

Supporting information for: ^1H NMR Global Diatropicity in Copper Hydride Complexes

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Supplementary Tables

Table SI-1. Experimental Cu- μ_3 -H distances (d_n) on top and bottom of the triangular cupola, average distance and standard deviation in Å for the Cu₂₀ and Cu₃₂ clusters.

	Top	Cu1	Cu2	Cu3	Average	STD	
Cu20^a	d ₁	1.62	1.58	1.63	1.61	0.03	
	d ₂	1.69	1.70	1.70	1.70	0.01	
	d ₃	1.75	1.73	1.70	1.73	0.02	
	Bottom						
	d ₁	1.64	1.59	1.54	1.59	0.05	
	d ₂	1.75	1.75	1.73	1.75	0.01	
	d ₃	1.77	1.81	1.80	1.79	0.02	
	Top						
	Cu32^b	d ₁	1.65	1.58	1.66	1.63	0.04
	d ₂	1.73	1.75	1.68	1.72	0.04	
	d ₃	1.77	1.87	1.71	1.78	0.08	
	Bottom						
	d ₁	1.65	1.58	1.66	1.63	0.04	
	d ₂	1.73	1.75	1.68	1.72	0.04	
	d ₃	1.77	1.87	1.71	1.78	0.08	

^a Ref.¹ ^b Ref.².

