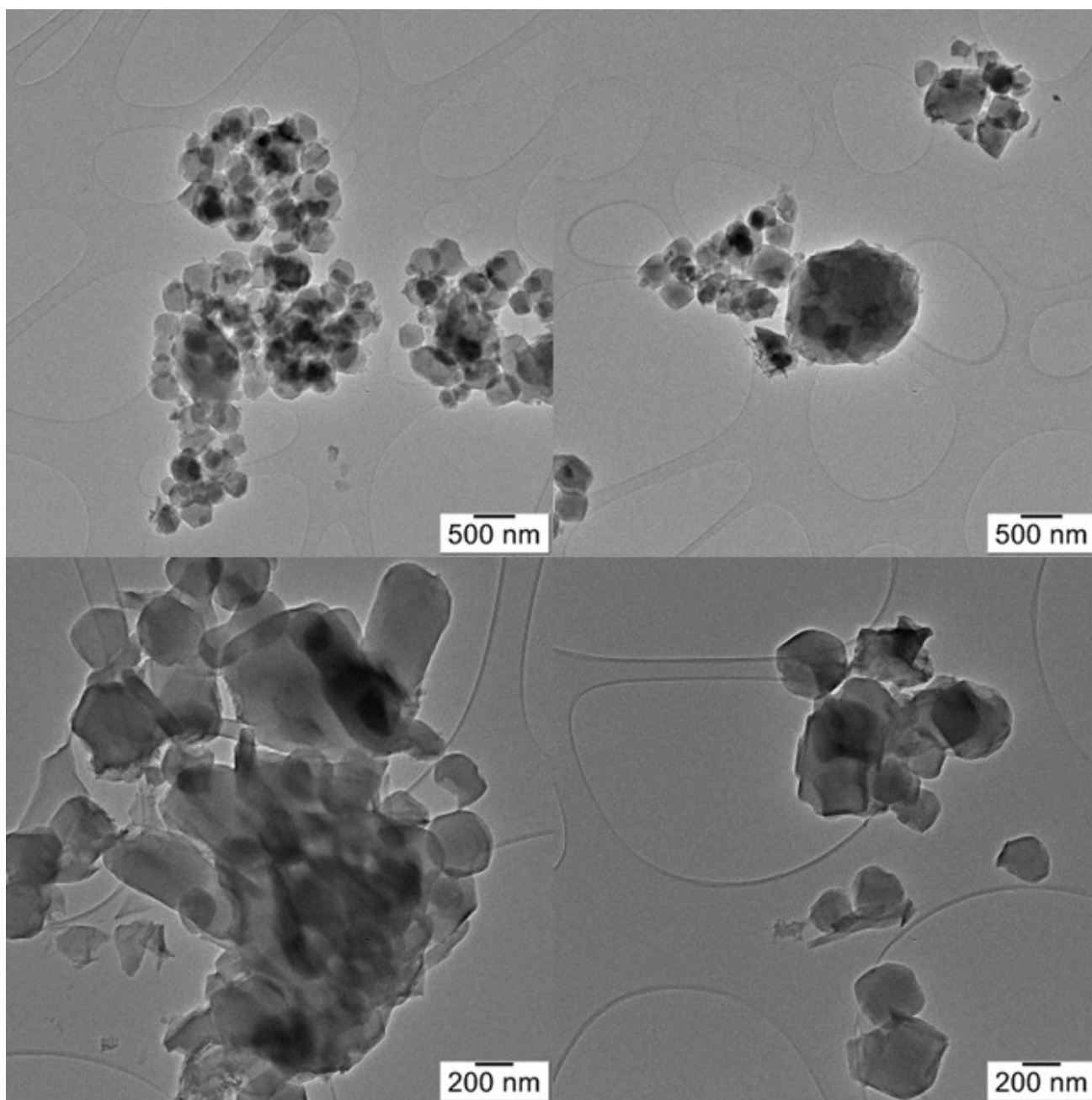
**Fig. S7** SEM images of the synthesized pristine  $\text{TiO}_2$  (a,b) and ZIF-67 (c,d) compounds



