

Supporting information for

Bifunctional TPE based Fluorescent Sensor for Liquid Viscosity and Amyloid β

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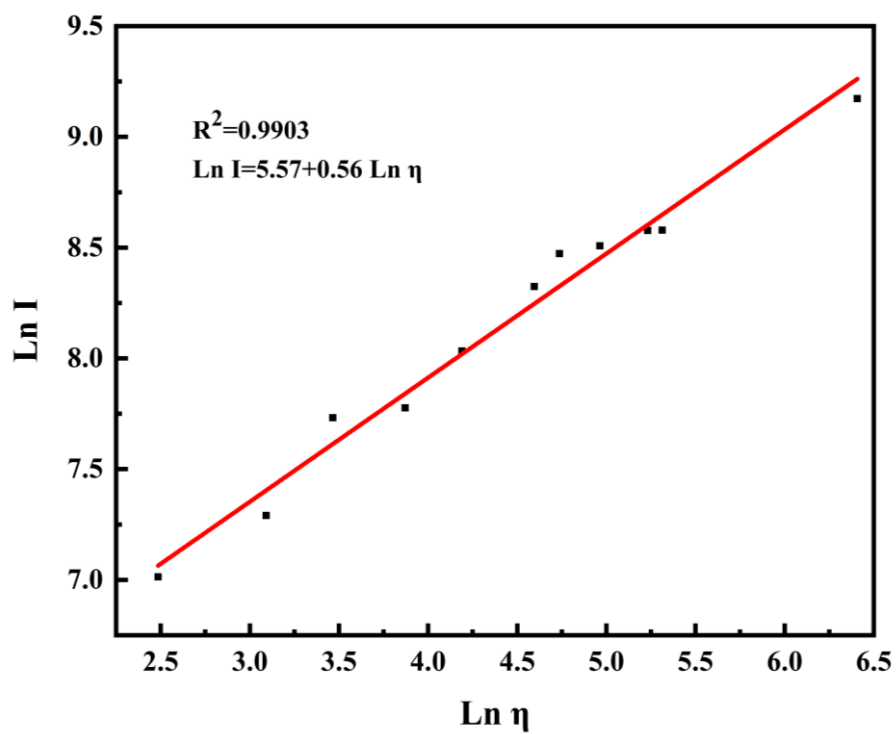


Figure S1. Repeated Plots of natural logarithmic fluorescent intensity of **TPE-Q** vs. natural logarithmic mixture viscosity. [**TPE-Q**] = 10 μM , excitation wavelength was 340 nm.