Supplementary Figures

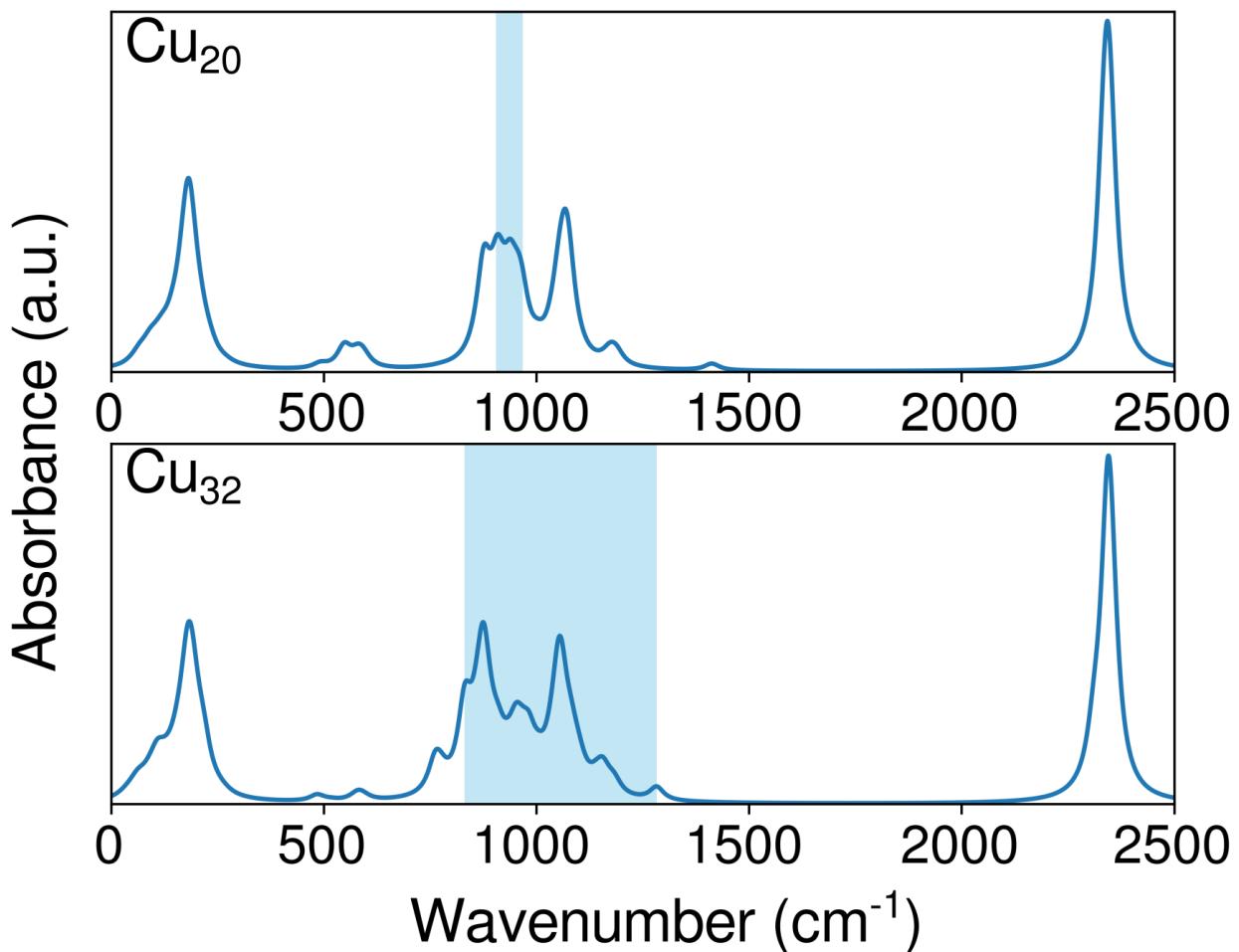


Fig. SI-1. IR spectra for $\text{Cu}_{20}\text{H}_{11}(\text{S}_2\text{PH}_2)_9$ (top row) and $\text{Cu}_{32}\text{H}_{20}(\text{S}_2\text{PH}_2)_{12}$ (bottom row). The blue shadow area indicates the range of the hydrides normal modes.

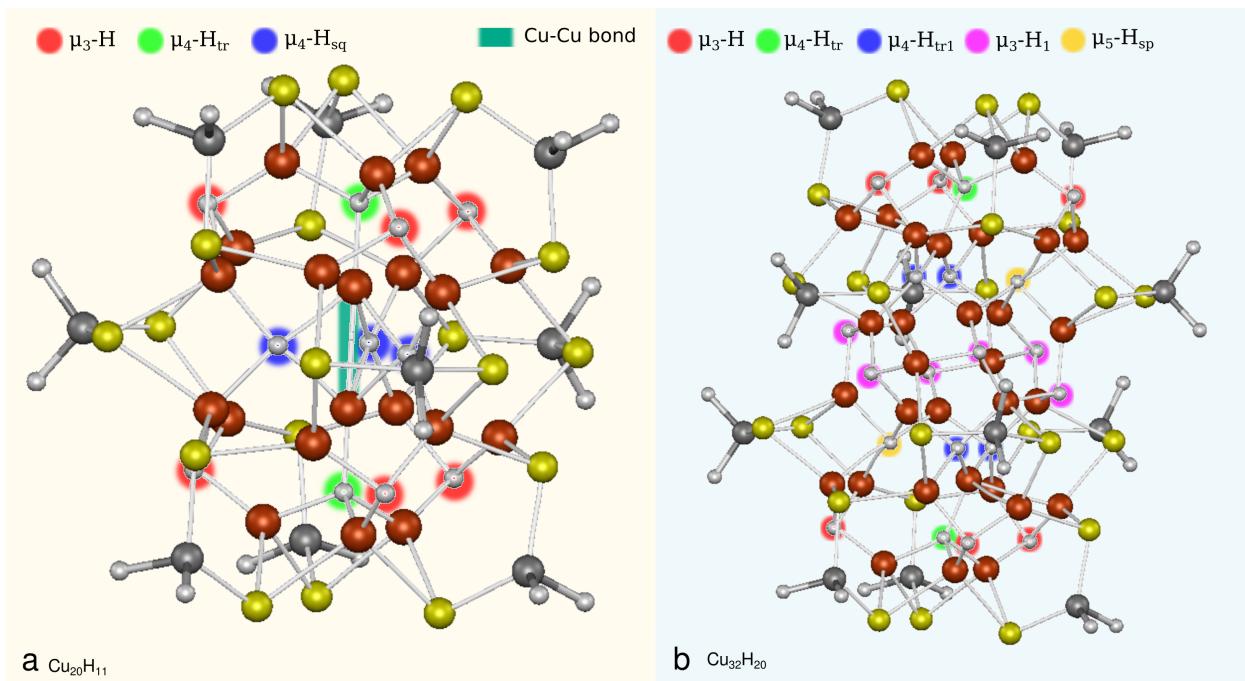


Fig. SI-2. Molecular graphs of **1** (left) and **2** (right). Bond path analyses shows that Cu-Cu bonds are absent with the exception of a Cu-Cu dimer at the centre of **1**. The different hydrides are highlighted.

