

Supplementary Data

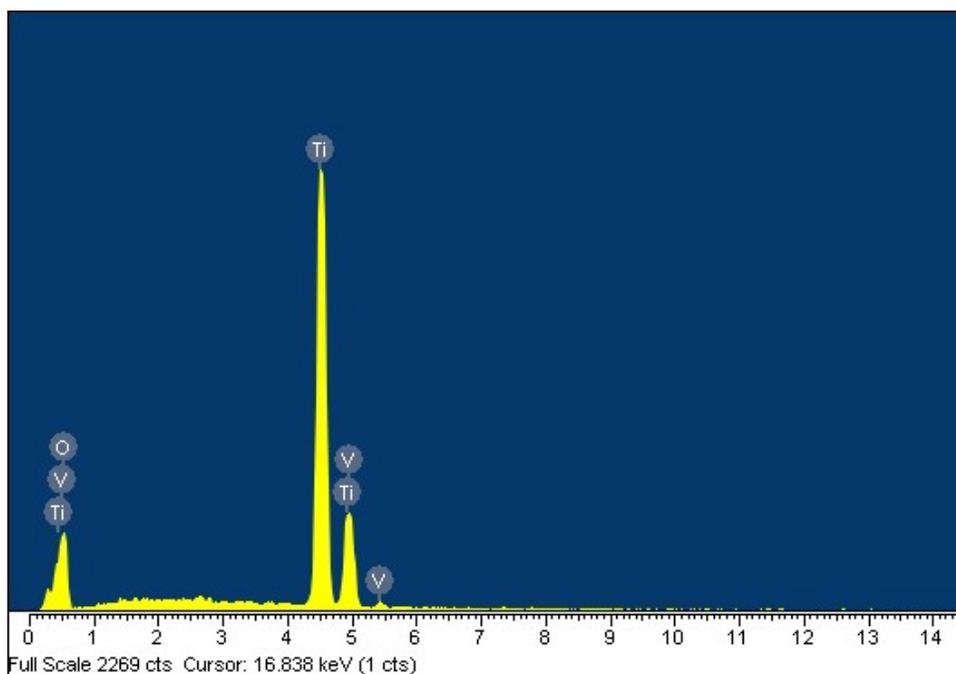


Figure S0. EDX spectrum of $\text{TiO}_2\text{-VO}_x$

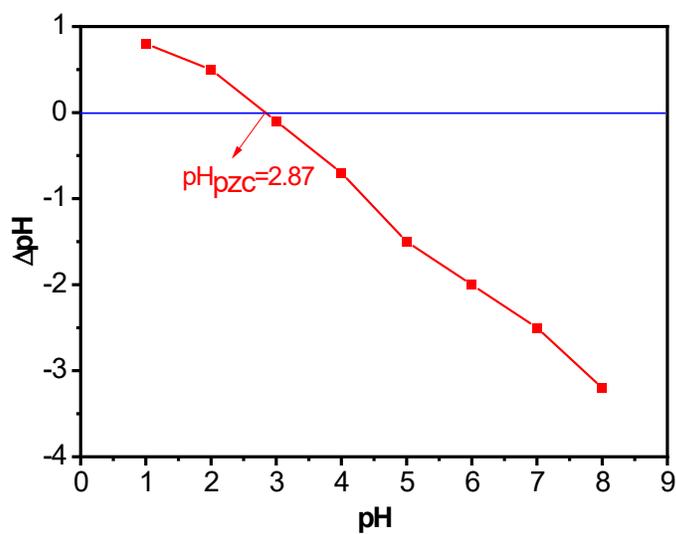


Figure S1. Isoelectric point (pH_{pzc}) of $\text{TiO}_2\text{-VO}_x/\text{PANi-PPy}$ composite

