

Supplementary Information

Zeolite imidazolate framework/g-C₃N₄ derived Co nanoparticles embedded in nitrogen doped carbon for efficient hydrogenation of phenol

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Table S1 Hydrogenation of phenol to cyclohexanol over various catalysts in references and this work

Catalysts	Reaction conditions				phenol	Cyclohexanol	Reference	Date
	T/°C	P/MPa	t/h	Solvent	Conversion/%	Selectivity/%		
Raney®Ni	220	2	2	water	87	52	[16]	2018
Ni/SiO ₂	200	1	4	water	100	100	[17]	2017
Ni/NiCaAlO _x	110	2	3	n-hexane	99.9	99.9	[18]	2021
Co/CeO ₂	150	3	16	water	82.5	100	[19]	2020
CoOx@C	150	3	16	water	98	77	[20]	2018
NiCo@C/ZrO ₂	200	2	4	water	96	91	[21]	2019
NiCo/γ-Al ₂ O ₃	250	2	4	water	82	85	[22]	2019
NiCo/SDSW	100	5	4	Isopropanol	99.9	99.9	[23]	2020
NiCo/Mg _x Ni _y O	150	2	2.5	n-Hexane	99.9	99.9	[24]	2020
This work	120	2	3	water	97.8	100		

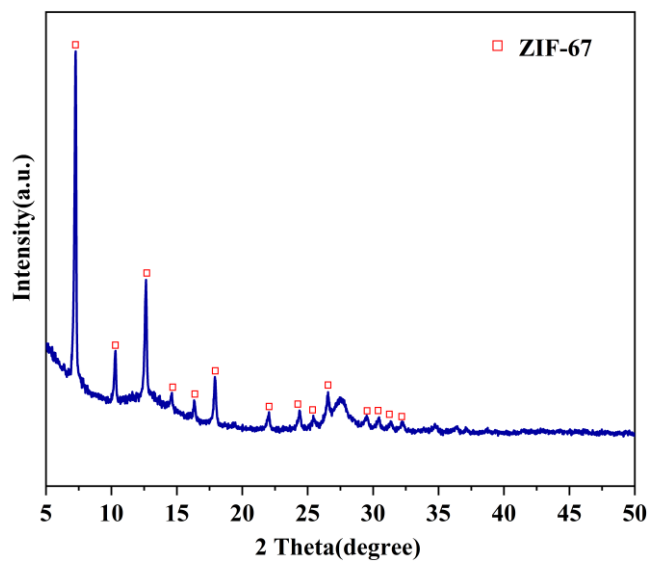


Fig. S1 XRD pattern of ZIF-67/g-C₃N₄

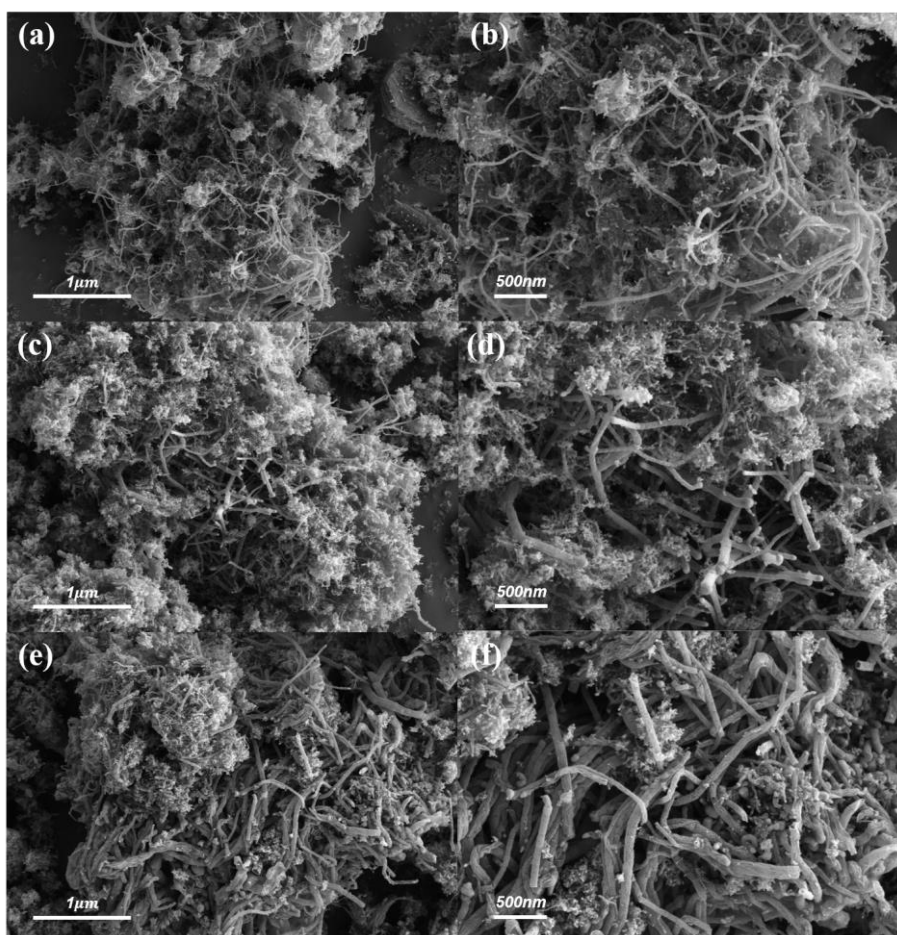


Fig. S2 SEM images of Co/NPCN_{0.25}, Co/NPCN_{1.0} and Co/NPCN_{1.5} catalysts

(a)(b) Co/NPCN_{0.25}; (c)(d) Co/NPCN_{1.0}; (e)(f) Co/NPCN_{1.5}