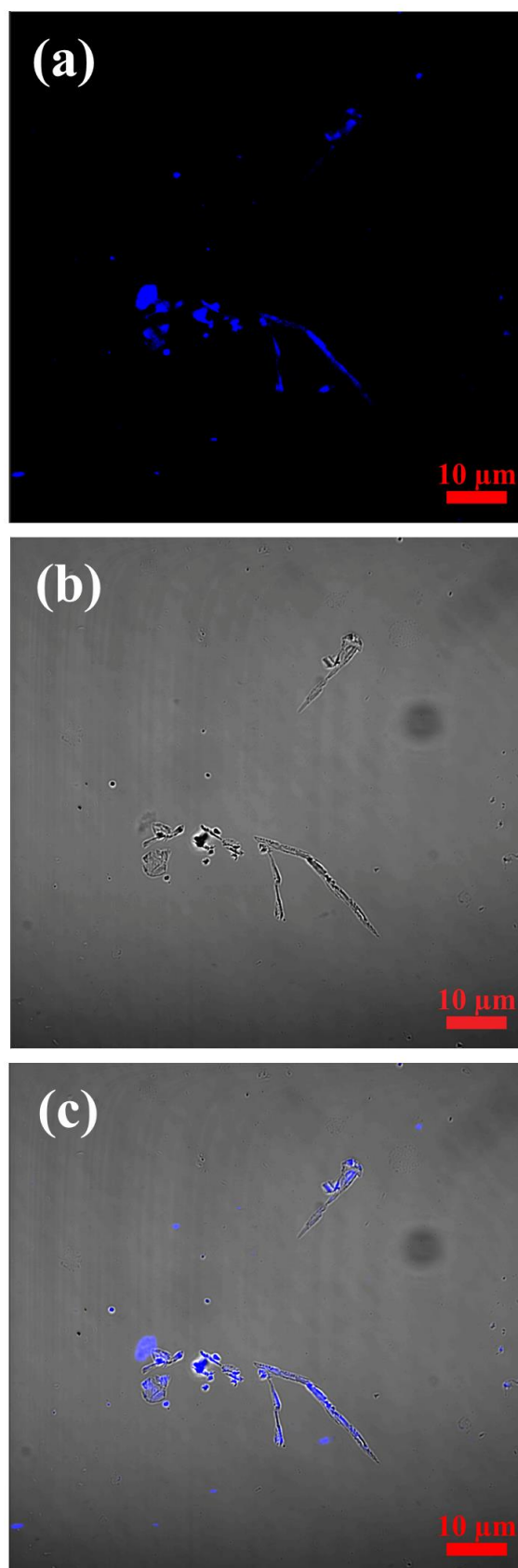


Figure S2. Fluorescent images of $A\beta_{1-42}$ stained with **TPE-Q**. Excitation wavelength was 405 nm. $[TPE-Q] = 1 \mu M$, $[A\beta_{1-42}] = 10 \mu M$. (a) Fluorescent field, (b) Bright field, (c) Merge