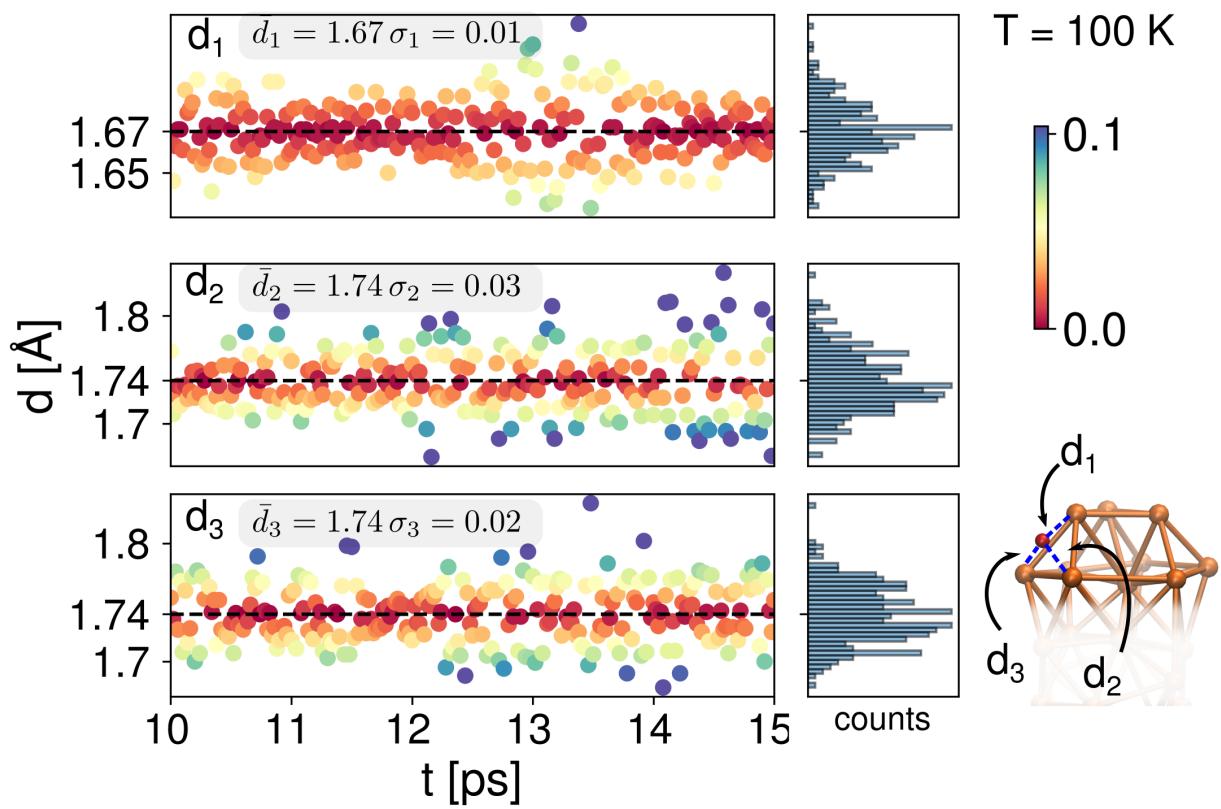


Fig. SI-3. Bond lengths analysis between the $\mu_3\text{-H}$ hydride and the three coordinated neighbor Cu atoms for the BOMD trajectory during 10–15 ps for the cluster **2** at 100 K.

