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New Journal of Chemistry

Supporting information

Heteroatoms preintercalated CI-terminated Ti₃C₂T_x MXene wrapped with

Mesoporous Fe₂O₃ Nanospheres for improved Sodium ion Storage

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Figure S1. SEM image of the densely packed Ti₃AlC₂ precursor.



Figure S2(a). SEM image of $Ti_3C_2T_x$ MXene nanosheets after borohydride treatment.



Figure S2(b). SEM image of Fe₂O₃-Ti₃C₂T_xcomposite.



Figure S3. EDS mapping images of Ti₃C₂T_x MXene nanosheets after borohydride treatment.



Figure S4. XRD pattern of pristine Ti₃AlC₂.



Figure S5. The corresponding EDS mapping images of Cl-terminated Ti₃C₂T_x MXene after APS solution treatment.



Figure S6. The high-resolution spectra of (a) the Cl 2p for Fe₂O₃@Ti₃C₂T_xcomposite, (b) and the Al 2p for Ti₃C₂T_x MXene.



Figure S7. The CV curves of (a) $Ti_3C_2T_x$ MXene, (b) and Fe₂O₃ for initial three cycles at a scan rate of 0.1 mV s⁻¹.



Figure S8. Cycling performance of MXene without heteroatoms pre-intercalation at a current density of 1.0 A g⁻¹.



Figure S9. Rate capability performance of MXene without heteroatoms pre-intercalation anode.



Figure S10. CV curves(a), Log (peak current) versus log (sweep rate) plots(b), and Contribution ratios of capacitive and diffusion-controlled behaviors(c) at different sweep rates of Fe2O3 electrode.

	Performance		
Materials	Cycling [Capacity(mA h g ⁻¹)/Current density(A g ⁻¹)/Cycles]	Rate [Capacity(mA h g-1)/Current density(A g ⁻¹)]	References
Fe ₂ O ₃ @MXene	350/1/200	188.5/4	This work
Fe ₂ O ₃ @mGC	377/0.1/100	267/4	S1 ¹
Hierarchical Fe ₃ O ₄ hollow nanostructures	150/0.1/50	80/1	S 2 ²
Fe ₂ O ₃ :Ge NFs	320/0.05/50	140/2	S3 ³
C@Fe ₂ O ₃	305/0.05/100	150/1	$S4^4$
Fe ₂ O ₃ @GNS	440/0.1/100	126/2	S5 ⁵
MFe ₂ O ₃ @N-HCNs	417/0.1/100	102/5	S6 ⁶

Table S1. Comparison of electrochemical performance for the similar anode materials fromthis work and previous reports

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