Supporting Information

Experimental Section

Hydrothermal synthesis of vanadium pentoxide nanowires.

 V_2O_5 NWs were synthesized through a hydrothermal method as previously described^{1,2}. VOSO₄.nH₂O (purity >99.9%), KBrO₃ (purity >98%), and nitric acid (purity ≥65%) were purchased from Alfa Aesar, Sigma Aldrich, and Trace SELECT® Ultra respectively. Briefly, 8 mmol of VOSO₄.nH₂O and 5 mmol of KBrO₃ were dissolved and stirred in 30 mL of Milli-Q water for about 30 mins at room temperature. Nitric acid was then added to the mixture solution dropwise until reaching pH of 1-2. Then the solution was transferred to a Teflon-lined stainlesssteel autoclave for a reaction period of 24 h at 180 °C. The autoclave was cooled down to room temperature naturally before the solution was filtered and washed with Milli-Q followed by ethanol several times. The resulting green nanomaterial was dried at 80 °C under vacuum overnight.

Synthesis of V₂O₅ NWs@DPA composite.

1 mg of V_2O_5 NWs were added into 5 mL of 10 mM Tris-HCl buffer (pH 8.0) with the aid of homogenizer for about 2 mins. Then 3.2 µg of dopamine (DPA) were added into the homogeneous V_2O_5 NWs solution and ultrasonicated for at least 2 hours to deposit DPA on the surface of the nanowire uniformly. The mixture was centrifuged at 10g for 30 mins and washed with Milli-Q water at least three times. The washing process would wash out any extra Tris-HCl buffer remaining in the mixture.

Synthesis of V₂O₅ NWs@DPA@AuNPs cascade nanozyme.

1 mg of V_2O_5 NWs@DPA was dispersed in 10 mL of Milli-Q water with the aid of homogenizer for about 3 mins. HAuCl₄ was added to reach the final concentration of 2 µM and ultrasonicated for 10 mins at room temperature. The freshly prepared NaBH₄ was added to reach the final concentration of 10 µM and further ultrasonicated for 15 mins at room temperature. The final mixture was then centrifuge at 10g for 30 mins and washed with Milli-Q water at least two times. The washing process would wash out free AuNPs in the solution.

To obtain the cascade catalytic activities, V_2O_5 NWs@DPA composites 1, 2, and 3, containing 1.6 µg, 3.2 µg, and 6.4 µg DPA per 1 mg of V_2O_5 NWs, respectively, were chosen to be decorated with AuNPs due to their high HPO-like activities (Fig. 1E). To obtain optimum GOx-like activity of AuNPs, we started with adding 100 µM AuNPs to the V_2O_5 NWs@DPA composites (at the V_2O_5 NWs concentration of 0.1 mg/ml), which we measured to provide about 2000 µM H₂O₂ in one hour (Fig. S1). The V₀ of the HPO-like activity of V₂O₅ NWs@DPA@AuNPs, using composites 1-3, was reduced by 41.0%, 44.5%, and 30.8% respectively in the presence of H₂O₂ and Br⁻. It was reduced by 88.5%, 81.4%, and 82.3% respectively in the presence of glucose, O₂, and Br⁻ (Fig. S7A). We subsequently attempted to lower the concentration of AuNPs to 3 µM to a V₂O₅ NWs@DPA solution with the V₂O₅ NWs

concentration of 0.1 mg/ml, which we measured to provide 12 μ M H₂O₂ in one hour (Fig. S1). The V₀ of the HPO-like activity of V₂O₅ NWs@DPA@AuNPs, using composites 1-3, was reduced by 29.5%, 15.5%, and 22.3% in the presence of H₂O₂ and Br⁻ and by 83.3%, 74.1%, and 77.6% in the presence of glucose, O₂, and Br⁻ (Fig. S7B). Therefore, lower concentrations of AuNPs led to greater cascade catalytic activities (likely due to less coverage of V₂O₅ NWs active sites). The composites 2 showed the highest catalytic activity among all three tested and was thus used in further characterizations as described below.