Table S1. Comparison of photocatalysts for PMS activation and RhB removal

Photocatalyst	Synthesis	Conditions	Remark	Reference
---------------	-----------	------------	--------	-----------

	method			
TiO ₂ -V ₂ O ₅ /PANi-PPy	Electrospinning, vapor-phase polymerization	Catalyst: 0.5 g/L; PMS: 0.65 mM; RhB: 20 mg/L; 25 °C; pH 7; Light: LED L4X 40W	97.03 % of RhB in 120 min Key ROS: ¹ O ₂	This work
CaFe ₂ O ₄	Sol-gel	Catalyst: 1 g/L; PMS: 0.65 mM; RhB: 10 mg/L; pH 4.8; 25 °C; Light: xenon 500 W with a 420 nm cut-off filter	99.2 % of RhB in 120 min Key ROS: h ⁺ and O ₂ [•]	[1]
BiVO ₄	Hydrothermal	Catalyst: 0.5 g/L; PMS: 1.0 mM; RhB: 10 mg/L; pH 3; 25 °C; Light: metal halogen 400 W with a 415 nm cut-off filter	90.9% of RhB removal in 60 min Key ROS: SO ₄ ^{•-} , HO [•] , and O ₂ [•] .	[2]
cPVC/Bi ₂ O ₃	Solvothermal	Catalyst: 0.4 g/L; PMS: 0.3 g/L; RhB: 10 mg/L; pH 5.18; 23 °C; Light: Osram Ultra-Vitalux 300 W	98% of RhB in 40 min Key ROS: SO ₄ ^{•-} and HO [•]	[3]
CoFe ₂ O ₄ @g-C ₃ N ₄	Sol-gel, urea thermal polymerization	Catalyst: 0.4 g/L; PMS: 0.09 g/L; RhB: 10 mg/L; pH: 9; Light source: Vonfram halogen 500 W	96% of RhB in 30 min Key ROS: SO ₄ ^{•-} , HO [•] , and O ₂ [•]	[4]
α-S	Wet chemical	Catalyst: 0.5 g/L; PMS: 0.4 g/L; RhB: 10 mg/L; pH 7; 40 °C; Light: 150 W Philips	100% of RhB in 50 min Key ROS: SO ₄ ^{•-} and HO [•]	[5]
BiFeO ₃ microsphere	Hydrothermal	Catalyst: 1 g/L; PMS: 5 mM; RhB: 5 mg/L; 25 °C; Light: xenon 500 W with a 420 nm cut-off filter	63% of RhB in 40 min Key ROS: HO [•] , SO ₄ ^{•-} , and O ₂ [•]	[6]

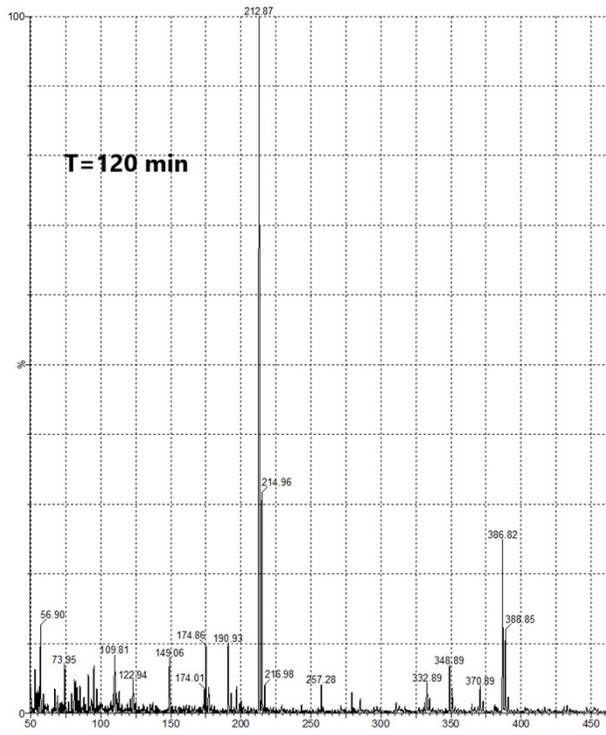
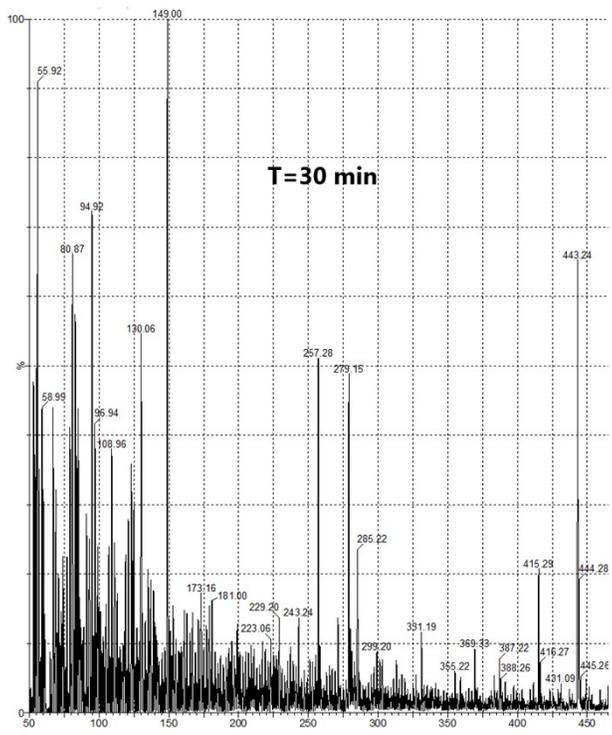