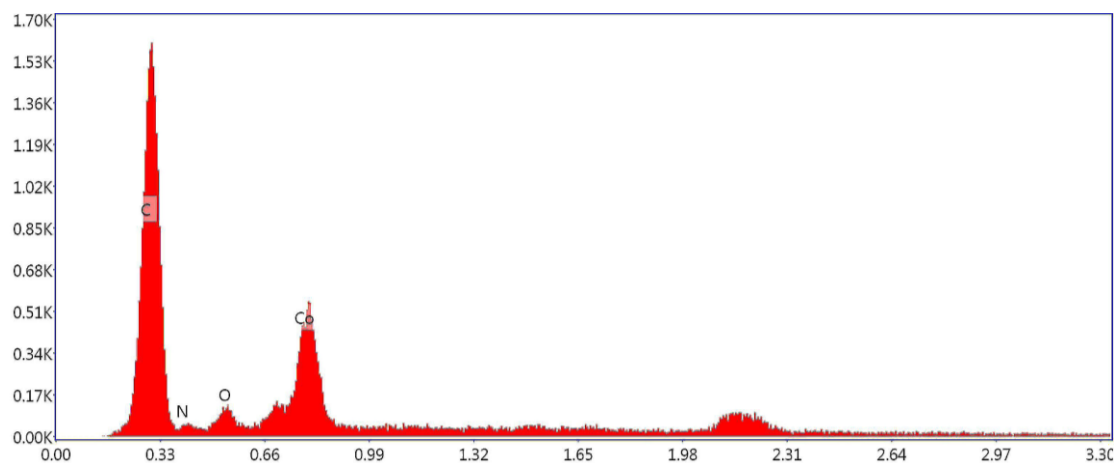


Fig. S3 EDS spectra of Co/NPCN_{0.5}

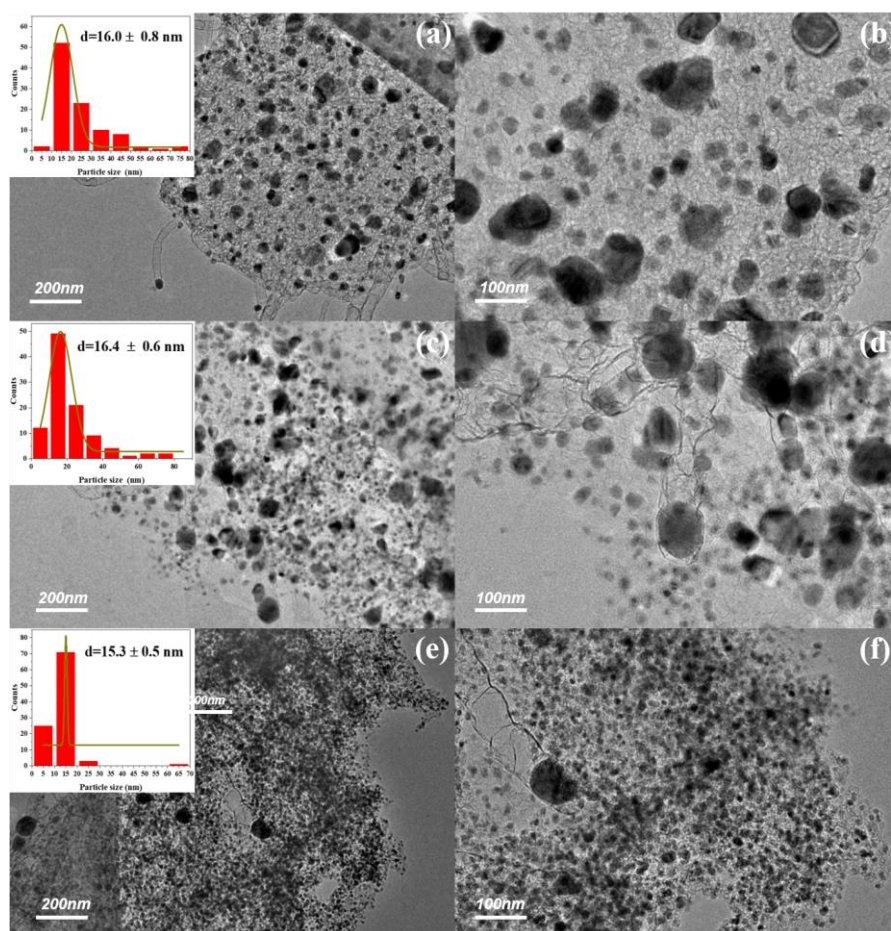


Fig. S4 TEM images of Co/NPCN catalysts with different g-C₃N₄ contents
 (a), (b): Co/NPCN_{0.25}; (c), (d): Co/NPCN_{1.0}; (e), (f): Co/NPCN_{1.5}

Table S2 Determination of cobalt content in Co/NPCN with different g-C₃N₄ content by ICP-AES

Catalysts	Co wt. (%)
Co/NPCN _{0.25}	38.6
Co/NPCN _{0.5}	40.3
Co/NPCN _{1.0}	40.4
Co/NPCN _{1.5}	40.7

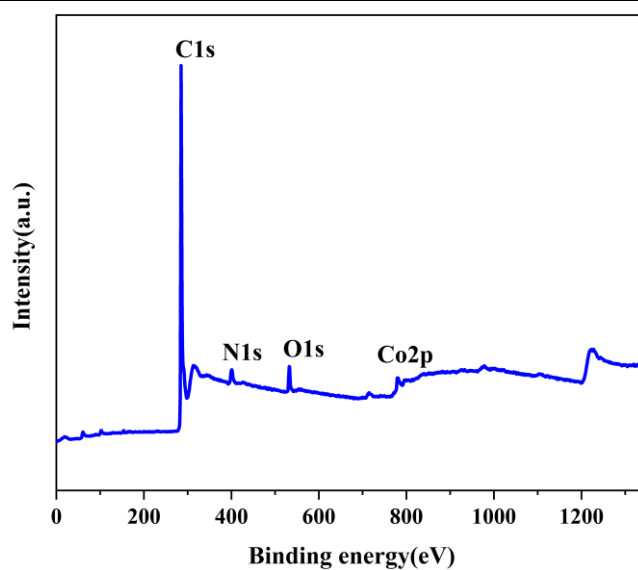
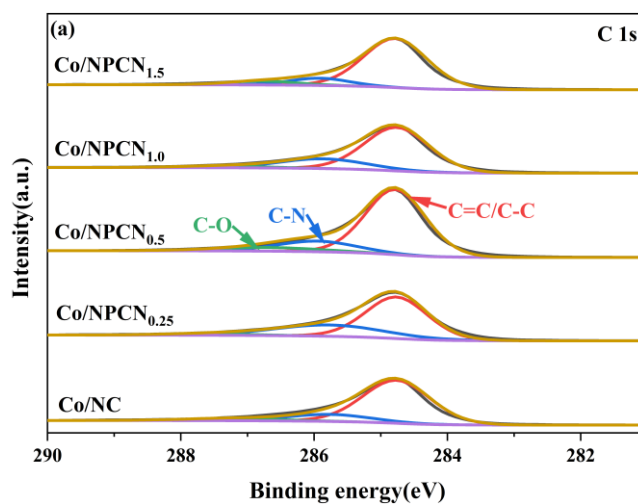