The concentration of AuNPs in V₂O₅ NWs@DPA@AuNPs was varied in the range of 1-3 μ M (in a V₂O₅ NWs@DPA solution with the V₂O₅ NWs concentration of 0.1 mg/ml) and characterized using the MCD method. The MCD results showed that V₂O₅ NWs@DPA@AuNPs with 2 μ M AuNPs had the best catalytic activities (Fig. S8), with the V₀ reduced by merely 2.27% in the presence of H₂O₂ and Br⁻ and by 63.7% in the presence of glucose, O₂, and Br⁻ (Fig. S8), compared to V₂O₅ NWs@DPA@AuNPs with 1 and 3 μ M AuNPs.

Transmission electron microscopy (TEM) analysis.

The TEM samples were prepared by dropping 5 μ L of nanowires/cascade nanozyme onto a holey-carbon coated copper grid for 10 s and suck the solution out using tissue. The low-magnification TEM and HRTEM images were acquired by using a field-emission-gun (FEG) FEI Tecnai F-20 microscope. Energy-dispersive X-ray spectroscopy (EDX) elemental mapping was obtained by using an Oxford X-Max detector. STEM-EDX mapping was set at a beam voltage of 200 keV and a pixel size of 128×128. Beam damage of STEM-EDX maps has been routinely observed before and after EDX mapping.

X-ray photoelectron spectroscopy (XPS).

Samples were analyzed using a Surface Science Instruments SSX-100 ESCA Spectrometer with operating pressure ca. $1x10^{-9}$ Torr. Monochromatic Al K α x rays (1486.6 eV) with photoelectrons collected from an 800µm diameter analysis spot. Photoelectrons were collected at a 55° emission angle with source to analyzer angle of 70°. A hemispherical analyzer determined electron kinetic energy, using a pass energy of 150 eV for wide/survey scans, and 50 eV for high resolution scans. A flood gun was used for charge neutralization of non-conductive samples.

Monochlorodimedone (MCD) method.

The bromination activity was measured for 180 s at room temperature in Milli-Q water with 0.02 mg/mL V₂O₅ NWs, 1 mM Br⁻, 10 μ M H₂O₂, and 50 μ M MCD (e_{290nm} =19.1 mM⁻¹cm⁻¹)¹. Potassium bromide (KBr) was used to provide Br⁻. And the concentration of H₂O₂ was calculated spectrophotometrically (e_{240nm} = 43.6 M⁻¹cm⁻¹).

 V_2O_5 NWs and MCD were added to the mixture last, prior to the measurements to trigger the reaction. In order to measure the Michaelis-Menten kinetics, two independent sets of experiments were performed: 1) Br⁻ concentration was varied (0-20 mM) while keeping

concentrations of V₂O₅ NWs/cascade nanozyme (0.02 mg/mL), H₂O₂ (10 μ M) and MCD (50 μ M) constant. 2) H₂O₂ or glucose concentration was varied (0-500 μ M) while keeping the concentration of V₂O₅ NWs/cascade nanozyme (0.02 mg/mL), Br⁻ (1 mM) and MCD (50 μ M) constant. For each measurement, blank tests and controls were performed, each measurement was carried out 4 times. The values were fitted to the Michaelis–Menten model and kinetic parameters were determined by Lineweaver–Burk linearization.

Detection of H₂O₂ production

The rate of producing H_2O_2 depending on the concentration of glucose in the range of 0-500 mM was measured with presence of 50 µM AuNPs and injected air (providing O_2) for one hour. The testing samples were then centrifuged at 10g for 2 mins. The suspension was tested using a H_2O_2 fluorometric assay kit named Hydrogen Peroxide Assay Kit from Cayman Chemical. Briefly, H_2O_2 was detected using 10-acetyl-3,7-dihydroxyphenoxazine (ADHP), a highly sensitive and stable probe for H_2O_2 . In a horseradish peroxidase catalyzed reaction, ADHP reacts with H_2O_2 with a 1:1 stoichiometry to produce highly fluorescent resorufin. The resorufin fluorescence can be detected using an excitation wavelength of 530 nm and an emission wavelength of 590 nm. The detected fluorescence was then compared with standard curve of H_2O_2 to get exact amount of H_2O_2 production.