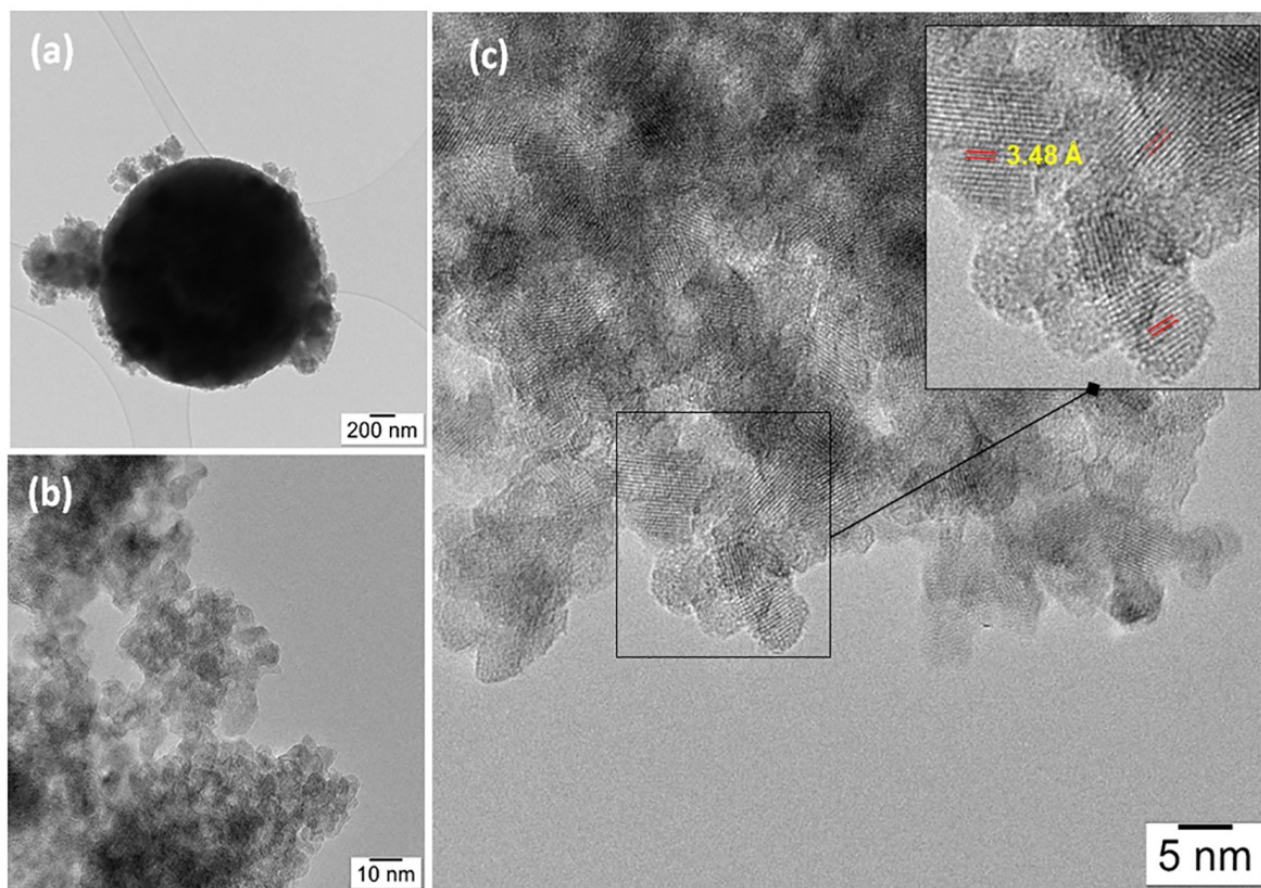


## TEM/HR-TEM analysis

**Fig. S8** TEM (images of the synthesized pristine ZIF-67 sample).

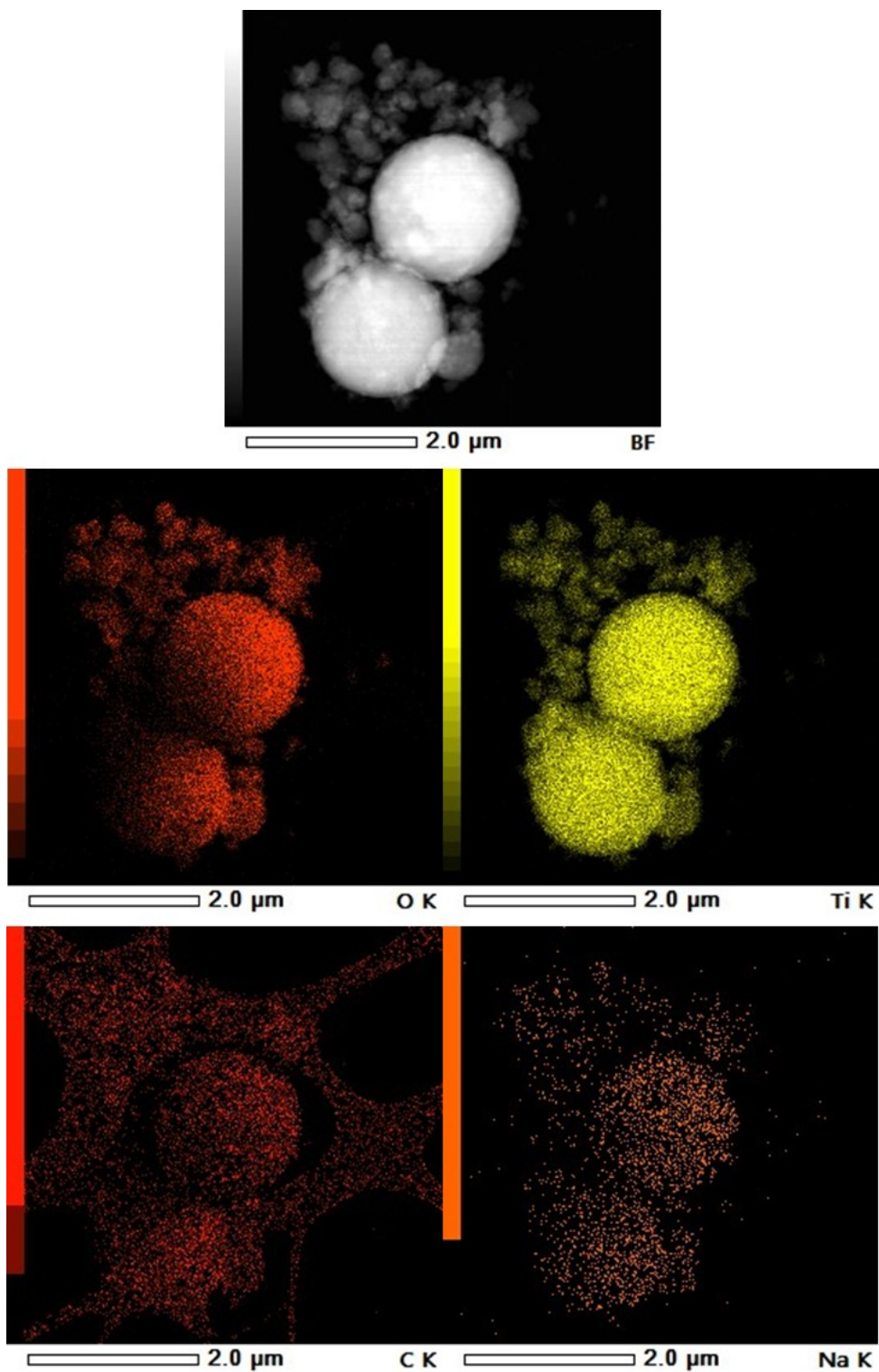


**Fig. S9** TEM (a) and HR-TEM (b,c) images of the synthesized