Fig. S5 XPS survey spectrum of Co/NPCN_{0.5}



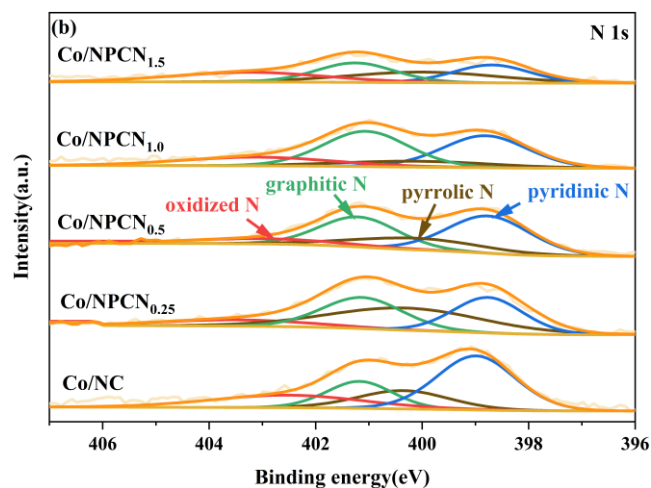


Fig. S6 XPS Spectra of Co/NPCN catalysts with different g-C₃N₄

(a) C 1s; (b) N 1s

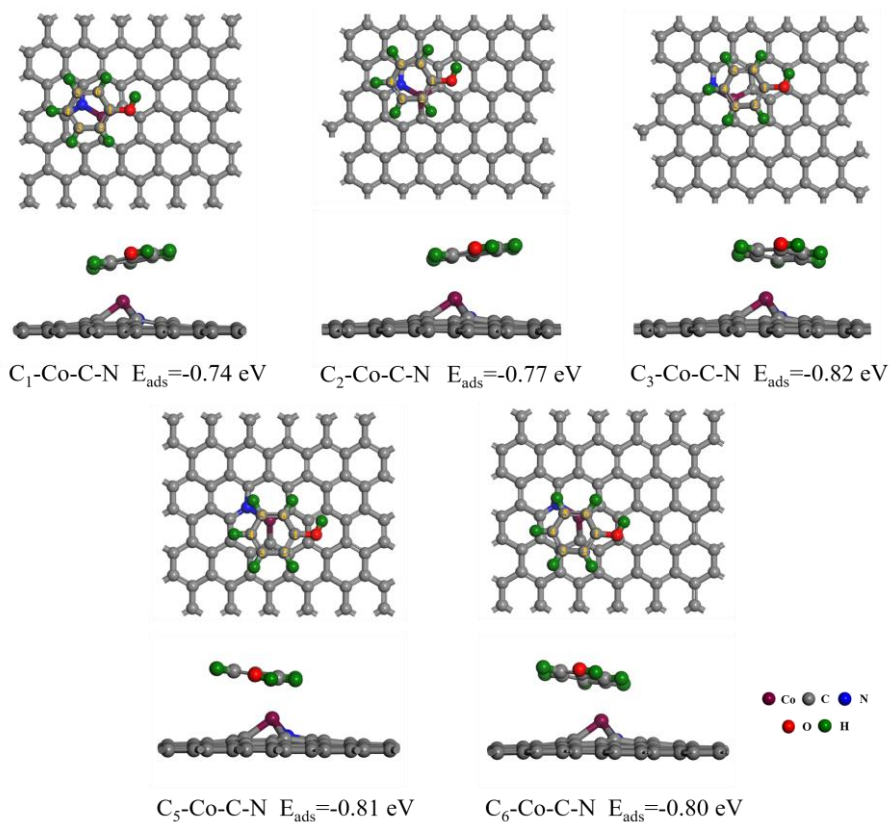


Fig. S7 Parallel stable adsorption configurations of C₁, C₂, C₃, C₅ and C₆ of phenol on Co sites of Co-C-N

The atomic colours of cobalt, carbon, nitrogen, oxygen and hydrogen are dark fuchsia, dark gray, blue, red and green, respectively.