Antimicrobial test.

Non-typable Haemophiles influenza (NTHi) was cultured in a Brain Heart Infusion (BHI) medium with defibrinated horse blood and nicotinamide adenine dinucleotide (NADH) in a humidified 5% CO₂-containting balanced-air incubator at 37°C until the mid-log phase was reached, according to established protocols³. Then the suspension assay was used to determine the minimum inhibitory concentration (MIC) of nanowires/cascade nanozymes/ciprofloxacin for eradicating NTHi completely in 7 hours. The first formulation contained 0.04 mg/mL V₂O₅ NWs, 1 mM Br-, and 20 µM H₂O₂. The second formulation contained 0.04-0.16 mg/mL of V₂O₅ NWs@DPA, 1 mM of Br-, and 20 µM of H₂O₂. The third formulation contained 0.04-0.16 mg/mL of V₂O₅ NWs@DPA@AuNPs, 1 mM of Br,62.5 mM of glucose with sustained air. The fourth formulation contained ciprofloxacin in the range of 0-0.8 µg/mL. The optical density at 600 nm (OD600) was measured as a function of time using a microplate reader (Infinite M1000 Tecan). Background from nanowires/cascade Pro, signals nanozyme/ciprofloxacin were subtracted from the final reading. Subsequently, the MICs were confirmed by plating on chocolate agar plates after 7-hour incubation and kept in 37°C incubator with 5% CO₂.

Antibiofilm test.

NTHi was cultured in a brain Hear Infusion (BHI) medium with defibrinated horse blood and nicotinamide adenine dinucleotide (NADH) in a humidified 5% CO₂-containing balanced-air incubator at 37°C until the mid-log phase was reached, according to established protocols³. Then the bacterial culture was diluted 10 times and 200 µL of the diluted culture was pipetted into 96-well round bottom plates and incubated at 37°C with 5% CO₂. Media were carefully

replaced with 200 μ L of the fresh media daily. The cascade nanozymes and ciprofloxacin (at their respective MICs against planktonic NTHi) were used to treat NTHi biofilms after 3 days of growth. The treatment lasted for 24 hours under the same conditions as the aforementioned antimicrobial test. Growth of biofilm was then evaluated by crystal violet staining, Bactiter-GloTM Microbial Viability Assay, and the LIVE/DEADTM BacLightTM Bacterial Viability Kit. For crystal violet staining, biofilms were stained with 0.5 %wt crystal violet for 10 mins. Then excess crystal violet was removed by washing each well several times until the liquid in each well became a clear solution. The plates were air dried and stained biofilms dissolved by adding 30 v/v% acetic acid solution, which released the absorbed crystal violet for quantification. The total cell count in biofilms was analyzed spectrophotometrically by measuring the optical density at 570 nm (OD570, corresponding to the absorbance of crystal violet) using a microplate reader (Infinite M1000 Pro, Tecan). Background signals from nanowires/cascade nanozyme and plates were subtracted from the final reading.

Bactiter-GloTM Microbial Viability Cell Assay provides a method to determine the number of live cells in a biofilm based on a quantification of ATP. The assay involves adding a single reagent (Bactiter-GloTM Reagent) directly to bacterial cells cultured in medium and measuring luminescence. The luminescent signal is proportional to the amount of ATP and thus the number of live cells. Background signals from nanowires/cascade nanozyme and plates were subtracted from the final reading.