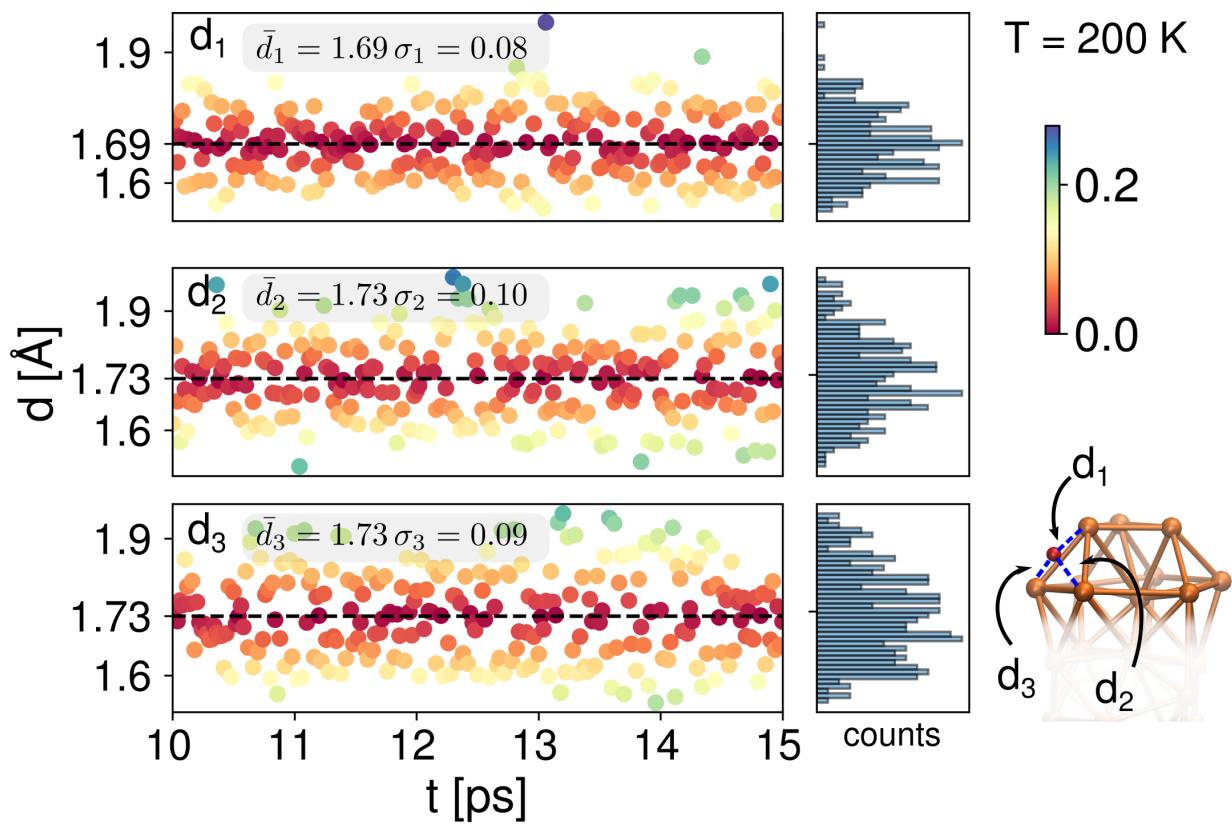


Fig. SI-4. Bond lengths analysis between the μ_3 -H hydride and the three coordinated neighbor Cu atoms for the BOMD trajectory during 10–15 ps for the cluster **2** at 200 K.