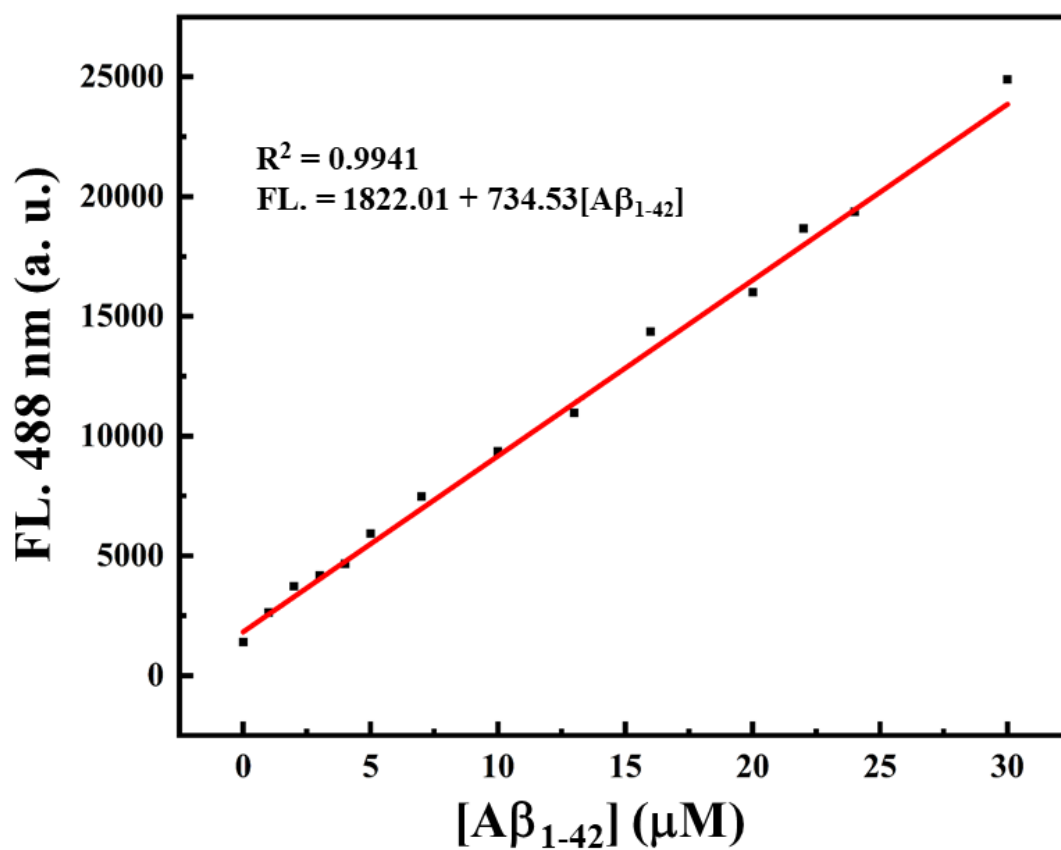


Figure S3. Repeated linear relationship of fluorescent emission intensity of TPE-Q and concentration of $A\beta_{1-42}$.

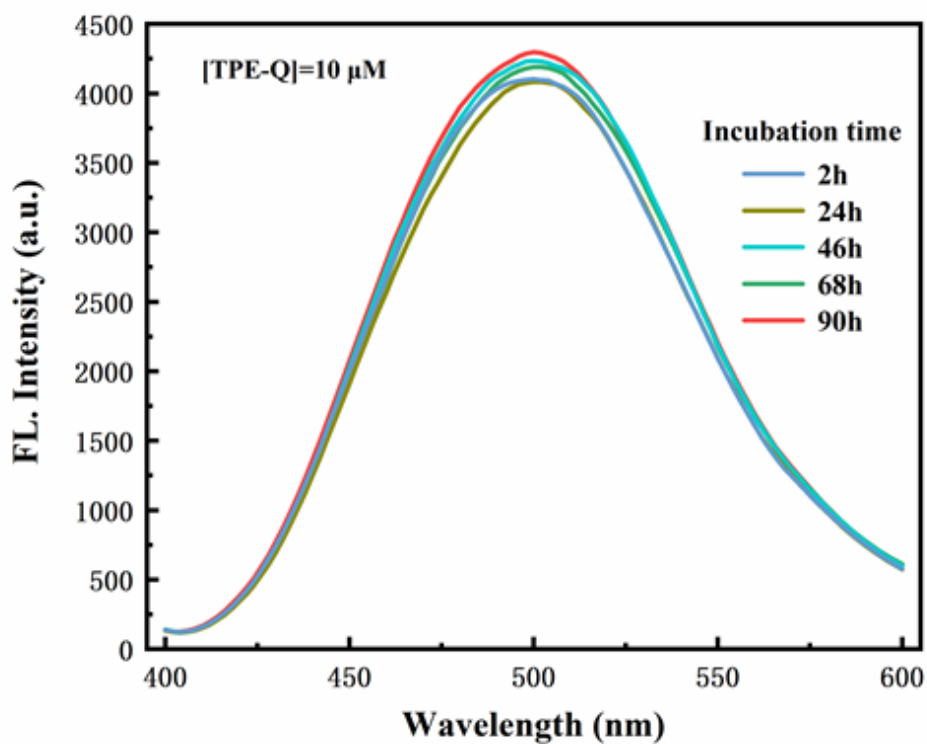


Figure S4. Repeated fluorescent spectra of TPE-Q in the absence of $A\beta_{1-42}$ with different incubation time, $[TPE-Q] = 10 \mu\text{M}$.

