

Supporting Information

Chloride ions directed synthesis of plate-like Cu₂O mesocrystals for effective nitrogen fixation

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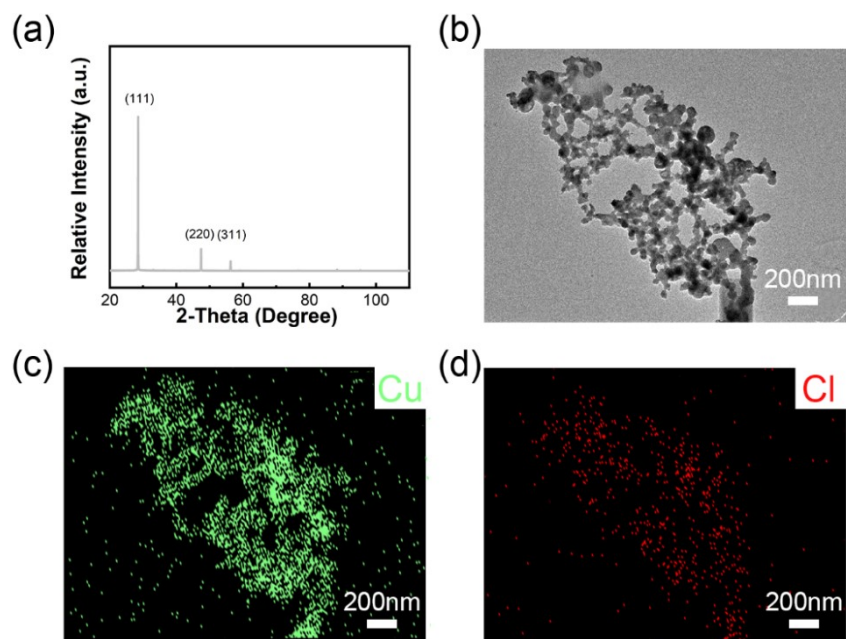


Fig. S1. (a) XRD patterns and (b) TEM image of the prepared CuCl precipitates. (c) Cu element and (d) Cl element mapping images of CuCl precipitates.

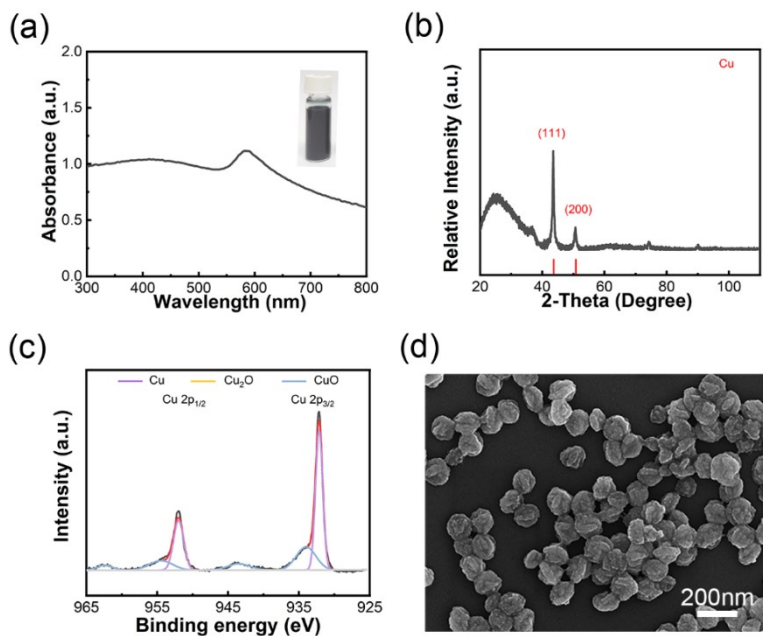


Fig. S2. (a) UV-vis absorption spectra, (b) XRD patterns (c) XPS patterns and (d) SEM images of Cu/CuO when ratio of Cu²⁺/Cl⁻ was 2:0.

