## Alkylation of benzene using CO<sub>2</sub> and H<sub>2</sub> over ZnZrOx/ZSM-5: the

## effect of Y doping

Guowei Bian<sup>a,b</sup>, Pengyu Niu<sup>a</sup>, Litao Jia<sup>a,c\*</sup>, Heqin Guo<sup>a</sup>, Debao Li<sup>a,c\*</sup>

<sup>a</sup>State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese

Academy of Sciences, Taiyuan 030001, PR China

<sup>b</sup>University of Chinese Academy of Sciences, Beijing 100039, PR China

<sup>c</sup>Dalian National Laboratory of Clean Energy, Dalian, 116023, PR China

\* Corresponding author. E-mail: jialitao910@163.com (Litao Jia), dbli@sxicc.ac.cn

(Debao Li).





Fig.S1 Y0, Y0.05, Y0.1, Y0.3, ZSM-5 SEM images

Table S1 d-spacings of prepared catalysts calculated using the Bragg equation from XRD

	20 / °	d(011) (nm)	20 / °	d(110) (nm)	20 / °	d(020) (nm)	20 / °	d(121) (nm)
Y0	30.53	0.2926	35.38	0.2535	50.83	0.1795	60.4	0.1531
Y0.05	30.46	0.2932	35.28	0.2542	50.78	0.1797	60.33	0.1533
Y0.1	30.45	0.2933	35.27	0.2544	50.70	0.1799	60.23	0.1535
Y0.3	30.41	0.2937	35.15	0.2551	50.44	0.1808	59.94	0.1542



Fig.S2 EDS-mapping of Y0.1

Flomont	Atomic	Quality /	Normalized	Atoms / %
Element	number	%	Mass / %	Atoms / 70
0	8	24.8	29.9	70.3
Zn	30	3.6	4.3	2.5
Y	39	3.8	4.6	1.9
Zr	40	50.8	61.3	25.3
		82.9	100	100

Table S2 EDS-mapping element content analysis of Y0.1.



Fig. S3 Y0.1TEM images of different magnifications

Table S3 Deconvolution results of O 1s XPS peaks.

Catalyst	O <sub>OH</sub> / %	O <sub>V</sub> / %	O <sub>L</sub> / %
Y0	13.1	17.8	69.1
Y0.05	11.0	24.7	64.3
Y0.1	13.6	25.7	60.7
Y0.3	17.3	20.2	62.5

Table S4 Y-doped ZnZrOx/HZSM-5 catalyzed alkylation of CO<sub>2</sub> and benzene.

Catalwat	Conv.	Conv.	Selectivity/%			
Catalyst	CO <sub>2</sub> /%	benz./%	M <sub>B</sub>	E <sub>B</sub>	X <sub>B</sub>	Other
Y0/ZSM-5	10.0	19.7	73.6	3.0	14.3	9.1
Y0.05/ZSM-5	10.2	22.7	61.8	2.6	12.4	23.2
Y0.1/ZSM-5	17.3	30.8	56.5	2.5	15.0	26.0
Y0.3/ZSM-5	14.5	18.3	73.3	3.0	11.5	12.2

Table S5 Liquid phase product distribution

Catalyst	Conv.	Liquid phase product distribution / %				
Catalyst	benz. / %	M <sub>B</sub>	EB	X <sub>B</sub>	C <sub>9+</sub>	
Y0/ZSM-5	19.7	80.2	3.3	15.5	1.0	
Y0.05/ZSM-5	22.7	79.8	3.3	16.1	0.7	
Y0.1/ZSM-5	30.8	74.8	3.3	20.0	1.9	
Y0.3/ZSM-5	18.3	81.5	3.3	12.8	2.3	
Zn0.1Ti/ZSM-5(30)	23.8	77.4	2.8	16.1	3.7	Ref.[1]

Table S6 Different mixing methods of Y0.1/HZSM-5 catalyzed alkylation of CO<sub>2</sub> and benzene.

	Conv.	Conv.	v. Selectivity/%			
	CO <sub>2</sub> /%	benz./%	M <sub>B</sub>	E <sub>B</sub>	X <sub>B</sub>	Other
dual-bed	12.8	9.33	56.8	5.9	6.2	31.1
granule-mixing	14.8	28.5	58.0	2.1	14.4	26.5
powder-mixing	17.3	30.8	56.5	2.5	15.1	25.9

Tuble 57 Gus phase products of 10.1725W 5 and 10.1101 CO2 hydrogenation to methanor								
	Conv.			Se	lectivity/%			
Catalyst	CO <sub>2</sub> /%	Methane	Ethylene	Ethane	Propylene	Propane	Butane	CO
Y0.1/ZS M-5	17.3	1.4	0.1	0.2	0.3	0.3	0.3	56.3
Y0.1	13.5	3.7	0	0	0	0	0	91.8

Table S7 Gas phase products of Y0.1/ZSM-5 and Y0.1 for CO<sub>2</sub> hydrogenation to methanol

Table S8 H <sub>2</sub> consumption amount of various catalysts.						
Catalysts	7nV O 7r / mmol/g	7rO / mmol/g	(ZnY-O-Zr)/(ZnY-			
Catalysis		$L(O_2)$ minol/g	O-Zr+ZrO <sub>2</sub> )			
YO	0.08	0.52	0.13			
Y0.05	0.18	0.58	0.24			
Y0.1	0.24	0.34	0.41			
Y0.3	0.11	0.54	0.17			



Fig. S4 IR spectra of the ZSM-5 after pyridine desorption

	BET Surface Area / m <sup>2</sup> /g	Pore Volume / cm <sup>3</sup> /g	Pore Size / nm
YO	28	0.03	3
Y0.05	33	0.04	3
Y0.1	29	0.04	3
Y0.3	14	0.01	3
ZSM-5	346	0.09	6

Table S9 Catalyst bet specific surface area, pore volume and pore size

 Liu, X.; Pan, Y.; Zhang, P.; Wang, Y.; Xu, G.; Su, Z. Frontiers of Chemical Science and Engineering 2021, 16 (3), 384-396.