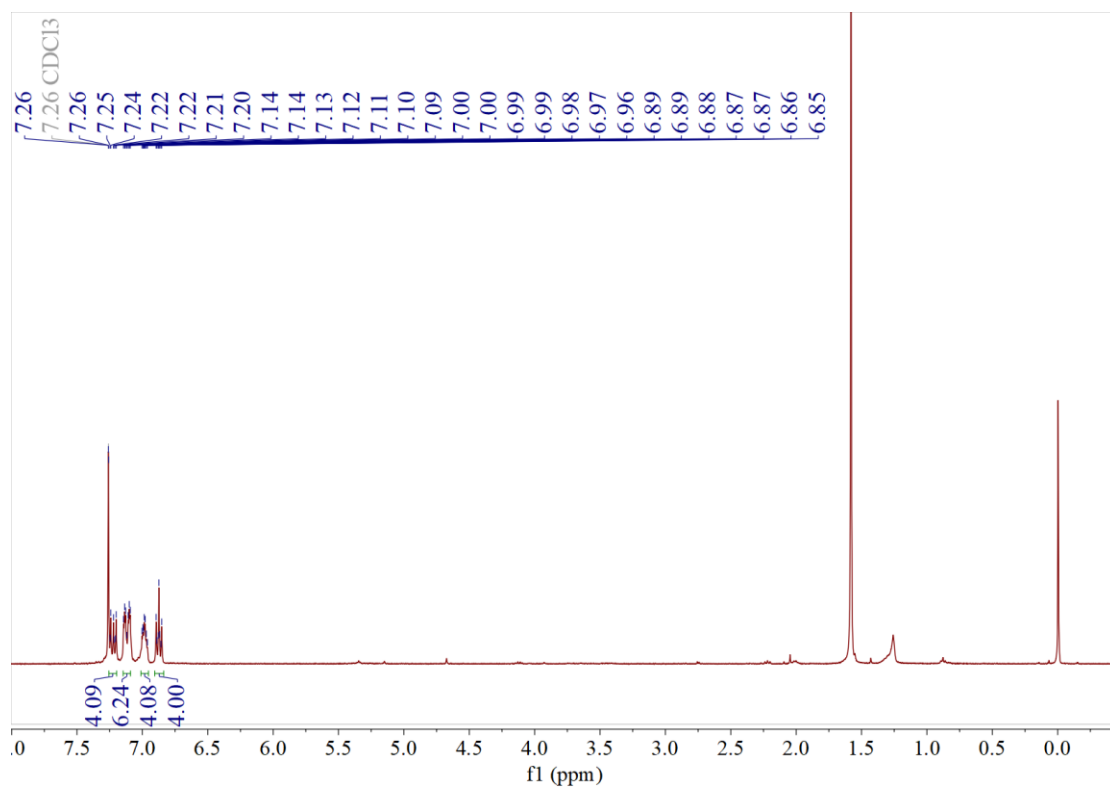


Figure S5. ^1H NMR of compound **1**. (CDCl_3 , 300 MHz)