Figure S2. MS spectra of RhB solutions degraded by TiO₂-VO_x/PANi-PPy/PMS/Vis system.

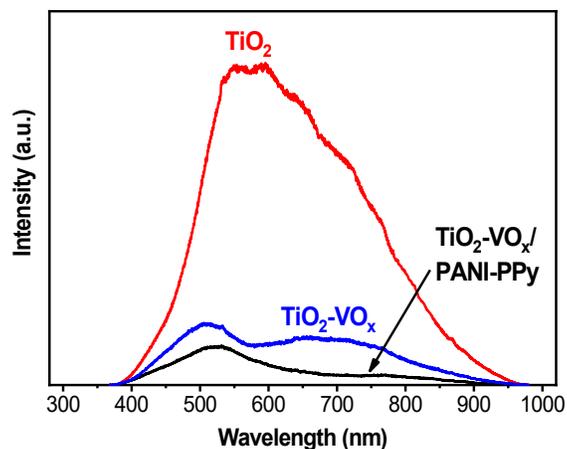


Figure S3. PL spectra of TiO₂ (red), TiO₂-VO_x (blue), and TiO₂-VO_x-PANi-PPy (black).

References

- [1] S. Guo, Z. Yang, H. Zhang, W. Yang, J. Li, K. Zhou, Enhanced photocatalytic degradation of organic contaminants over CaFe₂O₄ under visible LED light irradiation mediated by peroxymonosulfate, *Journal of Materials Science & Technology*, 62 (2021) 34-43.
- [2] Y. Liu, H. Guo, Y. Zhang, W. Tang, X. Cheng, H. Liu, Activation of peroxymonosulfate by BiVO₄ under visible light for degradation of Rhodamine B, *Chemical Physics Letters*, 653 (2016) 101-107.
- [3] H.-H. Pham, S.-J. You, Y.-F. Wang, M.T. Cao, V.-V. Pham, Activation of potassium peroxymonosulfate for rhodamine B photocatalytic degradation over visible-light-driven conjugated polyvinyl chloride/Bi₂O₃ hybrid structure, *Sustainable Chemistry and Pharmacy*, 19 (2021).
- [4] X. Guo, S. Ai, D. Yang, L. Zhao, H. Ding, Synergistic photocatalytic and Fenton-like degradation of organic contaminants using peroxymonosulfate activated by CoFe₂O₄@g-C₃N₄ composite, *Environ Technol*, 42 (2021) 2240-2253.
- [5] K.-Y. Andrew Lin, Z.-Y. Zhang, α -Sulfur as a metal-free catalyst to activate peroxymonosulfate under visible light irradiation for decolorization, *RSC Advances*, 6 (2016) 15027-15034.
- [6] F. Chi, B. Song, B. Yang, Y. Lv, S. Ran, Q. Huo, Activation of peroxymonosulfate by BiFeO₃ microspheres under visible light irradiation for decomposition of organic pollutants, *RSC Advances*, 5 (2015) 67412-67417.