TiO<sub>2</sub> nanoparticles



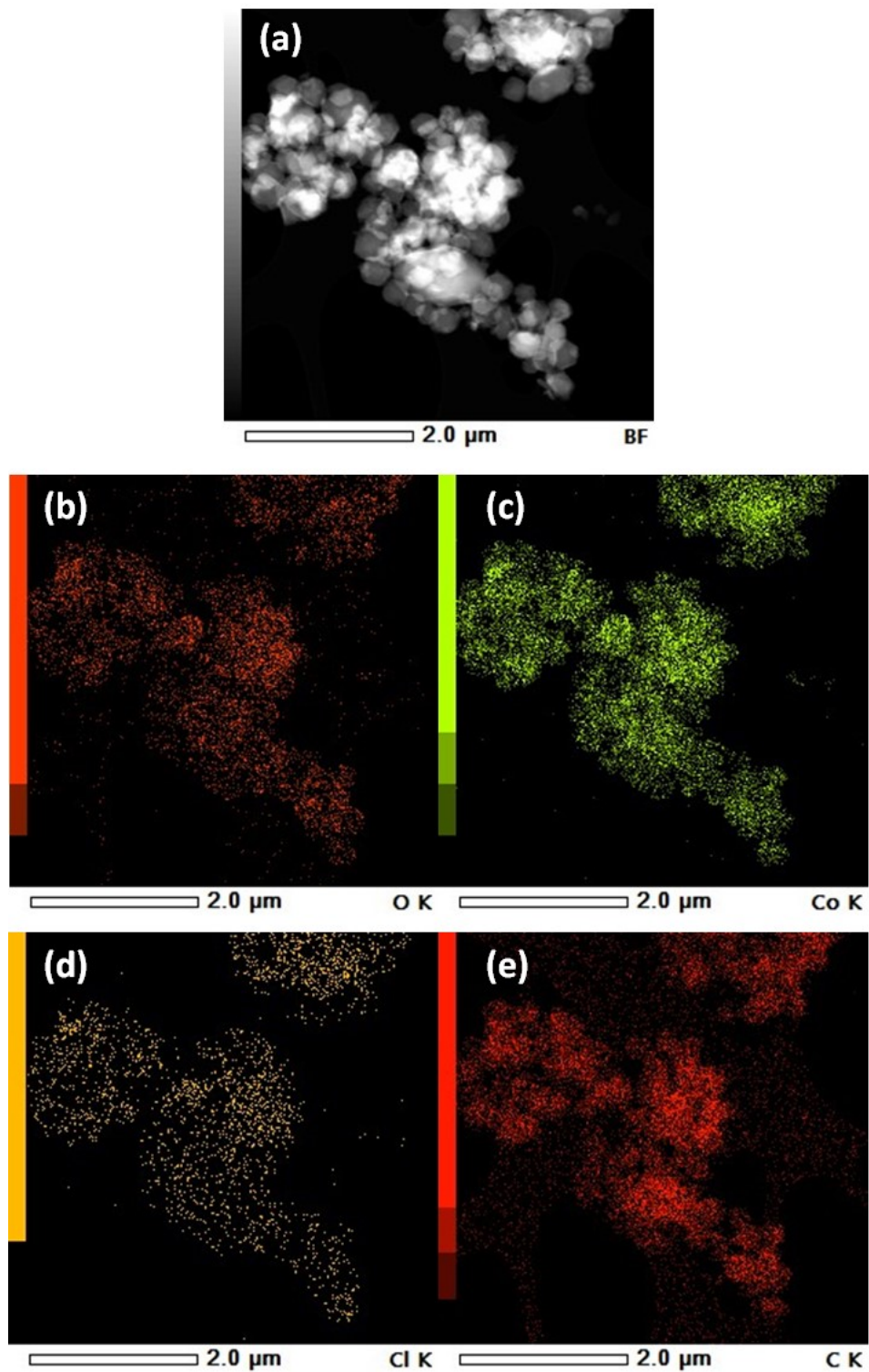
## TEM/EDS elemental mapping

**Fig. S10** Typical TEM and EDS elemental mapping images of O, Ti, C and Na for the TiO<sub>2</sub> nanoparticles.



## TEM/EDS mapping

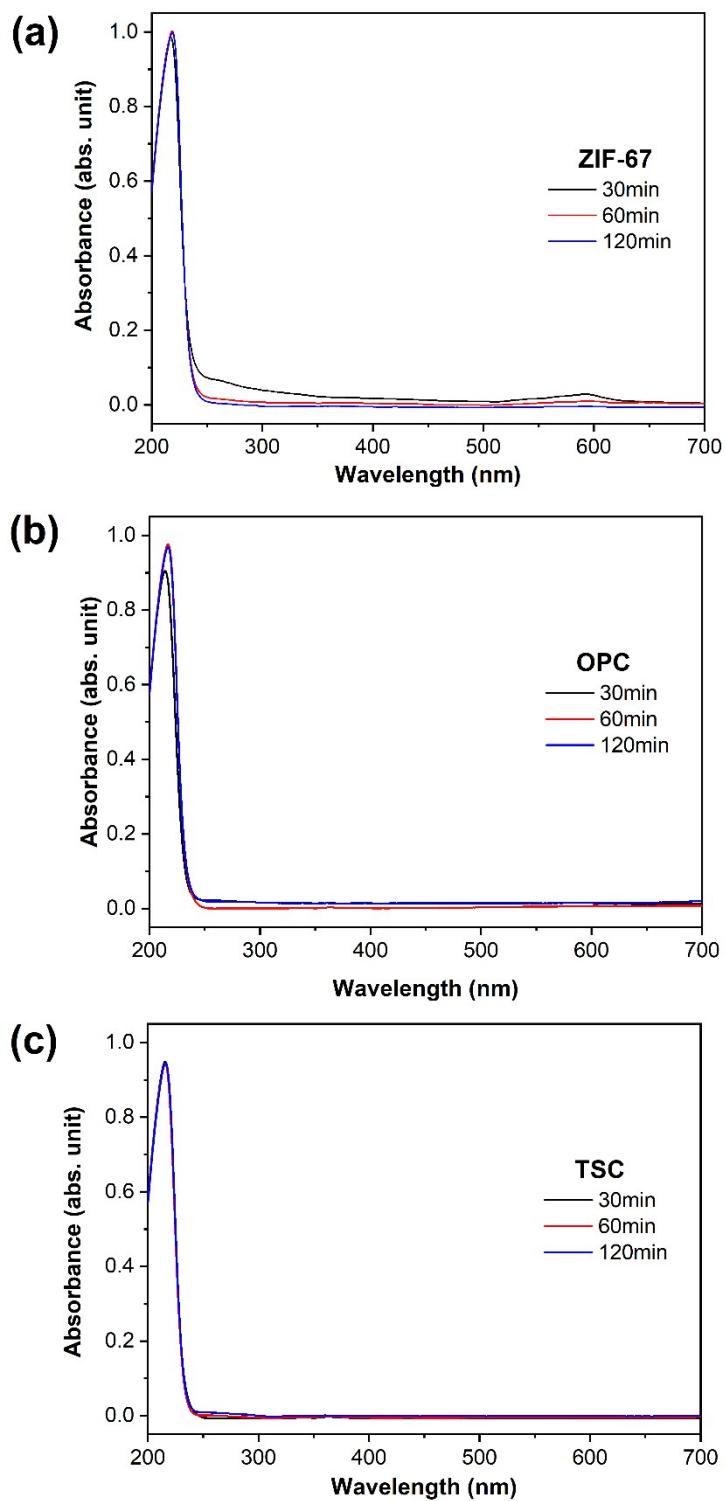
**Fig. S11** Typical TEM (a) and EDS elemental mapping of O (b), Co (c), Cl (d) and C (e) images for ZIF-67 nanoparticles (a-d).





## Cobalt/ligand release test by UV-vis spectroscopy

**Fig. S12** UV-vis absorption spectra for the Co ions and ligand release by ZIF-67 (a), OPC (b) and TSC (c) composites.



## Zeta Potential of TiO<sub>2</sub> nanoparticles

Fig. S13 Zeta potential variation of the TiO<sub>2</sub> nanoparticles as a function of pH.