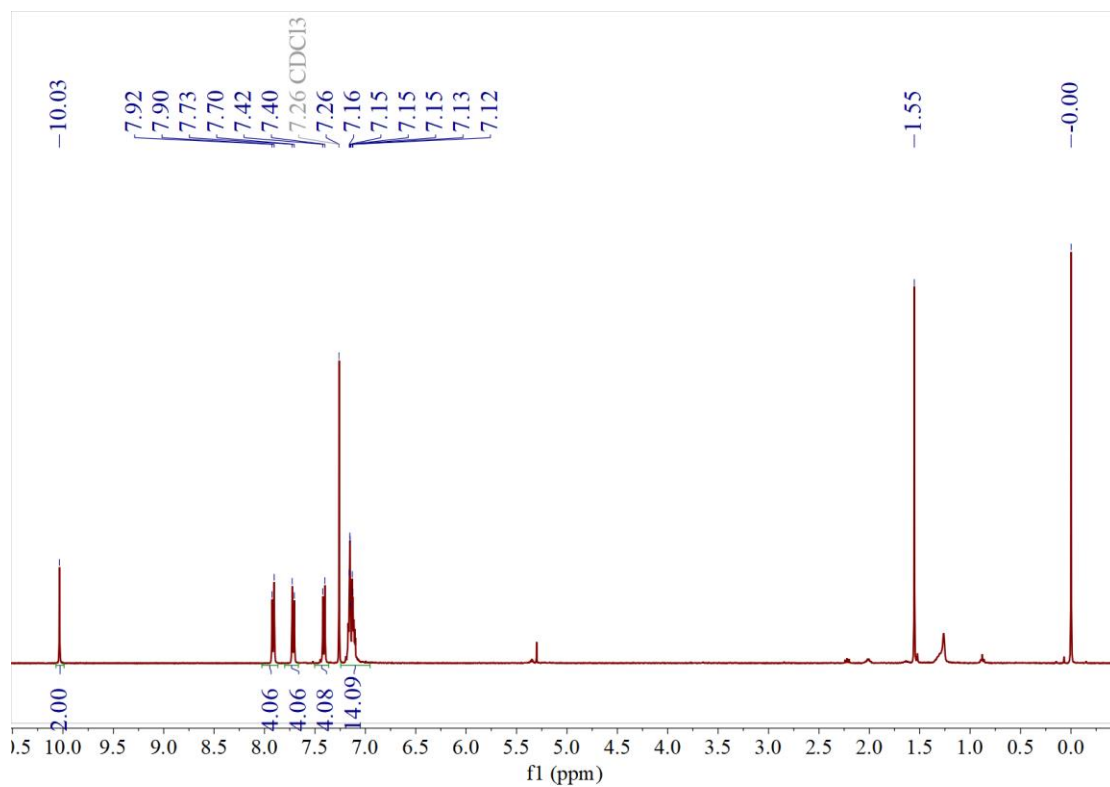


Figure S6. ^1H NMR of compound **2**. (CDCl_3 , 300 MHz)

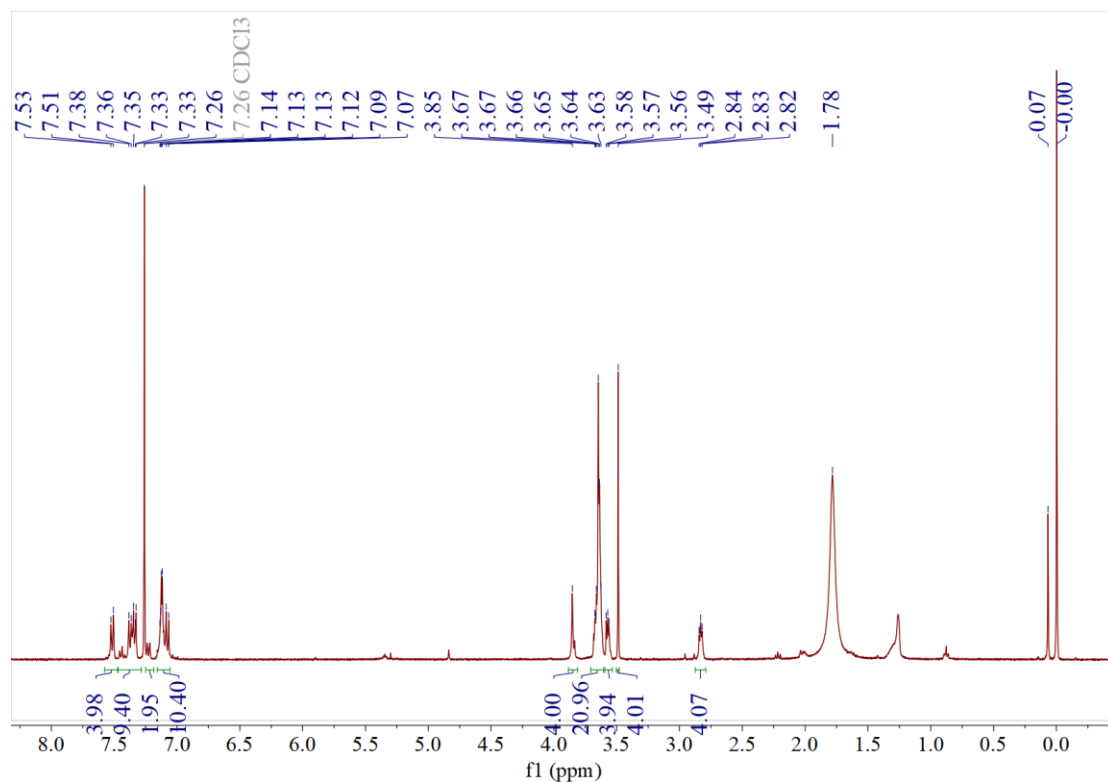


Figure S7. ¹H NMR of compound **3**. (CDCl₃, 300 MHz)

