

## Electronic Supplementary Information

### Experimental section

#### *Synthesis of materials*

0.8 g of LiF was dissolved in HCl (10 mL, 9 M) to obtain a uniform solution. 0.5 g of  $\text{Ti}_3\text{AlC}_2$  was added into the solution and left under stirring for 24 h at 40 °C to etch the Al element. Then the mixture was centrifugated and washed with deionized water many times until the pH was up to 6. The multilayered  $\text{Ti}_3\text{C}_2\text{T}_x$  precipitate was collected after removing the acidic supernatant.

The obtained  $\text{Ti}_3\text{C}_2\text{T}_x$  MXene was added into LiOH (20 mL, 1 M) and left to stand for 24 h. After washing with deionized water, the precipitate was transferred to  $\text{Mn}(\text{NO}_3)_2$  (20 mL, 1 M) solution, then stirred for 12 h in the oxygen atmosphere. Subsequently, the precipitate was centrifugated and washed using deionized water. Finally, the Mn- $\text{Ti}_3\text{C}_2\text{T}_x$  was obtained after drying in the vacuum at 60 °C.

#### *Material characterization*

The scanning electron microscopy (SEM) images were observed by JSM-7800F. The transmission electron microscopy (TEM) images were obtained by JEM-2100. The X-ray diffraction (XRD) tests were carried out on the Rigaku Ultima IV. The X-ray photoelectron spectroscopy (XPS) tests were conducted by Thermo Scientific ESCALAB Xi+.

#### *Electrochemical measurements*

The electrochemical measurements were performed using CR2025-type coin cells. The active materials, carbon black, and polyvinylidene fluoride were mixed in the N-Methyl-pyrrolidone solution at a mass ratio of 8:1:1, and they were stirred for 24 h. Subsequently, the mixture was coated on the titanium foil and dried in the vacuum at 60 °C for 12 h. The mass loading was 1 mg  $\text{cm}^{-2}$ . The 2 M

ZnSO<sub>4</sub> + 0.2 M MnSO<sub>4</sub> served as the aqueous electrolyte. The galvanostatic discharge-charge profiles and cycling performances were recorded on a Land CT2001A system. The Cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS) tests were carried out by the electrochemical workstation (CHI 660E).

#### *First-principles calculations*

The DFT calculations were performed using the Vienna Ab-initio Simulation Package (VASP) with the Projector Augmented Wave method. The energy cutoff was set to 450 eV. The *k*-point mesh of 3×3×1 was used for these calculations. The energy convergence threshold was set to 10<sup>-5</sup> eV.

**Table S1.** The comparison of the electrochemical performance between Mn–Ti<sub>3</sub>C<sub>2</sub>T<sub>x</sub> MXene and other MXene cathodes for RAZBs.

Cathode material	Discharge plateau (vs. Zn)	Capacity and cycling stability	Ref.
Mn–Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub> MXene	1.4 V	0.3 A g <sup>-1</sup> /100 cycle/143 mAh g <sup>-1</sup> 5 A g <sup>-1</sup> /1000 cycle/65 mAh g <sup>-1</sup>	This work
MnO <sub>x</sub> @Ti <sub>3</sub> C <sub>2</sub> T <sub>x</sub>	1.4 V	5 A g <sup>-1</sup> /400 cycle/50 mAh g <sup>-1</sup>	1
Sn <sup>4+</sup> -Ti <sub>2</sub> CT <sub>x</sub> /C	–	0.1 A g <sup>-1</sup> /130 cycles/138 mAh g <sup>-1</sup>	2
Ti <sub>3</sub> C <sub>2</sub> Br <sub>2</sub>	1.6 V	4 A g <sup>-1</sup> /1000 cycles/60 mAh g <sup>-1</sup>	3
Ti <sub>3</sub> C <sub>2</sub> I <sub>2</sub>	1.1 V	4 A g <sup>-1</sup> /700 cycles/70 mAh g <sup>-1</sup>	3
MoO <sub>3-x</sub> /MXene	0.8 V	4 A g <sup>-1</sup> /1600 cycles/52 mAh g <sup>-1</sup>	4

## References

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