The LIVE/DEADTM BacLightTM Bacterial Viability Kit was used to stain live and dead bacterial cells in a biofilm, and LSM710 Confocal microscope was used to take 2D and 3D images of the stained biofilms.

Cytotoxicity test in vitro.

Human primary dermal fibroblasts (hFBs, ATCC PCS-201-012) were cultured and maintained in Fibroblast Basal Medium (FBM, ATCC PCS-201-020) with Fibroblast Growth Kit – Low Serum (ATCC, PCS-201-041) and 1% penicillin/streptomycin (Gibco) at 37°C with 5 % CO₂. Once hFBs were seeded in sterile 96-well plates (Corning Costar) at a density of 10000 cells/well, the nanowires/cascade nanozymes formulations with 1 mM Br⁻ and 10 μ M H₂O₂ or 62.5 mM glucose with bubbling air were introduced into the cells and then incubated for 24 hours and 48 hours. CCK-8 kit for mammalian cells (Dojindo Molecular Technologies, Japan) was used to quantify the cell counts. Absorbance at 450 nm was measured after incubating cells with the CCK-8 kit reagents for 1–2 hours. Empty wells with CCK-8 assay reagents only were used as blanks and were subtracted from the final reading. Relative cell viability was calculated by normalizing the absorbance readings using that of nontreated cells. All assays were carried out in quadruplicates.

Statistical analysis.

Data were analyzed using GraphPad Prism 7.0 software and presented as mean \pm SD. Student's t test or one-way ANOVA followed by Tukey's HSD test was applied for comparisons between two groups or among multiple groups.

Figures



Fig. S1. H_2O_2 production based on different concentrations of AuNPs. (A) The H_2O_2 production as a function of AuNPs concentration (in range 0-100 μ M), collected with 62.5 mM glucose and bubbling air. (B) The H_2O_2 production as a function of AuNPs concentration (in range 1-3 μ M) in (A), collected with 62.5 mM glucose and bubbling air. Data are mean \pm SD. n=4.



Fig. S2. TEM image of naked AuNPs, showing unmodified AuNPs demonstrated a strong tendency to aggregate in solution.



Fig. S3. TEM of the V₂O₅ NWs.



Fig. S4. EDX spectrum of (A) V₂O₅ NWs and (B) V₂O₅ NWs@DPA@AuNPs cascade nanozyme. The peaks of V (0.453, 0.511,4.952,5.427 eV), O (0.525 eV), N (0.392 eV), and Au (1.648, 2.123, 2.410, 8.494, 10.308, 11.443, 11.585, 13.710, 13.382 eV) were analyzed in EDX element mapping.



Fig. S5. Scheme of the bromination of monochlorodimedone (MCD) to give dihalodimedone, used to quantify the reaction kinetics of HOBr generation.



Fig. S6. Antimicrobial effects of (A) physically mixed V_2O_5 NWs and AuNPs, (B) *in situ* grown AuNPs on the surface of V_2O_5 NWs, and (C) physically mixed V_2O_5 NWs@DPA and AuNPs of NTHi. Data are mean \pm SD. n=4.



Fig. S7. Scheme of the synthesis steps to obtain the cascade nanozyme.



