

Supporting Information

FAPbBr₃ Perovskite Nanocrystals Decorated on Graphitic Carbon Nitride (g-C₃N₄) Sheet for Interfacial Hole Transfer

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Fig. S1 PL spectra of FAPbBr₃ perovskite nanocrystals with solvent (0 μL to 620 μL).

Fig. S2 FT-IR spectra of PNC@g-C₃N₄ nanocomposites, PNC and g-C₃N₄.

Fig. S3 XPS analysis of PNC.

Fig. S4 XPS analysis of PNC@g-C₃N₄.

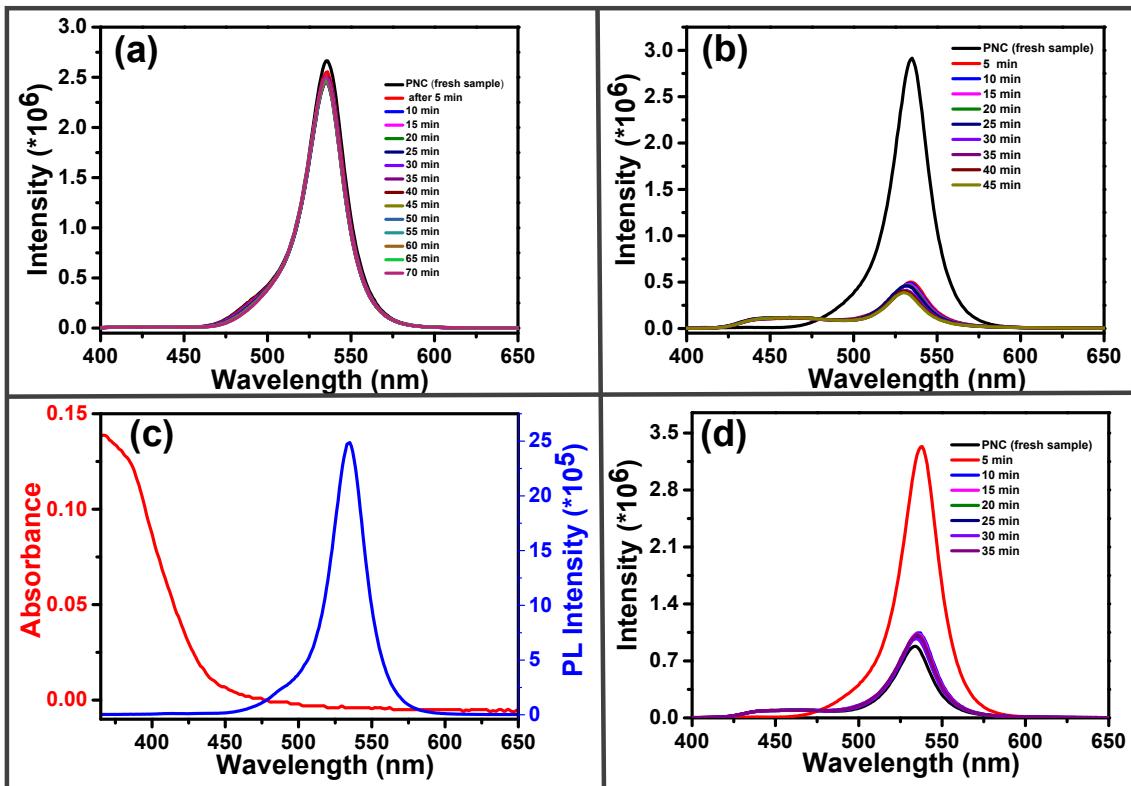


Fig. S1. (a) PL spectra of FAPbBr₃ perovskite nanocrystals with solvent (0 μ L to 620 μ L) show that no significant changes were observed. This represents that g-C₃N₄ is responsible for quenching the fluorescence of perovskite nanoparticle, (b & d) Fluorescence quenching of PNC in presence of (b) 620 μ L of the g-C₃N₄ solution, (d) with 300 μ L of a g-C₃N₄ solution, (c) No possibility of FRET.