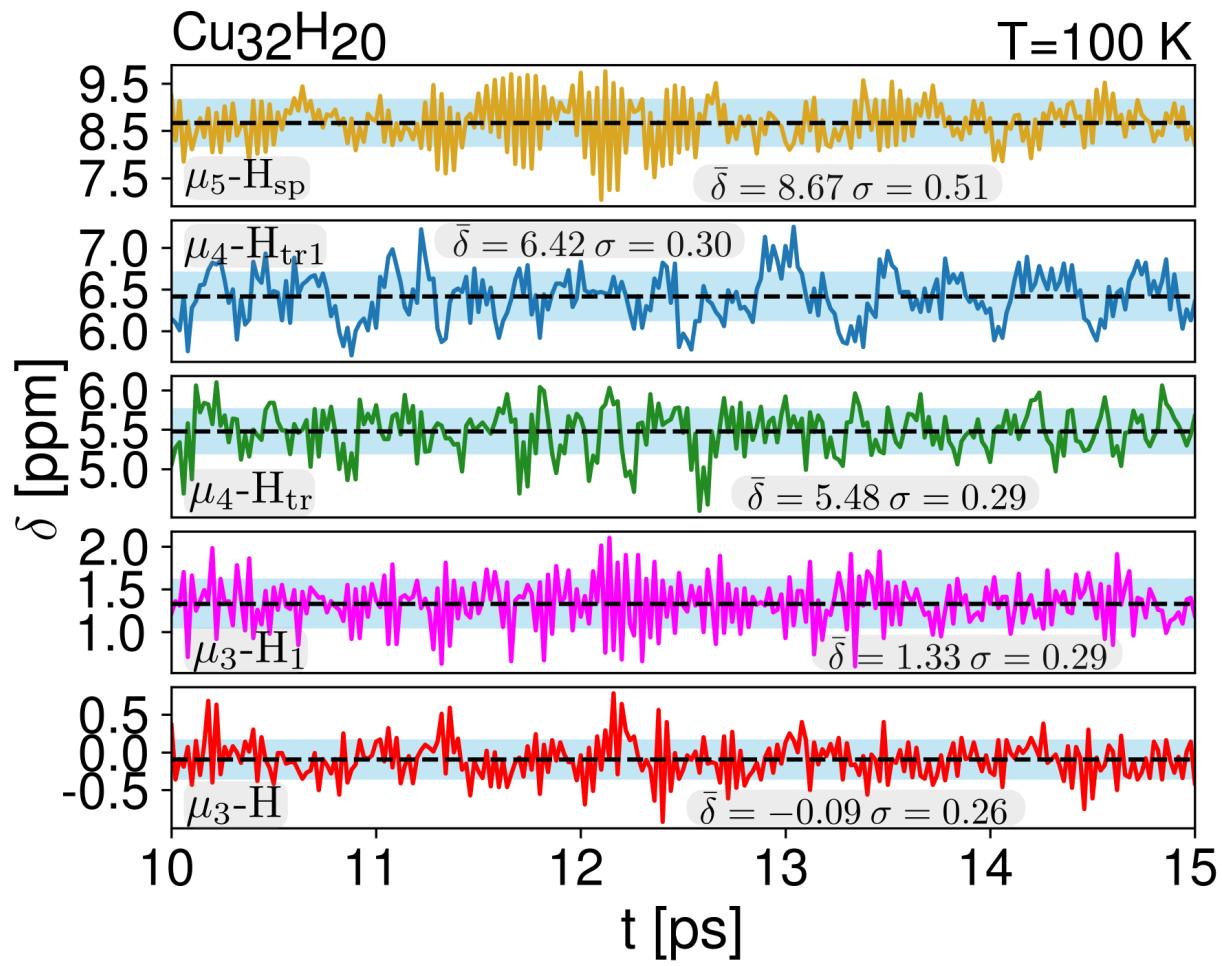


Fig. SI-5. ^1H NMR dynamics of one sample of each of the $\mu_3\text{-H}$, $\mu_3\text{-H}_1$, $\mu_4\text{-H}_{\text{tr}}$, $\mu_4\text{-H}_{\text{tr1}}$ and $\mu_5\text{-H}_{\text{sp}}$ hydride types for cluster **2** during 10 – 15 ps at 100 K. The average shifts are given by the dashed lines while the standard deviations are shown with the blue shading.