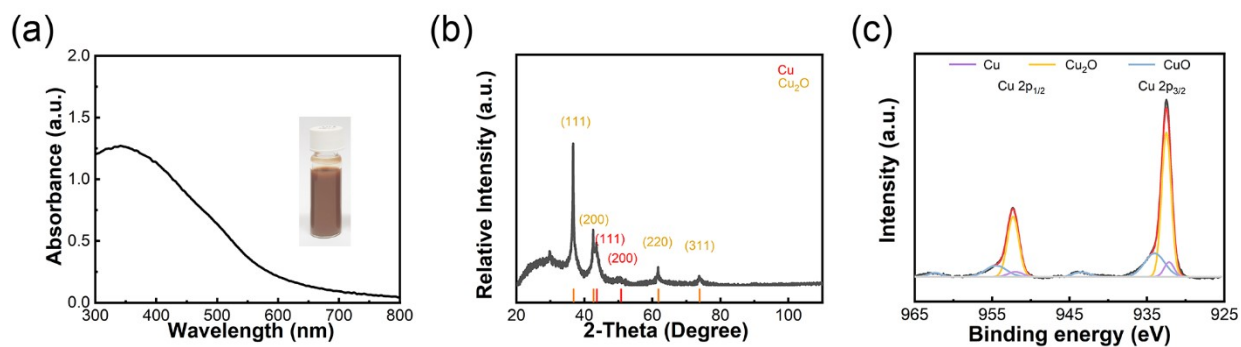


Fig. S3. (a) UV-vis absorption spectrum, (b) XRD patterns and (c) XPS patterns of the sample when ratio of $\text{Cu}^{2+}/\text{Cl}^-$ was 2:1. The inset in (a) was the photograph of the solution of the prepared sample.

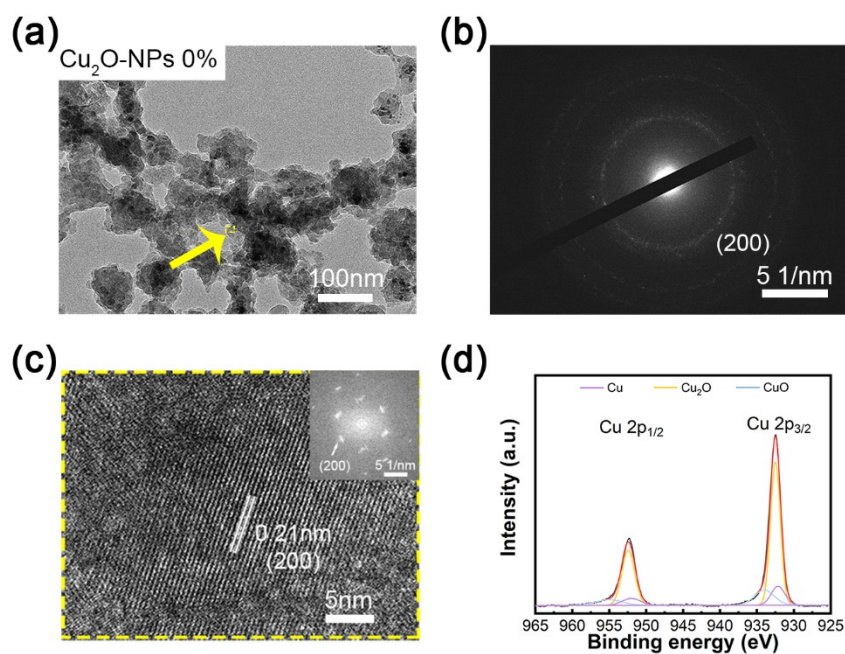


Fig. S4. (a) TEM image and (b) SAED pattern of Cu_2O -NPs prepared without addition of PAM. (c) HR-TEM image and FFT image of the selected area (yellow box) in (a), (d) XPS pattern of Cu_2O -NPs.

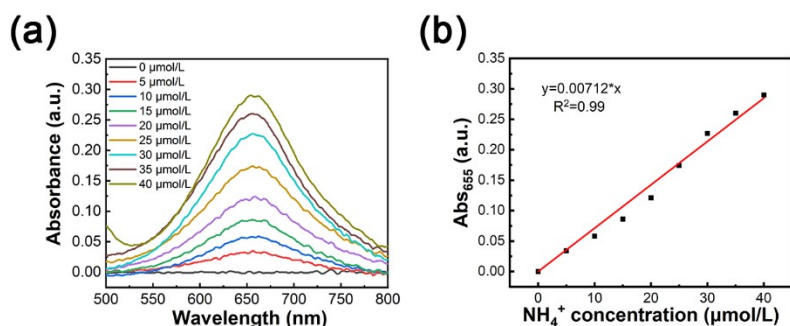


Fig. S5. (a) UV-vis absorption spectra of indophenol assays kept with different concentrations of NH_4^+ . (b) A calibration curve used to estimate the concentrations of NH_4^+ ions.