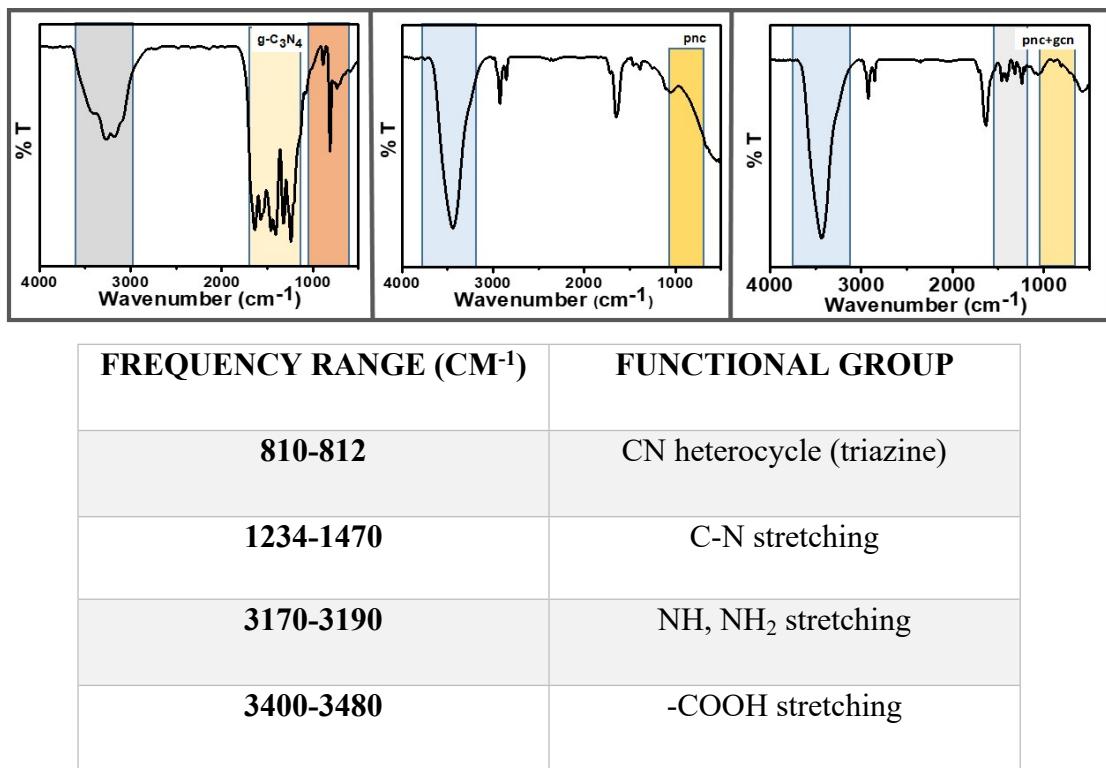


Fig. S2. FT-IR spectra of PNC@g-C₃N₄ nanocomposites, PNC and g-C₃N₄.

The FTIR spectra of FAPbBr₃@g-C₃N₄ and FAPbBr₃ show a negligible shift in stretching vibrational frequency. However, an additional peak present at 1313.27 cm⁻¹ indicates the g-C₃N₄ environment present around the FAPbBr₃ perovskite nanoparticle.