Table S3 Adsorption energies of parallel and vertical adsorption of phenol and cyclohexanol on Co sites of Co-C-N

Adsorption atom	Adsorption energy (eV)				
	Parallel adsorption		Vertical adsorption		
C ₃	phenol	-0.82	O	phenol	-0.79
	cyclohexanol	-0.36		cyclohexanol	-1.04
O	phenol	-0.79			
	cyclohexanol	-1.09			

Table S4 Catalytic performance of Co/NPCN catalysts at different pyrolysis temperatures

Catalysts	Phenol Conversion (%)	Cyclohexanol Selectivity (%)
Co/NPCN-600	0	--
Co/NPCN-700	72.8	100
Co/NPCN-800	97.8	100
Co/NPCN-900	29.6	100

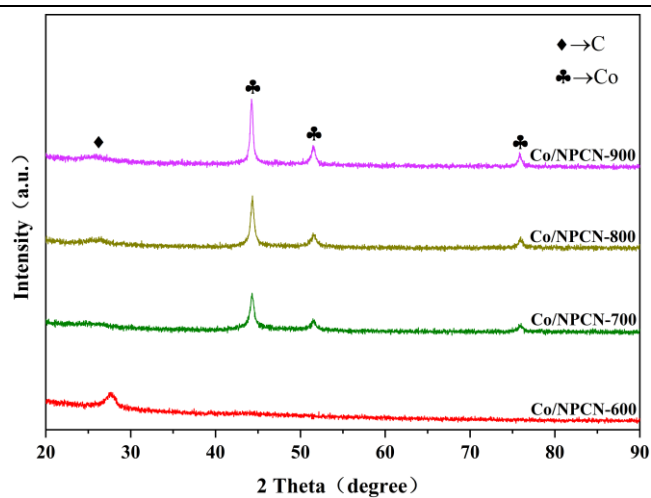


Fig. S8 XRD patterns of Co/NPCN catalysts at different pyrolysis temperatures

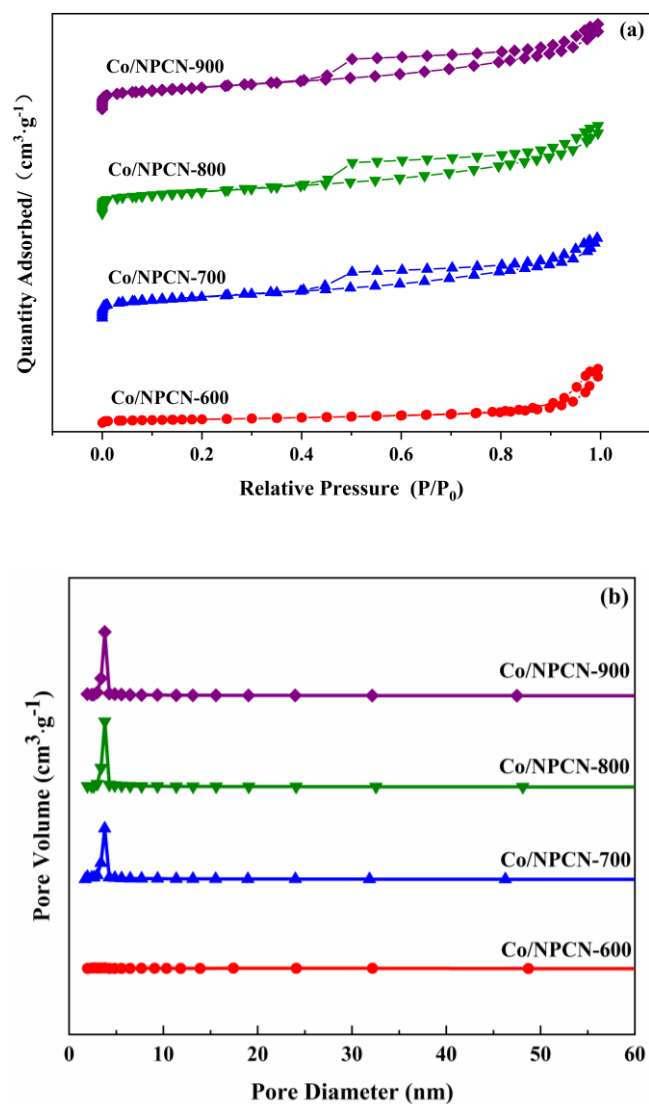


Fig. S9 N_2 adsorption-desorption isotherms (a) and pore size distribution curve (b)

Table S5 Structural parameters of Co/NPCN catalysts at different pyrolysis temperatures

Catalysts	S_{BET} (m^2/g)	V_t^b (cm^3/g)	D^c (nm)
Co/NPCN-600	79.1	0.3	18.1
Co/NPCN-700	320.1	0.5	6.1
Co/NPCN-800	342.6	0.8	8.7
Co/NPCN-900	330.3	0.5	6.3