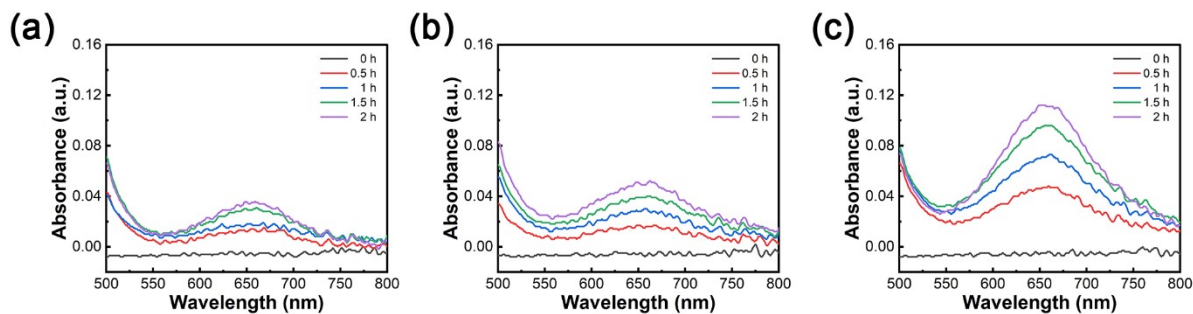


Fig. S6. UV-vis absorption spectra of indophenol assays kept with different times during photocatalytic nitrogen reduction for 2 h under visible light irradiation ($\lambda > 420 \text{ nm}$) at room temperature. (a) Cu_2O -NPs, (b) Cu_2O -MC-S and (c) Cu_2O -MC-P.

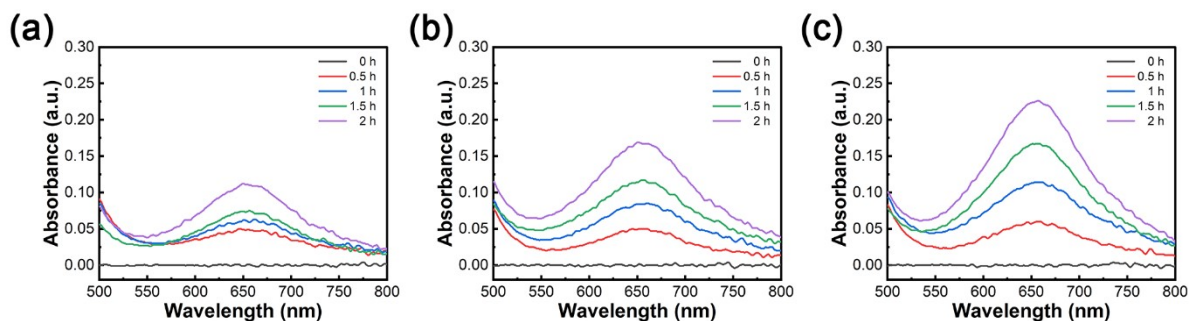


Fig. S7. UV-vis absorption spectra of indophenol assays kept with different times during photocatalytic nitrogen reduction for 2 h under 395 nm irradiation at room temperature. (a) Cu_2O -NPs, (b) Cu_2O -MC-S and (c) Cu_2O -MC-P.

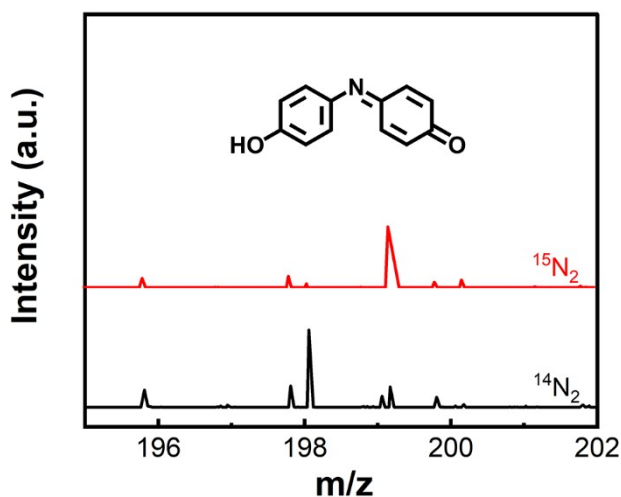


Fig. S8. Mass spectra of the indophenol products obtained by reaction of phenol with ammonia generated from photocatalytic $^{14}\text{N}_2$ or $^{15}\text{N}_2$ reduction. The inset showed the chemical structure of the indophenol product, with $m/z = 198$ (^{14}N) and $m/z = 199$ (^{15}N).