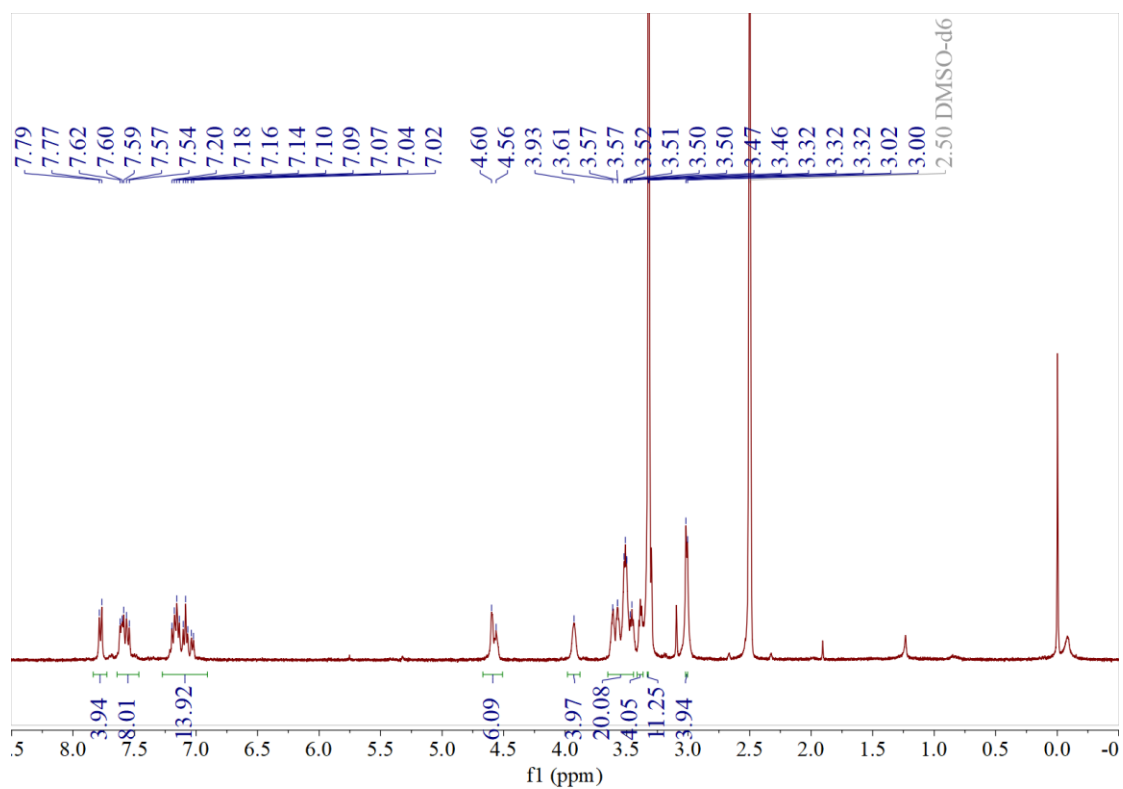


Figure S8. ¹H NMR of compound **TPE-Q**. (DMSO-*d*₆, 300 MHz)