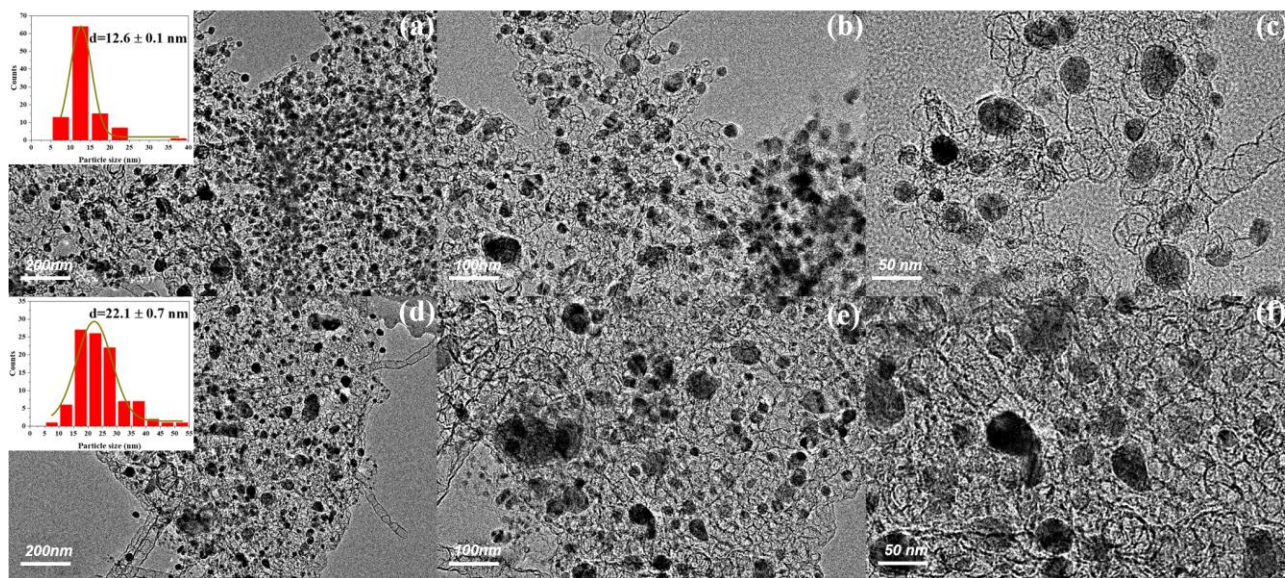


Fig. S10 TEM images of Co/NPCN catalysts at different pyrolysis temperatures

(a), (b), (c) Co/NPCN-700; (d), (e), (f) Co/NPCN-900

Table S6 Determination of cobalt content in Co/NPCN catalysts at different pyrolysis temperatures by ICP-

AES	
Catalysts	Co wt (%)
Co/NPCN-600	13.1
Co/NPCN-700	39.1
Co/NPCN-800	40.4
Co/NPCN-900	40.7

Table S7 Effects of different reaction pressures on selective hydrogenation of phenol

Pressure (MPa)	phenol Conversion (%)	cyclohexanol Selectivity (%)
1.0	96.4	100
1.5	96.9	100
2.0	97.8	100
2.5	97.8	100
3.0	97.9	100

Reaction conditions: catalyst 50mg, phenol 0.2mmol, H₂O 10mL, 120°C, 3h.

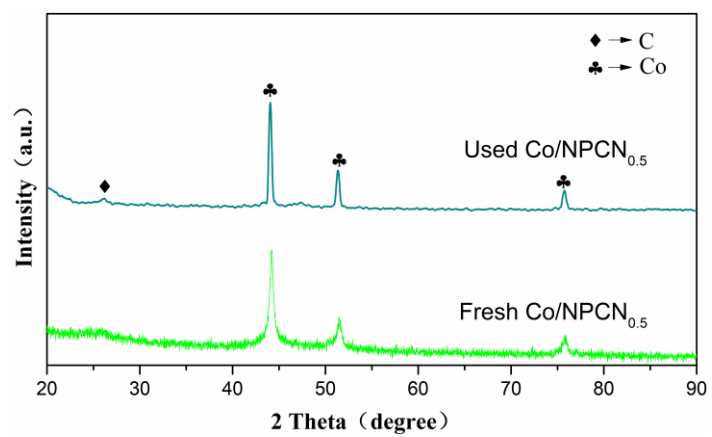


Fig. S11 XRD patterns of fresh and used Co/NPCN_{0.5} catalyst