Fig. S8. XRD patterns of AuNPs



Fig. S9. Catalytic activities of V_2O_5 NWs @DPA@AuNPs with different concentrations of DPA and AuNPs. (A) Solid circles represented the initial rate (V₀) of V₂O₅ NWs @DPA as a function of different DPA amount (µg), and solid squares and triangles represented the V₀ of V₂O₅ NWs @DPA@AuNPs decorated with 100 µM AuNPs as a function of different DPA amount (µg), collected at the V₂O₅ NWs @DPA /V₂O₅ NWs @DPA@AuNPs concentration of 0.02 mg/ml (in terms of the weight concentration of V₂O₅ NWs), the MCD concentration of 50µM, the Br⁻ concentration of 1 mM, and the H₂O₂ concentration of 10 µM or the glucose concentration of 62.5 mM. (B) Solid circles represented the V₀) of V₂O₅ NWs @DPA as a function of different DPA amount (µg), and solid squares and triangles represented the V₀ of V₂O₅ NWs @DPA@AuNPs decorated with 3 µM AuNPs as a function of different DPA amount (µg), collected at the V₂O₅ NWs @DPA /V₂O₅ NWs @DPA@AuNPs concentration of 0.02 mg/ml (in terms of the weight concentration of V₂O₅ NWs @DPA@AuNPs concentration of 0.02 mg/ml (in terms of the weight concentration of V₂O₅ NWs @DPA@AuNPs concentration of 50µM, the Br⁻ concentration of 1 mM, and the H₂O₂ concentration of 10 µM or the glucose concentration of 2.5 mM. Data are mean ± SD. n=4.







Fig. S11. NTHi biofilm growth curve. Data are mean \pm SD. n=5.



Fig. S12. Antibacterial effect of ciprofloxacin on NTHi. Minimum inhibitory concentration (MIC) was 0.4 μ g/mL. Data are mean \pm SD. n=4.



Fig. S13. 3D Confocal images of nontreated and treated NTHi biofilm. (A) nontreated NTHi biofilm, **(B)** NTHi biofilm treated with commercial antibiotic (ciprofloxacin), and **(C)** NTHi biofilm treated with V₂O₅ NWs@DPA@AuNPs cascade nanozyme.



Fig. S14. Characterization of V_2O_5 NWs. (A) XPS survey scan of the V_2O_5 NWs. (B) High-resolution XPS spectra of the V_2O_5 NWs in V_2O_5 NWs.



Fig. S15. Antibiofilm effects of V_2O_5 **NWs.** NTHi biofilm was treated with 0.08 mg/ml V_2O_5 NWs for 24 hours after three days of incubation. (A) Quantification of the crystal violet staining results of nontreated NTHi biofilm and NTHi biofilm treated with the 0.08 mg/ml V_2O_5 NWs. (B) Quantification of the BacTiter-GloTM Microbial Cell Viability Assay of nontreated NTHi biofilm treated with the 0.08 mg/ml V_2O_5 NWs. Data are mean \pm SD. n=5.



Fig. S16. In vivo therapeutic efficacy of the cascade nanozyme in a chinchilla OM model. (A) The timeline of the in vivo procedures to induce and treat OM in chinchillas. (B) Time course of the bacterial CFU count in the MEF of animals with OM. The formulations contained 0.16 mg/mL cascade nanozyme, 1 mM Br, and 62.5 mM glucose. Data are mean \pm SD. n=3.

Catalyst	Substrate	K _m (M)	V _{max} (M s ⁻¹)	Reference
V ₂ O ₅ NWs	Br	0.25×10^{-3}	5.17×10^{-10}	This work
	H ₂ O ₂	10.26×10^{-6}	7.26×10^{-10}	
V ₂ O ₅ NWs	Br	0.2×10^{-3} (Tris-HCl)	7.3×10^{-10}	F. Natalio et al.,
		0.6×10^{-3} (PBS)		Nature Nano.,
	H_2O_2	10 × 10 ⁻⁶ (Tris-HCl)	7.3×10^{-10}	2012.1
		$10 \times 10^{-6} (PBS)$		
Vanadium	Br⁻	3.1×10^{-3}		Z. Hasan, et al.,
chloroperoxidase (V-CPO)	H ₂ O ₂	16×10^{-6}		J. Biol. Chem.
from C. inaequalis	11202			2006.4
Vanadium bromoperoxidase (V-BPO) from <i>A. nodosum</i>	Br	18.1×10^{-3}		E. de Boer, J.
	H ₂ O ₂	22 × 10 ⁻⁶		Biol. Chem. 1988. ⁵

 TABLES

 Table S1. Kinetic parameters of different haloperoxidases.

References

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