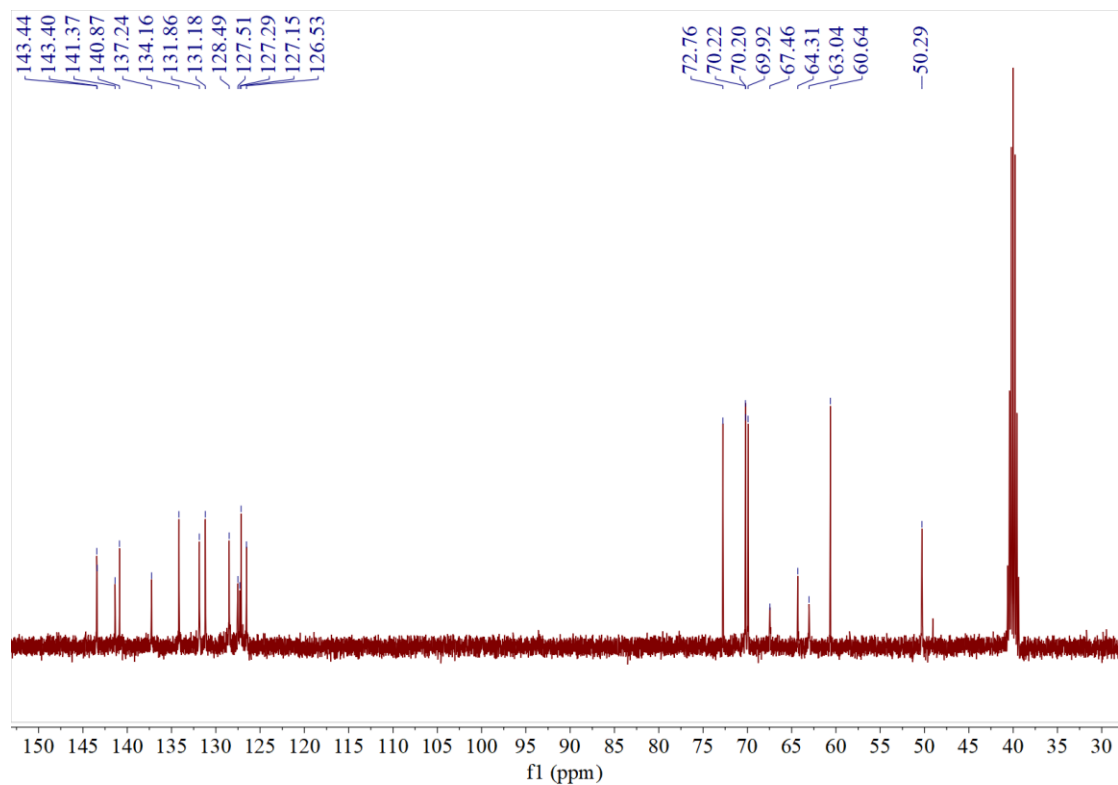


Figure S9. ^{13}C NMR of compound TPE-Q. (DMSO- d_6 , 150 MHz)

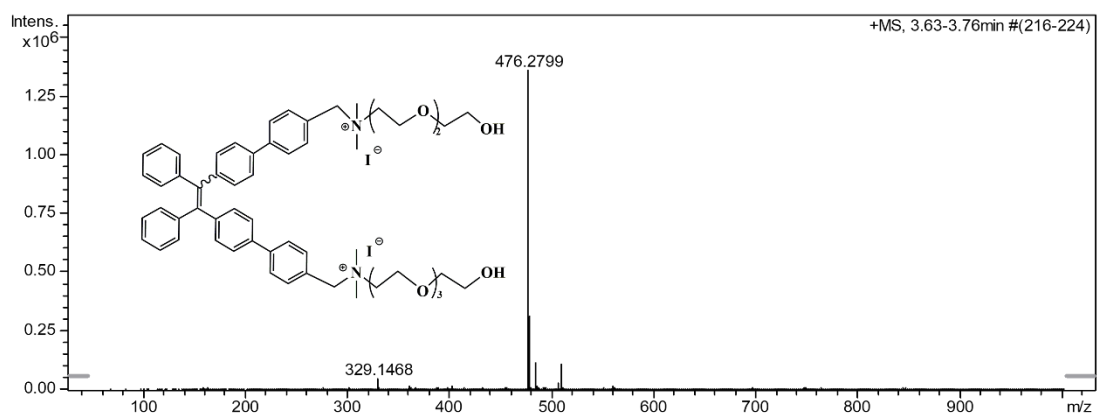


Figure S10. HRMS spectrum for TPE-Q.