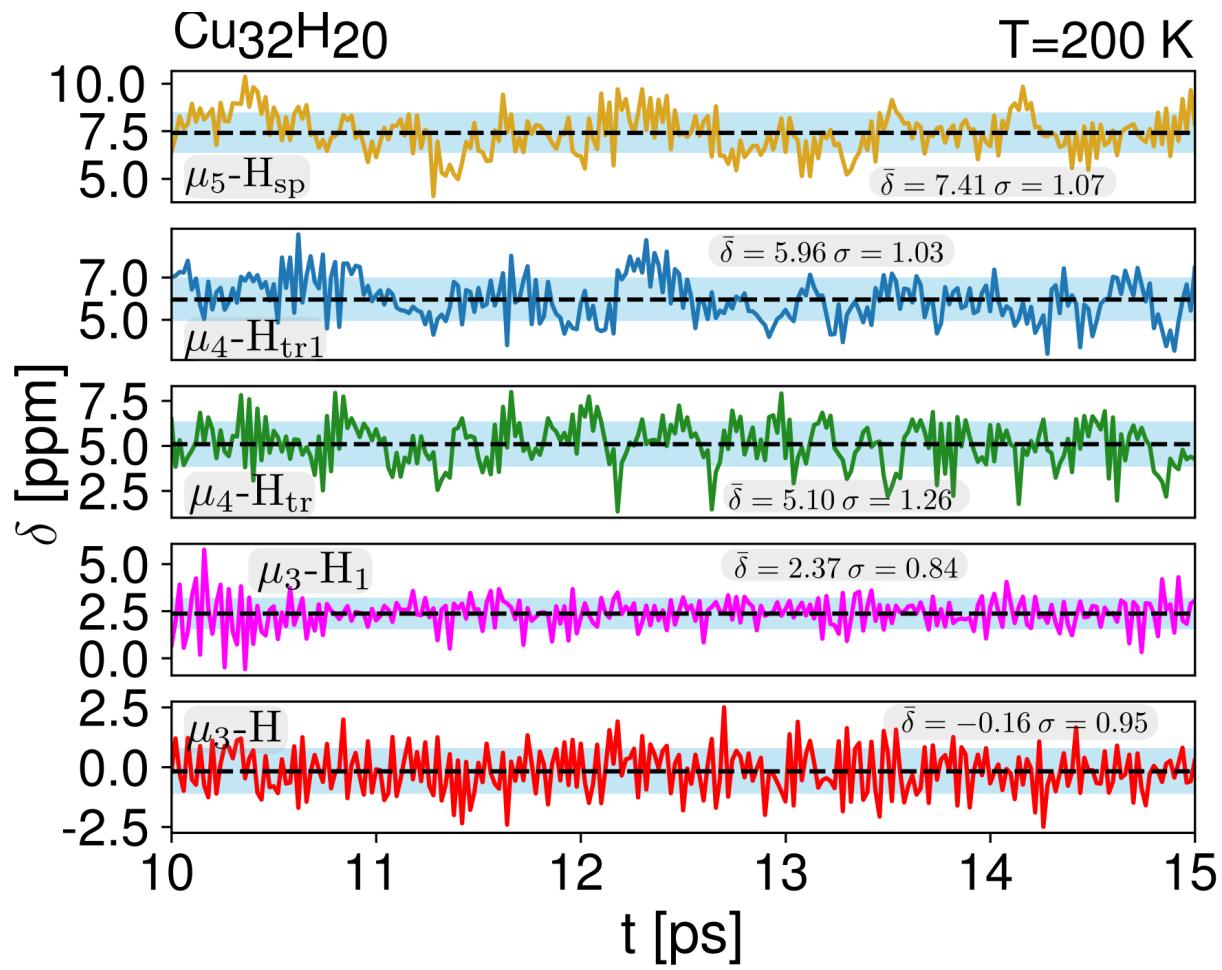


Fig. SI-6. ¹H NMR dynamics of one sample of each of the $\mu_3\text{-H}$, $\mu_3\text{-H}_1$, $\mu_4\text{-H}_{tr}$, $\mu_4\text{-H}_{tr1}$ and $\mu_5\text{-H}_{sp}$ hydride types for cluster **2** during 10 – 15 ps at 200 K. The average shifts are given by the dashed lines while the standard deviations are shown with the blue shading.

References

- (1) Dhayal, R. S.; Liao, J.-H.; Lin, Y.-R.; Liao, P.-K.; Kahlal, S.; Saillard, J.-Y.; Liu, C. W. A Nanospheric Polyhydrido Copper Cluster of Elongated Triangular Orthobicupola Array: Liberation of H₂ from Solar Energy. *Journal of the American Chemical Society* **2013**, *135*, 4704–4707.
- (2) Dhayal, R. S.; Liao, J.-H.; Kahlal, S.; Wang, X.; Liu, Y.-C.; Chiang, M.-H.; vanZyl, W. E.; Saillard, J.-Y.; Liu, C. W. [Cu₃₂(H)₂₀S₂P(O*i*Pr)₂₁₂]: The Largest Number of Hydrides Recorded in a Molecular Nanocluster by Neutron Diffraction. *Chemistry – A European Journal* **2015**, *21*, 8369–8374.