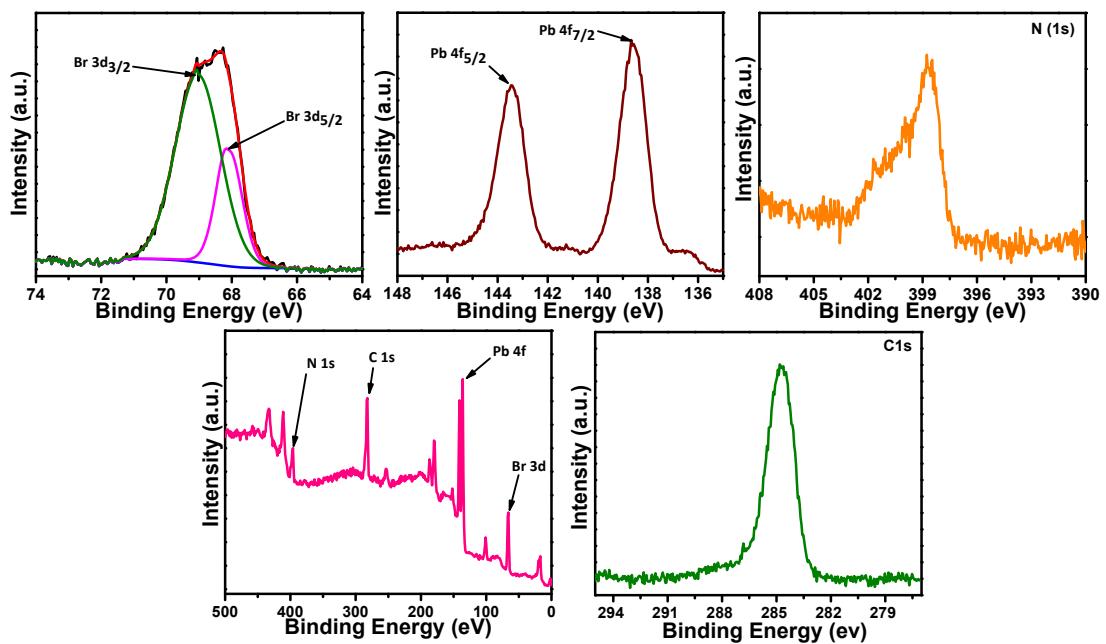


Fig. S3 XPS analysis of PNC.

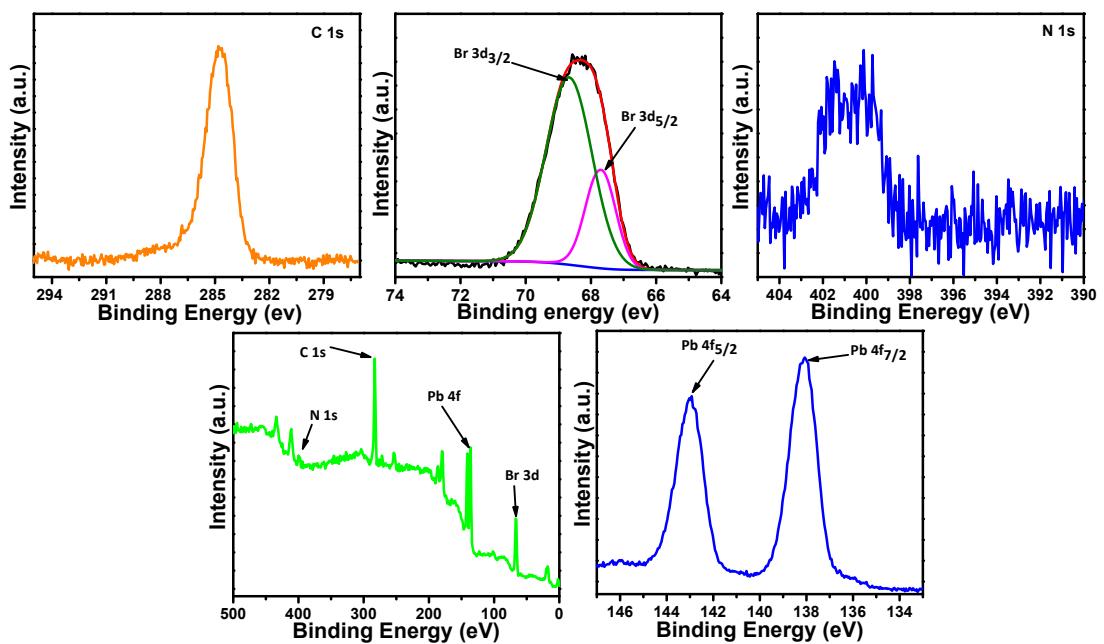


Fig. S4 XPS analysis PNC@g-C₃N₄.