1 SUPPORTING INFORMATION

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3 Spatially-directed Magnetic Molecularly Imprinted

4 Polymers with Good Anti-interference for Simultaneous

5 Enrichment and Detection of Dual Disease-related Bio 6 indicators

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22 1. Instrumentation

Images of transmission electron microscopy were obtained using a JEM-2100 23 transmission electron microscope (TEM, JEOL Co., Japan). Hysteresis loops were 24 acquired by a 9600-1 vibrating sample magnetometer (VSM, LDJ Co., USA) for testing 25 the magnetic properties of resultant materials. X-ray diffraction patterns were recorded 26 with a Rigaku D/max/2500v/pc X-ray diffractometer (Rigaku Co., Japan) with Cu Ka 27 radiation to identify the crystalline phase of nanomaterials. A Nicolet AVATAR-330 28 Fourier-transform infrared spectrophotometer (FTIR) (Thermo Electron Co. USA) was 29 utilized by the KBr pressing method. An automatic elemental analyzer (EL, 30 EUROVECTOR EA3000, Italy) was used to obtain elemental contents. The HPLC 31 analyses were conducted using a Hitachi L-2130 HPLC system (Japan) equipped with 32 a binary pump, an ultraviolet detector, and a C18 chromatographic column (150 33 mm×4.6 mm, 5 µm, Shimadzu, VP-ODS). The optimized mobile phase was 34 acetonitrile-ultrapure water (5:95, v/v) with regular flow rate of 0.5 mL/min and 35 injection volume of 10 µL. The column temperature was 40 °C, and the detection 36 wavelength was operated at 283 nm. 37



40 Fig. S1. The effect of the amount of PEI (A) and polymerization time (B) on the
41 imprinting performance of D-mag-MIPs and mag-NIPs towards DOPAC.
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44 Fig. S2. Scatchard plots for D-mag-MIPs towards DA (A) and DOPAC (B). Scatchard

- 45 plots for mag-NIPs towards DA (C) and DOPAC (D).
- 46



49 Fig. S3. Chemical structures of DA, DOPAC, and their coexisting competitors.



52 Fig. S4. Selective adsorption capacity (A) and *IF* (B) of D-mag-MIPs and mag-NIPs

53 towards different compounds in the single solutions.





Fig. S5. The reusability test of D-mag-MIPs after six adsorption-desorption cycles. 56 To assess the reusability, which is of great importance when it comes to transitioning 57 from lab scales to any realistic practical applications, we performed six adsorption-58 desorption cycles on the same D-mag-MIPs (Fig. S5). The adsorption capacities of D-59 mag-MIPs for DA and DOPAC show a slight decrease after 6 cycles, likely due to the 60 minor destruction of imprinted cavities caused by the ethanol-HAC (98:2, v/v) eluent 61 used to remove the adsorbed templates by disruption the interaction between templates 62 and functional monomer. However, it still maintains above 91% of the initial adsorption 63 amount, revealing that D-mag-MIPs are stable and possess good reusability in 64 applications. 65



Fig. S6. The reproducibility of D-mag-MIPs prepared in five batches.

To investigate the long-term utilization of D-mag-MIPs, the reproducibility of 68 obtained nanomaterials were tested. Five different batches of D-mag-MIPs were 69 synthesized on different days. The adsorption of DA and DOPAC on each batch of 70 nanomaterials was performed by five times independently (Fig. S6). The average Q71 values of the different batches of D-mag-MIPs towards DA and DOPAC are found to 72 be 7.35 and 8.24 mg/g with RSD <5.5% and 6.2%, respectively. These results 73 demonstrate that D-mag-MIPs possess satisfactory reproducibility, indicating the 74 satisfactory stability of the prepared nanomaterials. 75

77 Table S1 The adsorption kinetic constants for pseudo-first-order and pseudo-second-

Sorbents		Pseudo-	first-order	Pseudo-second-order				
	Targets	R^2	k_1 (/min)	<i>R</i> ²	k_2 (g/mg·min)	v₀ (mg/g·min)		
D-mag-MIPs	DA	0.8631	0.5977	0.9563	0.0296	3.292		
	DOPAC	0.9068	0.5370	0.9775	0.0473	5.030		

78 order rate kinetic models of D-mag-MIPs.

79

81 Table S2. Adsorption equilibrium constants for Langmuir and Freundlich isothermal

_			Q _{e,E} (mg/g)	Langmuir isothermal equation			Freundlich isothermal equation		
	Sorbents	Targets		R^2	Q _{m,L} (mg/g)	$K_{\rm L}({\rm mL/mg})$	R^2	$K_{\rm F}({\rm mL/mg})$	п
	D-mag-MIPs	DA	7.52	0.9936	8.7873	0.3671	0.9352	2.0422	0.5268
	0	DOPAC	8.16	0.9967	9.0334	0.6123	0.9396	2.8035	0.4473

82 fitting constants of D-mag-MIPs.

 $Q_{e,E}$ is the experimental value of Q_e ;

 $Q_{m,L}$ is the calculated value of Q_e by *Langmuir* isotherm equation.

88	DOPAC detection. (<i>n</i> =5)

87 Table S3. The performance parameters of D-mag-MIPs-HLPC method for DA and

Analyte	Linearity range	D ²	RS	SD (%)	LOD	LOQ
	(µg/mL)	Π-	intra-day	inter-day	(µg/mL)	$(\mu g/mL)$
DA	0.10-100	0.9981	1.5-3.3	2.8-3.9	0.020	0.065
DOPAC	0.10-100	0.9987	1.6-3.1	2.3-3.6	0.031	0.096

Analytical system	Linear	RSD (%)		LOD _{DA/DOPAC}	Spiking	S 1 -	Recovery	RSD	Reusabi	Magnetic	Dof
	range (μg/mL)	intra-day	inter-day	$(\mu g/mL)$	$(\mu g/mL)$	Sample	(%)	(%)	lity	Separation	Kel.
GEM ^a	0.61-15.3			0.404 /	/		/	/		No	[35]
CS/N,GODs@SPCE ^b	0.16-15.3	0.9		0.023/	4.5	Human urine	~ 100	2.0	94.5 (after 1 month)	No	[36]
DE-MIPs ^c	0-0.183			0.023/	3.0,1.2, 45.9, 91.8	Human serum	100.1-103.8	3.6-4.9		No	[37]
MISPE-HPLC-FL ^d	0.15-2.23	2.2-4.6	1.6-4.7	0.025/	0.15, 0.3, 0.76, 1.5, 2.3	Human urine	98.3-101.1			No	[38]
D-mag-MIPs-HPLC- UV	0.1-100	1.5-3.1	2.3-3.9	0.020/ 0.031	0.1, 1.0, 5.0	Human urine	95.5-98.6	2.7-5.1	91.0 (after 6 cycles)	Yes	This work

Table S4. Comparison with other methods for detection of DA or DOPAC.

a: Graphene modified electrode.

b: Nitrogen-doped graphene quantum dots-chitosan nanocomposite-modified nanostructured screen printed carbon electrode.

c: Dual-emission fluorescent molecularly imprinted polymers.

d: Molecularly	imprinted	solidphase	extraction.
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